

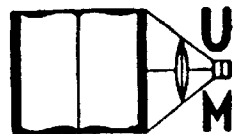
DOCTORAL DISSERTATION SERIES

TITLE A Study Of Yeast Growth-Promoting In
White Sugar

AUTHOR Harlow Homer Hall DATE 1943

UNIVERSITY Michigan State College

DEGREE Ph.D. PUBLICATION NO. 671



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Abstract

"A Study of Yeast Growth-Promoting Substances in White Sugar"*

Harlow H. Hall

The extent of multiplication of yeast cells in dilute solutions of white sucrose sugars is shown to be dependant upon the amount of a complex of yeast growth-promoting substances present. Relatively large amounts of substances in a given sugar may influence its use in the manufacture of beverages, mountain sirups, dairy products or other low sugar content liquids which are not heat processed and which are subject to spoilage by yeast. The extent of reproduction of an inoculum of cells of a known yeast culture in a 10-percent solution of the sugar is used as a means for determining the relative amount of yeast growth-promoting substances present.

The direct microscopic~~x~~ cell count technic, using a haemocytometer, is employed to determine the extent of yeast cell reproduction. Each sample of sugar to be tested is made up in amount sufficient to result in 100 ml. after sterilization in 300 ml. Erlenmeyer flasks. When cool 0.5 to 1.5 ml. of a water suspension of a 24-hour culture of the test yeast is added to result in an inoculation of approximately 50,000 cells per ml. of the test solution. After 72 hours incubation at 30° C. the number of cells per ml. is determined. The yeast inoculum multiple is recorded as the number of cells per ml. after incubation divided by the number of cells per ml. initially present.

Approximately 900 composite samples of sugar, representing the annual output of 29 to 87 domestic factories during a period of 11 years were examined. Many samples of sugar did not support the growth of yeast cells while others supported the growth of varying size crops. The average yeast inoculum multiples for yearly campaign composite samples was between 4.3 and 12.4 for the 11-year period. The results also show that the amount of growth-promoting substances may vary between suc-

* Thesis for the Degree of Doctor of Philosophy, Michigan State College, East Lansing, Michigan.

cessive strikes of sugar. It may also vary in sugar from the same factory from year to year.

In groups of samples a relationship exists between the relative amount of yeast growth-promoting substances and the amount of ash present. The ash of sugar does not, however, promote yeast growth. Although the growth-promoting factor is organic in nature, there is no correlation with known organic non-sugar impurities. There is likewise no correlation with the amount of the factor and the geographic areas in which the sugar is manufactured or, with the manufacturing process employed.

The growth-promoting factor is located at or near the surface of sugar crystals. It is almost completely eliminated by recrystallization of sugar from 80 percent alcohol and may be recovered from it by selective treatment. In the practice of sugar manufacture it is effectively removed by thoroughly washing the crystals of sugar to assure complete elimination of the sirup film surrounding each crystal. The growth-promoting substances are stable to heat and alkali treatment during the process of sugar manufacture and accumulate in the final molasses.

Biotin was found to be present in varying amounts (0.025✓ to 0.445✓ per gram of dry sugar) in alcohol-free extracts which were prepared by recrystallization of individual sugar samples from 80-percent alcohol.

A STUDY OF YEAST GROWTH-PROMOTING SUBSTANCES IN WHITE SUGAR

by

EARLEN BOWEN HALL

A THESIS

Submitted to the Graduate School of Michigan
State College of Agriculture and Applied
Science in partial fulfillment of the
requirements for the degree of

DOCTOR OF PHILOSOPHY

Department of Bacteriology
and
Hygiene

East Lansing, Michigan

1943

ACKNOWLEDGMENT

The writer wishes to express his sincere appreciation to Dr. F. E. Fabian for his interest and helpful guidance throughout this study. Also, the writer is deeply indebted to H. F. Price, Chief, Agricultural Chemical Research Division, Bureau of Agricultural and Industrial Chemistry, United States Department of Agriculture for his continued interest and many invaluable suggestions. Acknowledgment is also made to L. S. Stanton and Sam Byall, Bureau of Agricultural and Industrial Chemistry for their excellent cooperation while conducting this study. The writer is especially grateful to the management and personnel of the Michigan beet sugar factories for their cooperation in supplying a large number of special samples of finished sugar and intermediate products.

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I. INTRODUCTION

The white sucrose sugars, which include beet and cane, are perhaps the nearest approach to chemically pure substances that are included in the daily diet. As a food commodity, sugar is the cheapest, on a caloric basis, on the American market and is used in quantity in more food products than any other prepared ingredient. Because sugar-containing plants lend themselves to easy cultivation and bulk handling, manufacturing interests have been able, for many years, to produce and sell sugar in sufficiently large quantities to make it an important item in world commerce.

The value of use of sugar as an ingredient in prepared foods is not limited to its energy-giving content, nor to its sweetening property. As a preservative in water-containing foods, it has no equal from the standpoint of maintaining flavor, color, turgidity or keeping quality. **Sugar is truly the master preserver.**

The utility of sugar, like many other food ingredients, has certain limitations which are governed by the presence or absence of foreign substances. It has been observed that impurities may often be present in food ingredients in relatively large quantities and yet not be of concern when the ingredient is used for a selected food product. However, a trace of another impurity may greatly impair its utility when used in the same product. Sugar is no exception to this observation. The presence of bacteria, yeast and molds in sugar and substances which stimulate their growth in solutions are illustrative. For instance,

the presence of a few million of viable yeast cells in sugar is of no concern when the sugar is for table use. Such a sugar would not, however, be acceptable for use in the manufacture of beverages. Likewise, the presence of a relatively large amount of yeast growth-promoting substance in sugar used for table sweetening would be of no concern. However, it would not be desirable to use such a sugar for the manufacture of beverages or similar products which might become contaminated with the organism either the sugar itself or from other sources. Because of the extensive use of sugar in beverages, fruit-flavored, flavoring color and sweetened condensed milk, all of which are subject to spoilage by yeast, the presence of yeast growth-promoting substances in sugar is of vital importance. It is a problem that is attracting considerable attention among the manufacturers, as well as the users, of sugar.

Scientific workers have, for many years or longer, delved into the mysteries attending the growth or failure of growth of yeasts in nutrient solutions. No previous workers have been concerned with the ~~effect of growth-promoting substances and sugar impurities from the~~ standpoint of yeast nutrition requirements. Hall, James and Stuart (17) were the first to consider subject from the standpoint of the commercial manufacture and use of sugar, when it became apparent that there were differences between the biological quality of sugars of the same type, i.e., beet or cane sugar, so that these differences were noticeable in their use in food products, especially beverages.

The data presented here as a result of this research were obtained in the laboratory of the Bureau of Agricultural and Industrial Chemistry of the United States Department of Agriculture. During the course

of the sugar, the sugar was analyzed, and the sugar was of
certified pure sugar were received and examined. It was conducted in
conjunction with the study of the chemical impurities in white sugar. The
majority of the sugar samples were obtained from individual factory
and large commercial companies, or those which constitute a portion of each
sample of sugar being during a given sugar campaign. They were furnished
in part from some of the best sugar industry which actively participated
in the United States Department of Agriculture in a program of manu-
facturing sugar of uniform biological and chemical quality. Signifi-
cant factory operating data were made available by the manufacturer of
each sample of sugar used in the evaluation and interpretation of
analytical data. The results of the analysis were, in turn, made avail-
able to the manufacturers as a record of factory performance and as a
guide for future operations.

The major objectives of this research were:

1. To determine the presence or absence of yeast growth-promot-
ing substances in commercially manufactured white sugar.
2. To develop a method for evaluating the biological quality of
white sugar based on the relative amount of yeast growth-promoting
substances present.
3. To study the properties of, and identify the growth-promoting
substances found.
4. To determine the relationship of the presence of such sub-
stances with other known non-sugar substances, especially in sugar of
known origin.
5. To develop methods for the removal of such substances during
the manufacturing process.

II. HISTORICAL DEVELOPMENT OF THE SUGAR INDUSTRY

CANE SUGAR

During the early centuries, man's chief sweet food was honey. Its replacement by cane-sugar began in the Far East when India developed sugar cane, Saccharum officinarum, a source of sweets, which Alexander's army in 327 B.C. discovered its excellence. From the East country to Europe sugar cane; the Arabs and the Arabs took it to the Mediterranean countries in the Middle Ages; Columbus and his followers introduced its culture in Cuba and the West Indies; and by 1650, it had been introduced into all the West Indian islands. In 1704, its culture spread to Louisiana where it has continued to the present time.

There was a sugar refinery in New York City as early as 1659, which refined raw sugar imported from the West Indies and Cuba. In the year 1730, Nicholas Bayard erected a refinery on Wall Street, where an European expert was able to make, for the first time in America, several types of sugar; that is, single- and double-refined loaf sugar, crushed sugar, brown sugar and a form of rock candy. It is believed that Bayard was the first to establish sugar refining as a separate enterprise in America. His first newspaper advertisement appeared in the New York Gazette for August 17, 1730, as follows:

"PUBLISHED NOTICE is hereby given that NICHOLAS BAYARD of the City of New York has erected a Refining House for refining all sorts of sugar and Sugar-Candy, and has procured from Europe an experienced Artist in that mystery. At which Refining House all Persons in City and Country may be supplied by whole-sale and Retail with both double and single-Refined Loaf

Sugar, as also Powder and Shop-Sugars and Sugar-Candy, at reasonable rates."

The initial success of Bayard prompted the construction of additional refineries: so that by 1755, New York, Philadelphia and Boston were refining 1,130,000 pounds of sugar yearly or about 2 percent of the total amount which was consumed annually.

Many new innovations in refining technique and machinery were introduced into the industry during the nineteenth century which enabled the industry further to increase its output of refined sugar manyfold so that the industry supplied three-fourths of the 4.3 billion pounds of sucrose consumed in the United States in 1941. Some of the inventions which led to the establishment of sound processing practices were Watt's steam engine which supplanted man and animal power, and the vacuum pan by Howard in 1813. The introduction of modern cane-crushing machinery, juice filtration, carbonation, sulfitation, crystallizers, centrifugals and granulators, which are discussed by Perist (30) in his book "The Manufacture of Sugar from the Cane and Beet", were other innovations of importance. Also contributing to the success and expansion of the industry was the development of modern planting and harvesting machinery.

To supply our cane-sugar needs, there are about 100 refineries and factories which refine imported raw sugar and/or process domestic sugar cane which is grown in Louisiana and Florida.

BETTER SUGAR

Sugar beets, Beta vulgaris, were cultured in Europe during the eighteenth century and the first beet-sugar factory was established in Austria in 1769. Napoleon heavily subsidized the industry in France



Illustration 1. BEET SUGAR FACTORY SMOKING PILES OF BEET IN POLYMERINE

in 1810 in order to alleviate the shortage of cane sugar created during the blockade of French ports by the British. The production of beet sugar in the United States began in 1811 in Massachusetts. In 1812, the first beet sugar factory was established in California, Illinois and Wisconsin, between 1863 and 1876. The first successful factory in the United States was built at Livermore, California in 1870. There were only 30 beet sugar factories in operation in mid-1870. By the year 1890, the number had increased to 47, producing an average of 100 tons of sugar per day for an operating season of about 100 days.

The success and development of the beet sugar industry was likewise dependent upon the development of much of the same modern equipment and methods used by the cane sugar industry. Differences in the methods for processing sugar cane and sugar beets, such as, the diffusion of sugar from beets and the Steffen method for recovering sugar from beet sugar molasses are described in detail by Harriot (39) in his book "The Manufacture of Sugar from the Cane and Beet". Of especial importance is the recent development of farm machinery which aids fairly to supplant much of the labor formerly required in thinning, blocking and towing of beets. Still another important development is the domestic production of beet seeds. Although the seed producing area is limited to Pacific Northwestern states, there is much evidence available to support the belief that seeds from this area will produce better quality beets throughout the beet-growing states than seeds previously imported from Europe. The capacity of the existing beet sugar factories is limited only by the production of sugar beets.

DOMESTIC PRODUCTION AND CONSUMPTION OF SUGAR

During the year 1941, there were 5 operating beet sugar factories which produced over one and three-fourths million tons of sugar. In the same year there were 50 cane sugar factories in operation in Louisiana and in Florida which produced a total of 22,000 tons of cane or direct-consumption sugar.

The production of sugar by the beet sugar industry is characterized by the production of sugar by the cane sugar industry. This is a result of the relatively small area of beet sugar production throughout the larger geographic area. The production of beet and cane sugar by the United States is given in Table 1 for the period 1929-1939 and annually for 1939, 1940 and 1941.

The value of sugar consumed by the several branches of the food industry for 1939 is estimated to be \$11,637,919. This includes corn sugar with a valuation of \$47,373. The amount of each type of sugar and its value is given for the several branches of the industry in Table 1.

The largest user of sugar is the bakery industry which used in excess of one and one-fourth billion pounds in 1939. Following this industry are the confectionery, canned and preserved foods, beverage, and dairy industries and large users. It is estimated that about 40 percent of the sugar is consumed by industry, while the remaining 60 percent is consumed in the household and in commercial eating establishments.

The annual distribution per capita consumption in short tons of sugar, respectively, of sugar in the United States for the period

TABLE 1. SUGAR PRODUCTION IN STATES, AVERAGE 1939-33
ANNUALLY 1939, 1940 AND 1941*

STATE		P R O D U C T I O N			
		Average 1939-33	1939	1940	1941
<u>Beet sugar:</u>	<u>No. of Factories**</u>	<u>1,000 short tons</u>	<u>1,000 short tons</u>	<u>1,000 short tons</u>	<u>1,000 short tons</u>
California	10	231	452	454	315
Colorado	17	232	162	313	299
Idaho	9	43	127	145	107
Michigan	12	113	162	163	158
Montana	5	99	140	163	118
Nebraska	7	116	136	115	121
Ohio	4	30	42	45	46
Utah	6	83	100	74	82
Wyoming	5	90	92	93	79
<u>Other States:***</u>	<u>10</u>	<u>108</u>	<u>159</u>	<u>191</u>	<u>161</u>
<u>Cane Sugar:</u>					
Florida	2	39	65	94	96
Louisiana	65	167	406	210	323
Total Beet sugar			1,643	1,761	1,454
Total Cane Sugar			471	314	419
United States		1,605	1,114	1,075	1,903

*Agricultural Statistics, 1941, United States Department of Agriculture

**Number of operating factories in state, 1942

***Including: Iowa and Oregon, 2 each, and Washington, Indiana,
S. Dakota, Kansas and Wisconsin, 1 each.

TABLE 2. SUGAR CONSUMED IN SELECTED FOOD INDUSTRIES, 1939*

<u>Industry</u>	<u>Total Consumption (lbs.)</u>	<u>Beet (lbs.)</u>	<u>Cane (lbs.)</u>	<u>Corn (lbs.)</u>	<u>Total Cost</u>
Meat Products	52,290,600	14,627,595	35,523,413	4,134,592	\$ 2,344,936
Dairy Products	341,148,674	113,014,935	214,236,863	13,926,876	15,949,016
Canned and Preserved Products	624,571,391	201,284,058	397,728,035	25,569,298	28,158,853
Grain-Mill Products	51,536,837	21,923,405	26,713,602	1,869,830	2,349,901
Bakery Products	1,256,012,386	383,127,668	742,210,826	130,673,892	58,948,204
Confectionery and Related Products	1,052,129,634	290,732,537	736,268,104	25,128,993	47,506,288
Beverages	567,423,543	24,346,816	458,825,909	83,800,818	26,151,764
Miscellaneous**	447,866,044	8,458,899	424,115,139	15,292,006	19,549,993
Aggregate Total	4,634,711,357	1,379,012,498	3,246,632,656	309,066,703	\$211,622,919

*Bureau of the Census, U. S. Department of Commerce, Report of March 14, 1941

**Flavorings, vinegar, cider, pharmaceuticals, etc.

1923-1941 are given in Table 3.

The maximum total distribution of 6,098,311 tons in 1941 is the highest on record. This high value, in comparison with previous years, is attributed to the accumulation of large stock piles by some industrial users, and is not the actual amount used during the year. On the basis of the amount distributed in 1941 the per capita consumption was 121.2 pounds. This figure is misleading; the true consumption was not materially greater than that shown for 1940.

TABLE 3. ANNUAL DISTRIBUTION AND PER CAPITA CONSUMPTION OF SUGAR FOR THE PERIOD 1923-1941

<u>Year</u>	<u>Total Distribution (tons)</u>	<u>Per Capita (lbs.)</u>	<u>Year</u>	<u>Total Distribution (tons)</u>	<u>Per Capita (lbs.)</u>
1923*	4,677,332	94.5	1933**	6,377,500	101.4
1924	5,041,760	99.9	1934	6,349,090	100.1
1925	5,417,770	107.0	1935	6,633,928	104.0
1926	5,641,513	108.0	1936	6,706,195	104.4
1927	5,249,865	101.5	1937	6,671,401	103.1
1928	5,547,439	103.5	1938	6,642,153	101.0
1929	5,746,474	106.4	1939	6,367,518	104.7
1930	5,693,481	102.8	1940	6,890,667	104.6
1931	5,527,091	99.7	1941	8,098,321	122.2
1932	5,435,134	93.4			

*Facts About Sugar, vol. 22, No. 1, 1933, p. 71

**Values for 1933 to 1941 obtained from Sugar Section, Agricultural Conservation and Adjustment Administration, U. S. Department of Agriculture

III. LITERATURE

YEAST GROWTH IN MINERAL-SALT SUGAR SOLUTIONS

Yeasts have been propagated in synthetic nutrient solutions since 1860, when Pasteur (40) first reported the results of his classic experiments on yeast growth in mineral-salt sugar solutions. Pasteur observed that if yeast the size of a pin head was introduced into a solution of sugar, containing phosphate and the minerals in yeast ash, multiplication and fermentation resulted. Almost from the beginning of these studies, until a few years ago, there existed among many workers a controversy regarding the nutritional requirements of yeast. Among the first workers to oppose Pasteur's beliefs was the German chemist, Liebig (41) who, in 1871, contested the statement of Pasteur with regard to the multiplication and fermentation of yeast in a medium free from nitrogenous organic matter. He failed to get the same results when he repeated Pasteur's experiments and vigorously denied the possibility of obtaining either growth or fermentation in a mineral-salt solution.

Other workers also questioned Pasteur's results, although for the following quarter of a century much valuable work was done on synthetic media based on the results of Pasteur. Few papers were published, however, following the controversy between Pasteur and Liebig, each of whom had either formulated or supported separate theories regarding alcoholic fermentation.

EFFECT OF BIOS ON YEAST GROWTH

In 1901, Hildiers, (51) renewed the interest in this field of investigation by suggesting that a synthetical substance, which he first

found to be present in yeast water, and designated as "bios", was necessary for the fullest development of yeast in synthetic nutrient solutions. His observation that growth was activated by the addition of a filtrate from boiled yeast led him to believe that, although yeast would grow in a synthetic medium without this substance, its development was much better when present. He found that small seedings with Saccharomyces cerevisiae produced no fermentation in his medium composed of water, sucrose, magnesium sulfate, potassium chloride and calcium chloride, while large seedings did produce fermentation. He concluded that growth was stimulated, not because of an increase in the number of cells used, but by some growth-promoting substance which was added when large seedings were employed. Growth of yeast in this medium was activated by the addition of a filtrate from boiled yeast; the evolution of carbon dioxide gas serving as a measure of activation. Wildiers' data on the effect of yeast extract upon the fermentation with small seedings of yeast are as follows:

cc. boiled yeast emulsion

Time days	1	2	3	4	5
	cc.CO ₂	cc.CO ₂	cc.CO ₂	cc.CO ₂	cc.CO ₂
2	0	0	0.5	1.2	2.5
3	0	0	1.0	2.1	4.7
4	0	0	1.2	3.0	5.6

Some criticism has been offered regarding the title of Wildiers' original paper which was "Une nouvelle substance indispensable au développement de la levure". Those criticising this title believed

the words "normal development" should have been used, since it has been found that bios is not absolutely necessary for the fullest development of yeasts but when absent, small, unhealthy cells result. He proposed the name bios because the substance was unknown chemically and because this name would be satisfactory until the factor was better understood. The Greek meaning of the word bios is life. Filialani described bios as follows:

1. Soluble in water.
2. Insoluble in absolute alcohol and ether; 80 percent alcohol, however, permitting a good extraction.
3. Not present in yeast cell. It is then not an inorganic substance.
4. Not destroyed by boiling for a half hour in a 5 percent solution of sulfuric acid. To destroy it, a 20 percent solution was necessary.
5. Bios seemed to be changed by one-half hour boiling in 1.0 percent solutions of NaOH.
6. Not precipitated by lead acetate.
7. Filterable.
8. Contained in Liebig's meat extract, commercial peptone and in beer wort.
9. Bios is not present in such substances as urea, isoargin, aniline, tyrosine, nucleic bases, adenine, quinine, thymus nucleic acid, creatine, or peptic and tryptic digestion products of albumin.

Fildiers' explanation for the differences that existed between the results obtained by Pasteur and Liebig was based on the amount of yeast used as the inoculum. Whereas Pasteur used a portion of yeast the size

of a pin head, Liebig used a smaller amount. This difference in the size of the inoculum seemed to Liebig to be sufficient to account for the failure of growth in Liebig's experiments because, when he repeated Pasteur's experiments, he found that there was a minimum limit to the number of cells that would give growth and that much depended upon the nutrient medium from which the cells were taken.

Liebig's work, like Pasteur's, was soon to become the subject of such controversy based on the amount of inoculum used to seed synthetic media, the carry-over of stimulants in the inoculum to fresh media and, also, the influence of impurities in the media ingredients. Fernbach (14) was the first worker to challenge Liebig's conclusions by suggesting that his results were due to the presence of toxic substances in the culture media. This opinion was later shared by Kindisch (70) who believed that the presence of copper salts introduced into the media through distilled water, as suggested by Fernbach (14), would inhibit yeast growth. Although Kindisch was one of Liebig's active opponents, he was able to confirm Liebig's fundamental observation. However, he preferred not to offer an explanation for it until after further investigation. The possibility that Liebig's lies counteracted the toxic effect of copper and other salts, thus permitting growth, was expressed by certain workers. Kindisch's opinion regarding the influence of impurities on yeast growth was furthered when he observed that yeast failed to develop in cane sugar solutions because of the presence of traces of ultramarine which had been used for blueing the sugar.

Liebig was not without supporters of his views for around (4), 1907, defended him against Kindisch and other German workers who believed that impure water was responsible for some of the results obtained. He

showed that water, which was an ingredient of the medium, was free of toxic substances and, therefore, was not a factor. Winisch suggested that the sugar used by Brand contained substances which were poisonous to yeast.

The first reference to sugar as an ingredient inhibiting the development of yeast appears to be by Winisch. Koszowicz, (39) in 1903, was another worker to recognize the possible influence of impurities in sugar on yeast growth, although unlike Winisch, he recognized a stimulative effect of sugar impurities on yeast development. He was able to determine that a difference existed in the stimulative ability of sugar by carefully determining the size of the yeast crop in duplicate experiments in which the sugar was the only variable. The composition of the medium for one of his experiments (third) was as follows:

Water	100 ml.
KCl	0.2 gram
$(\text{NH}_4)_2\text{HPO}_4$	0.1 "
MgSO_4	0.02 "
$\text{Ca}_2\text{H}(\text{PO}_4)_2$	0.02 "
Commercial refined sugar	5.0 "

The composition of the medium of another (fourth) was the same, except that pure saccharose was used. In the third experiment there was no development of cells during the first two weeks, but after forty days there were 364 million cells in 100 ml. of the medium. In the fourth experiment, in which the same strength inoculum was used, there was no development during the first three weeks, but after sixty days there were 220 million cells in 100 ml. of the medium. In addition to demonstrating the differences in the biological quality of sugar, Koszowicz contributed otherwise to the subject not only by improving on the methods for estimating the progress of yeast development, but by his studies of the extent

of growth which resulted from variable numbers of cells when seeded into nutrient solutions. Kossovic concluded his paper as follows:

"We have seen that commercial sugar retards yeast growth and fermentation. Also, in these cases the slight traces of organic impurities in the crude sugar may be responsible for the retardation. The facts mentioned above do not justify the assertion that for the multiplication of yeast certain organic compounds, aside from sugar, are necessary, although it must be admitted that traces of organic compounds are of vast influence on the capability of yeast growth. This work emphasizes the symbiotic nature of the yeasts."

A number of papers were published following Kossovic's work on various problems relating to the nature of bios, its origin and its role in yeast nutrition, but workers had apparently lost interest in the subject until the vitamin question was opened. Williams' work immediately took on a new significance because it was suggested that bios might be a vitamin-like substance. A number of investigators subscribed to this belief; Funk and Lucin (16) later suggested that the extent of yeast growth in test solutions be used as a measure of the vitamin B content of vitamin-containing materials.

Williams (59), in 1919, was the first to suggest the possible relationship that existed between the vitamin and bios complex, as a result of his observations on the requirements for an unknown organic constituent (or constituents) of wort or yeast extracts. It was shown:

"that the substance promoting the growth of yeast occurs in the same materials as those in which vitamins have been found; namely protein-free milk, wheat germ, lactose (Kilbourn), yeast, egg yolk and pancreatin. The substance is none of the commoner amino acids contained in an acid digest of casein with tryptophane added. It has the same properties of solubility, precipitation by phosphotungstic acid, absorption on fuller's earth, heat stability, and behavior toward acid and alkali, as nearly as we know these properties, as the water soluble vitamins and in addition the two substances so far as we know have no divergent properties.....From the cumulative evidence offered we believe we are justified in concluding that as far as present knowledge is concerned the substance

of these factors is not sufficient to explain the growth of the animals. It is the influence of substances which in animal nutrition present biotin-like properties."

Prior to the association of biotin to the biotin factor, the "vitamin" was supposed to designate a single vitamin substance, the identification of other substances was described as the "vitamin B complex." The existence of two water-soluble "B" vitamins was suggested by Vitonchi (47) in the same year and shortly after the publication of the paper referred to above. Until 1936, in the literature, the other substances described pertaining to the effect by nutritive substances on the growth of yeast and certain other organisms was located in yeast extract; however, there was no mention of either biotin or vitamin B by these writers. In the same paper, Schramm (9) indicated that yeast fermentation was stimulated by vitamin-like material and concluded that:

"The amount of organic material required is so small and the effect upon the growth and activity of the yeast is so striking that a marked similarity exists between these organic substances and the vitamins required for the normal development of animals."

~~Prior to these publications which associated vitamins with biotin,~~
there were innumerable papers on the biotin question; many others followed later.

Tramer (56) in 1935, reviewed the biotin question and included a bibliography of 144 references. Miller, (44) and Perrett, (50) also reviewed the subject. Many of the papers concerned the indispensability of biotin to the fullest development of yeast, while others described new media designed to test the significance of biotin. Still others attempted to decide on the identity or non-identity of biotin and the vitamin B complex based on studies on crude plant or animal materials. As a result of these studies, it has become known that the entire biotin effect cannot be ascribed to any single substance, and that "vitamin B" activity is

likely to be a single substance but rather to a complex of organic compounds. There is a great overlapping of compounds in the two extracts. One of the authors, however, the multiple nature of the extract obtained by Adams, (42) and obtained also by Bior II by fractionation after Adams. Miller, McCusker and Darling (45) and Miller, McCusker and Darling (46) also contributed to further fractionating Bior II into Bior II and Bior IIB.

More recently, Williams, (41) in 1941, reviewed the subject in consideration of the most recent knowledge available on the identification of growth-inhibiting nutrients for yeasts. In addition to reviewing the specific growth inhibitors which have been identified and studied, he discussed the causes for confusion which have arisen regarding interpretation of results previously obtained. Among these are (1) the different strains of yeast which have been used by various investigators; (2) diversity of experimental procedures; (3) differences in the source of nutrients; and (4) unrecognized potency of some nutrients.

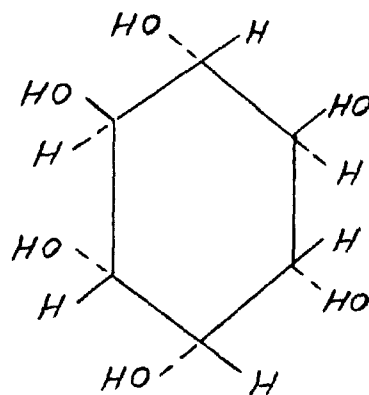
~~number of specific compounds are now known which influence the~~
growth of yeast. The role of vitamin activities are well determined; many of them comprising the vitamin B complex. The most important of the substances are: inositol, thiamine, pantothenic acid, biotin and pyridoxine. While it is not the purpose of this review to discuss in detail each of the above compounds, brief mention is made of the origin, composition and occurrence, and role of these factors known to influence yeast growth.

EFFECT OF INOSITOL ON YEAST GROWTH

Inositol, which was isolated by L. J. McCusker (13) from yeast, was the first yeast substance found to possess bioactivity. It is the physio-

In the first instance, it is to be noted that the
 (19) is a derivative of the parent compound, and is
 derived from it by the substitution of a hydroxyl
 group for a hydrogen atom. The parent compound is
 a cyclic ether, and the hydroxyl group is attached
 to the carbon atom adjacent to the oxygen atom.
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 and is derived from it by the substitution of a
 hydroxyl group for a hydrogen atom. The parent
 compound is a cyclic ether, and the hydroxyl group
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 of the parent compound, and is derived from it by
 the substitution of a hydroxyl group for a hydrogen
 atom. The parent compound is a cyclic ether, and
 the hydroxyl group is attached to the carbon atom
 adjacent to the oxygen atom.



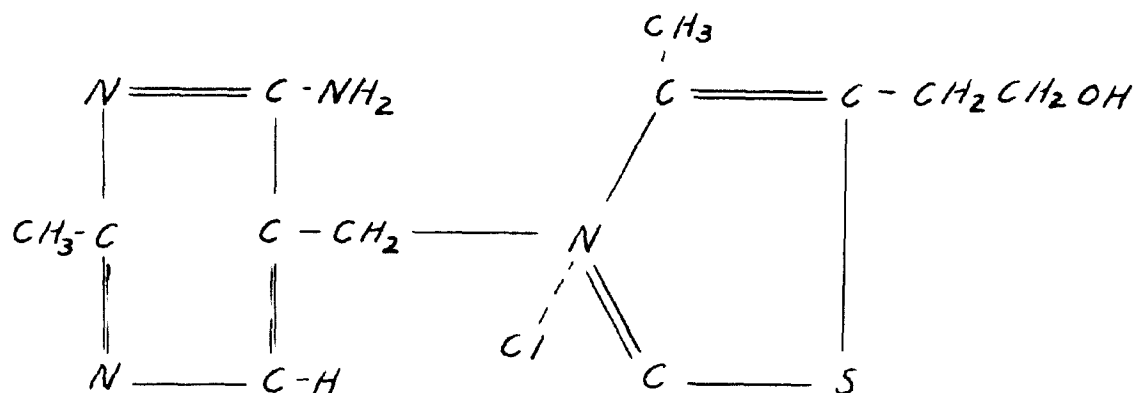
The (19) is a derivative of the parent compound,
 and is derived from it by the substitution of a
 hydroxyl group for a hydrogen atom. The parent
 compound is a cyclic ether, and the hydroxyl group
 is attached to the carbon atom adjacent to the
 oxygen atom. The (19) is a derivative of the
 parent compound, and is derived from it by the
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 derivative of the parent compound, and is derived
 from it by the substitution of a hydroxyl group
 for a hydrogen atom. The parent compound is a
 cyclic ether, and the hydroxyl group is attached
 to the carbon atom adjacent to the oxygen atom.

physiology. Janssens (32) found inositol to be indispensable for the growth of "Milliers' yeast" (Milliers, 31) and Inell (63) indicate that "old process" yeast may require inositol for continued production. Loebel and Landstein (41) found inositol to be necessary for the growth of one of two of 13 strains of osmophilic yeast.

EFFECT OF THIAMINE ON YEAST GROWTH

Thiamine (vitamin B₁) was first isolated from natural sources by Jansen and Donath, (33) in 1926. The effect of this antineuritic vitamin on yeast growth was conclusively demonstrated by Williams, Wierman and Kereczuk, (69) who tested the crystalline vitamin as prepared by Jansen and Donath. They found that it was effective in concentrations as low as 0.00001 mg. per ml. of culture medium, when supplemented by a residue of rice bran which was unabsorbed by fuller's earth. It was found in a wide variety of foods, but there are few foods of either plant or animal origin that can be considered a potent source of thiamine. Its importance in nutrition is well recognized, some foods being enriched with thiamine chloride.

The knowledge regarding the chemical identification of thiamine resulted largely from the efforts of Williams and coworkers (7, 63, 69) who determined its empirical formula to be C₁₂H₁₈N₄SOCl₂. The synthesis of vitamin B₁ was effected by Williams and coworkers soon after determination of the structure of thiamine. It was found to contain a pyrimidine and a thiazole nucleus, the structural formula of thiamine chloride being determined as shown.



The response of yeast to thiamine varies with different cultures. Williams and Soehn (64) recorded varied response by "old process" and yeast No. 574 of the American Type Culture Collection, while several other yeasts failed to respond similarly under the same conditions (52, 63). Schultz, Stein and Frey (51, 53) observed that yeast fermentation is greatly enhanced by the presence of thiamine and developed a quantitative method for determining small amounts of thiamine used in a fermentation test. By their original method they were able to estimate the presence of 0.5 to 4.0 micrograms of vitamin B₁.

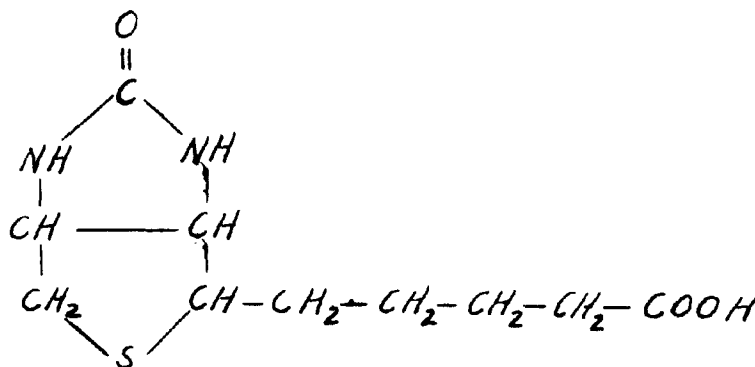
EFFECT OF BIOTIN ON YEAST GROWTH

Crystalline biotin was first isolated in the form of its methyl ester by Kogl and Tanaka (36), in 1936. Its isolation from the yolk of duck eggs was effected by a lengthy series of fractionations which resulted in the recovery of less than five percent of the total amount present. Because of the extremely small amount present in natural sources and the difficulty attending its recovery, Kogl (36) was able to obtain but 70 milligrams of crystals during five years of work. Only a few more have been isolated since development of a recovery process.

Three groups of workers studied and named a growth-promoting substance which later became identified as biotin. Allison and coworkers (2, 3, 31) found that several species of root-nodule bacteria, Rhizobium, required the presence of a growth factor which was obtainable from yeast, molasses and commercial saccharose. This factor, which was found to have a pronounced effect on cell respiration, was designated as a coenzyme of oxidation, namely, coenzyme R. Subsequent work by East and Allison (37) in which comparisons were made on the effect of growth factor, biotin and coenzyme R indicate that they are identical.

Second group, György and coworkers (6, 11, 11, 12, 13, 14, 15) studied a substance occurring in yeast and liver which was capable of preventing the fetal syndrome resulting from feeding large amounts of raw egg white, a syndrome found to occur in all species studied. This substance, which was originally designated as vitamin H, was later recognized as a member of the vitamin B complex, namely, biotin. The third group of workers included Miller and coworkers (45, 46) who fractionated Biotin II into Biotin IIA and Biotin IIB. Kogl and Tönnis (38) indicated the identity of Biotin IIB as biotin while studying the properties of crystalline biotin from egg yolk.

Kogl and Price (37) showed the empirical formula of biotin to be $C_{21}H_{31}O_6N_2S$ and Du Vigneaud (10) believes its structural formula to be



vidence to support the determination of the structural formula is presented by the latter writer and need not be reproduced here.

Biotin is present in low concentration in its natural state; the vitamin activity is extraordinarily in minute amounts as compared with most other substances. Shall, and Williams (55) determined that Baker's yeast contained about 0.12ug. of biotin per gram as compared with 10ug. or more of thiamine per gram. Caldwell and Williams (7) found the biotin content of dried eggs to be 0.35ug. and 0.4ug. per gram, respectively. They determined the biotin content of a sample of sugar to be 0.004ug. per gram.

1891 showed that only 0.0004ug. of biotin is necessary in 2 ml. of culture medium to cause a 100 percent increase in the growth of *Saccharomyces* yeast in 5 hours. Shall and coworkers, (55) believe that biotin is the limiting factor in the growth of yeast and showed that the presence of a minute amount of pyridoxine greatly increased its growth-promoting effect. They also believe that yeast is able to synthesize biotin during growth. Lockwood and Langer (4) found biotin to be essential to the growth of all 17 strains of *Aspergillus* yeast studied.

EFFECT OF PANTOTHENIC ACID ON YEAST GROWTH

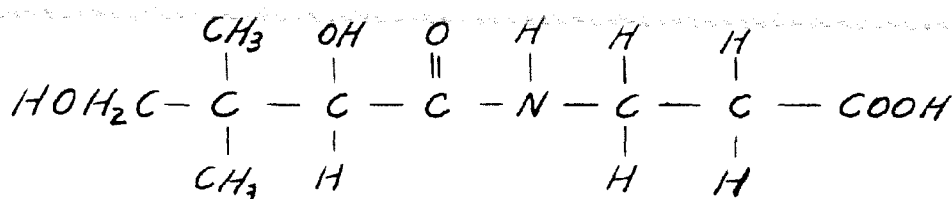
Williams and Gregory (61) and Williams and Gross (6), in 1931, were the first to indicate that the material for the "factor X" was a single substance. Later, (1934), Hoverson, Largent and Williams (51) determined its universal presence in plant and animal tissues and designated the substance "pantothemic acid", the name being derived from the Greek meaning from everywhere. Demonstration of its acid properties was made while studying the identity of "Soldiers'

sies by a new tool, namely, fractional electrical transport in a A.P. potential field without the use of membranes. In 1935, Williams (60) reviewed the results of this method for the separation of relatively weak acids and bases of low molecular weight. In 1941 he reviewed (61) the entire subject of pantothenic acid.

Williams and his coworkers were solely responsible for the identification of pantothenic acid. Considerable interest is attached to its identification by Williams (61) who states that:

"The outstanding point of interest in the pantothenic acid investigation is the fact that so much regarding its structure was determined before it was obtained in pure condition. In fact, the complete structure was known and the physiologically active substance was synthesized before pantothenic acid or its salts or other derivatives (excluding cleavage products) were obtained in condition such as to yield the correct analyses."

The acidic form for calcium pantothenate (the most stable form) is $(C_{12}H_{16}O_5N)_2Ca$ and the structural formula for the physiologically active acid (condensed with pyrimine) is:

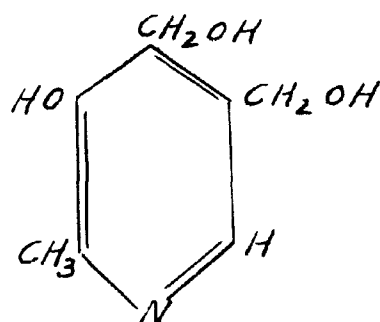


During the course of the investigation of the influence of pantothenic acid on yeast growth the "Gebrüder Mayer" strain was usually used, however, it was found to be an inept nutrient for all of the strains tested by Williams and Lounsbury (65) and Williams, in 1942, Baell (62). "Gebrüder Mayer" yeast does not appear to be able to synthesise its nutrilitic although the rate of synthesis by other

strains may be a limiting factor for multiplication.

EFFECT OF PYRIDOXINE ON YEAST GROWTH

Pyridoxine (vitamin B₆) was isolated in crystalline form by Armstrong and Stevens (35) and by Kossy (40) in 1938. Its vitamin activity, as representative of yeast title in rats, was recognized by G. B. Py, (39) in 1935. Its stimulatory effect on yeast growth was recognized simultaneously by Dembits, Itin and Lee (54) and by Williams (15) in 1939. The richest natural source of it is vitamin B₆ rich vegetables. It can be synthesized by bacteria and fungi (14) in 1935. Its structural formula is shown as:



The crystalline vitamin has shown by Dembits and co-workers (54) in India and Williams (15) to be active in amounts down to 0.0001 mg. per ml. In order for the B₆ response to be obtained, the medium must contain other nutrients, particularly pantothenic acid and biotin. Vitamin B₆ can be synthesized by yeast during growth, so that its presence is dispensable.

EFFECT OF OTHER NUTRIENTS ON YEAST GROWTH

Numerous other compounds are reported (61) which affect the growth of yeast in solutions. Some are known to be chemical constituents of yeast so that their role in yeast nutrition is assured. They

are: ascorbic acid, leucine and other amino acids; uracil, which probably enters into the synthesis of nucleic acid; ethanolamine; nicotinic and nicotinic acid, and riboflavin. Some other compounds showing slight effects are: indole butyric acid, carotene, choline, acetyl choline, colchicine, deuterium oxide, dibenzanthracene derivatives, nitrobenzene and follicular hormone. None of these compounds shows activity in the small amounts in which described more fully.

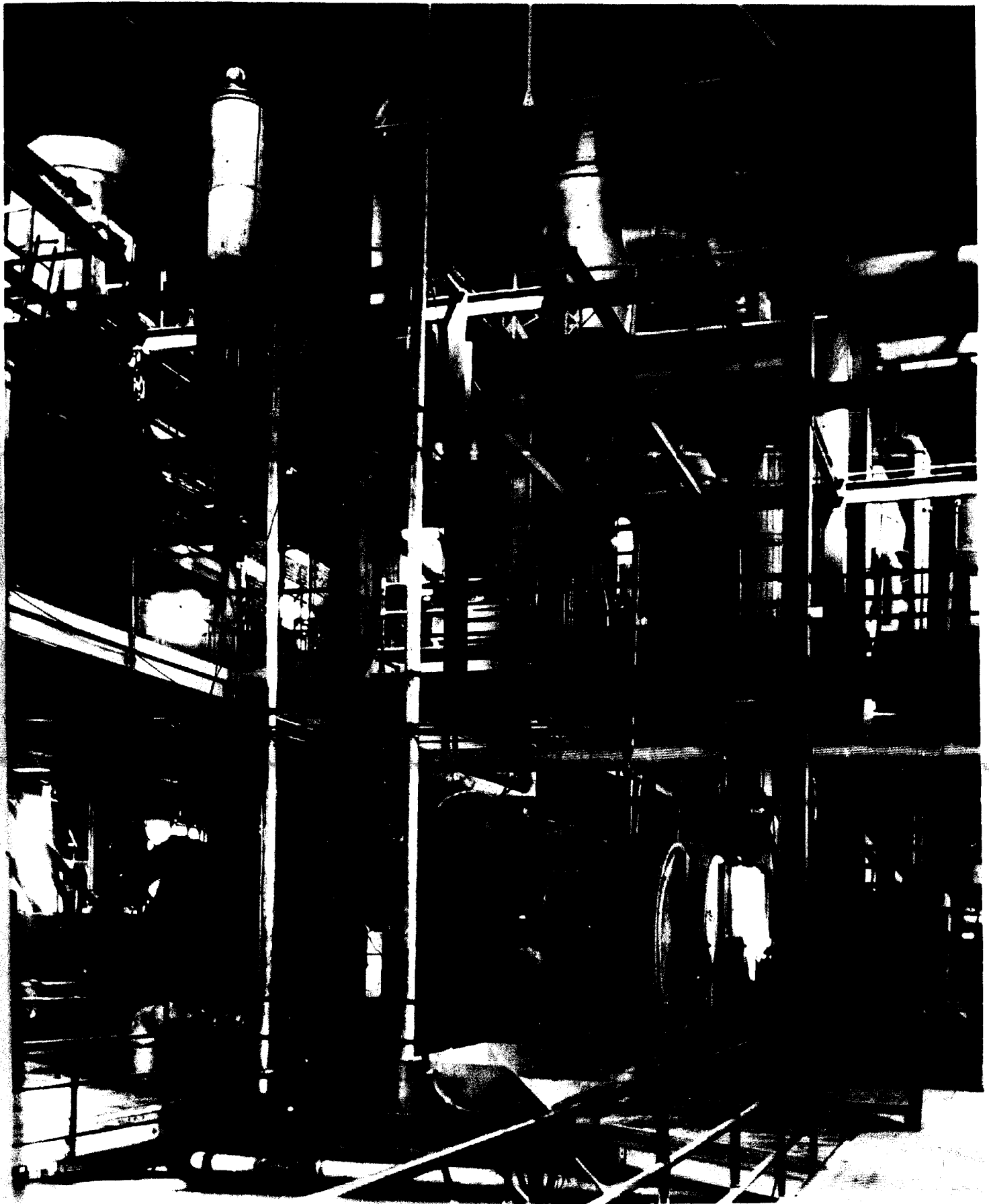


Illustration 2. INTERIOR VIEW OF MODERN SUGAR FACTORY

IV. EXPERIMENTAL

THE YEAST

Preliminary studies were made during which a culture of yeast was selected for the test. The culture was isolated from a bottle of a fermented beverage in which it had grown and produced a visible sediment. It is identified as a species of the genus Saccharomyces. The culture which is design here as "beverage yeast" in this study was chosen because of its source and also because of its growth characteristics in response to stimulants in sugar solutions. Soon after isolation, its cultivation in a number of sugar solutions indicated that the extent of multiplication varied in different sugars and, first, all cultures, dependent on the presence or amount of unknown substances in the sugar. The results also indicated that a good differentiation was obtained in the size of the yeast cross from solutions made from several sugars, when they were inoculated with approximately 50,000 yeast cells per ml. of test solution and incubated at 30° C. for 72 hours. There was greater reproduction of the inoculum in solutions containing 10 percent sugar than in solutions containing lesser amounts as shown in Table 4, Figure 1. The 10 percent concentration of sugar, which supported maximum growth, was chosen for this study, since it approximated that of beverages and also because numerous other workers have used this concentration of sugar in yeast nutrition studies.

Table 4. DEVELOPMENT OF YEAST IN MULTIPLES OF INOCULUM
IN 1-15 PERCENT SUCROSE SOLUTIONS

SUCROSE CONCENTRATION PERCENT	YEAST INOCULUM MULTIPLE
1.0	1.4
2.0	10.8
3.0	8.3
4.0	19.1
5.0	19.0
6.0	17.8
7.0	22.8
8.0	26.7
9.0	28.2
10.0	31.4
15.0	30.8

Inoculated with 50,000 cells
per ml.

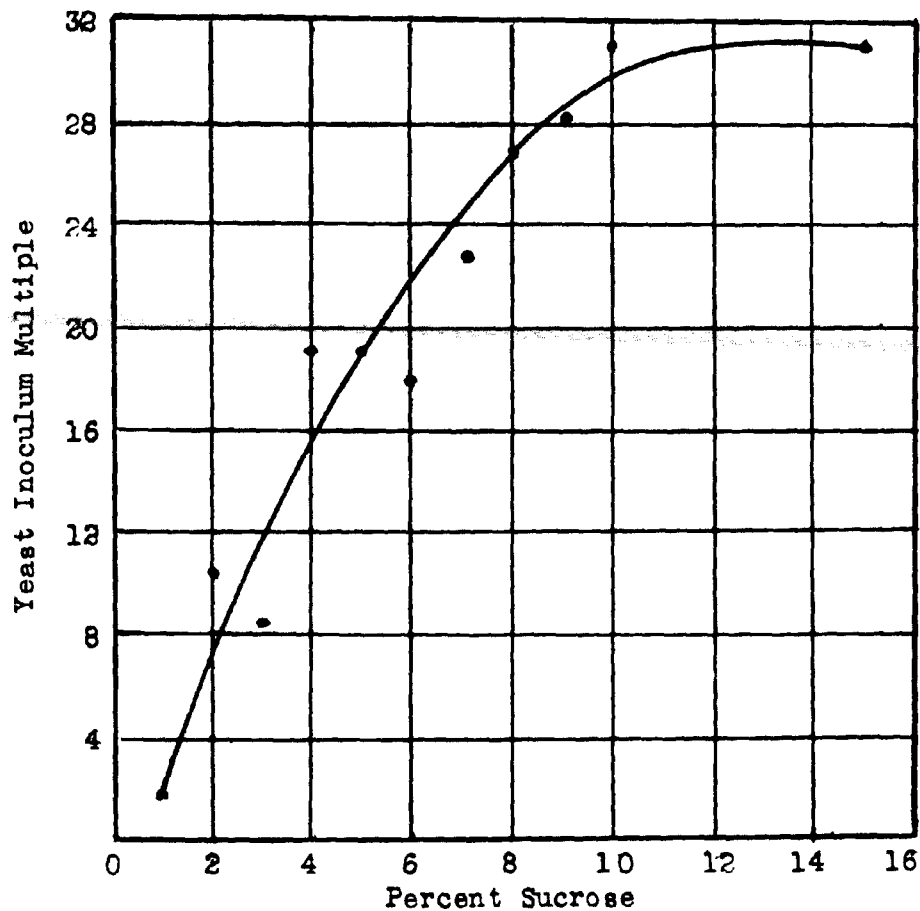


Figure 1. Development of yeast in multiples of inoculum in
1-15 percent sucrose solutions. Inoculated with
50,000 cells per ml.

METHOD FOR ESTIMATING YEAST INOCULUM EFFICIENCY

On the basis of the above observations, a standard test technique has been developed for estimating the relative amount of yeast growth-promoting activity in a sample of yeast. A detailed description of the technique is as follows: 1) grams of each sample of yeast to be tested are weighed into a 300 ml. volumetric flask fitted with a cotton plug. Sufficient distilled water is added to give 100 ml. of 10 percent solution after sterilization in the autoclave at 15 pounds pressure for 20 minutes. Following sterilization and cooling, the solutions are inoculated with the yeast culture. For the purpose of this report, the inoculating suspension is designated as the "inoculum" and is the number of cells introduced per ml. of test solution. The inoculum is prepared from a 24-hour culture grown on a yeast agar slant. The mass of cells is transferred into a sterile transfer media to about 100 ml. of sterile, distilled water. After vigorous shaking, the number of cells per ml. is determined by counting with a hemacytometer. The amount of inoculum necessary to add about 50,000 cells per ml. of test solution is calculated. For extreme accuracy in counting, the cells and diluting the inoculum, it should be adjusted so that from 0.5 to 1.5 cells are used. The inoculated solutions are incubated at 30° C. for 72 hours during which time they are gently agitated at 24-hour intervals. The number of cells that develop are determined by again counting 400 cubic mm. fields with a hemacytometer. A few glass beads are added to the flask to assist in breaking up clusters of cells. The number of cells that develop per ml. divided by the number of cells per ml. used by the inoculum gives the efficiency

TABLE 5. RELATIVE RISES OF 1 - 2 AND 3-4% AT 30°C IN
DIFFERENT SUGAR MEDIA INITIATED BY YEAST INOCULUM AT 1-
10%.

Type Sugar	Yeast Inoculum Multiple	Type Sugar	Yeast Inoculum Multiple	Type Sugar	Yeast Inoculum Multiple
Distilled Water	1.0				
B	1.0	B	2.2	B	10.0
B	1.0	B	3.0	B	10.1
C	1.0	B	3.7	C	10.8
B	1.0	B	4.0	B	10.1
B	1.0	B	4.0	B	10.1
B	1.0	B	6.1	B	10.8
B	1.1	B	6.5	PC	11.4
C	1.1	B	6.3	B	11.7
B	1.2	B	8.4	B	13.1
B	1.4	B	10.4	C	14.1
B	1.4	B	11.1	PC	13.8
B	1.5	B	12.5	B	15.3
B	1.8	C	13.1	PC	15.8
B	2.1	C	14.5	PC	15.5
B	2.2	PC	15.5		

Inoculum 35,000 cells per ml.

B - Beet Sugar

C - Refined cane sugar

PC - Plantation cane sugar

the presence of growth-promoting substances in some of the refined cane sugar is believed to be due to the very low dilution ratios. The wide range of growth-promoting activity which is obtained (see Table 22) between the two dilutions, however, is not sufficient, in supporting evidence for this belief. There is no knowledge of the mechanism by which impurities in cane sugar could originate in the sugar. Indeed, it is known that the results from the action of factory operations, i.e., effluents, are not sufficient to account for both the growth-promoting and the inhibiting effects of both the refined and the unrefined sugar.

The failure of the inoculum to develop in some of the cultures suggested the possibility of the presence of toxic substances in the sugar, rather than of growth-promoting substances. This possibility was investigated by diluting the sugar in varying proportions, and also by adding to it known amounts of growth-promoting substances. It was assumed that if toxic substances were present, multiplication of cells would not occur, regardless of the amount of substances added by the inoculum. If toxic substances were not present, then multiplication should occur, the extent depending on the amount of growth-promoting substances added through the inoculum.

The results obtained fail to support the idea of the possibility of the existence of toxic substances. Other tests, in which refined sucrose (C. P. grade) was used, failed to give any indication of the presence of toxic substances in the sugar. Yeast cells, which were known to be several days in solution but failed to support their multiplication, were not stained by methylene blue, thus indicating that death of the cells had not occurred.

examined were more numerous in 1933 than in 1931, and extended throughout the 1941 season. Data are available for the samples collected in 1931, 1933, or 1933 are given in Tables 6, 7 and 8. The exact location within the water body is given in preceding paper.

Twenty-nine samples were collected and received from the lake in 1931. These samples, although they were in general not far from the shore, were not collected in the same manner as in preceding years. The range of total inorganic phosphorus is from 1.0 to 13.0, average for 6.6. The range of total phosphorus is from 1.1 to 17.3, average for 6.5 for 6 samples received in 1931. It should be pointed out, however, that the maximum value of 17.3 for sample No. 61 greatly exceeds the value of 19.1 for sample No. 47. A study of the factory operation of the lake and results of the chemical analysis (in constituents, organic and inorganic compounds) of the irrigation of the lake and the possibility of a situation that would account for the high value. The maximum maximum values of 1.0 to 13.0, which were obtained for the 6 samples received in 1932, are in the same order of magnitude as the previous years. The average multiple value of 6.9, as compared with 6.5 for the previous year, indicates a decreased amount of phosphorus in the water in 1933.

The results obtained by the examination of the individual organic composite materials presented in Tables 6, 7 and 8 are typical of those obtained during the entire year of this study. Therefore, in view of the results of the examination of the samples, which are outlined for the individual samples for each of the remaining samples from 1934 to 1941, the results are given with pre-

TABLE 6. RELATIVE AMOUNT OF YEAST GROWTH PROMOTING SUBSTANCE IN
MAY 1951 IN 1951 ON HIGH CARBON DIOXIDE ATTEMPTED BY
YEAST INOCULUM MULTIPLES

<u>Sample</u>	<u>Yeast Inoculum Multiple</u>	<u>Sample</u>	<u>Yeast Inoculum Multiple</u>
1	1.0	16	4.0
2	1.0	17	4.0
3	1.0	18	6.1
4	1.0	19	6.5
5	1.0	20	6.9
6	1.1	21	8.4
7	1.4	22	10.4
8	1.4	23	11.1
9	1.5	24	11.6
10	1.8	25	15.7
11	2.1	26	18.1
12	2.3	27	20.1
13	2.4	28	20.1
14	3.0	29	23.0
15	3.7		

Number of cells in inoculum, 35,000 per ml.

TABLE 7. RELATIVE AMOUNT OF SULFUR DIOXIDE-PRODUCING SUBSTANCES IN BEET SUGAR IN 1932 CAMPAIGN DEPOSITS
 AS ESTIMATED BY YEAST INOCULUM MULTIPLES

<u>Sample</u>	<u>Yeast Inoculum Multiple</u>	<u>Sample</u>	<u>Yeast Inoculum Multiple</u>	<u>Sample</u>	<u>Yeast Inoculum Multiple</u>	<u>Sample</u>	<u>Yeast Inoculum Multiple</u>	<u>Sample</u>	<u>Yeast Inoculum Multiple</u>
1	1.0	15	1.3	29	3.4	43	6.2	57	17.0
2	1.0	16	1.5	30	3.4	44	6.2	58	12.4
3	1.0	17	2.0	31	3.6	45	7.0	59	15.0
4	1.0	18	2.4	32	3.8	46	7.5	60	17.6
5	1.0	19	2.6	33	3.3	47	7.4	61	21.2
6	1.0	20	2.6	34	3.8	48	7.8	62	23.0
7	1.0	21	2.8	35	4.2	49	8.2	63	25.0
8	1.0	22	2.8	36	4.2	50	8.4	64	25.0
9	1.2	23	3.0	37	4.6	51	8.8	65	32.0
10	1.2	24	3.0	38	4.6	52	9.6	66	35.8
11	1.4	25	3.0	39	4.8	53	10.8	67	39.2
12	1.4	26	3.2	40	4.8	54	11.2	68	40.2
13	1.6	27	3.2	41	5.2	55	11.6		
14	1.6	28	3.4	42	5.4	56	12.0		

Number of cells in inoculum, 47,000 per ml.

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TABLE 8. GROWTH OF *STREPTOCOCCUS* IN BROTH CULTURE IN 1973. (Data from 1973)

Sample	Yeast Inoculum	Yeast Inoculum	Yeast Inoculum	Yeast Inoculum	Yeast Inoculum	Yeast Inoculum	Yeast Inoculum	Yeast Inoculum	Yeast Inoculum
1	1.0	1.0	1.0	1.0	1.0	1.0	1.0	1.0	1.0
2	1.0	1.0	1.0	1.0	1.0	1.0	1.0	1.0	1.0
3	1.0	1.0	1.0	1.0	1.0	1.0	1.0	1.0	1.0
4	1.0	1.0	1.0	1.0	1.0	1.0	1.0	1.0	1.0
5	1.0	1.0	1.0	1.0	1.0	1.0	1.0	1.0	1.0
6	1.0	1.0	1.0	1.0	1.0	1.0	1.0	1.0	1.0
7	1.0	1.0	1.0	1.0	1.0	1.0	1.0	1.0	1.0
8	1.0	1.0	1.0	1.0	1.0	1.0	1.0	1.0	1.0
9	1.0	1.0	1.0	1.0	1.0	1.0	1.0	1.0	1.0
10	1.0	1.0	1.0	1.0	1.0	1.0	1.0	1.0	1.0
11	1.0	1.0	1.0	1.0	1.0	1.0	1.0	1.0	1.0
12	1.0	1.0	1.0	1.0	1.0	1.0	1.0	1.0	1.0

Number of cells in inoculum, 50,000 per ml.

vious results for each year in Table 9.

The average yearly multiple values show a varied variation over the 11-year period. Beginning with the campaign year 1938, there are three distinct periods during which the average multiple values showed yearly increases not only to be followed by sudden increases which are again followed by yearly decreases. It would appear that the average multiple values for each year are determined by the number of fish caught in the fishery. However, it is not clear from the data whether the multiple values are directly proportional to the number of fish caught. It is significant that the multiple values for each year are not exactly the same but are close enough to each other to permit the use of the sugar in the fishery for each year. The multiple values for each year are close to each other, with the exception of the 1940, 1941, and 1942 years, which are slightly higher than the other years. The multiple values for each year are close to each other, with the exception of the 1940, 1941, and 1942 years, which are slightly higher than the other years.

Heavily increased production of fish, the result of increasing production, which is a result of the selection of operation of the fishery, or of other factors, have to be located. The fishery is heavily increased. During this period of time, however, it is not clear from the data whether the multiple values are directly proportional to the number of fish caught. It is significant that the multiple values for each year are not exactly the same but are close enough to each other to permit the use of the sugar in the fishery for each year. The multiple values for each year are close to each other, with the exception of the 1940, 1941, and 1942 years, which are slightly higher than the other years.

TABLE 2. AV. NO. OF INSECTS PER PLANT SAMPLE TO WHICH
100 G. OF FERTILIZER WAS APPLIED IN 1931-1941

<u>Year</u>	<u>No. Samples</u>	<u>No. Plants Inspected per A.</u>	<u>No. of Insects per Sample</u>		
			<u>Average</u>	<u>Minimum</u>	<u>Maximum</u>
1931	29	35,000	6.6	1.0	23.0
1932	63	47,200	7.5	1.0	39.8*
1933	56	50,000	6.9	1.0	36.4
1934	59	48,000	4.8	1.0	13.4
1935	65	49,700	14.4	1.0	59.6
1936	70	47,000	8.6	1.1	27.4
1937	70	51,000	6.0	1.6	19.2
1938	77	50,000	7.6	1.4	25.1***
1939	75	53,400	9.1	1.9	40.1**
1940	37	55,000	8.0	1.1	18.4
1941	37	50,000	6.6	1.0	16.6

*only one sample above 39.8

**only one sample above 40.1

***only one sample above 13.2

amount of growth-promoting substance in the diet. This value is divided into two parts, i.e., a lower and an upper composite value, as the lower and upper percent of the samples in the group is obtained. It is also, however, possible to obtain for the group of samples a lower and an upper composite value, i.e., the lower and upper percent of the samples in the group is obtained. The results are given in Table 10.

The relationship which exists between the amount in samples and the amount of growth-promoting substance is directly related to the amount of the diet. There is also a direct relationship between the amount of growth-promoting substance and the amount of the diet. This is not the case, however, because the amount of growth-promoting substance is not directly related to the amount of the diet.

Since the relationship which exists between the amount and the growth-promoting substance for the samples composite values examined, this relationship is stated on the basis of individual samples over the 11-year period. It is shown that a direct correlation might be found between these values and the amount of growth-promoting substance or the ash content of the samples. By using the amount of the diet, this is usually estimated daily during the construction, and only for estimating the relative amount of growth-promoting substance, information would be available immediately regarding this quality of the diet. In order to show this relationship, the results of the composite values and the percent of individual secondary composite values were obtained on a daily basis. The

TABLE 10. PERCENTAGE OF INSECTS BEING COLLECTED IN THE YEAR 1931 - 1941

Year	Insects collected	Percent collected	Percent of insects collected		Percent of insects collected	
			Average collected	Average collected	Average collected	Average collected
1931	6.6	.0074	.0044	.0043	19	10
1932	8.5	.0070	.0049	.0046	50	18
1933	6.9	.0097	.0036	.0033	40	10
1934	4.8	.0040	.0030	.0029	36	12
1935	11.4	.0032	.0016	.0020	27	18
1936	4.6	.0035	.0030	.0021	30	31
1937	7.0	.0016	.0025	.0021	37	13
1938	7.0	.0039	.0006	.0016	47	30
1939	4.1	.0010	.0004	.0017	27	33
1940	4.0	.0039	.0025	.0024	45	42
1941	6.6	.0091	.0032	.0022	49	33

*The percentage of insects collected in the year 1931 - 1941, based on the total number of insects collected in the year 1931 - 1941.

and the other is given in Table II.

[illegible]

It is also possible that the amount of growth-promoting substance in the growth-promoting medium is insufficient for constant stimulation of the terminal individual tubules. Therefore, stimulation, resulting in a reduction in the ash content of the samples over the three periods, may be due to the relatively small amount of growth-promoting substance which is obtained for the weekly average for the entire series. There is no correlation or connection throughout the series between the ash content of the tubules and corresponding ash content. With the exception of the first historical G. H. French, there is no correlation of the values for the years 1902, 1903, 1904 and 1905. In view of the values of the percentage ash content of the tubules and so far as is in comparison with the ash content for the remaining years, a correlation is not particularly significant. It is concluded, therefore, that the ash content of the samples does not depend on the amount of growth-promoting substance present. Therefore the second conclusion for is: (1) The stimulative substance in the medium, therefore, does not affect the ash, and (2) the amount of stimulating substance is too small to be quantitatively correlated with other properties, at least as estimated by this method.

TABLE 11. FACTORIES WITH MULTIPLE "MULTI" AND "MULTI" FACTORIES WITH MULTIPLE "MULTI" FACTORIES

Factory	1931	1932	1933	1934	1935	1936	1937	1938	1939	1940	1941
A* Multiple ash	X	10.8 .036	6.8 .013	1.2 .042	11.4 .025	4.7 .035	5.4 .035	7.8 .035	6.5 .036	5.0 .032	5.0 .032
B Multiple ash	X	7.1 .026	4.7 .014	1.6 .045	1.1 .022	4.2 .012	7.1 .017	6.6 .035	5.0 .032	10.0 .035	10.0 .035
C Multiple ash	X	12.8 .031	14.1 .031	7.8 .025	15.1 .030	11.0 .014	9.6 .017	14.4 .030	14.0 .017	9.4 .012	6.4 .012
D Multiple ash	11.1 .016	5.1 .016	4.1 .016	3.6 .065	5.3 .053	3.7 .016	1.2 .055	6.4 .037	5.0 .035	1.1 .033	5.0 .033
E Multiple ash	8.4 .013	4.6 .025	4.7 .025	3.0 .010	12.0 .010	6.1 .025	8.8 .015	3.4 .012	2.2 .035	6.7 .035	6.0 .035
F Multiple ash	1.4 .013	3.4 .013	4.4 .015	3.4 .040	3.1 .060	14.4 .037	1.2 .045	5.0 .037	3.1 .035	10.9 .027	4.2 .032
G* Multiple ash	X	10.0 .024	15.3 .037	1.4 .075	4.1 .010	4.1 .035	3.2 .040	7.6 .031	11.6 .037	12.5 .033	5.6 .037
H Multiple ash	1.4 .013	3.0 .013	5.7 .015	1.6 .037	3.3 .030	1.1 .013	1.6 .035	4.0 .030	14.2 .036	6.6 .032	5.4 .035
I Multiple ash	X	3.2 .014	1.2 .035	1.0 .040	1.2 .016	2.8 .040	9.4 .037	4.8 .030	4.4 .034	10.2 .034	7.2 .030
J Multiple ash	1.0 .013	5.4 .016	3.2 .035	X	X	7.1 .013	6.0 .035	5.0 .016	6.5 .035	10.5 .017	X
K Multiple ash	11.1 .017	1.8 .013	6.1 .017	3.1 .025	13.3 .012	1.2 .025	1.1 .010	7.6 .010	7.3 .016	X	X
L Multiple ash	1.0 .010	2.1 .017	5.0 .017	1.1 .013	2.5 .035	1.1 .011	6.6 .037	4.2 .035	17.4 .034	1.5 .032	3.6 .032
M* Multiple ash	6.0 .024	1.1 .014	6.2 .014	4.3 .014	1.3 .013	3.0 .013	1.0 .016	7.6 .013	7.1 .013	6.0 .013	6.6 .013

*-5, tuffan factories; *80-1, non-tuffan factories; 1, 10, 20 - the received in 1931, 1932, 1933, 1934, 1935, 1936, 1937, 1938, 1939, 1940, 1941.

Results of these tests obtained from individual consecutive measurements of the factorials in which the standard deviation of the amount of variation within the series is consecutive studies. In this way it is possible to obtain information regarding the uniformity of operation of the factorial in the detection of a change of the level of deviation for the series from the consecutive factorials. Fifty-six samples were collected for each study. The exact incubation periods were determined for the individual samples and from these results the average (mean) of the deviation values were calculated. The results are given in Table 12.

The average values of individual multiple values for the series from Factorials 1, 2 and 3 are 2.9, 5.3 and 4.7, respectively. Corresponding to these values are the mean deviation values of 1.4, 1.8 and 1.3, respectively. Grouping these values in a numerical index of the uniformity of operation, in these values zero for a perfect result, if every 2 is given to have the least operation. The operation of Factorial 2 is the most uniform, since it has the least deviation.

STUDY OF THE EFFECT OF SUGAR ON CHYTRIDIA IN THE MOUTH

When it is observed that most sugars do not support a multiplication of yeast cells in 10 percent solutions, while others stimulate extensive multiplication, attention was directed to a study of the nature and properties of the stimulants. A number of sugars, some of which promote yeast growth, on the other hand, were extracted with alcohol in a manner similar to that described by Levenstein (1930), (6) who also observes high activity in Yeast's (highest purity) medium. The test sugars were dissolved in hot, distilled water until saturated solutions resulted. After the solutions had cooled,

TABLE 12. RELATIVE AMOUNT OF GROWTH-PRODUCING SUBSTANCES IN COMBINATION WITH SUGAR

FACTORY					
A		B		C	
Strike No.	Yeast Inoculum Multiple	Strike No.	Yeast Inoculum Multiple	Strike No.	Yeast Inoculum Multiple
221	8.3	394	5.8	184	4.2
222	6.4	395	6.2	185	3.8
223	8.0	396	6.9	186	4.3
224	10.1	397	5.8	187	5.3
225	8.1	398	3.2	188	5.1
229	12.9	399	5.6	189	4.3
241	11.6	400	3.3	190	5.0
242	9.3	401	5.6	191	4.4
244	15.1	402	6.4	192	5.4
245	15.9	403	3.2	193	5.4
246	11.0	404	4.1	194	5.8
247	6.0	405	4.9	195	5.2
248	5.6	406	4.8	196	5.8
249	9.3	407	8.0	197	4.7
250	11.5	408	4.1	198	4.8
		409	7.1	199	3.6
		410	6.0	200	3.4
		411	5.6	201	4.4
		412	6.3	202	4.9
		413	5.1	203	4.3
		414	7.1	204	4.9
		415	6.4	205	4.6
		416	7.3	206	5.0
		417	5.8	207	4.4
		418	7.5	208	5.4
Average (Mean) 9.9		Average (Mean) 5.8		Average (Mean) 4.7	
Mean Deviation 2.4		Mean Deviation 1.0		Mean Deviation 0.53	

the sugar the solids were removed, and to them 95 percent ethyl alcohol was added to result in an alcoholic concentration of 80 percent. The sugar crystallized out from these solutions and was usually complete within 96 hours. The alcoholic liquor was then removed from the crystallized sugar by decantation and reserved for future studies. The sugar was dried, powdered, and bottled as alcohol-purified sugars. These were then compared with those of the unpurified sugars in 80 percent solutions. The results given in Table 12 show the influence of alcohol purification in decreasing the amount of stimulants.

A substantial reduction was obtained in the amount of stimulants in all media which originally stimulated the growth of yeast. These reductions are similar to those obtained by Fink and Freedman (13, 14), who showed that the size of the yeast crop in their synthetic medium was reduced from 6.1 mm. to 2.5 mm. when ordinary sugar was replaced with sugar recrystallized from alcohol. These workers believed that the growth-promoting substance was vitamin B, and that without its presence, the strain of yeast used in their studies was unable to synthesize vitamin B and was therefore unable to grow.

Palmer and Nelson (11) and Barnes and Silve (20) did not obtain reductions in the size of yeast crops when alcohol-purified sugar was substituted for unpurified sugar in synthetic media. It is apparent that they were using a higher grade of sugar, such as A and B, Table 12, for their experiments than that used by Fink and Freedman.

FRACTIONATION OF ALCOHOLIC EXTRACT

The solubility of the growth-promoting substance in 80-percent alcohol, its absence in alcohol, resistance to alkaline treatment

TABLE 13. PURIFICATION OF YEAST BY CRYSTALLIZATION FROM 80-PERCENT ALCOHOL

<u>Sample</u>	<u>Yeast Inoculum</u> <u>Unpurified</u> <u>Sugar</u>	<u>Culture</u> <u>Purified</u> <u>Sugar</u>	<u>Sample</u>	<u>Yeast Inoculum</u> <u>Unpurified</u> <u>Sugar</u>	<u>Culture</u> <u>Purified</u> <u>Sugar</u>
A	1.0	1.0	G	27.0	9.5
B	1.0	1.0	H	31.0	10.0
C	3.0	1.0	I	34.5	1.0
D	11.3	6.6	J	44.0	3.0
E	22.6	3.0	K	59.0	12.0
F	22.0	9.6	L	37.3	1.0

(during the manufacturing process of sugar) suggested that it might be bios. The alcoholic extract which remained after crystallization of the sugar was purified according to the method described by Lucas (15) and the two fractions of fraction from salt combinations. Two fractions were thus prepared and the response of the test yeast to the fractions was observed to determine if the stimulant possessed bios properties similar to those in previously studied plant extracts. Two fractions designated as Bios I and Bios II were thus prepared from the alcoholic liquors in the following manner:

Fractionation of Bios

The alcoholic liquor from 1,100 grams of crystallized sugar is distilled in vacuo to remove the alcohol. The volume of the remaining concentrate is 150 ml. of a sirupy, straw-colored liquid. To it a concentrate is added 200 ml. of a hot, saturated aqueous solution of barium hydroxide $\text{Ba}(\text{OH})_2 \cdot 8\text{H}_2\text{O}$ to effect complete precipitation of uncrystallized sugar remaining in the concentrate. The quantity of barium hydroxide required is tested at a minimum by a previous determination on an aliquot sample. After precipitation is forced which is not removed. Alcohol, 95 percent, is then added in volume equal to twice the sum (350 ml.) of the volume of the concentrate and the barium hydroxide solution; at first a heavy, sticky precipitate forms which adheres to the walls of the distiller. As to the stirring rod, but which later becomes loose and granular. A few drops of concentrated barium hydroxide are added to a small quantity of the clear liquid to determine if a adequate amount of barium hydroxide has been added. The precipitate, after it is filtered off, is washed with dilute alcohol (2 volumes of 95 percent alcohol to one volume of water) and the filtrate, together with the washings, is poured out for Bios II as described later. (Smaller quantities of sugar may be crystallized from alcohol, thus requiring proportionally smaller quantities of reagents).

Bios I

The barium precipitate, after washing, is stirred with three or four successive portions of water at room temperature, filtering between each. The undissolved precipitate which contains no bios is rejected. Bios I liquor (Lucas) is prepared from the combined filtrates by saturation with carbon dioxide, and then boiled in vacuo to remove the excess of the gas, after which it is filtered. The precipitate con-

precipitation. To the filtrate, heated to about 65° C., is added 2.5 N sulfuric acid until no further precipitate forms. There is to be an excess of acid to remove the excess of acid in the solution, although no oil is obtained. The precipitate, which contains no bios, is rejected and the filtrate is retained as "Verde Bios I solution", (Lugar). For the purpose of this study this fraction is considered as Bios I. The volume is about 10 percent of the original concentrate. (Note: In the preparation of Bios I from left coatings, but a small amount of purification of the "Verde Bios I solution" with less acetone in order to eliminate glucose. This step is not necessary when the fractions containing multiple fractions obtained from crystalline sucrose).

Bios II

The Bios II liquor is obtained by removing the excess oil, which is boiled off in a cup. The precipitate is filtered off and discarded and contains none of the Bios II fraction. The filtrate is evaporated in vacuum at the lowest possible temperature until about 10 percent of the original volume remains. At this concentration, large bubbles form which nearly fill the container. The viscous liquid is treated with about 1,500 ml. of acetone or sufficient to effect complete precipitation. The acetone solution is decanted off, the precipitate washed with acetone and the washings added to the solution. The solution is concentrated in vacuum until a heavy viscous residue which is taken up in water. This solution, which contains Bios II, is made up in volume to that of the Bios I.

A number of alcoholic extracts from purified sugar were fractionated by this method and were added back in 10 percent solutions of the purified sugars from which they were obtained. The amounts of individual fractions were used which showed maximum stimulation of the yeast inoculum. These were 10 ml. and 5 ml. of Bios I and Bios II solutions, respectively. The results obtained from four typical samples showing initial yeast inoculum multipliers from 1.4 to 57.4 are given in Table 14.

Crystallization of samples 1 and 2 from 50 percent alcohol resulted in complete removal of the stimulant. Since only a slight amount of stimulant was present in the original sugar, very little

TABLE 16. - COMPARISON OF THE EFFECTS OF DIFFERENT TYPES OF PESTICIDES ON THE MORTALITY OF PESTS IN THE FIELD

NOTE: Figures are percentages of mortality.

Time of day	1	2	3	4
Unsprayed sugar	1.5	1.4	41.8	57.4
Unsprayed sugar	1.3	.8	3.1	5.1
Unsprayed sugar (10 ml.)	1.1	1.1	7.1	17.1
Unsprayed sugar plus (3 ml.)	1.0	1.7	12.6	31.0
Unsprayed sugar plus (3 ml.)	1.1	1.7	13.3	47.5
Unsprayed sugar plus (3 ml.)	50.750	46.803	21.050	50.750

effect was expected from the various disc fractions. There was a similarity to the effect of yeast cells introduced into the host solution after 72 hours incubation. The results obtained with the yeast at residues C and B show clearly the effect of the inoculum on sugar containing larger amounts of the host solution.

Optimally, then, of residues C and B collected 94.3 and 89.7 per-cent ethanol, was actively, or the yeast inoculum utilized. The stimulation of the inoculum by the disc fraction alone was relatively small, but was from 1.5 times up to 2.5 times greater with disc II solution alone. When combined in the inoculated medium, a still greater increase in the yeast inoculum utilization was obtained. The results are shown graphically in Figure 1, in which the scale of yeast inoculum utilization values for residues C and B is reduced 15 times that for residues A and B. Although the utilization values are lower than those obtained with the uninoculated sugar, the trend is the same in the inoculated as in the uninoculated sugar for yeast utilization. It is, therefore, concluded that the sugar shows high activity similar to that of other plant extracts. The results also indicate that there was no stimulatory substance present in sugar. Further work is being done to identify the nature of the stimulatory substance in 1973 in view of receiving the results of the analysis of sugar free from yeast growth-promoting substances.

IDENTIFICATION OF THE STIMULANT IN INTERMEDIATE PRODUCTS FROM THE SUGAR

The reason for the failure of some of the sugar growth-promoting activity, while others showed widespread variations, was investigated from the standpoint of the manufacturing process in the

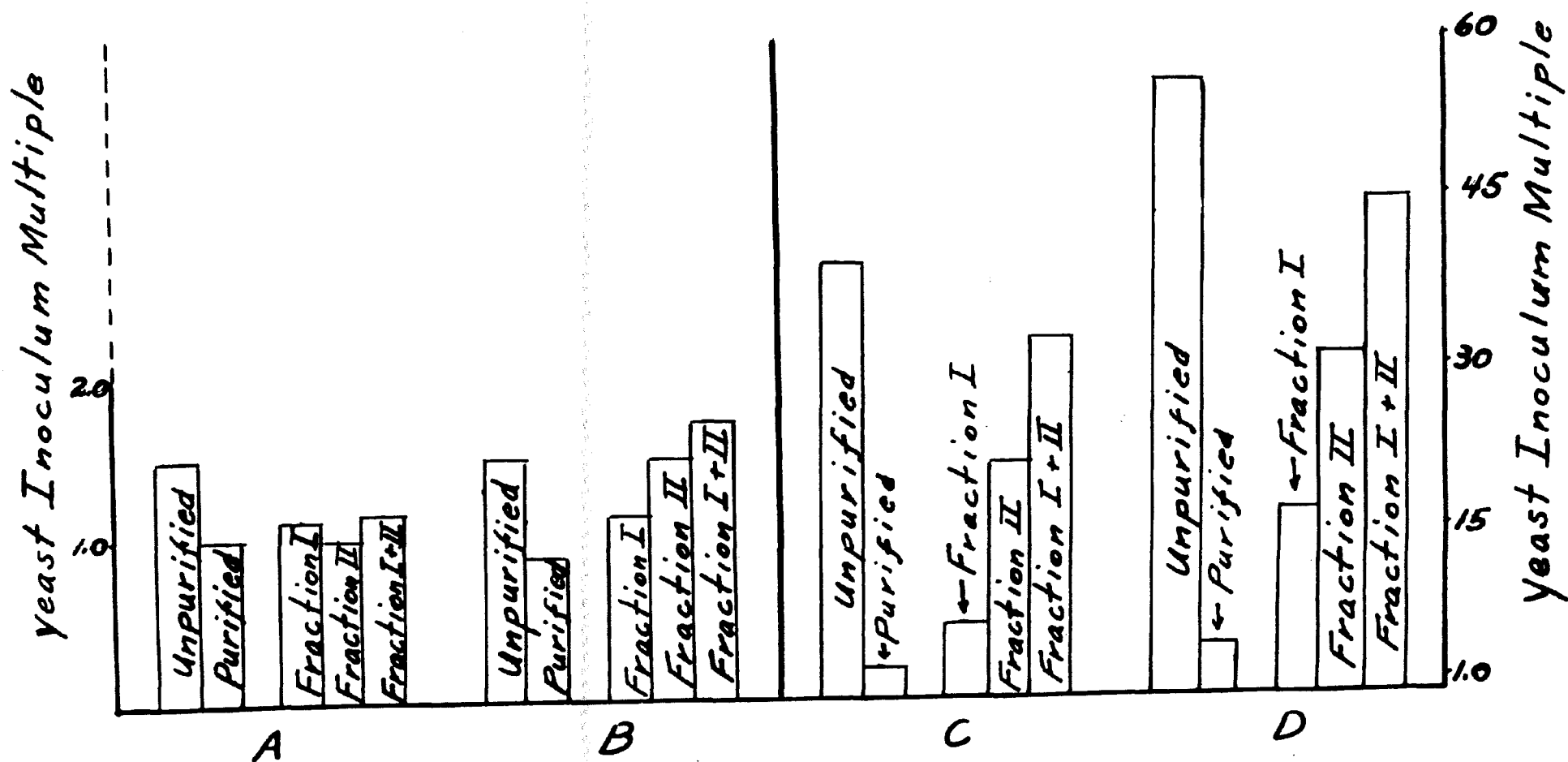


Figure 2. Purification of Sugars by Crystallization From Alcohol and Addition of Purified Fractions to Solutions of Purified Sugar.

hope that some of the food for cell in the cell stimulant from all the different factories. However, the cell stimulant, which is a mixture of the cell stimulant and the cell stimulant, is a mixture of the cell stimulant and the cell stimulant. The extent of yeast growth obtained by the cell stimulant is different for different factories and it seems to vary. The stimulant substances are believed to have their origin in the sugar and are not destroyed during the sugar recovery process.

Since the most extensive multiplication of the yeast occurred in the white-sugar process, which is the most concentrated form of sugar and which obtained in the process, it is evident that there is a substantial concentration of the stimulant in the product. The extent of the relative amount of growth-promoting substances in white sugar versus sucrose is extended to include a composite number of each from 14 different factories and 15 non-different factories. A composite number is used in 10 percent solutions of sugar and the yeast inoculum multiples determined. The results obtained are given in Table 15.

There is no uniform relationship between the relative amount of stimulant in the sugar and sucrose. An example of the discrepancy which exists is illustrated between samples A and B from the Steffen factories. The sugar from factory A has a multiple value of 10.0 as compared with 3.3 for that from factory B, but its respective sucrose value of 130.0 as compared with 24.0 from factory A. Similar relationships also exist between the samples obtained from the non-Steffen factories. The failure to eliminate the stimulant from the

TABLE 15. COMPARISON OF ALKYLATION OF STEROL-PRODUCING STRAINS IN VITRO WITH STRAINS PRODUCING STEROLS AND NON-PRODUCING STRAINS

STRAIN	STEROL-PRODUCING			NON-PRODUCING	
	Alkylated Sterol	Multiple		Alkylated Sterol	Multiple
A	10.0	10.0	:	A	10.0
B	104.0	11.6	:	B	171.0
C	156.0	1.0	:	C	301.0
D	162.0	0.8	:	D	316.0
E	180.0	6.6	:	E	354.0
F	184.0	4.7	:	F	356.0
G	224.0	16.6	:	G	360.0
H	230.0	3.8	:	H	401.0
I	231.0	6.3	:	I	455.0
J	260.0	0.0	:	J	457.0
K	250.0	2.6	:	K	506.0
L	282.0	0.0	:	L	660.0
M	378.0	5.0	:	M	842.0
N	414.0	11.2	:		

ADDITIONAL INFORMATION IS CONTAINED
ON THE FOLLOWING PAGES

little video exists between the two.

Substantial reductions were obtained in all of the test samples. The reduction was 50.0 percent or greater for 12 of the 14 samples examined, and in general, the greater the amount of stimulant present

Substantial reductions were obtained in all of the test samples. The reduction was 61.0 percent on average for 12 of the 14 samples examined, and in general, the greater the amount of stimulant present,

Table 16. Determination of the Effect of Heat Treatment on the Rate of Growth of Bacteria on Media Containing 10% Crystalline Cellulose

Sample	Heat Treatment		Percent Reduction
	Untreated	Treated	
1	11.6	2.0	84.1
2	12.4	2.4	80.6
3	10.0	1.0	89.0
4	6.4	1.6	75.0
5	6.0	2.2	63.3
6	4.0	2.2	60.7
7	3.6	1.0	71.2
8	3.4	1.0	70.6
9	2.2	1.0	54.5
10	2.0	1.0	50.0
11	1.8	1.0	44.4
12	1.4	1.0	28.6

4

[illegible]

Attending the elimination of the mother liquor and of clinging small-grain sugar with a minimum amount of water in the centrifugals is related not so much to that of coarse grain sugar. In these observations it is evident that in order to manufacture a "better" grade sugar, the film of mother liquor surrounding each crystal is possible. Small part of each crystal itself can't be eliminated. Also, it is less efficient to be experienced in eliminating these portions of film of crystals of uniform size than from smaller crystals.

INFLUENCE OF QUANTITY OF CENTRIFUGAL WASH WATER ON GRAIN-PRESSING SUBSTANCES

In order to determine if greater elimination of non-sugar substances could be obtained from crystal surfaces varying centrifugation or increased quantities of centrifugal wash water, the quantity of water was changed in successive baskets of centrifuge while draining a strike of white sugar. One machine was selected in each of four factories, and for four successive baskets the quantity of water normally was increased or decreased by two thirds. A sample of finished sugar was collected from the machine after each change while the sugar was being loaded out so that the sample was representative of the entire basket. The yeast inoculum multipliers were determined for each sample of sugar. The results, together with the quantity and temperature of water used, are given in Table 17.

There was a decrease in the yeast inoculum multiplier values for each sample (except one sample from factory C) when the quantity of wash water was increased by two thirds during the successive baskets of sugar. Conversely, when the normal quantity of water was decreased by two thirds at least, the yeast inoculum multiplier values

TABLE 17. INFLUENCE OF AMOUNT OF CO_2 TO YEASTING. INFLUENCE OF QUANTITY OF FERMENTED U.S. WATER

<u>Factory A</u>		<u>Factory B</u>	
<u>Quarts Water</u> <u>at 38° C</u>	<u>Yeast Inoculum</u> <u>Multiple</u>	<u>Quarts Water</u> <u>at 72° C</u>	<u>Yeast Inoculum</u> <u>Multiple</u>
14*	22.7	16*	13.4
16	20.5	18	9.6
18	19.4	20	7.4
20	14.8	22	3.6

<u>Factory C</u>		<u>Factory D</u>	
<u>Quarts Water</u> <u>at 36° C</u>	<u>Yeast Inoculum</u> <u>Multiple</u>	<u>Quarts Water</u> <u>at 36° C</u>	<u>Yeast Inoculum</u> <u>Multiple</u>
15*	3.8	24*	3.8
17	6.8	22	3.4
19	6.4	20	6.0
21	4.8	18	7.8

*quarts normally used.

(except one change) increased. That results of the change in the growth-promoting substance is located at or near the surface of the crystal is indicated by the ability of it to be eliminated by increased evaporation of the water. The effect of the size of the crystal itself on the behavior of the substance were not studied. Since the effect was not seen, however, many factories have installed one, and some countries are still to be sure (55° C. to 60° C. or higher) and the water to be 100° C. or higher. Both improvements are reported to reduce in the elimination of ice crystal surface, non-ionic impurities. The other is to be used to include you a growth-promoting substance, so it may be presumed that similar improvement in elimination likewise resulted.

It is suggested that for the manufacture of "beverage" grade sugar, coarse grain be utilized because of the nature of the surface which the crystal surface, and, therefore, a large percentage of surface irregularities, may be removed. Although it is felt that the sugar produced by the present manufacturing methods is generally satisfactory for use in beverage and food products, an extra effort should be made to insure the elimination of the most objectionable amount of non-sugar irregularities. It is recognized that their presence in sufficient quantities in any sugar can satisfy the description of "beverage" of contaminating agents and permit the development of a sufficiently large to produce turbidity or sediment in the solution.

V. 12742

Early in 1943, shortly before completion of this financial year, number of boats in service was 10. There were 1000 tons of cloth, un-

Walter R. Rife and Mrs. D. L. Rife and Father Robert A. Rife, Past and
 Present Minister, Federal Security Administration, Washington, D.C.

are removed from the recrystallization process.

5. They are also used in the organic phase and are largely removed by washing with water or by extracting the organic phase with water, centrifugation.

6. One of the stimulative components is identified as blatin.

VITAMIN B12 - COBALT

- (1) Abelson, P. H., and Tamm, L.
1919. Über die Vitaminwirkung einer die Bakterien von Germany
bestimmenden, an Vitamin B12 reicheren Substanz aus
Hefe auf andere Bakterien. *Mittg. Arch. Ges. Bakt.*
176: 209.
- (2) Allen, F. E., Clever, A. S., and Burk, L.
1933. A new vitamin. *Science*. 75: 317.
- (3) Allen, F. E., and Tabor, F. W.
1935. Cobaltous chloride as a cobalt source. *Can. J. Chem.* 13: 473.
- (4) Bann, J.
1937. Le "Vitamin" de l'Helium ne joue pas le rôle d'un facteur de
la cellulose. 23: 125.
- (5) Bann, J.
1919. Vitamin requirements of certain yeasts. *J. Biol. Chem.*
22: 435.
- (6) Birch, E. F., and Gray, G.
1937. Physiological properties of the factor (Vitamin B12) Curative
of egg yolk injury. *J. Biol. Chem.* 131: 761.
- (7) Christen, V. J., and Williams, E. J.
1941. The Vitamin Content of Yeast. The University of Texas
Publication. Number 4137, October 1.
- (8) Cline, J. E., Williams, E. J., Ingle, A. S., and Tabor, F. E.
1937. Studies of Crystalline Vitamin B12. XVI. Identification of
the Pyrimidine Portion. *J. Am. Chem. Soc.* 59: 520.
- (9) Clever, F. E., and Tabor, F. W.
1937. Observations on the Growth of Yeasts in Pure Solutions.
J. Biol. Chem. 14: 317.
- (10) de Vigneau, V.
1941. The Structure of Biotin. *Science*. 96: 455.
- (11) de Vigneau, V., Melville, B. S., Gray, G., and Rose, C. J.
1940. On the Identity of Vitamin B12 and Biotin. *Science* 91: 61.
- (12) Eakin, E. F., and Williams, E. J.
1939. Vitamin B12 in Yeast Culture. *J. Am. Chem. Soc.* 61: 1931.
- (13) Hartcott, Edna. V.
1913. The Isolation and Identification of Biotin. *J. Phys. Chem.*
32: 1094.

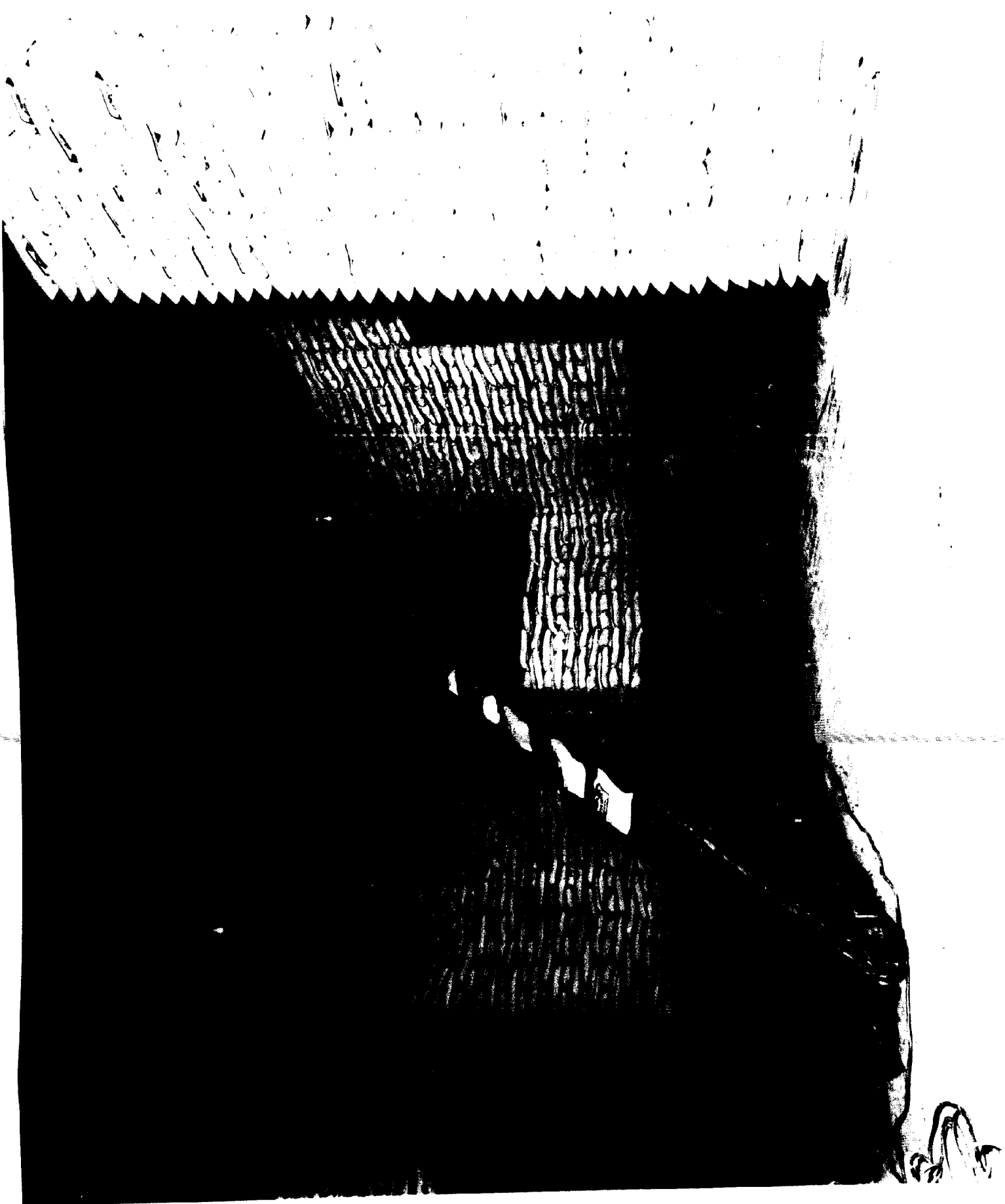
- (12) Fernsten, C.
1931. Die Entfaltung der Hefe in einem mineralischen Medium.
Ann. d. Pflanzenern. u. Bodenk. 11: 217.
- (13) Fisher, H. E., and Johnson, H. E.
1923. Studies on Yeast. VI. Growth Continuous Growth of
Saccharomyces cerevisiae in Synthetic Medium. J. Biol.
Chem. 22: 150.
- (14) Funk, C., and Johnson, H. E.
1930. Yeast for Utilitarian-Born Vitamins for its Physiological
Application. J. Biol. Chem. 44: 437.
- (15) Funk, C., and Freedman, E.
1933. On Yeast Growth in a Synthetic Medium. J. Biol. Chem. 100: 311.
- (16) Funk, C., and Freedman, E.
1933. The Presence of a Yeast Growth-Inhibiting Vitamin in
Jama Sugar. J. Biol. Chem. 56: 661.
- (17) Gordon, R. A.
1930. Outline of Biochemistry. John Wiley and Sons, Inc. N. Y.
Second Edition.
- (18) Gregg, P.
1935. Vitamin B_2 Complex. I. Differentiation of Riboflavin
and the Anti-chilgry Factor. Biochem. J. 29: 741.
- (19) Gregg, P.
1937. Attempts to Isolate the Anti-egg White Injury Factor
(Vitamin B). J. Biol. Chem. 119: 53.
- (20) Gregg, P.
1939. The Curative Factor (Vitamin B) For Egg White Injury,
with Particular Reference to its Presence in Different
Foodstuffs and in Yeast. J. Biol. Chem. 131: 753.
- (21) Gregg, P., Allen, R., and Lecherer, F.
1939. Attempts to Isolate the Factor (Vitamin B) Curative of
Egg White Injury. J. Biol. Chem. 131: 745.
- (22) Gregg, P., Melville, D. B., Burr, L., and Vignaud, V.
1940. The Possible Identity of Vitamin B with Biotin and
Coenzyme R. Science. 91: 143.
- (23) Gregg, P., Rose, C. E., Johnson, H., Melville, D. B. and du
Vigneaud, V.
1940. A Further Note on the Identity of Vitamin B with Biotin
and Coenzyme R. Science. 92: 609.
- (24) Hall, H. A., and Jones, G. H.
1936. Crystal Surface Contaminants and Biological Activity of
Sugar. Food and Sugar. 31: 221.

- (7) Hill, L. A., Jones, J. A., and Short, L. A.
1933. Plant Growth Substances in Soda Sugar. Ind. Eng. Chem.
25: 1938.
- (8) Jones, L. A., and Hill, L. A.
1934. The Isolation of Vitamin B₁ from Soda. Biochem. J. 28: 430.
- (9) Harris, E. V., and Johnson, E.
1933. Growth of Plants. J. Am. Chem. Soc. 55: 1413.
- (10) Hill, L. A.
1933. The Isolation of Vitamin B₁ from Soda. Biochem. J. 28: 430.
- (11) Hill, L. A., and Johnson, E. F.
1935. The Isolation of Vitamin B₁ from Soda. Biochem. J. 29: 131.
- (12) Hill, L. A.
1935. The Isolation of Vitamin B₁ from Soda. Biochem. J. 29: 131.
- (13) Hill, L. A., and Johnson, E. F.
1935. The Isolation of Vitamin B₁ from Soda. Biochem. J. 29: 131.
- (14) Hill, L. A., and Johnson, E. F.
1935. The Isolation of Vitamin B₁ from Soda. Biochem. J. 29: 131.
- (15) Hill, L. A., and Johnson, E. F.
1935. The Isolation of Vitamin B₁ from Soda. Biochem. J. 29: 131.
- (16) Hill, L. A., and Johnson, E. F.
1935. The Isolation of Vitamin B₁ from Soda. Biochem. J. 29: 131.
- (17) Hill, L. A., and Johnson, E. F.
1935. The Isolation of Vitamin B₁ from Soda. Biochem. J. 29: 131.
- (18) Hill, L. A., and Johnson, E. F.
1935. The Isolation of Vitamin B₁ from Soda. Biochem. J. 29: 131.
- (19) Hill, L. A., and Johnson, E. F.
1935. The Isolation of Vitamin B₁ from Soda. Biochem. J. 29: 131.
- (20) Hill, L. A., and Johnson, E. F.
1935. The Isolation of Vitamin B₁ from Soda. Biochem. J. 29: 131.

- (41) Williams, R. L., and Williams, R. L.
1947. Nitrilite as a substitute for Vitamin B₁₂. *J. Biol. Chem.* 172: 177.
- (42) Williams, R. L., and Williams, R. L.
1947. The identification of the anti-leukemic activity of Vitamin B₁₂. *J. Biol. Chem.* 172: 177.
- (43) Williams, R. L., and Williams, R. L.
1947. "Nitrilite" as a substitute for Vitamin B₁₂. *J. Biol. Chem.* 172: 177.
- (44) Williams, R. L., and Williams, R. L., and Williams, R. L.
1947. The identification of the anti-leukemic activity of Vitamin B₁₂. *J. Biol. Chem.* 172: 177.
- (45) Williams, R. L., and Williams, R. L., and Williams, R. L.
1947. The identification of the anti-leukemic activity of Vitamin B₁₂. *J. Biol. Chem.* 172: 177.
- (46) Williams, R. L., and Williams, R. L., and Williams, R. L.
1947. The identification of the anti-leukemic activity of Vitamin B₁₂. *J. Biol. Chem.* 172: 177.
- (47) Williams, R. L.
1947. The identification of the anti-leukemic activity of Vitamin B₁₂. *J. Biol. Chem.* 172: 177.
- (48) Williams, R. L., and Williams, R. L., and Williams, R. L.
1947. The identification of the anti-leukemic activity of Vitamin B₁₂. *J. Biol. Chem.* 172: 177.
- (49) Williams, R. L.
1947. The identification of the anti-leukemic activity of Vitamin B₁₂. *J. Biol. Chem.* 172: 177.
- (50) Williams, R. L.
1947. The identification of the anti-leukemic activity of Vitamin B₁₂. *J. Biol. Chem.* 172: 177.
- (51) Williams, R. L., and Williams, R. L., and Williams, R. L.
1947. The identification of the anti-leukemic activity of Vitamin B₁₂. *J. Biol. Chem.* 172: 177.
- (52) Williams, R. L., and Williams, R. L., and Williams, R. L.
1947. The identification of the anti-leukemic activity of Vitamin B₁₂. *J. Biol. Chem.* 172: 177.
- (53) Williams, R. L., and Williams, R. L., and Williams, R. L.
1947. The identification of the anti-leukemic activity of Vitamin B₁₂. *J. Biol. Chem.* 172: 177.
- (54) Williams, R. L., and Williams, R. L., and Williams, R. L.
1947. The identification of the anti-leukemic activity of Vitamin B₁₂. *J. Biol. Chem.* 172: 177.
- (55) Williams, R. L., and Williams, R. L., and Williams, R. L.
1947. The identification of the anti-leukemic activity of Vitamin B₁₂. *J. Biol. Chem.* 172: 177.

- (56) Tamm, F. J.
1915. The Vitamin Question. *Chem. Rev.* 1: 397.
- (57) Tamm, F. J., and Tamm, F. J.
1939. The Relation of "Cysteine" to Vitamin Science. 29: 687.
- (58) Williams, E.
1921. Une nouvelle substance indispensable au développement de la levure. *La Cellule*. 18: 813.
- (59) Williams, E. J.
1919. The Vitamin Requirement of Yeast. *J. Biol. Chem.* 38: 465.
- (60) Williams, E. J.
1935. An Artificial Medium for Growth of Yeast in Biochemical Research. *J. Biol. Chem.* 52: 157.
- (61) Williams, E. J.
1941. Growth-Requiring Nutrients for Yeasts. *Biol. Rev.* 16: 49.
- (62) Williams, E. J., and Tamm, F. J.
1931. Further Fractionation of Yeast Nutrients and Their Relation to Vitamin B₁. *Biochem. J.* 25: 703.
- (63) Williams, E. J., Tamm, F. J., and Tamm, F. J.
1942. The Relationship of Inositol, Thiamin, Biotin, Pantothenic Acid, and Vitamin B₆ to the Growth of Yeast. *J. Am. Chem. Soc.* 64: 1204.
- (64) Williams, E. J., and Bach, L. P.
1930. The Effects of Antineuritic Vitamin Preparations on the Growth of Yeasts. *J. Biol. Chem.* 87: 591.
- (65) Williams, E. J., and Bach, L. P.
1934. The Effects of Inositol, Crystalline Vitamin B₁ and Pantothenic Acid on the Growth of Different Strains of Yeast. *Biochem. J.* 22: 1647.
- (66) Williams, E. J., and Truesdale, J. H.
1932. The Use of Fractional Electrolysis in the Fractionation of the Vits of Yeasts. *J. Am. Chem. Soc.* 54: 4171.
- (67) Williams, E. J.
1935. Structure of Vitamin B₁. *J. Am. Chem. Soc.* 57: 179.
- (68) Williams, E. J.
1936. Structure of Vitamin B₂. *J. Am. Chem. Soc.* 58: 1063.
- (69) Williams, E. J., Tamm, F. J., and Tamm, F. J.
1934. Larger Yields of Crystalline Anti-Neuritic Vitamin. *J. Am. Chem. Soc.* 56: 1157.

- (7) in Isen, E.
1946. Herr "Der Herr von Isen" spielt nicht die Rolle eines
"Gegenspieler". Ein solch. General. 17. 5.7.



VITA

Winton Marshall Hall, the only son of Fred Marshall and George Bushnell Hall, was born October 1, 1904, at East Leroy, Michigan. He received his education at the University of East Leroy (1918), and the high school education at Ft. Cass, Michigan (1921). Five years later he completed his undergraduate work at the Michigan State College, earning the Bachelor of Science degree in the School of Applied Science. During the following two years he worked as a Junior Chemist and Junior Plant Pathologist, in the laboratories of the Michigan Department of Agriculture, Lansing, Michigan. On August 18, 1930 he became employed by the present Agricultural Research Administration, Bureau of Agricultural and Industrial Chemistry, United States Department of Agriculture, Washington, D. C., as a Junior bacteriologist, where he has since received promotion in bacteriology on sugar, sirups, molasses, starch, silage, and egg products. Promotions were granted to Associate bacteriologist (1936) and Senior (1941) to bacteriologist, the position being held at present. As the result of part time studies at the Michigan State College, the University of Maryland, College Park, Maryland, and the United States Department of Agriculture Graduate School, Washington, D. C., the degree of Doctor of Science was obtained at the latter school in 1936. Fraternity and society memberships are held as follows: Sigma Xi, Society of American Bacteriologists, American Public Health Association and the Institute of Food Technologists.

LIST OF PUBLICATIONS

1. Yeast Growth-Stimulants in White Sugar.
Hall, E. H., J. H. J. A., and Stuart, L. F.
Ind. Eng. Chem. 25: 1931, 1933.
2. Yeasts Found in Fermented White Sugar.
Hall, E. H., and J. H. J. A.
Bacter. Rev. 11: 31, 1933.
3. The Yeast Sterilizer Honey in Culture Media.
Hall, E. H., and J. H. J. A.
B. Act. 27: 240, 1934.
4. Yeast Growth-Stimulants in White Sugar.
Hall, E. H., and J. H. J. A.
Bacter. Rev. 31: 275, 1936.
5. Yeast Growth-Stimulants in White Sugar.
Hall, E. H., and J. H. J. A.
B. Act. 32: 377, 1937.
6. Effect on Yeast Growth-Stimulants of Anticommunicable
Anticommunicable, Isolated from Yeast, and Anticommunicable
Anticommunicable in the 1937-38 crop.
Hall, E. H., and J. H. J. A.
Proc. of 6th. Conference of Sugar Cane Soc. 1938, 1939.
7. Survival of Thermophilic Bacteria Spores in
Fermented White Sugar.
Hall, E. H.
Ind. Eng. Chem. 4: 400, 1932.
8. Effect of Radiant Energy on Thermophilic Bacteria
Spores in Fermented White Sugar.
Hall, E. H., and J. H. J. A.
Ind. Eng. Chem. 31: 1165, 1939.
9. Studies on the Efficiency of Control in
Control of Microorganisms in Sterile.
Robinson, J. H., and Hall, E. H.
Food Research. 6: 621, 1941.
10. Storage Changes in Air-Dried White Sugar Powder.
Stuart, L. F., Hall, E. H., and Hicks, E. H.
The U. S. Egg and Poultry Magazine. 43: 629, 1944.
11. Notes on the Microbiological Separation and Preservation
of Glycerine Egg Yolk.
McFarland, V. H., and Hall, E. H.
The U. S. Egg and Poultry Magazine. In Press.

11. Esterification of White Sugar.

I. Characteristics of Yeasts found in Esterified Sugar.

II. Case Studies on Esterified Sugar.

III. Summary.

Master of Science Thesis, University of Michigan, 1936.