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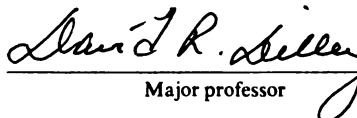
Mg²⁺ Induced Bitter Pit-Like Symptoms

presented by

Douglas M. Burmeister

has been accepted towards fulfillment
of the requirements for

Ph.D. degree in Horticulture


Major professor

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PREDICTION AND STUDY OF BITTER PIT ON APPLES USING
 Mg^{2+} INDUCED BITTER PIT-LIKE SYMPTOMS

By

Douglas M. Burmeister

A DISSERTATION

Submitted to
Michigan State University
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ABSTRACT

PREDICTION AND STUDY OF BITTER PIT ON APPLES USING Mg^{2+} INDUCED BITTER PIT-LIKE SYMPTOMS

By

Douglas M. Burmeister

Bitter pit is a physiological disorder affecting apples. Apple fruit vacuum infiltrated with Mg^{2+} develop bitter pit-like symptoms [Mg^{2+} induced pits (MgIP)] that are similar to naturally occurring bitter pit. Addition of Ca^{2+} to the infiltration media attenuates Mg^{2+} induced pit and the number of bitter pit-like lesions induced by Mg^{2+} is inversely related to the endogenous (native) Ca^{2+} concentration of individual fruits. MgIP was studied as a predictor of bitter pit development in storage and as a system to study bitter pit initiation and development.

Induction of MgIP on Northern Spy apples (*Malus domestica* Borkh.) by infiltrating Mg^{2+} salt solutions into the fruits was positively correlated with bitter pit that developed naturally during storage. The endogenous (native) fruit Ca^{2+} concentration was inversely related to the MgIP and to bitter pit development following storage.

Golden Delicious apple fruit were infiltrated with various concentrations of Ca^{2+} and Mg^{2+} with and without Ca^{2+} -affecting reagents or other cations. Including Trifluoperazine with Mg^{2+} increased pitting over Mg^{2+} alone.

Verapamil and nifedipine had no effect on MgIP or its attenuation by Ca^{2+} . Cyclopiazonic acid attenuated MgIP. Ethyleneglycol-bis(B amino ethyl ether)-N,N'-tetra acetic acid (EGTA), and 2,3,5-triiodobenzoic acid (TIBA) attenuated MgIP. Cycloheximide inhibited Mg^{2+} induced pit. Heating at 38°C prior to infiltrating the fruits attenuated MgIP. Cations Ba^{2+} , La^{3+} , Co^{2+} , Sr^{2+} included at 20.0mM all completely arrested MgIP induced by 0.18M Mg^{2+} . K^{+} and Na^{+} partially inhibited MgIP. We have demonstrated that treatments affecting calcium homeostasis or cellular metabolism can alter MgIP development. We conclude that MgIP may be useful in studying bitter pit initiation and development.

An experiment was conducted to determine if prestorage heat treatments could attenuate bitter pit development on Northern Spy and superficial scald on Red Delicious and Law Rome apples. Attenuation of physiological disorders by heat treatment beyond that attainable by CA and low temperature was not apparent. Heat treatments were effective in improving flesh firmness retention.

Dedicated to the memory of my Mom,
Helen C. Burmeister

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Guidance Committee:

The journal paper format was chosen for this thesis in accordance with departmental and university regulations. The thesis is divided into two chapters in which the first is accepted for publication in *Postharvest Biology and Technology*, the second has been accepted for publication in the *Journal of Agricultural and Food Chemistry*.

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LITERATURE REVIEW

Description of Bitter Pit.

Bitter pit is in a group of physiological disorders affecting apples and pears known as corking disorders. These include bitter pit, corky core, cork spot, and crinkle. Faust and Shear (1968) have reviewed corking disorders. Bitter pit is characterized by sunken lesions that may develop on fruits as they mature on the tree or after harvest during storage. The tissue below the skin in the pitted area becomes discolored and dehydrated, resembling cork. When the lesions develop while the fruit are on the tree, the disorder is sometimes referred to as tree pit. Bitter pit is distinguished from other corking disorders in that lesions are mostly located in the outer cortex just below the skin. The affected tissue is softer than adjacent healthy tissue and it occurs at harvest or during storage.

Factors affecting bitter pit incidence.

The first reports of bitter pit were over 100 years ago by Jager (1889) as cited by Faust and Shear (1968). The disorder has been of commercial concern since the early 20th century and factors affecting its occurrence have been well documented (Faust and Shear, 1968; Ferguson and Watkins, 1989). Environmental factors and cultural practices that contribute to increased bitter pit incidence include: light cropping, Ca^{2+} deficiency, excessive tree vigor, excessive N nutrition, and moisture stress. Fruit harvested immature are also prone to the disorder.

Susceptibility to bitter pit varies among cultivars and geographically. A cultivar can be considered susceptible in one region of a country or state and not

in another. For example, Smock and Neubert (1950) reported that Golden Delicious when grown in the U.S. had no susceptibility. It is now known that immature Golden Delicious can be prone to develop bitter pit and this is common in the Pacific NW. Bitter pit has also been reported on Golden Delicious in Australia, New Zealand, and South Africa (Ferguson and Watkins, 1989 and references therein).

Mineral Content of the Fruit.

Of the factors contributing to bitter pit incidence, Ca^{2+} deficiency of the fruit has emerged as the crucial factor. Delong (1936) was the first to observe that low fruit Ca^{2+} levels correlated with bitter pit development in storage. Garman and Mathis (1956) determined that spraying trees with Ca^{2+} salts to fruit during development reduced bitter pit incidence. Since that time, the effects of environment, cultural, and chemical treatments on the mineral content of apple fruit in relation to bitter pit have been thoroughly investigated. In addition, high levels of Mg^{2+} and/or K^+ may also predispose fruit to bitter pit. Perring (1986) found that few apples were pitted in years when Ca^{2+} levels were high even if their Mg^{2+} and K^+ levels were high. When Ca^{2+} levels were marginal, more of the bitter pit afflicted fruit had high K^+ and Mg^{2+} levels. Thus, a direct relationship between K^+ or Mg^{2+} alone and bitter pit is often not detected especially if native calcium levels are high, but high ratios of K^+ , and/or Mg^{2+} , to Ca^{2+} are associated with bitter pit frequently (Webster and Forsyth, 1979; Hopfinger and Pooviah, 1979; Martin *et al.* 1960).

Studies attempting to relate levels of N, P, B, Na, Zn, and Cu to bitter pit incidence have been reviewed in detail (Faust and Shear, 1968; Ferguson and Watkins, 1989). No consistent relationships have been found indicating a critical role for these elements in bitter pit.

Distribution of minerals within fruit.

Complicating the Ca^{2+} /bitter pit relationship is the fact that mineral constituents of apple fruit are not uniformly distributed (Wilkinson and Perring, 1961; Perring and Wilkinson 1965; Faust *et al.* 1967). The Ca^{2+} concentration is higher in the stem than in the calyx end of the fruit, higher in the core and skin than in the cortex, and concentrations can vary around the fruit axis. Transverse distribution of K^{+} and Mg^{2+} exhibits a similar pattern of distribution but there is no longitudinal gradient.

Considerable effort has been expended to relate changes in mineral distribution during storage with bitter pit development. Translocation of Ca^{2+} from the core to the outer cortex of apples during storage has been reported (Ferguson and Watkins, 1983; Bramlage *et al.* 1979; Perring, 1986). Information concerning the movement of other ions is conflicting. In Spartan apples, an accumulation of K^{+} P and Mg^{2+} in the inner cortical zones was recorded (Perring, 1984), whereas an outward movement of K^{+} and Mg^{2+} has been recorded for Cox,s Orange Pippin (Ferguson and Watkins, 1983). In the latter case, the authors surmised that ions may have been concentrated in the outer cortex by water losses resulting in the appearance of migration. Perring (1989) proposed that the

pattern of redistribution of minerals may depend on variety. Perring (1986) and Perring and Pearson (1986) associated an irregular transient withdrawal of Ca^{2+} from the mid- and outer-cortical regions during storage just prior to the appearance of bitter pit lesions. This withdrawal of Ca^{2+} was less pronounced in samples that developed less bitter pit.

Prediction.

Another approach to reduce losses is to predict the potential of fruits to develop bitter pit using mineral analysis. This is a useful commercial practice (Ferguson and Watkins, 1989). Analysis of fruit Ca^{2+} or the ratio of K^+ , Mg^{2+} , $\text{K}^+ + \text{Mg}^{2+}$ to Ca^{2+} is commonly employed in these predictive methods. Fruits are sampled from orchards prior to harvest and analyzed for Ca^{2+} . Lots of low Ca^{2+} fruits can be identified and marketed or processed prior to development of bitter pit. Fruits from orchards with a high Ca^{2+} status can normally be safely held for long durations in storage.

Although the relationship between bitter pit and Ca^{2+} is highly variable, thresholds for Ca^{2+} where minimal or no bitter pit may be expected have been established (Wills *et al.* 1976). In one study, the correlation coefficient (r) ranged from 0.37-0.90 for the Ca/bitter pit relationship (Wills *et al.* 1976). Thresholds vary depending on the tissue (eg. whole fruit, cortical plugs, or skin), variety, growing region and season. Some of the variation in thresholds may reflect different sampling and analysis procedures. Bitter pit incidence can vary greatly, even at Ca^{2+} levels below threshold. This makes accurate prediction impossible.

Attempts to reduce variability by expressing Ca^{2+} on a different basis (eg. mg Ca per cell surface) or use of fractionating tissue into soluble and insoluble fractions have been unsuccessful (Perring, 1986). Incorporation of other orchard factors (eg. fruit size, tree age, cropping level) with mineral analysis is used to improve prediction of bitter pit occurrence in storage (Holland, 1980). The practicality of such predictive methods is limited due to the inherent variability of bitter pit incidence within and among orchards, varieties and growing seasons (Wills *et al.* 1976).

For a predictive system to be practical, results must be known well enough in advance of the harvest to allow time for management decisions. The relationship of Ca^{2+} to bitter pit has been shown to be reliable up to 3 weeks before harvest (Ferguson *et al.*, 1979). Analysis of young fruitlet midway in the growing season has also proven to be useful (Marcelle, 1989). However, fruit Ca^{2+} analysis is expensive, time consuming, and requires specialized equipment. Also, since there is a high variability among fruits within an orchard, large sample sizes are necessary to assess bitter pit potential. Waller, (1980) suggested a minimum of 30 fruit, preferably 80 to accurately assess calcium levels in orchards. An alternative method of assessing fruit Ca levels would be useful.

Control.

Generally, attempts to control bitter pit under commercial conditions by cultural practices such as soil Ca^{2+} application, summer pruning (to increase fruit/leaf) ratio, and delayed harvesting, are only marginally effective (Perring,

1979; Boon, 1980). The primary methods employed in attempts to control the disorder are: applying Ca^{2+} sprays to trees at intervals throughout the growing season (Jackson, 1962; Cooper and Bangerth, 1976) using post-harvest dips or drenches in Ca^{2+} solutions (Mason and Drought, 1975), or infiltrating Ca^{2+} solutions into fruit (Scott and Wills, 1979; Cooper and Bangerth, 1976). Control of bitter pit by Ca^{2+} treatments is highly variable and not always complete (Ferguson and Watkins, 1989).

Storage Conditions.

Bitter pit have been reduced by low temperature (Perring, 1986) and CA storage (Hewett, 1984; Sharples and Johnson, 1987). High water loss during storage is sometimes associated with increased bitter pit (Hewett, 1984; Scott and Wills, 1979). The benefits of CA storage and postharvest Ca^{2+} treatments are additive (Hewett, 1984; Webster and Forsyth, 1979; Johnson, 1979). CA storage lowered the threshold Ca^{2+} level where no bitter pit was expected (Sharples and Johnson, 1987). If ripening and metabolism are suppressed for long periods by low O_2 storage, irreversible attenuation of bitter pit may result (Ferguson and Watkins, 1989; Perring, 1986).

Structural and Biochemical Differences in Affected Tissue.

The site of bitter pit is associated with the ends of the vascular bundles (Smock and Van Doren, 1937). Its structural development is distinguished from other corking disorders by absence of abnormal growth of cells such as observed in cork spot (Simons *et al.*, 1971). It is believed this difference could be in the

timing of the disorder. Bitter pit occurs just before harvest, or during storage when cells may be unable to initiate cell division (Faust and Shear, 1968). Pectin proturbances into intercellular spaces have been observed (Simons, 1962). The few structural studies of bitter pit development are of fruit that develop symptoms while still on the tree (Ferguson and Watkins, 1989). A systematic study of structural development of the anatomical aspects of bitter pit in storage has not been accomplished.

During bitter pit development, there appears to be a general migration of mineral and organic constituents into the pitted tissue. Pitted tissue is higher in starch, glucose, and fructose, but low in sucrose and higher in N and pectin than adjacent healthy tissue (Askew *et al.*, 1960; Faust, *et al.*, 1968a). Citric acid is the main organic acid in pitted tissue, while malic acid was the predominant acid in healthy tissue (Steenkamp and de Villiers, 1983; Faust and Shear, 1968). In studies of Jonathan spot, another Ca^{2+} related disorder, Richmond *et al.*, (1964) demonstrated movement of minerals into the affected area which was associated with a higher level of total organic acids, mainly malic acid, in the affected tissue.

Consistently, the pitted zones of apple fruit are higher in Ca^{2+} and Mg^{2+} than non-pitted tissue (Steenkamp and de Villiers, 1983; Meyer *et al.* 1979). High $\text{Mg}^{2+}/\text{Ca}^{2+}$ (Hopfinger and Poovaiah, 1979; Hopfinger *et al.* 1984) have been reported. Ford (1979) demonstrated that $^{45}\text{Ca}^{2+}$ moved into the pitted area as symptoms developed.

Cork spot, unlike bitter pit has a red color formation on the skin above the

incipient lesion (Faust and Shear, 1968b). In etiological studies of cork spot, ethylene production followed by a respiration rise were the first signs of the onset of the disorder. This was followed by an increase protein synthesis, pectin synthesis, and movement of ions into afflicted tissues. Acetate was the major respiratory substrate utilized in the pitted tissue. They hypothesized the acetate and ethylene were breakdown products of membrane fatty acids.

The mineral and biochemical changes documented for bitter pit and other corking disorders are symptomatic of other diseased and mechanically injured tissue. It is hypothesized that these differences between pitted and healthy tissue are not related to the initiation, but are the result of the metabolic disturbance and subsequent tissue breakdown (Ferguson and Watkins, 1989; Faust and Shear, 1968).

Calcium and Fruit Ripening.

Calcium's influence on apple ripening is acknowledged, but not understood. Fruit with a high Ca^{2+} content ripen at a slower rate than fruits with lower flesh Ca^{2+} levels. This has been demonstrated by comparisons of respiration rates of fruits within a range of Ca^{2+} contents (Faust and Shear 1972; Bramlage *et al.*, 1974), or between fruit with raised Ca^{2+} content by sprays during development on the tree (Cooper and Bangerth 1976). Bramlage *et al.* (1974) suggested that Ca^{2+} reduced the rate, but not the timing of the respiratory climacteric in apples.

Red Delicious apples of low Ca^{+2} levels entered the ethylene climacteric earlier than fruits with high Ca^{+2} (Tomala and Dilley, 1989). Avocado treated

with Ca^{+2} had both a depressed and a delayed ethylene climacteric (Tingwa and Young, 1974). Ca^{+2} reduces ethylene production, and delays ripening and subsequent softening of apples. This effect was mimicked by Sr^{+2} , but was not seen with K^{+2} , or Mg^{+2} infiltration. Ida Red apples infiltrated with Ca^{+2} had reduced ethylene, ACC content and ethylene forming enzyme (EFE) activity (Tomala and Dilley, 1990). It is not known how Ca^{2+} or other cations affect respiration, ethylene production and ripening. Ca^{+2} may diminish respiration by decreasing membrane permeability thereby reducing the diffusion of respiratory substrates from the vacuole to centers of metabolic activity in the cytoplasm (Faust and Klein, 1974; Bangerth *et al.* 1972; and Cooper and Bangerth 1976). Direct effects of Ca^{2+} on respiration and ethylene production have not been demonstrated (Ferguson, 1984). It has been suggested that calcium may be required for some ethylene-dependent plant processes (Raz and Fluhr, 1992).

Calcium as a second messenger.

Ca^{2+} is as an essential element in plant nutrition, cell wall structure, and membrane structure and function (Bangerth, 1979; Ferguson and Drobak, 1988; Pooviah, 1988). Also Ca^{2+} 's role as a second messenger mediating many plant responses to external stimuli is now well accepted (Heplar and Wayne, 1985; Poovaiah and Reddy, 1987). Ca^{2+} mediates plant responses through changes in cytosolic Ca^{2+} concentration ($[\text{Ca}^{2+}]_{\text{cyt}}$). External stimuli (e.g. light, growth regulators, gravity) imposes an action potential that results in the release of Ca^{2+} into the cytosol either by an influx through voltage-gated Ca^{2+} channels in the

plasma membrane, or by releasing Ca^{2+} from intracellular organelles such as the vacuole, endoplasmic reticulum, chloroplast, and mitochondria (Wayne, 1993). This increase in $[\text{Ca}^{2+}]_{\text{cyt}}$ is transient and can be very localized (Cheek, 1989). Ca^{2+} then binds to calmodulin, or other Ca^{2+} binding proteins which then leads to a physiological response. This may involve the phosphorylation or dephosphorylation of certain enzymes and possibly gene expression.

Low $[\text{Ca}^{2+}]_{\text{cyt}}$ is maintained against a concentration gradient (1000 fold) between the cytosol and other cell compartments (vacuole and cell exterior). Resting values for $[\text{Ca}^{2+}]_{\text{cyt}}$ are between $0.1\text{-}1.0\mu\text{M}$ for all plants in which this has been measured (Poovaiah and Reddy, 1987). Maintenance of low $[\text{Ca}^{2+}]_{\text{cyt}}$ is believed to be primarily accomplished by an active efflux of Ca^{2+} from the cytosol via Mg^{2+} dependent Ca^{2+} activated ATPase located in the plasma membrane, and by the similar uptake ATPases located in the membranes of the intracellular organelles (Moore *et al.* 1984). The Ca^{2+} ATPase of the plasma membrane is believed to play a major role in plant homeostasis (Briskin, 1990; Carfoli, 1987).

In plants, measuring $[\text{Ca}^{2+}]_{\text{cyt}}$ is problematic because of the presence of cell walls, large vacuoles, chloroplasts, and high cellular turgor pressure. Harker and Venis (1991) have measured intracellular and extracellular free calcium in apple fruit using calcium-sensitive microelectrodes. Because of the highly vacuolated nature of apple cells only one successful $[\text{Ca}^{2+}]_{\text{cyt}}$ measurement was accomplished. The $[\text{Ca}^{2+}]_{\text{cyt}}$ was verified to be $0.05\mu\text{M}$. The apoplastic and vacuolar $[\text{Ca}^{2+}]$ ranged from $0.02\text{-}1.3\text{mM}$ and $0.06\text{-}1.0\text{mM}$, respectively. The total calcium in the

apple tissue was positively correlated with the apoplastic free calcium. Assuming that 90% of the cell weight is vacuole, they calculated that the vacuole accounted for 40% of the total calcium in apple tissue. As much as 60% is thought to be in the cell wall and associated apoplastic space (Demarty *et al.* 1984). These results concur with other measurements (Miller and Sanders, 1987; Felle, 1988).

Initiation and Development

Although mineral composition is a major factor, it is not considered the cause of the bitter pit disorder. Early theories concerning the cause of bitter pit include excessive transpiration, necrosis of immature cells filled with starch, toxicity to sprays, and viral infection (Faust and Shear, 1968). The cause(s) and mechanism of initiation and development of bitter pit are not known. Currently, bitter pit is thought to result from a localized Ca^{2+} deficiency or mineral imbalance, but there is no direct evidence for this (Ferguson and Watkins, 1989; Perring, 1986). Mineral analysis has only been conducted at the tissue level. Since it has not been possible to identify sites on fruit where pits might develop and measurements of the relevant pools of Ca^{2+} in apple fruit are scarce, the etiology of bitter pit is unclear. Many theories have been proposed to account for development of the disorder.

Simon (1978) proposed that exogenous water in the intercellular spaces of the fruit may cause cells to swell. Because of Ca^{2+} deficiency they lose permeability burst and dry out. This is not compatible with the observation that high humidity conditions attenuate bitter pit (Hewett, 1984). Ferguson and

Watkins (1989) point out that the discrete nature of the disorder would require localized changes in water relations.

Bangerth (1979) proposed that Mg^{2+} , K^+ , H^+ and organic acid are antagonistic to membrane function. That high levels of these ions compete for Ca^{2+} binding sites thereby increasing membrane permeability and leading to loss of cell function. Steenkamp and de Villiers (1983) suggested the role of Ca^{2+} in preventing bitter pit was to protect membranes by chelating citric and oxalic acid which they demonstrated dissolved the middle lamellae of apple fruit cells.

The suggestion of Perring (1986) that a localized Ca^{2+} deficiency resulting from withdrawal of Ca^{2+} from the outer cortex accounts for development in storage needs further study (Ferguson and Watkins, 1989). Due to the high variation between samples, data are often expressed as percent of the total so quantitative relationship between redistribution of minerals and bitter pit has not been established (Fergusson and Watkins, 1989). Transverse sections used do not account for longitudinal movements of Ca^{2+} . Storage conditions that would attenuate bitter pit such as CA storage did not alter Ca^{2+} redistribution (Perring, 1984; Perring and Pearson, 1987).

Ferguson and Drobak, (1988), Ferguson and Watkins (1989), and Ferguson (1990) have considered the various cellular compartments of Ca^{2+} . Given: 1) reliable relationships between Ca^{2+} and bitter pit are detectable; and 2) the major compartments of Ca^{2+} in cells are the apoplast and vacuole (Harker and Venis, 1991), it would follow that one or both of these compartments may be critical to

bitter pit development. It is thought that the vacuole due to its low surface area and large volume may provide the rapid transient rises of $[Ca^{2+}]_{\text{cyt}}$ necessary for cellular function (Ferguson, 1990). It has been proposed (Ferguson and Drobak, 1988) that the extracellular Ca^{2+} directly accessible to the plasma membrane is the critical pool involved in bitter pit development. This pool of extracellular Ca^{2+} directly accessible to the plasma membrane is where exogenous Ca^{2+} is suspected to exert its effect (Ferguson and Watkins, 1981). It is believed that enough Ca^{2+} is usually present to maintain the structure and function of membranes. However, during certain conditions that would require transient rises in $[Ca^{2+}]_{\text{cyt}}$ for signal transduction, the Ca^{2+} availability in this pool may become limiting. The cell may then be unable to respond to external stimuli, which in turn could lead to cellular dysfunction (e.g. bitter pit). This could involve the failure to maintain or to prevent certain metabolic events.

In this scheme, low native $[Ca^{2+}]$ is not always detrimental unless exacerbating conditions such as drought or cold shock are present. Thus, a particular Ca^{2+} level in fruit may be adequate under certain conditions, but may not be adequate when there is an increased demand for Ca^{2+} for cellular activity (eg. ripening). This proposal includes the role of Ca^{2+} as a second messenger and accounts for several of the observations concerning bitter pit development in apple including: 1) the variation in disorder development due to the factors previously discussed; 2) that fruits of low total (native) Ca^{2+} content do not necessarily develop the disorder; 3) fruits of varying maturities with similar Ca^{2+}

contents differ in susceptibility; 4) irreversible attenuation can be affected by application of CA; 5) increasing Ca^{2+} contents artificially in the extracellular space can attenuate bitter pit and affect intracellular events associated with ripening and senescence (Glenn *et al.*, 1988).

The apparent antagonism of bitter pit by Mg^{2+} has been previously discussed. Fruit with bitter pit are often found to have high $\text{Mg}^{2+}:\text{Ca}^{2+}$ ratios (Perring, 1986). Mg^{2+} is a necessary cofactor for many enzymatic reactions. There are examples of Ca/Mg interactions in cellular functions. The intracellular levels of Mg^{2+} are thought to be maintained in the mM range. Measurements of Mg^{2+} specific current (I_{Mg}) in *Paramecium* indicate free Mg^{2+} concentrations were in the range of 0.1-0.7mM. It is suggested that most of the intracellular Mg^{2+} exists in a bound form (White and Hartzell, 1989). Activation of the I_{Mg} was found to be Ca^{2+} dependent (Preston, 1990). Sensitivity of Ca^{2+} channels to blockage by extracellular Mg^{2+} has been demonstrated in N-methyl-D-Aspartate (NMDA) receptor of nerve cells in vertebrates (Burnashev *et al.*, 1992). In apple tissue, Harker *et al.* (1989) found that Mg^{2+} inhibited transport of Ca^{2+} in the apoplast. Thus, Mg^{2+} may have a cellular role for in bitter pit development.

Bitter pit-like symptoms (Mg^{2+} induced pit) have been induced on apples by Mg^{2+} by infiltration treatments whereas other divalent and monovalent cations have not shown this effect (Conway and Sams, 1987; Hopfinger *et al.*, 1984; Fallahi *et al.*, 1987). Researchers disagree as to whether Mg^{2+} induced pits are the same as the natural bitter pit disorder. It has been suggested that the Mg^{2+}

induced pits result from breakdown caused by Mg^{2+} toxicity at the entry point into the fruit (Ferguson and Watkins, 1989). Also, Mg^{2+} is known to stimulate polyphenol oxidase activity (an enzyme involved in the browning reactions associated with the death of cells) in fruits vacuum infiltrated with Mg^{2+} (Hopfinger *et al.*, 1984). Fallahi *et al.*, 1987 compared Mg^{2+} induced pits to natural bitter pit and concluded that induced pits lacked corky tissue. We have demonstrated that these Mg^{2+} induced pits are similar to naturally occurring bitter pit (Burmeister and Dilley, 1991). Addition of Ca^{2+} to the infiltration media attenuates Mg^{2+} induced pit and the number of bitter pit-like lesions induced by Mg^{+2} is inversely related to the endogenous (native) Ca^{2+} concentration of individual fruits. Using these Mg^{2+} induced bitter pit-like lesions, as a model for bitter pit may provide new insight into the nature of the bitter pit disorder and provide means to assess bitter pit potential.

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Chapter 1

**Correlation of Bitter Pit on Northern Spy Apples with Bitter Pit-like Symptoms
Induced by Mg^{2+} Salt Infiltration.**

ABSTRACT:

Induction of bitter pit-like symptoms (Mg^{2+} induced pits [MgIP]) on Northern Spy apples (*Malus domestica* Borkh.) by infiltrating Mg^{2+} salt solutions into the fruit was positively correlated with bitter pit that developed naturally during storage. Fruit at harvest were infiltrated with 0.1M MgCl_2 in 0.3M sorbitol with 0.1% Tween 20 and placed at 20°C for 10 days (d) after which the number of MgIP was determined on individual fruits. A parallel sample of fruits from each orchard was stored at 5°C in air (both years) and at 3°C in air or controlled atmosphere (CA) storage in the second year. MgIP was positively correlated with bitter pit development in storage for fruits harvested 20d and 10d before and at optimal maturity for long term storage. The endogenous (native) fruit Ca^{2+} concentration was inversely related to MgIP and to bitter pit development following storage.

INTRODUCTION:

Bitter pit is a physiological disorder affecting apples. It is characterized by sunken lesions that may develop on fruits as they mature on the tree or during storage. The tissue below the skin in the pitted area becomes discolored and dehydrated (Faust and Shear, 1968). Certain environmental and cultural conditions contribute to increased bitter pit incidence including: light cropping, Ca^{2+} deficiency, excessive tree vigor, excessive N nutrition, and moisture stress. Of these, calcium deficiency is the major factor that has been associated with bitter pit (DeLong 1936; Garman and Mathis, 1956; Ferguson and Watkins, 1989).

Low fruit Ca levels consistently correlate with high levels of bitter pit (Ferguson *et al.*, 1979; Wills *et al.*, 1976; Waller, 1980). The primary methods employed in attempts to control the disorder are: applying Ca^{2+} sprays to trees at intervals throughout the growing season, using post-harvest dips or drenches in Ca^{2+} solutions, or infiltrating Ca^{2+} solutions into fruit (Scott and Wills, 1979; Cooper and Bangerth, 1976). Control of bitter pit by Ca^{2+} treatments is highly variable and not always complete.

Another approach employed to reduce losses from bitter pit is to use predictive methods based upon mineral analysis to determine the fruits potential to be afflicted with the disorder. This has been shown to be useful and is practiced commercially (Ferguson and Watkins, 1989). Analysis of fruit Ca^{2+} is commonly employed for these predictive methods. Fruit are sampled from orchards prior to harvest and analyzed for Ca. Orchards with low Ca fruits can thereby be segregated for early marketing or the fruits can be processed prior to development of the bitter pit disorder. Fruit from orchards with a high Ca^{2+} status can normally be safely held for long durations in storage. Predictive systems are in practice that use fruit Ca^{2+} analysis alone, or in combination with other orchard factors (eg. fruit size, tree age, cropping level), to predict bitter pit occurrence in storage (Holland, 1980). The practicality of such predictive methods is limited due to the inherent variability of bitter pit incidence within and among orchards, varieties and growing seasons (Wills *et al.*, 1976). Timeliness is important. For any predictive system to be practical, the result must be known

well enough in advance of the harvest to allow time for management decisions. The relationship of Ca^{2+} to bitter pit has been shown to be reliable up to 3 weeks before harvest (Ferguson *et al.*, 1979). Analysis of young fruitlets midway in the growing season has also proven to be useful (Marcelle *et al.*, 1989). However, fruit Ca^{2+} analysis is expensive, time consuming, and requires specialized equipment. Also, because of the high variability among fruits within an orchard, large sample sizes are necessary to assess bitter pit potential (Waller, 1980). An alternative method of assessing fruit Ca levels would be useful. Researchers have reported the induction of bitter pit-like symptoms (MgIP) on apples within 7d of Mg^{2+} treatment by infiltration (Conway and Sams, 1987; Hopfinger *et al.*, 1984; Fallahi *et al.*, 1987). We have demonstrated that MgIP are similar to naturally occurring bitter pit (Burmeister and Dilley, 1991). Addition of Ca^{2+} to the infiltration media attenuates MgIP and the number of MgIP is inversely related to the endogenous (native) Ca^{2+} concentration of individual fruits. If orchards of a low Ca^{2+} status have fruit more susceptible to MgIP then it may be possible to assess bitter pit potential in a relatively quick and inexpensive manner. The results reported herein are from a 2-year study correlating susceptibility to MgIP with the occurrence of bitter pit in storage.

MATERIALS AND METHODS:

Orchards were selected that exhibited a range of bitter pit susceptibilities based on orchard history. The number of orchards sampled, locations and harvest dates varied between years. Orchards from the eastern, western, and

southwestern districts in Michigan were used in this study.

1990 Studies. Northern Spy and Red Delicious cultivars were employed. Apples from 5 locations were harvested at ca. 1 week prior to, and from 11 locations at maturity (ca. 10% of fruits at $> 0.2 \mu\text{l l}^{-1}$ internal ethylene concentration). Samples of 10-15 fruit from each of 20 trees were randomly selected at each location and were divided into two lots. One lot was vacuum infiltrated (absolute pressure of 100mmHg for 2 min.) while submersed in 0.1M MgCl_2 (50 fruit) in 0.3M sorbitol as osmoticum with 0.1% Tween 20 as surfactant. The other lot was infiltrated with 0.3M sorbitol 0.1% Tween 20 which served as the control (30 fruit). The fruits were held for 10d at 20° C, and the number of MgIP lesions on individual fruits was recorded. None of the fruits infiltrated with 0.3M sorbitol with Tween 20 developed pits (data not shown). Parallel samples of fruit not infiltrated were stored in perforated polyethylene bags in air at 5° C for 5 months after which the number of naturally-occurring bitter pit lesions on each individual fruit was recorded. Individual fruits minus the core were analyzed for calcium by atomic absorption spectrometry according to Tomala and Dilley, (1989). Calcium data are expressed as ppm dwt ($\mu\text{g g}^{-1}$ dry wt).

1991 Studies. Only the Northern Spy cultivar was employed. Five orchards were sampled ca. 20d, and 6 orchards at 10d prior to and at optimal maturity for long term storage (Dilley and Dilley, 1985). Procedures were as previously described except calcium analysis was conducted using a bulked sample consisting of longitudinal sections (1/16 of each fruit excluding seeds) taken from 40 fruit

from each location/harvest combination. Storage treatments included air at 3°C, and controlled atmosphere (CA) of 3.0% O₂ + 3.0% CO₂ at 3°C for 6 months. Fruit were held in 3°C air for 7d before CA was applied. Following storage fruit were held for 7d at 20°C before assessment of bitter pit. These are the commercially recommended storage conditions for Northern Spy in Michigan.

Data were analyzed by regression of MgIP vs bitter pit occurring during storage. MgIP and bitter pit were assessed by determining the average number of pits/fruit and the % of fruits with bitter pit. Calcium data were analyzed by regression of endogenous (native) Ca vs mean number of MgIP, or % of fruit with bitter pit that developed during storage.

RESULTS AND DISCUSSION:

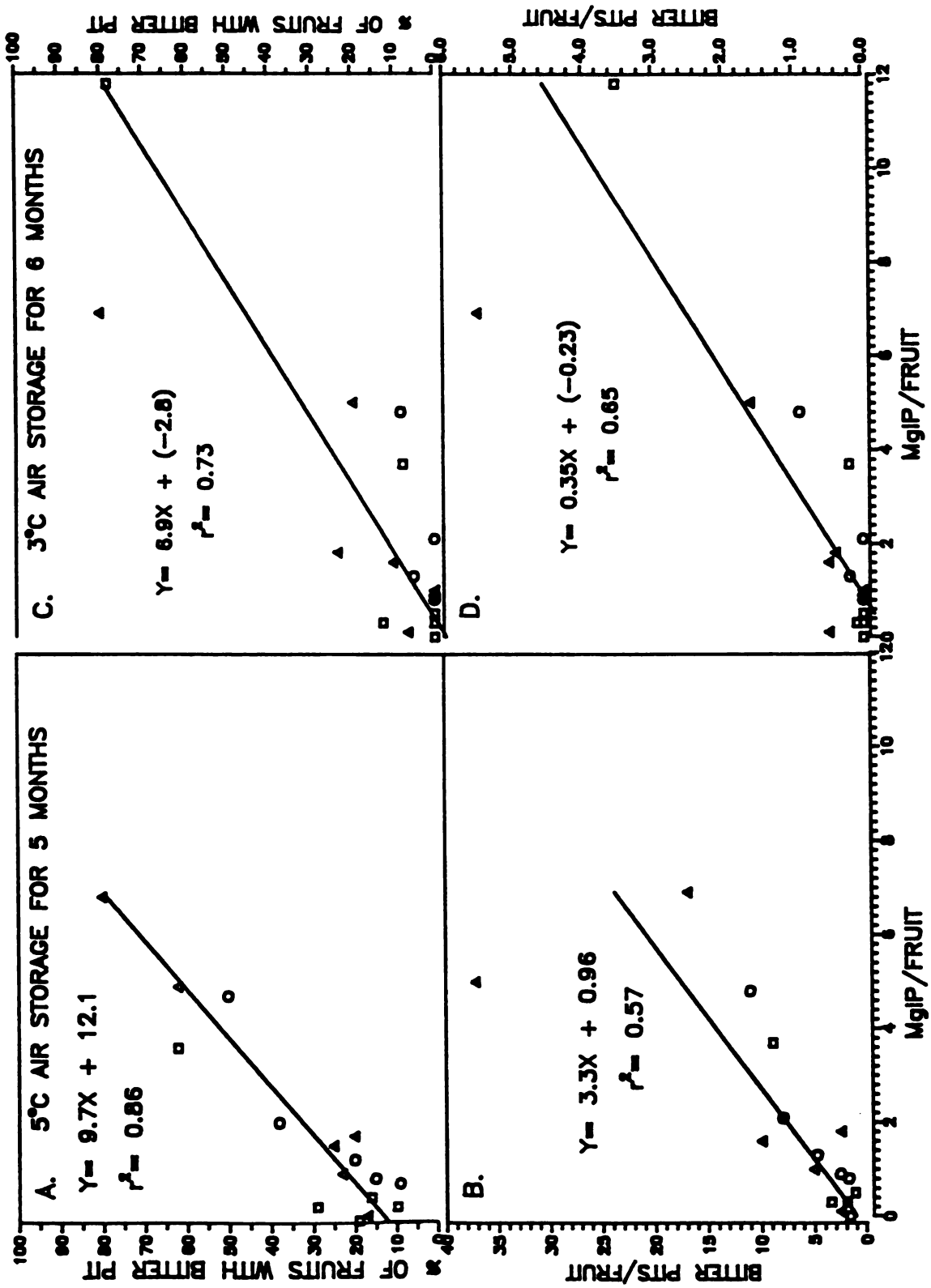
For Red Delicious, incidence of MgIP and natural bitter pit development were low for all locations sampled. This precluded development of prediction equations for this variety. This may reflect a high calcium status of these blocks, but this was not verified. Only the results for Northern Spy are presented.

1990. For the Northern Spy harvested early there was no correlation between MgIP and natural bitter pit (see appendix 2, Fig. 1). For late harvest 1990, severity (bitter pits/fruit) and incidence (% of fruits with bitter pit) was positively correlated with bitter pit after storage; $Y = 1.76X + 1.2$, $r^2 = 0.69$ and $Y = 6.7X + 20.0$, $r^2 = 0.57$, respectively. In 1991, severity and incidence was positively correlated with the severity of MgIP for all harvests for both the 5°C (Fig. 1, A and B) and 3°C (Fig. 1 A and C) storage temperatures. Maturity of

fruit at harvest can be an important factor in subsequent development of bitter pit (Ferguson and Watkins, 1989). We monitored fruit maturity in the 1991 season to determine if maturity at harvest affected the correlation of bitter pit with MgIP. An effect of harvest date on the correlation was not apparent (Fig. 1) but due to weather conditions there was not as much variation in maturity between districts as normally would be expected in a typical season (data not shown). Bitter pit levels were higher for fruits stored in air at 5°C (Fig. 1, A and B) than stored in air stored at 3°C (Fig. 1, C and D). Regressions for the 3°C CA yielded similar prediction equations; severity vs severity and incidence were; $Y = 0.32X + (-0.27)$, $r^2 = 0.69$ and $Y = 5.7X + (-5.6)$, $r^2 = 0.73$, respectively. All regressions were significant at $P < 0.01$. In most instances, severity and incidence of bitter pit were lower in the 3°C CA treatment. We attribute this to a lower rate of metabolism of fruits in CA. It has been demonstrated that postharvest conditions that slow ripening and metabolism can reduce bitter pit levels (Perring, 1986).

The endogenous (native) Ca was inversely related to the average number of MgIP/fruit and bitter pits/fruit that developed during storage (data not shown). This type of relationship is considered typical when relating bitter pit to fruit calcium levels. The exceptions to the relationship being fruit that are low in Ca do not necessarily develop bitter pit (Ferguson and Watkins, 1989). In this study, endogenous Ca levels below ca. 200 $\mu\text{g/g}$ dwt., a range of bitter pit levels can be expected from 0 - >35 pits/ fruit depending on storage temperature (see appendix 2, Fig. 2 and 3, respectively).

Figure 1. Correlation between MgIP/fruit at harvest and natural bitter pit development on Northern Spy apples during storage in air. Fruit were harvested 20d (O) and 10d (□) prior to and at optimal maturity (Δ). Each data point represents the average of 50 fruits. All regressions were significant at $P < 0.01$. For 5°C storage, A) MgIP/fruit vs % of fruits with bitter pit and B) MgIP/fruit vs bitter pits/fruit and for 3°C storage, C) MgIP/fruit vs % of fruits with bitter pit and D) MgIP/fruit vs bitter pits/fruit.



In utilizing MgIP to predict bitter pit some of the same problems encountered with the use of Ca analysis are evident. The relationship between bitter pit and Ca is highly variable. This makes accurate prediction of levels of bitter pit impossible (Perring, 1986). However, threshold levels of Ca where minimal or no bitter pit may be expected have been established (Wills *et al.*, 1976). Therefore, high variation does not necessarily preclude the use of MgIP as a method to assess bitter pit potential. The correlation coefficient (r^2) values presented here are similar in magnitude to others used for bitter pit prediction (Holland, 1980; Waller, 1980). Moreover, in no instance was a high level of MgIP associated with a low level of bitter pit. For fruits stored at 3°C, our data indicate when MgIP is below ca. 5 pits/fruit, bitter pit incidence was below 25% and number of bitter pits/fruit 2.0 (Fig. 1, C and D). Prediction of % incidence of bitter pit from % incidence of MgIP was highly correlated ($r^2 = 0.81$) (see appendix 2, Fig. 4).

Developing a practical prediction scheme for physiological disorders such as bitter pit is problematic. Many issues need to be considered. We sampled fruit from orchards with a wide range of tree ages, fruit maturities, and environmental, and cultural conditions. We observed a good correlation between MgIP and natural bitter pit occurrence up to 20d prior to the commercially recommended harvest date. This time frame is desirable to provide lead time to segregate fruit lots to be processed immediately from those to be processed after long-term storage. A MgIP/bitter pit relationship could also be useful to indicate

the need for additional Ca^{2+} treatment by postharvest dips. In order to determine if the relationship we have found between the propensity of fruits for MgIP and occurrence of bitter pit during storage may be generally applicable, experiments need to be conducted with various cultivars under a variety of storage temperature and atmosphere conditions. Postharvest handling and storage practices can affect bitter pit occurrence. For example, a delay in establishing CA conditions of greater than 7d promotes ripening during storage and can have a detrimental effect by promoting the development of physiological disorders such as bitter pit (Fica *et al.*, 1985).

Assessment and prediction of bitter pit has been reviewed (Ferguson and Watkins, 1989). It is unclear from the published reports whether low % incidence of bitter pit represent a few fruit with light or severe symptoms or, if a high average number of pits/fruit is the result of a few fruit with severe symptoms. Any amount of pitting on an individual fruit destroys its fresh market (dessert) value. In the case of the pie slice industry in Michigan both bitter pit incidence and severity are considered important in regard to processing Northern Spy apples (Geisler, 1992, personal communication, Coloma Fruit, Inc.). Fruits visually estimated to have >15 pits after peeling are culled. Apple slices affected with bitter pit can be removed during processing, or the pitted area peeled away, but extremely high incidence of pitting slows the processing of apples destined to secondary markets.

The correlation between MgIP and development of bitter pit in storage

extends our previous observations (Burmeister and Dilley, 1991) that MgIP is similar to the bitter pit disorder. We believe that the method described has the potential to be a reliable and inexpensive means to assess fruits according to their potential for bitter pit. Can the assessment of MgIP at harvest be related to bitter pit development potential in a practical manner? Counting the number of fruit affected in a sample can be accomplished easily, rather than counting the number of pits on each individual fruit. Perhaps an index that relates incidence and severity such as used for superficial scald (Lurie *et al.*, 1989) would be useful.

We have recently demonstrated that treatments affecting calcium homeostasis (eg. calmodulin antagonists, Ca^{2+} chelators) can alter MgIP development (Burmeister and Dilley, 1993). We speculate that the MgIP development and the bitter pit disorder may share common mechanisms. Calcium is recognized to be a second messenger in plant cells in mediating cellular metabolism (eg. calmodulin, and Ca-ATPase) (Heplar and Wayne, 1985). Models have been proposed to explain how calcium might regulate cell metabolism (Ferguson and Drobak, 1988; Poovaiah, 1988). The cytosolic concentrations of free Ca^{2+} and Mg^{2+} are known to be maintained at the submicromolar and millimolar range, respectively. Ferguson (1990) has suggested that depletion of the extracellular pool of Ca^{2+} results in loss of the cells' capacity to respond to external stimulus resulting in cellular dysfunctions (eg. bitter pit).

Apple tissue susceptible to bitter pit has been shown to be low in Ca^{2+} , especially in relation to the Mg^{2+} level (Garman and Mathis, 1956). Harker *et al.*,

(1989) found that Mg^{2+} inhibits $^{45}Ca^{2+}$ transport across discs of cortical flesh of apple fruit. Gilroy *et al.*, (1989) reported that high extracellular Mg^{2+} and K^+ concentrations resulted in rapid breakdown of Ca^{2+} homeostasis of carrot protoplasts. Sensitivity of Ca^{2+} channels to blockage by extracellular Mg^{2+} has been demonstrated in animal systems (Burnashev *et al.*, 1992). Infiltration of Ca^{2+} in the extracellular space of apple fruit can attenuate bitter pit and affect intracellular events associated with ripening and senescence (Glenn *et al.*, 1988). Combined with these observations, our data suggest a more specific role for Ca^{2+} and Mg^{2+} in bitter pit development. We speculate that extracellular Mg^{2+} supplied by infiltration may affect the supply of Ca^{2+} in the apoplast of apple fruit influencing the cells ability to regulate cytosolic Ca^{2+} ; either by perturbation of a voltage-regulated Ca^{2+} channel or displacing Ca^{2+} from ionic binding sites in the apoplast. This could interfere with the role of Ca^{2+} as a second messenger involving a Ca^{2+} -ATPase, or Ca^{2+} /calmodulin linked phosphorylation of an enzyme, or a regulatory protein involved in cellular homeostasis, or metabolism. We believe Mg^{2+} exacerbates the potential for apple fruit to initiate the chain of reactions involved in expressing this Ca^{2+} related disorder.

The Relationship Between the Number of Bitter Pits per Fruit and % Incidence of the Bitter Pit Disorder.

Interpretation of research data on bitter pit and how this may relate to the practical significance of bitter pit in the commercial sector may yield different opinions. Assessment of bitter pit and other fruit storage disorders is in question. It is sometimes not clear from the published reports whether low % incidence of bitter pit represent a few fruit with light or severe symptoms or, if a high number of pits/fruit is the result of a few fruit with severe symptoms. Any amount of pitting on an individual fruit destroys its fresh market (dessert) value. In the case of processing apples as fresh slices for the bakery industry both bitter pit incidence and severity are considered important (Geisler, personal communication, Coloma Fruit, Inc. 1992). Fruits visually estimated to have >15 pits after peeling are culled. Apple slices affected with bitter pit can be removed during processing, or the pitted area pared away, but extremely high incidence of pitting slows the processing of apples destined to the bakery industry. It is important to have a basis to measure the bitter pit disorder for both commercial and research purposes.

Meaningful assessment of bitter pit is important in terms of any prediction scheme that might be employed. Counting the number of fruit affected in a sample can be accomplished easily, rather than counting the number of pits on each individual fruit. An numerical index that relates incidence and severity such

as used for superficial scald may be useful.

The relationship between number of pits per fruit and % incidence of bitter pit for Northern Spy stored at 3°C appears to be linear (Fig. 2). If this relationship is proven to be consistent between seasons severity could be predicted on the basis of incidence.

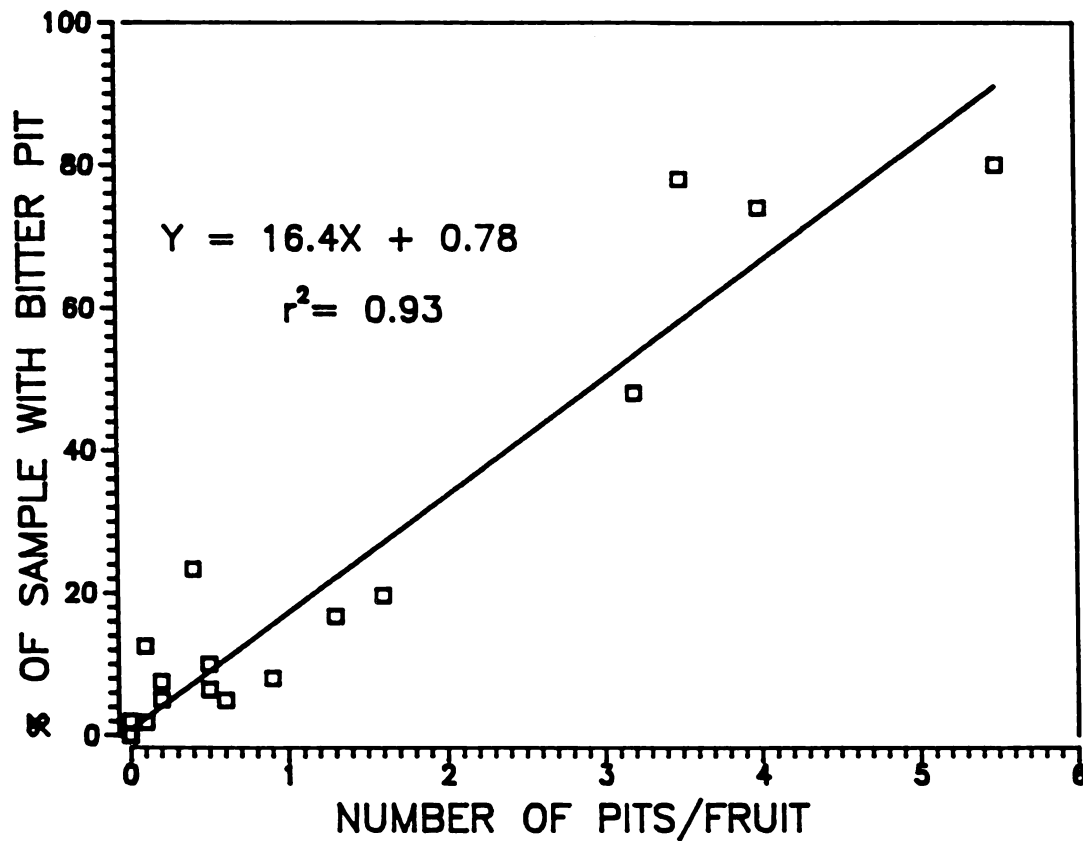


Figure 2. The relationship between the number of bitter pits per fruit and % incidence of bitter pit for Northern Spy stored at 3°C.

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Chapter 2

Characterization of Mg^{2+} Induced Bitter Pit-like Symptoms on Apples: A Model System to Study Bitter Pit Initiation and Development.

ABSTRACT

Vacuum infiltration of MgCl_2 solutions into intact apple fruits induces bitter pit-like symptoms [Mg^{2+} induced pits, (MgIP)]. Including Ca^{2+} in the infiltration media prevents MgIP. Golden Delicious apple fruit were infiltrated with various concentrations of Ca^{2+} and Mg^{2+} with and without Ca^{2+} -affecting reagents or other cations. Including trifluoperazine (TFP) with Mg^{2+} increased pitting over Mg^{2+} alone. Verapamil and nifedipine had no effect on MgIP or its attenuation by Ca^{2+} . Cyclopiazonic acid (CPA) attenuated MgIP. Ethyleneglycol-bis(B-amino-ethyl-ether)-N,N'-tetra acetic acid (EGTA), and 2,3,5-triodobenzoic acid (TIBA), attenuated MgIP. Cycloheximide and actinomycin D inhibited MgIP, while puromycin had no effect. Heating fruits at 38°C prior to infiltrating the fruits with MgCl_2 attenuated MgIP. Cations Ba^{2+} , La^{3+} , Co^{2+} , Sr^{2+} included at 20.0mM prevented MgIP (induced by 0.18M Mg^{2+}). Ca^{2+} (3.0mM) included with 0.18M Mg^{2+} inhibited MgIP 50%. K^+ and Na^+ partially inhibited MgIP. We have demonstrated that treatments affecting calcium homeostasis or cellular metabolism can alter MgIP development. We conclude that MgIP may be a useful tool to understanding natural bitter pit development.

Key words: apple, bitter pit, calcium, channel blocker, homeostasis

INTRODUCTION

Bitter pit is a corking disorder in apples characterized by sunken lesions that develop just prior to harvest or during storage. The tissue below the skin in the pitted area becomes discolored and dehydrated (Faust and Shear, 1968). Susceptibility to bitter pit varies among cultivars and geographic regions. Disorder incidence has been associated with environmental and cultural conditions. Excessive tree vigor, light cropping, calcium deficiency, and moisture stress are among the factors that predispose the fruits to bitter pit (Faust and Shear, 1968; Perring, 1986; Ferguson and Watkins, 1989). Fruits which are immature at harvest are also prone to develop bitter pit. The relationship between elemental nutrition and bitter pit development has been studied extensively (Garman and Mathis, 1956; Martin *et al.*, 1960; Jackson, 1962; Cooper and Bangerth, 1976). Fruits with bitter pit are generally low in Ca, especially in relation to high Mg levels. Treating fruits with Ca^{2+} reduces pitting, while treatments with Mg^{2+} increases the incidence of pitting.

Pitted tissue contains high concentrations of Ca^{2+} and Mg^{2+} (Garman and Mathis, 1956; Hopfinger and Poovaiah, 1979; Askew *et al.*, 1960; Meyer *et al.*, 1979). Ford (1979) demonstrated that $^{45}\text{Ca}^{2+}$ moved into the pitted area as the tissue symptoms developed. Pitted tissue and normal tissue differ in many organic and mineral constituents (Faust and Shear, 1968). Studies of Jonathan spot, Richmond *et al.* (1964), another Ca^{2+} related disorder, demonstrated movement of minerals into the affected area and this was associated with a higher level of total

organic acids, mainly malic acid, in the affected tissue. It is hypothesized that these differences between pitted and healthy tissue are not related to the initiation, but are the result of the metabolic disturbance and subsequent tissue breakdown (Ferguson and Watkins, 1989).

The cause and mechanism of initiation and development of bitter pit are not known. Bitter pit is thought to result from a localized Ca^{2+} deficiency or mineral imbalance, but there is no direct evidence for this (Ferguson and Watkins, 1989; Perring, 1986). Since it has not been possible to identify sites on fruit where pits might develop, studies of bitter pit are normally conducted on fruit tissue showing visible symptoms of the disorder.

Bitter pit-like lesions were induced on apples after Mg^{2+} treatment (Hopfinger *et al.*, 1984; Conway and Sams, 1987; Fallahi *et al.*, 1987; K. Tomala, 1988 in our laboratory, unpublished data). Mg^{2+} induced pits were synonymous to bitter pit as indicated by: MgIP is counteracted by including Ca^{2+} in the infiltration media, and pitting incidence was inversely related to the native fruit Ca^{2+} level (Burmeister and Dilley, 1991). Further, we have correlated susceptibility to Mg^{2+} induced pitting with bitter pit occurring in storage (Burmeister and Dilley, in press). Here we have used MgIP as a model for investigating the physiology and biochemistry of bitter pit initiation and development.

EXPERIMENTAL

Experiments were conducted with Golden Delicious apples (*Malus domestica* borkh.), harvested in 1991 from the Michigan State University Clarksville Horticultural Experiment Station, Clarksville, MI. Fruit (preclimacteric at harvest) were held in controlled atmosphere storage (3% CO₂ + 1.5% O₂ at 0°C) for approximately 7 months. Randomly selected blemish-free fruits (6-9 cm diameter) were vacuum infiltrated with various solutions by submersing them at an absolute pressure of 100mm Hg for 2 min. All solutions contained 0.3M sorbitol as an isotonic osmoticum, and 0.1% Tween 20 as a surfactant. Sorbitol/surfactant solutions were included as controls. Thirty-six fruits were employed in each treatment. After infiltration, fruits were stored for 10 days in air at 20°C and the number of bitter pit-like lesions MgIP on individual fruits were then recorded. Analysis of variance was used to test for main effects and interactions, or treatment sum of squares was partitioned into single degree of freedom contrasts as appropriate for each experiment (Little, 1981).

Various chemical agents known to affect Ca²⁺ availability, transport, binding, or action were investigated to learn how perturbations of [Ca²⁺] might be involved in bitter pit development. These included: EGTA, a chelator; TIBA, an auxin/Ca²⁺ transport inhibitor; verapamil (Vp) and nifedipine (Nf), Ca²⁺ channel blockers; trifluoperazine (TFP), a calmodulin antagonist; cyclopiazonic acid (CPA), a Ca²⁺-ATPase inhibitor; cycloheximide and puromycin, protein synthesis inhibitors; actinomycin-D, an inhibitor of RNA synthesis and several cations.

Calcium channel blockers verapamil (Vp), nifedipine (Nf) and calmodulin antagonist trifluoperazine (TFP). Treatments were factorially arranged with 3 levels of Ca^{2+} (0.0M, 0.01M, 0.02M), and 3 levels of Mg^{2+} (0.0M, 0.04M, 0.18M), alone or with Vp (100 μM) or TFP (100 μM). Fruits were infiltrated with 100 μM Nf alone, or with 0.18M Mg^{2+} . Nf was dissolved in dimethylsulfoxide (DMSO) and added to the solutions resulting in a final concentration of 0.1% DMSO. All controls contained 0.1% DMSO.

Cyclopiazonic Acid (CPA). Fruits were infiltrated with CPA (40 μM) or 0.18M Mg^{2+} , or CPA (40 μM) with Mg^{2+} .

2,3,5-triiodobenzoic acid (TIBA) and ethyleneglycol-bis(B amino ethyl ether)-N,N,N',N'-tetra acetic acid (EGTA). Fruits were treated with EGTA (100 μM), and TIBA (100 μM) alone or with 0.18M Mg^{2+} .

Protein synthesis inhibitors and antibiotics. Fruits were infiltrated with 0.18M Mg^{2+} alone, or with cycloheximide (25 $\mu\text{g ml}^{-1}$), or puromycin (6.25 $\mu\text{g ml}^{-1}$), or actinomycin D (25 $\mu\text{g ml}^{-1}$).

Heat treatments. Experiment 1. Fruits were heated for 0, 1, 2 or 3 days at 38°C then infiltrated with 0.18M Mg^{2+} . During the heat treatments the fruits were in perforated polyethylene bags to prevent desiccation. Experiment 2. Fruits were infiltrated with 0.18M Mg^{2+} and placed at 20°C for 7 days, or 3 days at 38°C followed by 4 days at 20°C, or 3 days at 20°C then 3 days at 38°C and then to 20°C for 7 days. Fruits were enclosed in polyethylene bags as in experiment 1.

Divalent and monovalent cations. Fruits were infiltrated with 0.18M Mg^{2+} alone or with 1.25mM, 2.5mM, 5.0mM, 10mM, or 20mM Ca^{2+} . Fruits were infiltrated with either 0.18M Mg^{2+} , or 0.18M Mg^{2+} + 20mM Ca^{2+} alone, or with La^{3+} , Sr^{2+} , Co^{2+} or Ba^{2+} at 20 mM as chlorides. Fruits were also infiltrated with these cations alone as control. Fruits were infiltrated with 0.18M Mg^{2+} alone or including K^{+} or Na^{+} at 40 mM.

RESULTS AND DISCUSSION

Including TFP with Mg^{2+} increased pitting over Mg^{2+} alone, and Ca^{2+} attenuated MgIP in the treatments that included TFP (Table I). All main effects and interactions were significant ($P = 0.05$) except for Mg^{2+} *TFP, and Mg^{2+} * Ca^{2+} *TFP interactions. TFP is a calmodulin antagonist of the phenothiazine series. Cytosolic [Ca^{2+}] increased in carrot protoplasts treated with TFP (Gilroy *et al.*, 1987). Inhibitors of this class have induced bitter pit-like symptoms (Fukumoto and Nagai, 1983). Results of experiments using TFP must be interpreted with caution because these drugs are suspected to have general, non-specific detergent properties (Personal communication, Dr. Ian Ferguson, DSIR Auckland, NZ). However, in more recent studies of Ca^{2+} fluxes across the plasma membrane of *Commelina commuis L.*, Siebers *et al.* (1990) concluded that the effect of TFP was to mobilize membrane associated Ca^{2+} and trigger release of Ca^{2+} from vesicles. They suggest that TFP induces Ca^{2+} influx and/or inhibits Ca^{2+} efflux across the plasma membrane. No evidence of a detergent effect of TFP was found. TFP treated plasma membrane-rich vesicles were still able to

import $^{45}\text{Ca}^{2+}$ after being washed of excess TFP. Since Ca^{2+} attenuates pitting induced by TFP, we believe that TFP may act through specific binding rather than by a detergent effect. However, binding of TFP is not specific to calmodulin. There are many examples of TFP binding to other Ca^{2+} related proteins such as troponin C, S-100, and Ca^{2+} activated phospholipid-dependent protein kinase (Hartshorne, 1985 and ref. therein). Roufogalis *et al.* (1983) suggested that TFP binds to the activated state of the Ca^{2+} and Mg^{2+} stimulated ATPase of erythrocytes. Therefore, TFP may cause several effects on calcium-linked metabolism rather than affecting a single event by binding to one specific site.

Vp and Nf are members of the dihydropyridine series of Ca^{2+} channel blockers and are believed to block voltage gated Ca^{2+} channels from the inner side of the plasma membrane; they enter the Ca^{2+} channel while it is in the open state (Carfoli, 1987). These drugs affect many plant systems (Heplar and Wayne, 1985). Vp at 100 μM reduced pitting induced by 0.18 M Mg^{2+} (Table I). However, the main effects of Vp and Vp* Mg^{2+} interaction were only significant at $P = 0.1$. Analysis of variance showed no other significant effects. Vp did not affect Ca^{2+} attenuation of MgIP. Another Ca^{2+} channel blocker Nf at 100 μM also did not significantly reduce pitting induced by 0.18M Mg^{2+} (data not shown). Similar experiments with Vp and Nf included at 500 μM showed no consistent reduction in MgIP (data not shown). These results suggest that Ca^{2+} entry into the cell is not required for induction of pitting by Mg^{2+} .

Table I

The effects of Ca^{2+} and Ca^{2+} channel blockers on mean number of Mg^{2+} induced pits per fruit.

Mg^{2+} Conc'n	Ca^{2+} concn. M		
	0.0	0.01	0.2
Average number of MgIP per fruit			
0.0	0.0	0.0	0.0
0.04	1.0	0.0	0.0
0.18	18.7	0.6	0.0
<u>100μM TFP</u>			
0.0	4.6	1.0	0.0
0.04	3.8	0.2	1.0
0.18	22.7	3.5	1.0
<u>100 μM Verapamil</u>			
0.0	0.0	0.0	0.0
0.04	1.7	0.2	0.2
0.18	12.0	0.1	0.4

CPA, an inhibitor of Ca^{2+} -ATPase significantly attenuated MgIP (Table II). Including CPA ($10\mu\text{M}$) (a Ca^{2+} -ATPase inhibitor) with Ca^{2+} did not affect the Ca^{2+} attenuation of MgIP (data not shown). CPA is a specific inhibitor of Ca^{2+} -ATPase and is believed to act by preventing the conformational change (E1 to E2) of the enzyme that is necessary for Ca^{2+} transport (Seidler *et al.* 1989). In animal systems, CPA inhibits P-type calcium dependent ATPases of the endoplasmic and sarcoplasmic reticulum that do not require calmodulin for activation. CPA treated vesicles have a reduced rate of Ca^{2+}

Table II

The effect of cyclopiazonic acid (CPA) on Mg^{2+} induced pitting.

Treatment	Mean number of pits per fruit	Sig. ¹
0.18M Mg^{2+}	14.5	
0.18M Mg^{2+} + CPA ($40\mu\text{M}$)	6.4 vs 0.18M Mg^{+2}	***
CONTROLS:		
0.3M Sorbitol Alone	0.0	
0.3M Sorbitol + CPA ($40\mu\text{M}$)	0.0	

¹Treatments were partitioned into single degree of freedom contrasts as indicated.

* = $p \leq 0.05$, ** = $p \leq 0.01$, *** = $p \leq 0.001$

efflux (Riley and Goeger, 1990).

EGTA at 100 μM significantly attenuated pitting induced by 0.18M Mg^{2+} (Table III). EGTA is a specific chelator of Ca^{2+} (Heplar and Wayne, 1985). Its effect may have been exerted by sequestering Ca^{2+} in the apoplast thereby reducing the amount of Ca^{2+} available to be transported across the plasmalemma. TIBA at 100 μM also significantly reduced the amount of pitting induced by 0.18M Mg^{2+} (Table III). TIBA has been demonstrated to decrease calcium accumulation in apple fruits (Tomala and Dilley, 1989). Ca^{2+} transport in plants has been linked to the polar transport of auxin and the latter is known to be inhibited by TIBA (dela Fuente and Leopold, 1973; Banelos *et al.*, 1987). TIBA, by blocking auxin efflux from the cell could prevent Ca^{2+} entry into the cell enhanced by 0.18M Mg^{2+} treatment.

Table III

The effect of EGTA and TIBA on Mg^{2+} induced pit.

Treatment	Mean number of pits per fruit	Sig. ¹
0.18M Mg^{2+}	21.5	
0.18M Mg^{2+} + EGTA (100 μ M)	6.8 vs 0.18M Mg^{2+}	***
0.18M Mg^{2+} + TIBA (100 μ M)	4.1 vs 0.18M Mg^{2+}	***
CONTROLS:		
EGTA (10 ⁻⁴ M) Alone	0.0	
TIBA (10 ⁻⁴ M)	0.5	
0.3M Sorbitol	0.0	

¹see footnote to table 2.

Cycloheximide at $25\mu\text{M}$ totally inhibited MgIP development (Table IV). This was also found in another experiment with cycloheximide at $10\mu\text{g ml}^{-1}$ (data not shown). Cycloheximide blocks protein synthesis by inhibiting aminoacyl transferase in peptide bond formation in the ribosome. Puromycin at $6.25\mu\text{g ml}^{-1}$ did not significantly reduce MgIP development. This antibiotic is also an inhibitor of normal protein synthesis by causing the cell to produce an abnormal polypeptide. The reason puromycin was less inhibitory than cycloheximide in reducing MgIP may be because the concentration employed was too low. Actinomycin-D at $25\mu\text{g ml}^{-1}$ significantly inhibited MgIP development (Table IV). This antibiotic inhibits RNA synthesis by binding to DNA. Fruit of treatments that included antibiotics had decay symptoms that were distinguishable from MgIP. Collectively, the results with the antibiotics suggest that the induction of pitting by Mg^{2+} may involve mRNA and proteins synthesized *de novo* subsequent to Mg^{2+} treatment.

Table IV

The effect of protein synthesis inhibitors and antibiotics on Mg^{2+} induced pitting.

Treatment	Mean number of pits per fruit		Sig. ¹
0.18M Mg^{2+}	25.8		
0.18M Mg^{2+} + Puromycin ($6.25\mu g\ ml^{-1}$)	21.7	vs Mg^{2+} Alone	n.s.
0.18M Mg^{2+} + Actinomycin D ($25\mu g\ ml^{-1}$)	9.4	vs Mg^{2+} Alone	***
0.18M Mg^{2+} + Cycloheximide ($25\mu g\ ml^{-1}$)	0.0	vs Mg^{2+} Alone	***
CONTROLS:			
Puromycin ($6.25\mu g\ ml^{-1}$)	0.0		
Actinomycin D ($25\mu g\ ml^{-1}$)	0.0		
Cycloheximide ($25\mu g\ ml^{-1}$)	0.0		
0.3m Sorbitol alone	0.0		

¹see footnote to table 2.

Heating apples at 38°C for 1 to 3 days prior to infiltrating them with Mg^{2+} markedly reduced the amount of MgIP (Table V). Heating fruits immediately after Mg^{2+} infiltration more than doubled in number of pits that developed (Table VI), whereas heat treatment applied 3 days following infiltration with Mg^{2+} pitted to the same degree as fruits not heated. Collectively, these data indicate that heating fruits prior to subjecting them to the stress of Mg^{2+} infiltration lessens their susceptibility to MgIP but exacerbates pitting when applied immediately after Mg^{2+} infiltration. Exposure of plants and harvested plant organs to temperatures in the range of 35 to 40°C can profoundly affect physiological and biochemical processes during and subsequent to the heat stress. Alteration of transcription and translation is a response common to all plants heated in the range of 35 to 40°C and this is known as the heat shock (HS) response (Nagao *et al.*, 1986). Heating induces the formation of a complex family of heat shock proteins (HSP) ranging in molecular weight from about 10 kD to nearly 100 kD (Kimpel and Key, 1985). Prestorage heat treatments of apples have been found to inhibit ripening (although not irreversibly), reduce the rate of subsequent softening of apples, and attenuate storage disorders (Porritt and Lidster, 1978; Klein and Lurie, 1992). We hypothesize that heating ameliorated MgIP as a consequence of evoking the heat shock response.

Table V

The effect of heat treatment at 38°C prior to Mg^{2+} infiltration.

Heat Treatment	$[\text{Mg}^{2+}]$	Mean number of pits per fruit	Sig. ¹
0d	0.18M	8.4	
1d	0.18M	1.9 vs 0d, 0.18M Mg^{2+}	***
2d	0.18M	3.7 vs 0d, 0.18M Mg^{2+}	***
3d	0.18M	2.9 vs 0d, 0.18M Mg^{2+}	***
CONTROLS:			
0d	0.00M	0.0	
1d	0.00M	0.0	
2d	0.00M	0.0	
3d	0.00M	0.0	

¹see footnote to table 2.

Table VI

The effect of heat treatment at 38°C following Mg^{2+} infiltration.

Heat Treatment	[Mg^{2+}]	Mean number of pits per fruit	Sig. ¹
20°C Continuous	0.18M	18.3	
3d Heat 20°C	0.18M	46.2 vs 20°C Continuous	***
3d 20°C 3d Heat	0.18M	21.0 vs 20°C Continuous	n.s.
CONTROLS:			
20°C Continuous	0.00M	0.0	
3d Heat 20°C	0.00M	1.6	
3d 20°C 3d Heat	0.00M	0.0	

¹see footnote to table 2.

A relatively low Ca^{2+} concentration was found to counteract pitting induced by Mg^{2+} (Burmeister and Dilley, 1991). The concentration of Ca^{2+} necessary to attenuate pitting induced by 0.18 M Mg^{2+} by 50% was $\approx 3.0\text{mM}$ (data not shown). La^{3+} , Co^{2+} , Sr^{2+} and Ba^{2+} at 20.0mM all completely attenuated Mg^{2+} pitting induced by 0.18M Mg^{2+} (data not shown). La^{3+} is a Ca^{2+} channel blocker (Heplar and Wayne, 1985). La^{3+} may not be able to cross plant membranes (Thompson *et al.*, 1973). It has been demonstrated to block turnover of the phosphorylated intermediate of the Ca-ATPase in the microsomal fraction of maize coleoptiles (Briars and Evans, 1989). Co^{2+} is known to block Ca^{2+} induced seed germination (Wayne and Heplar, 1984) and also to inhibit ethylene production by inhibiting ACC oxidase (Kuai and Dilley, 1992). Ba^{2+} and Sr^{2+} ions can often substitute for Ca^{2+} in the binding of ligands such as membrane proteins (Heplar and Wayne, 1985). Ca^{2+} channels of charophytes do not transport Ba^{2+} and only slowly transport Sr^{2+} . Our results suggest that these cations may act on Ca^{2+} binding sites in the extracellular space.

K^+ (40.0mM) and Na^+ (40.mM) included with 0.18M Mg^{2+} only partially attenuated MgIP (Table VII). This is about twice the concentration of Ca^{2+} that completely inhibited the induction of MgIP by 0.18M Mg^{2+} . The effect of Na^+ could be on the Ca^{2+} - Na^+ antiport (Darnell *et al.*, 1990). K^+ has been shown to stimulate ATPase activity in microsomal preparations of apple fruit (Lurie and Ben-Arie, 1983).

Table VII

The effect of Na^+ and K^+ on Mg^{2+} induced pitting.

Treatment	Mean number of pits per fruit	Sig. ¹
0.18M Mg^{2+} Alone	17.4	
0.18M Mg^{2+} + 0.04M K^+	10.7 vs 0.18M Mg^{2+}	*
0.18M Mg^{2+} + 0.04M Na^+	5.8 vs 0.18M Mg^{2+}	***

¹see footnote to table 2.

We have demonstrated that we can alter the development of MgIP with treatments that are known to affect Ca^{2+} homeostasis and cellular metabolism. Given the similarities between MgIP and bitter pit, our results may imply specific roles for Ca^{2+} and Mg^{2+} in bitter pit initiation and development. Ca^{2+} mediates many responses in plants and animals (Carafoli, 1987; Heplar and Wayne, 1985). The concentration of Ca^{2+} in the cytosol ($[\text{Ca}^{2+}]_{\text{cyt}}$) is maintained in the submicromolar range (Ferguson and Drobak, 1988; Poovaiah and Reddy, 1987; Poovaiah, 1988) by the sequestering of calcium into organelles and export of Ca^{2+} across the plasmalemma by ATPases. Extracellular signals give rise to transient increases in $[\text{Ca}^{2+}]_{\text{cyt}}$ either by release from cellular organelles, or by the opening of specific Ca^{2+} channels in the plasma membrane. $[\text{Ca}^{2+}]_{\text{cyt}}$ affects cellular processes by binding to enzymes, or to Ca^{2+} binding proteins such as calmodulin. Ferguson (1990) suggested that the critical pool of Ca^{2+} involved in bitter pit initiation and development is the extracellular compartment directly accessible to the plasma membrane. Sufficient extracellular Ca^{2+} would be needed for cells to respond to environmental signals. Insufficient Ca^{2+} would prevent cells from responding and cause cell disfunction (e.g. bitter pit). This explanation may account for the fact that fruits of low Ca may not develop bitter pit unless conditions (eg. drought, excessive tree vigor, immaturity of fruit at harvest) trigger the cellular response.

The role of Mg^{2+} in bitter pit initiation is not understood. The concentrations of Mg^{2+} and K^{+} are generally high in relationship to calcium in

fruits with bitter pit (Perring, 1986). Harker *et al.*, (1989) found that Mg^{2+} inhibits $^{45}Ca^{2+}$ transport across discs of cortical flesh of apple fruit. Our data suggest that high levels of Mg^{2+} in the extracellular space may be an important factor in bitter pit initiation perhaps by preventing the influx of extracellular Ca^{2+} to the cytoplasm via specific Ca^{2+} channels.

Gilroy *et al.*, (1989) reported that high extracellular Mg^{2+} and K^+ concentrations resulted in rapid breakdown of Ca^{2+} homeostasis of carrot protoplasts. There is evidence for a Ca^{2+}/Mg^{2+} antagonistic relationship in the activation and inhibition of the Mg^{2+} dependent Ca^{2+} ATPases (Kawaski *et al.*, 1979; Kylin and Kähr, 1973; Vianna, 1975) in the microsomal fractions of plants and animals. This seems to vary among species and tissues. In apple fruit, Lurie and Ben-Arie (1983) found both Mg^{2+} and, to a lesser degree, Ca^{2+} inhibited ATPase activity of the plasma membrane. We speculate that extracellular Mg^{2+} supplied by infiltration could disrupt cellular homeostasis via the key enzyme(s) that regulate intracellular Ca^{2+} . This in turn results in the chain of reactions that result in MgIP. The high levels of Mg^{2+} infiltrated into the extracellular space presumably are akin to the high Mg^{2+} levels often found in fruits with bitter pit. We believe that the pitting symptoms induced by Mg^{2+} infiltration are physiologically synonymous with the bitter pit disorder.

Additional Experiment.

O₂ Dependency of MgIP. Controlled atmosphere storage is known to reduce the incidence and severity of bitter pit in apple fruit (Perring, 1986). This may be attributed to retardation of ripening and senescence development and/or the lower rate of metabolism at the lowered O₂ level. If the effect of Mg²⁺ infiltration in MgIP is a consequence of reducing the rate of O₂-dependent metabolism, MgIP should be lowered as the O₂ supply to the fruit becomes limiting. The O₂ dependency was determined for Golden Delicious apple fruit.

Golden Delicious apple fruit were harvested preclimacteric (ca. 10% of fruits at $>0.2\mu\text{l l}^{-1}$ internal ethylene concentration) from the Clarksville Michigan State University and held in air at 1°C for 4 weeks. A preliminary experiment was conducted in which fruits were infiltrated with 0.0M, 0.09M, 0.18M, and 0.36M Mg²⁺ in 0.4M sorbitol: 0.1% Tween 20 and stored at 20°C in air or 3.0% CO₂ with 1.5% or 3.0% O₂ for 1 month. More MgIP was found in air than at the reduced O₂ atmospheres but symptoms of fermentation damage were evident at the reduced O₂ levels (data not presented).

In a second experiment, fruits were infiltrated with 0.0M, 0.18M as before and stored at 20°C ventilating with 2.5, 5.0, 7.5, 10.0, or 21.0% O₂ for 12d. MgIP increased for fruits infiltrated with 0.18M Mg²⁺ as the O₂ concentration increased to 7.5% with no further increase (Fig 1). There was no apparent fermentation damage evident at any of the O₂ concentration employed. Since ripening changes were not at issue in this experiment it appears that MgIP is an O₂ dependent

metabolic process showing a similar O_2 saturation to respiration.

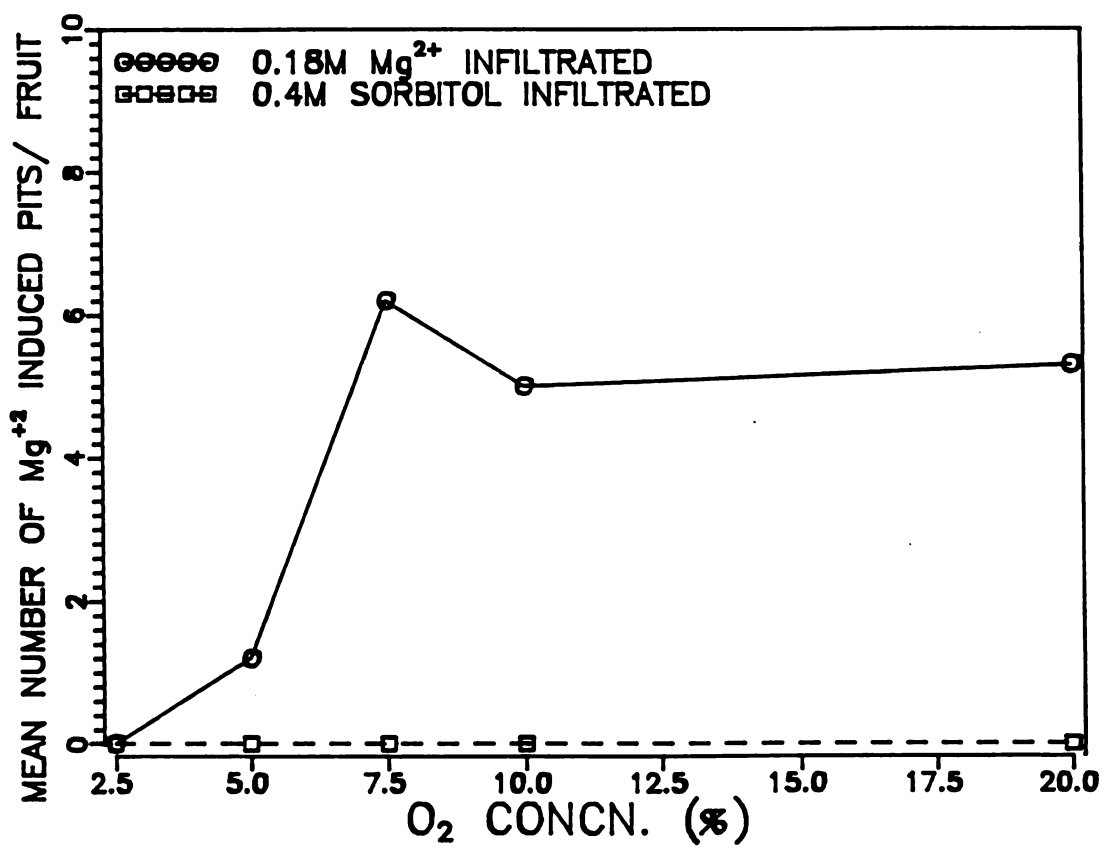


Fig 1. Dependency of MgIP on O_2 concentration (%).

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SUMMARY, CONCLUSIONS AND FUTURE DIRECTIONS:

Bitter pit is a commercially important physiological disorder of apples. The factors affecting its occurrence are well characterized . However, the causes and mechanisms of bitter pit initiation and development, respectively are not known. The disorder has been mainly associated with low fruit Ca and often appears to be exacerbated by high Mg levels (Ferguson and Watkins, 1989). Apple fruit vacuum infiltrated with isotonic sorbitol solutions containing Mg^{2+} develop bitter pit-like symptoms [Mg^{2+} induced pits (MgIP)]. We have demonstrated that MgIP is similar to naturally occurring bitter pit. Addition of Ca^{2+} to the infiltration media attenuates MgIP and the number of lesions induced by Mg^{2+} is inversely related to the endogenous (native) Ca^{2+} concentration of individual fruits (Burmeister and Dilley, 1991). Studies were conducted using MgIP as a means to assess bitter pit potential and as a model for bitter pit initiation and development. Induction of MgIP on Northern Spy apples 10 days after infiltrating 0.1M $MgCl_2$ salt solutions into the fruits was positively correlated with bitter pit that developed naturally in nontreated fruits during five and seven months in storage at 5°C and 3°C , respectively. The endogenous (native) fruit Ca^{2+} concentration was inversely related to the number of pits induced by Mg^{2+} and to bitter pit development following storage.

Golden Delicious apple fruit were infiltrated with various concentrations of Ca^{2+} and Mg^{2+} with and without Ca^{2+} -affecting reagents or other cations.

Including Trifluoperazine (TFP), a calmodulin antagonist with Mg^{2+} increased pitting over Mg^{2+} alone. Verapamil and nifedipine (calcium channel blockers) had no effect on MgIP or its attenuation by Ca^{2+} . Cyclopiazonic acid (Ca^{2+} -ATPase inhibitor) attenuated MgIP. Ethyleneglycol-bis(B amino ethyl ether)-N,N',N'-tetra acetic acid (EGTA), and 2,3,5-triiodobenzoic acid (TIBA) attenuated MgIP. Cycloheximide (a protein synthesis inhibitor) inhibited Mg^{2+} induced pit. Heating at $38^{\circ}C$ prior to infiltrating the fruits attenuated MgIP. Cations Ba^{2+} , La^{3+} , Co^{2+} , Sr^{2+} included at 0.02M all completely arrested MgIP induced by 0.18M Mg^{2+} . K^{+} and Na^{+} partially inhibited MgIP.

Many models could be proposed to explain MgIP development. From the data presented it is difficult to draw specific conclusions. In general reagents, or cations that would ameliorate influx of Ca^{2+} across the plasma membrane or sequester Ca^{2+} attenuate MgIP. Cyclopiazonic acid would be expected to increase MgIP by preventing Ca-ATPase from maintaining low cytosolic $[Ca^{2+}]$. Its attenuation of MgIP could be the result of reducing membrane permeability. In any case factors that would decrease $[Ca^{2+}]$ attenuated MgIP. The multivalent cations Ba^{2+} and Sr^{2+} and channel blockers Co^{2+} and La^{3+} substitute for Ca^{2+} in attenuation of MgIP. La^{3+} is not believed to cross the plasma membrane. Therefore Ca^{2+} may be acting at the plasma membrane in preventing MgIP development. Na^{+} and K^{+} may attenuate of MgIP by displacing Ca^{2+} from ion exchange sites in the cell wall. This would increase $[Ca^{2+}]$ in the apoplastic solution adjacent to the plasma membrane. Trifluoperazine may act directly on

calmodulin or calcium-modulated proteins interfering with cell function resulting in MgIP. Cycloheximide may inhibit protein synthesis necessary for MgIP development.

There is a paucity of information available concerning the roles of extracellular Ca^{2+} and Mg^{2+} in the function of plant cells, especially bulky storage organs such as apple fruit. In animal systems, recent evidence suggests that both the extracellular and intracellular concentrations of Ca^{2+} and Mg^{2+} are important in maintenance of cell Ca^{2+} homeostasis. Intracellular Ca^{2+} is suspected to be modulated by both extra and intracellular $[\text{Mg}^{2+}]$ in cardiac cells (White and Hartzell, 1989). We speculate that high extracellular Mg^{2+} may interfere with a Ca^{2+} channel in the plasma membrane. The presence of extracellular Ca^{2+} may alter membrane permeability to Mg^{2+} thereby allowing the channel to be inactivated. Little is known about Mg^{2+} transport across cell membranes and a regulatory role for Mg^{2+} in cells. Our data indicate that an effect akin to this may be occurring in development of MgIP. Thus, high $[\text{Mg}^{2+}]$ in apple fruit may result in inability of a Ca^{2+} channel to close efficiently resulting in impairment of the cells ability to regulate cytosolic $[\text{Ca}^{2+}]$. High levels of Mg^{2+} and low levels of Ca^{2+} may in themselves not be detrimental, but when such a fruit is confronted with a stress such as cold storage then the regulation of $[\text{Ca}^{2+}]$ becomes critical. This would lead to the event(s) (eg. protein synthesis) that result in bitter pit.

We have demonstrated that susceptibility to MgIP at harvest is correlated with bitter pit development in storage and that treatments affecting calcium

homeostasis or cellular metabolism can alter MgIP development. These studies indicate that extracellular Ca^{2+} and Mg^{2+} concentrations are important in bitter pit development. This may be due to the influence of apoplastic Ca^{2+} and Mg^{2+} concentrations on maintenance of the cytosolic Ca^{2+} concentration.

MgIP may be a useful to study bitter pit. Bitter pit is usually studied after the disorder has already developed. MgIP can be induced in a matter of days after Mg^{2+} infiltration and this is attenuated with Ca^{2+} . MgIP may be useful as model system to study bitter pit at the molecular level. Further research on the effects of exogenous $[\text{Ca}^{2+}]$ and $[\text{Mg}^{2+}]$ on intracellular events such as protein synthesis and phosphorylation following infiltration is warranted. Changes in specific proteins could then be correlated with natural bitter pit development. If changes in specific proteins occurred were identified then *in vitro* translation of mRNA extracted at various times could be employed to detect mRNAs novel to MgIP and bitter pit development.

The use of MgIP to study bitter pit is limited by the necessity of using whole fruit. It would be desirable to use a tissue disc or cell culture to study the effects of Ca^{2+} and Mg^{2+} . Ideally, measurement of cytosolic Ca^{2+} with fluorescent dyes or Ca-selective electrodes would be measured. However, apple tissue is not amiable to these techniques due to highly vacuolated cells and sensitivity to wounding. Future effort should focus on development of an isolated system to study the effects of Ca^{2+} and Mg^{2+} on apple fruit cells.

It is apparent that MgIP may be useful in understanding of the etiology of

bitter pit initiation and development as well as a practical means of assessing bitter pit potential.

APPENDIX ONE: Table of means for Chapter 1.

Table 1. Parameters measured for 1990, early harvest. Each number represents an average of a minimum of 30 fruit.

Location	DIAM. WT. (cm)	SEED COUNT	MgIP (Avg)(%)	Ca ($\mu\text{g g}^{-1} \text{ dwt}^{-1}$)	BITTER PIT AFTER STORAGE	
					(Pits/fruit)	(%)
Epple Orchards	7.6	172	7.2	3.6	50	208
						24.8
						81
Molter Farms	8.5	230	9.2	3.7	36	237
						17.2
						59
Rosenburg Farms	7.8	192	5.0	1.3	29	264
						1.5
						44
WmB. Inc. A.	8.0	209	8.6	1.3	17	193
						2.6
						37
Wittenbach Orch.	8.0	289	4.0	5.8	17	289
						22.9
						88

Table 2. Parameters measured for 1990, late harvest. Each number represents an average of a minimum of 30 fruit.

Location	DIAM. WT. (cm)	SEED COUNT	MgIP (Avg)(%)	Ca ($\mu\text{g g}^{-1}$ dwt ⁻¹)	BITTER PIT AFTER STORAGE (Avg) (%)			
Epple Orchards	7.5	169	6.7	7.7	58	219	9.6	70
Molter Farms	8.7	260	7.2	9.6	51	195	22.2	76
Rosenburg Farms	7.8	189	3.4	2.2	16	238	9.1	54
WmB. Inc. A.	8.4	236	9.8	6.6	60	201	11.9	81
WmB. Inc. B.	8.1	208	13.0	5.7	30	201	9.1	40
Hort. Res. Cent.	8.7	257	6.0	2.4	18	182	9.2	64
Clarksville Expt.	8.6	258	9.0	2.4	26	198	9.9	52
Mueller Farms A.	8.3	230	11.0	2.1	20	256	2.9	24
Mueller Farms B.	7.5	181	6.8	1.9	20	321	1.7	14
Mueller Farms C.	7.8	182	10.0	2.2	16	256	2.8	32
Peabody Orchards.	7.2	155	8.7	0.0	0	357	0.0	0

Table 3. Parameters measured for fruit harvested August 22, 1991. Each number is the average of 50 fruit.

Location	WT. (grams)	SEED COUNT	MgIP (Avg) (%)	Ca ($\mu\text{g g}^{-1} \text{ dwt}^{-1}$)	BITTER		PIT		AFTER		STORAGE	
					5°C (Avg) (%)	3°C Air (Avg) (%)	3°C Air (Avg) (%)	3°C CA (Avg) (%)				
Clarksville Expt.	170	10.0	0.8	14	355	1.6	9.0	0.0	0.0	0.0	0.0	0.0
Peabody Orchards.	146	11.2	2.1	27	260	7.9	38	0.0	0.0	0.0	0.0	0.0
Hort. Res. Cent.	154	9.0	4.8	58	221	10.9	50	0.9	8.0	0.0	0.0	0.0
Blossum Orchard.	144	9.0	0.9	20	240	11.9	15	81	0.0	0.0	0.0	0.0
Mueller Farms.	141	11.5	1.3	28	366	9.1	20	40	5.0	0.0	0.0	0.0

Table 4. Parameters measured for fruit harvest Sept. 3, 1991. Each number represents the mean of 50 fruit.

Location	<u>BITTER PIT AFTER STORAGE</u>									
	WT. (grams)	SEED COUNT	MgIP (Avg)	(%)	Ca ($\mu\text{g g}^{-1} \text{ dwt}^{-1}$)	5°C (Avg)	(%)	3°C Air (Avg)	(%)	3°C CA (Avg)
Clarksville Expt.	168	9.7	0.3	12	342	3.4	29	0.0	0.0	0.0
Peabody Orchards.	173	11.6	0.0	0.0	251	1.6	19	0.0	0.0	0.0
Hort. Res. Cent.	165	9.3	3.7	48	208	8.8	62	0.2	7.5	0.0
Blossum Orchard.	148	8.9	0.5	22	231	1.1	16	0.0	0.0	0.0
Mueller Farms.	166	9.5	0.3	10	236	1.9	9.7	0.1	12.5	0.0
WmB. Inc.	186	11.2	11.8	79	195	-	-	3.5	78.0	4.0
										74.0

Table 5. Parameters measured for fruit harvested Sept. 11, 1991. Each number represents an average of 50 fruit.

Location	WT. (grams)	SEED COUNT	MgIP (Avg) (%)	Ca ($\mu\text{g g}^{-1} \text{ dwt}^{-1}$)	BITTER PIT AFTER STORAGE						
					5°C (Avg) (%)	3°C Air (Avg) (%)	3°C CA (Avg) (%)	3°C CA (Avg) (%)			
Clarksville Expt.	172	9.6	1.0	16	286	5.0	23	0.0	0.0	0.0	
Peabody Orchards.	177	11.2	1.6	10	194	9.9	25	0.5	10.0	0.6	4.9
Hort. Res. Cent.	178	8.0	5.0	58	180	37.1	62	1.6	19.6	0.1	1.8
Blossum Orchard.	163	9.4	1.8	20	210	2.4	20	0.4	23.3	1.3	16.7
Mueller Farms.	142	9.6	0.1	2	295	2.5	17	0.5	6.4	0.0	1.9
WmB. Inc.	194	11.2	6.9	70	167	16.8	80	5.5	80.0	3.2	48

APPENDIX TWO: Additional figures for Chapter 1.

Fig. 1. Plot of MgIP/Fruit vs % of fruit with bitter pit that developed during storage at 5°C for 5 months. Data are cumulative for both the 1990 and 1991 seasons.

For MgIP/fruit vs bitter pits/fruit that developed in storage, $Y = 1.9X + 3.3$, $r^2 = 0.31$. For % of fruit with MgIP vs % of fruit with bitter pit after storage, $Y = 0.92X + 14.5$, $r^2 = 0.51$.

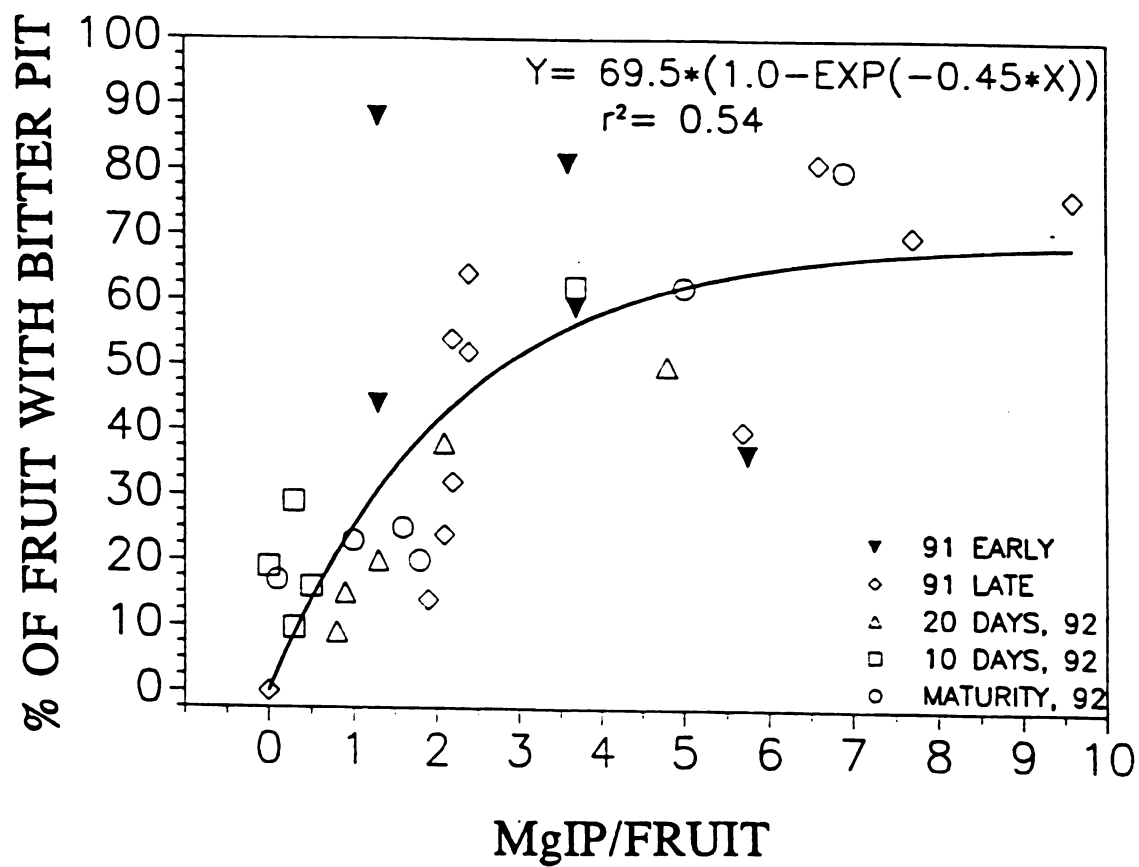


Fig. 2. Plot of endogenous (native) calcium ($\mu\text{g g}^{-1}$ dwt.) vs **MgIP/fruit**.

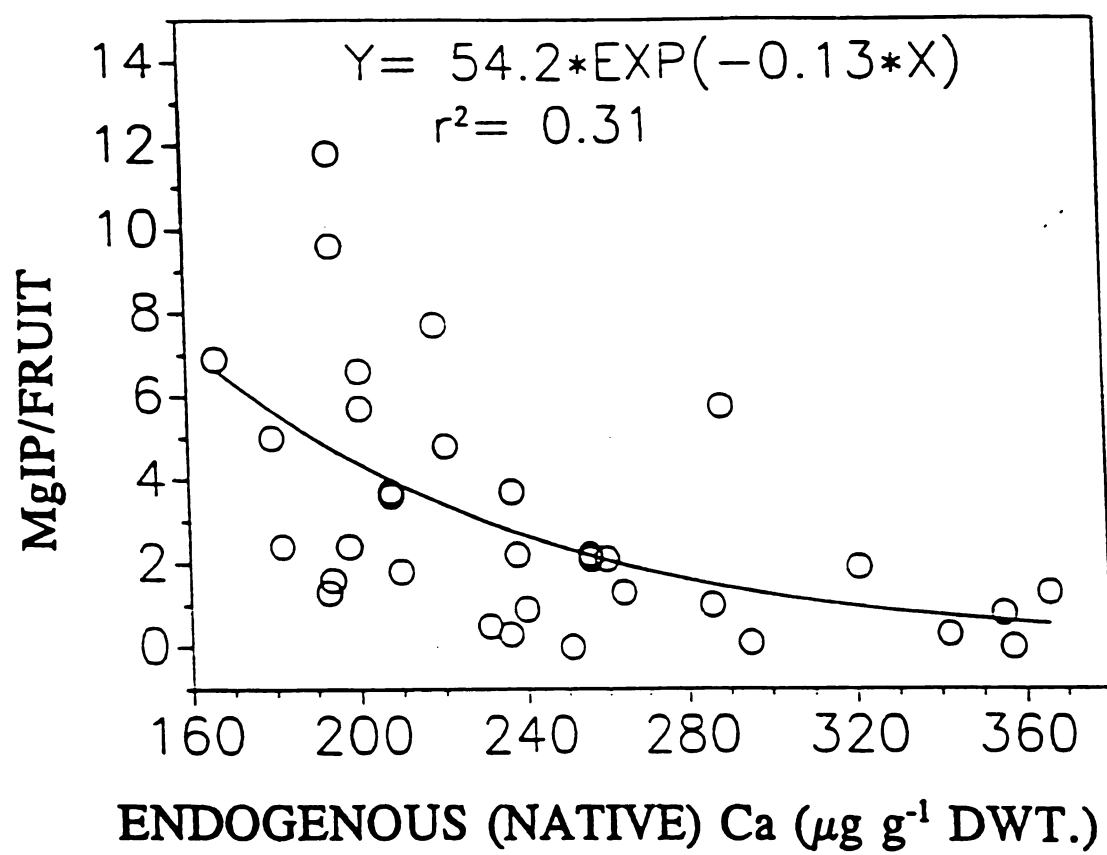


Fig. 3. Plot of endogenous (native) calcium ($\mu\text{g g}^{-1}$ dwt.) vs bitter pits/fruit that developed in storage. For 5°C and 3°C storage $Y = 473.6 \cdot \text{EXP}(-0.02 \cdot X)$, $r^2 = 0.49$, and $Y = 502.0 \cdot \text{EXP}(-0.043 \cdot X)$, $r^2 = 0.59$, respectively.

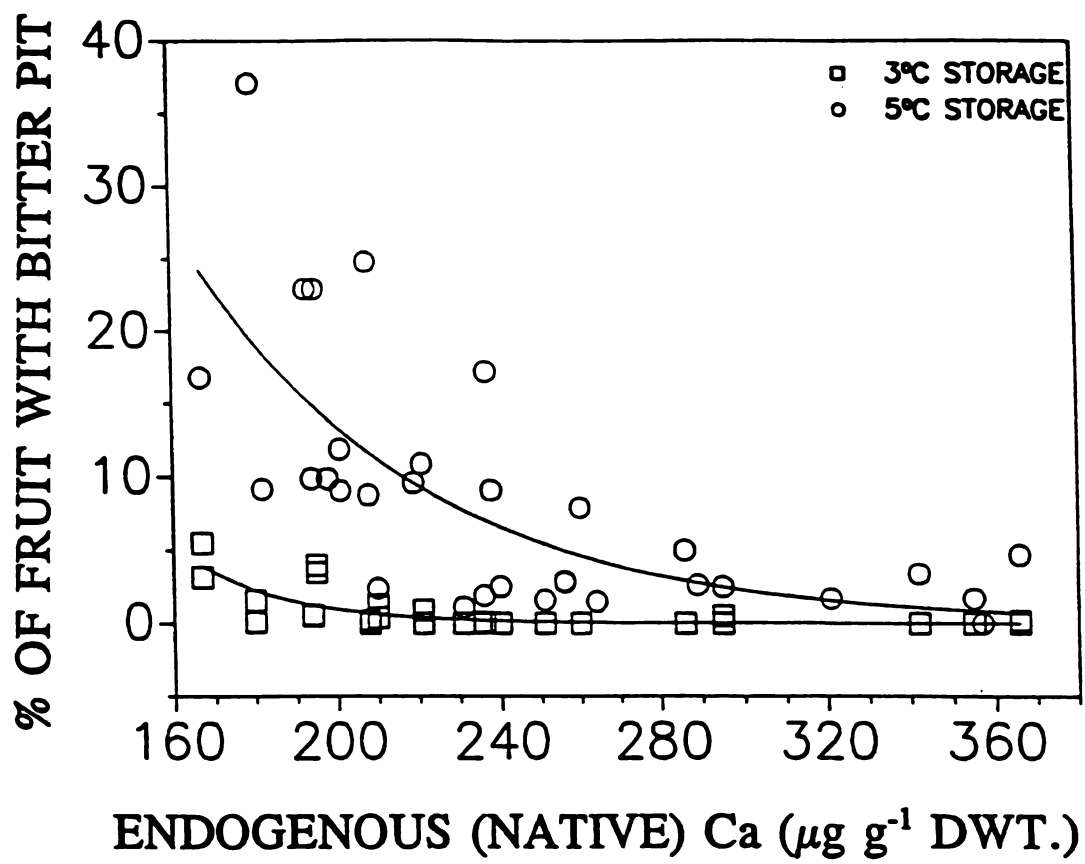
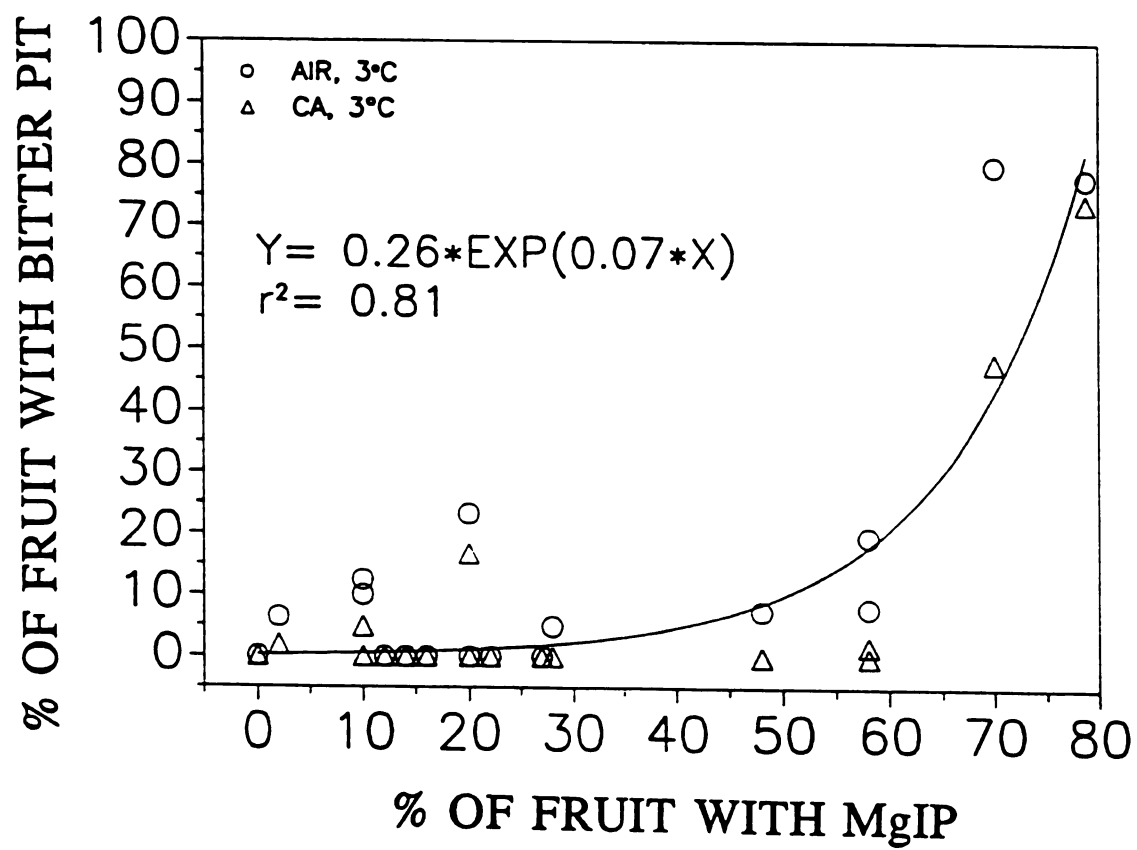


Fig. 4. Plot of % of fruit with MgIP vs % of fruit with bitter pit after 3°C air and CA storage for 7 months.



APPENDIX THREE: The Effects of Prestorage Heat Treatments on Physiological Disorder Development and Softening of Northern Spy, Red Delicious and Law Rome Apples.

INTRODUCTION

Exposure of plants and harvested plant organs to 35 to 40°C temperatures profoundly affects physiological and biochemical processes during and subsequent to the heat stress. Work in our laboratory as well as others has shown heat treated tomatoes to be less sensitive to chilling injury (Picton and Grierson, 1988; Lurie and Klein, 1990 and 1991; Burmeister *et al.*, in preparation). Other work with apples has shown an inhibition of ripening and a decrease in softening rate of heat treated fruit (Maxie *et al.* 1974; Porritt and Lidster, 1978). In addition, superficial scald on apples has been shown to be attenuated by prestorage heat treatments, although not completely (Lurie *et al.*, 1991; Klein and Lurie, 1992a). Interest in use of physical rather than chemical means to control postharvest disorders has been prompted by concern of the public over the safety of these chemicals (Klein and Lurie, 1992b). Environmental Protection Agency registration of some fungicides, and antioxidants may be withdrawn and the cost of re-registration of existing, or registration of new chemicals is prohibitive. In addition, the storage and disposal of the waste solution from drenching facilities is a problem. The objective of this study was to determine if prestorage heat treatments in combination with

subsequent controlled atmosphere (CA) storage would attenuate physiological disorder development on apples without sacrificing quality.

MATERIALS AND METHODS

Preclimacteric Northern Spy, Red Delicious, and Law Rome apples were obtained from the Michigan State University Clarksville Experiment Station.

Purge CA system: Fruits (40-60) were randomized to fill into 20L CA chambers by cultivar and held in air at 38°C for 0, 2, 3, or 4d and then placed in air at 1 or 3°C for the balance of 7d before establishing CA conditions. Chambers were ventilated with 3.0% CO₂:1.5% O₂, or 3.0% CO₂:3.0% O₂ at 75ml min⁻¹.

Chambers left in air with loosely fitting lids served as controls. Each atmosphere/heat treatment was done in triplicate. **Quasi-static system:** Law Rome were randomized and loaded into CA chambers as before and held at 38°C for 0, 1, 2, 3, or 4d then placed in air at 1 or 3°C for the balance of 7d before establishing CA. CA was maintained employing a David Bishop

Instrument Oxystat II CA system. Fruit were stored for 8 months and examined for disorders after a subsequent 7d period in air at 20°C. Northern Spy were evaluated for bitter pit while, those of Red Delicious and Law Rome were examined for superficial scald. Flesh firmness was determined with Effegi penetrometer (11mm dia.) for 10 fruits upon removal from storage and after 7 days at 20°C in air.

RESULTS AND DISCUSSION

Purge system (3°C). Heating delayed softening of Northern Spy fruits stored in air and with 3:3 atmosphere. The effect of heat was evident for fruits on removal from storage and after holding them in air at 20°C for 7d (Table 1). Attenuation of softening was significantly greater ($P < 0.01$) as the duration of heating increased. Fruits stored at 1.5 O₂ retained the greatest firmness during storage. Heating diminished the percentage of apples stored at 3:1.5 which developed bitter pit but increased bitter pit on fruits stored at 3:3.

Red Delicious. Retention of flesh firmness and scald attenuation were greatest in the 2d heat treatment at the 3.0:1.5 atmosphere (Table 2). However, the effect of heat treatments on retention of flesh firmness and scald attenuation for Red Delicious was not significant.

Law Rome. The effect of heat treatments on firmness upon removal from storage was significant ($P < 0.01$), but not for fruits held for 7d at 20°C. Heat treatments significantly increased scald for this cultivar.

Law Rome, quasi static system 1° and 3°C. Attenuation of softening for heated fruit upon removal and after 7d at 20°C was significant for both 1° and 3°C storage temperatures ($P < 0.01$). The effects of heat treatment on scald were significant in both the 1°C and 3°C storage temperatures ($P < 0.05$ and $P < 0.01$, respectively). Scald appeared to increase with duration of heating for fruit stored at 1°C while decreasing for the 3°C storage temperature (Table 4 and 5).

These results indicate that heat treatments to these varieties do attenuate softening but may not be a viable alternative to control physiological storage disorders in long term storage beyond appropriate application of low temperature and CA. These results concur with those of Klein and Lurie (1992a) who found no benefit on storage firmness following the heat treatment of Granny Smith apples held in CA for 8 months. They did not evaluate these long term CA fruit for physiological disorder development. In previous studies, storage durations have been relatively short (Lurie *et al.*, 1991). It may be that positive effects of heat treatment on scald prevention may only be realized following short storage durations or when application of appropriate CA conditions is not feasible. Further investigations will be necessary to determine conditions where prestorage heat treatment may be beneficial.

Table 1. Effect of prestorage heat treatments on flesh firmness and bitter pit on Northern Spy apples stored at 3°C for 8 months in purge mode.

<u>Atmosphere</u> (CO ₂ :O ₂)	<u>Flesh firmness (N)</u>							
	<u>0d</u> <u>Days of heat</u>				<u>7d at 20°C</u> <u>Days of heat</u>			
	<u>0d</u>	<u>2d</u>	<u>3d</u>	<u>4d</u>	<u>0d</u>	<u>2d</u>	<u>3d</u>	<u>4d</u>
Air	51.2	61.8	61.7	64.2	58.7	61.8	64.9	65.8
3.0:3.0	62.8	68.7	70.8	69.8	64.1	71.2	70.9	73.9
3.0:1.5	74.6	75.9	74.7	72.7	75.4	81.3	79.4	81.4
	LSD _{0.05} = 1.7				LSD _{0.05} = 1.6			
	<u>Bitter pit (%)</u>							
Air	17.6				10.7	12.8	12.8	
3.0:3.0	0.7				0.0	9.0	6.8	
3.0:1.5	11.1				1.5	2.6	0.0	
	LSD _{0.05} = 2.3							

Table 2. Effect of prestorage heat treatments on flesh firmness and superficial scald on Red Delicious apples stored at 3°C for 8 months in purge mode.

<u>Atmosphere</u> (CO ₂ :O ₂)	<u>Flesh firmness (N)</u>							
	<u>0d</u> <u>Days of heat</u>				<u>7d at 20°C</u> <u>Days of heat</u>			
	<u>0d</u>	<u>2d</u>	<u>3d</u>	<u>4d</u>	<u>0d</u>	<u>2d</u>	<u>3d</u>	<u>4d</u>
Air	54.6	49.5	47.9	52.8	55.0	46.3	45.8	51.5
3.0:3.0	53.2	50.6	55.8	54.5	52.2	45.1	53.1	46.0
3.0:1.5	61.2	64.7	60.1	57.0	60.4	63.5	57.5	58.2
	LSD _{0.05} = 1.5				LSD _{0.05} = 1.8			

	<u>Scald (% of surface area)</u>			
Air	51.7	57.4	55.8	52.8
3.0:3.0	47.2	60.7	60.1	49.2
3.0:1.5	39.4	18.2	32.1	32.1
	LSD _{0.05} = 4.3			

Table 3. Effect of prestorage heat treatments on flesh firmness and superficial scald on Law Rome apples stored at 3°C for 8 months in purge mode.

<u>Atmosphere</u> <u>(CO₂:O₂)</u>	<u>Flesh firmness (N)</u>							
	<u>0d</u> <u>Days of heat</u>				<u>7d at 20°C</u> <u>Days of heat</u>			
	<u>0d</u>	<u>2d</u>	<u>3d</u>	<u>4d</u>	<u>0d</u>	<u>2d</u>	<u>3d</u>	<u>4d</u>
Air	56.9	67.8	73.7	70.1	52.0	59.7	71.6	66.1
3.0:3.0	65.6	71.7	73.5	73.0	65.5	67.8	64.6	60.5
3.0:1.5	75.4	74.6	73.3	72.4	67.2	73.3	67.2	67.7
	LSD _{0.05} = 2.1				LSD _{0.05} = 2.6			
	<u>Scald (% of surface area)</u>							
AIR	75.7 69.9 74.4 80.1							
3.0:3.0	16.2 25.6 37.0 34.0							
3.0:1.5	10.7 21.3 16.4 27.4							
	LSD _{0.05} = 3.8							

Table 4. Effect of prestorage heat treatments on flesh firmness and superficial scald on Law Rome apples stored at 1°C for 8 months in quasi-static CA system.

Atmosphere (CO ₂ :O ₂)	Flesh firmness (N)									
	<u>0d</u>					<u>7d at 20°C</u>				
	<u>Days of heat</u>					<u>Days of heat</u>				
	<u>0d</u>	<u>1d</u>	<u>2d</u>	<u>3d</u>	<u>4d</u>	<u>0d</u>	<u>1d</u>	<u>2d</u>	<u>3d</u>	<u>4d</u>
Air	52.9	57.2	61.9	66.7	66.5	51.5	52.9	60.4	64.5	65.1
3.0:3.0	59.1	58.2	73.5	73.0	66.8	54.2	54.3	60.9	62.3	63.4
3.0:1.5	59.7	60.4	73.3	72.4	63.2	57.5	57.1	59.2	59.8	61.8
	LSD _{0.05} = 1.3					LSD _{0.05} = 1.7				
	<u>Scald (% of surface area)</u>									
Air	72.0 72.6 75.1 71.2 80.0									
3.0:3.0	34.3 45.4 47.6 58.1 54.7									
3.0:1.5	20.9 41.4 49.3 41.3 43.8									
	LSD _{0.05} = 5.3									

Table 5. Effect of prestorage heat treatments on flesh firmness and superficial scald on Law Rome apples stored at 3°C for 8 months in quasi-static CA system.

<u>Atmosphere</u> (CO ₂ :O ₂)	<u>Flesh firmness (N)</u>									
	<u>0d</u>					<u>7d at 20°C</u>				
	<u>Days of heat</u>					<u>Days of heat</u>				
	<u>0d</u>	<u>1d</u>	<u>2d</u>	<u>3d</u>	<u>4d</u>	<u>0d</u>	<u>1d</u>	<u>2d</u>	<u>3d</u>	<u>4d</u>
AIR	44.0	47.9	53.6	57.3	56.4	42.3	47.7	46.6	55.3	50.9
3.0:3.0	49.5	53.2	53.1	54.2	55.8	48.1	48.3	51.2	52.3	51.5
3.0:1.5	53.5	53.4	53.8	55.5	57.5	50.7	50.8	50.8	52.6	54.1
	LSD _{0.05} = 1.5					LSD _{0.05} = 1.2				
	<u>Scald (% of surface area)</u>									
Air						78.5	66.1	68.3	81.6	73.5
3.0:3.0						81.0	81.8	63.3	76.0	82.7
3.0:1.5						77.7	72.6	65.0	64.4	74.8
						LSD _{0.05} = 2.8				

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