WITTIG REARRANGEMENTS OF SILYL ALLYLIC CYCLIC ETHERS AND [1,2] CARBON TO CARBON SILYL MIGRATION OF $\alpha$-HYDROXY ALLYL SILANES

## By

Emmanuel Wesonga Maloba

## A DISSERTATION

Submitted to
Michigan State University in partial fulfillment of the requirements for the degree of Chemistry - Doctor of Philosophy


#### Abstract

The [1,2]- and [1,4]-Wittig rearrangements of cyclic $\alpha / \gamma$-silyl allyl ethers have been studied. These rearrangements are based on the ability of the silicon atom to stabilize an $\alpha$-anion. Treatment of diastereomeric 2-silyl-6-aryl-5,6-dihydro-2H-pyrans with $n$-butyllithium (sec-butyllithium for cis diastereomer) result in stereoconvergent ring contraction to corresponding $\alpha$-silylcyclopentenols and/or ( $\beta$-cyclopropyl)acylsilanes via [1,2]- and [1,4]-Wittig rearrangements respectively. Studies on expanded ring size, 2-silyl-7-aryl-2,5,6,7-tetrahydrooxepins, led to the formation of $\alpha$ silylcyclohexenols and ( $\beta$-cyclobutyl)acylsilanes. On the other hand, moving the silyl group to the 4-position of the dihydropyran moiety led predominantly to [1,4]-Wittig rearrangement.

Concurrently, 2-silyl-3-hydroxy-5-arylcyclopentan-1-one and 2-silyl-3-hydrox-6-arylcycloyhexan-1-one can be accessed by subjecting 1-silyl-5-arylcyclopent-2-en-1-ol and 1-silyl-6-arylcyclohex-2-en-1-ol, respectively, to $m$-CPBA and $\mathrm{NaHCO}_{3}$. In this transformation epoxidation triggers a [1,2]-carbon-to-carbon silyl migration. The stereochemical orientation of the resulting product is such that the $\alpha$-silyl and $\beta$-hydroxy groups are trans to one another. Furthermore, the hydroxy group in the starting silane directs the epoxide formation, which in turn dictates the stereochemical outcome of the resulting $\beta$-hydroxy making this reaction stereospecific in nature. Lastly, the synthesis of 2-alkyl-2-silylalkanals can be achieved by conditions that effect spontaneous [1,2]-carbon-to-carbon silyl migration in 1-silyl-2-alkyl-2-alken-1-ols. These conditions are independent of the substituents on silicon but require an alkyl substituent at the olefin position next to the $\mathrm{sp}^{3}$ carbon bearing the silyl group.


To my family: My dad Syverio; my mom Evalyne (Monica); my wife Harriet; my daughter Trixie; and my siblings; Andrew, Hildah, and Simon

Thank you for your patience and all the support

## ACKNOWLEDGEMENTS

I would like to express my sincere gratitude to my PhD advisor, Professor Robert E. Maleczka, Jr., for his guidance and support throughout my PhD program. I came to graduate school with zero knowledge on setting up and running organic synthesis reactions. His patience and diplomatic approach to my questions and concerns helped me grow both as a scientist and a person. Even though he was the department chair at the time I joined his research group, he always found time out of his busy schedule to follow my research progress and give me valuable suggestions. Since the purpose of PhD program is to be an independent researcher, he gave me freedom and resources to explore my own ideas. I am grateful to have had him as my mentor.

I would also like to thank my PhD committee members; Professor Babak Borhan, Professor William D. Wulff and Professor Melanie Cooper for their advice and helpful suggestions during my second-year oral examination and my final defense. I appreciate the lectures given by Professor Borhan as the course instructor for CEM 845 which provided the necessary knowledge I needed to conduct my research. I thank Professor Wulff for leading the discussion in CEM 850 where I learnt the basics of proposing plausible organic synthesis reaction mechanisms. I would also like to thank him for the lectures during CEM 852 where I gained valuable knowledge of organic synthesis and contributed to his database. I would also like to thank Professor Xuefei Huang for CEM 851 which also gave me the opportunity to learn advanced physical organic chemistry, and which was applicable to my research. I would also like to thank Professor Jetze Tepe and Professor Milton R. Smith, III for CEM 956 (heterocyclic chemistry) and CEM 820 (organometallics) respectively.

I would like to thank the staff at the Department of Chemistry, Michigan State University for all their support. I would like to thank Dr. Daniel Holmes and Dr. Li Xie for making sure the NMR instruments work well, Dr. Richard Staples for solving my crystal structures, Dr. Anthony

Schilmiller for the high-resolution mass spectroscopy assistance. I would also like to thank Anna Osborn for the guidance on PhD program in chemistry always making sure we don't miss the deadlines. I would like to thank Heidi Wardin for her work in the graduate office and the chair's office. I would like to thank Dawn Kuhn, Bill Flick and Eric Smariege for all their help with chemical and instrument ordering. I would like to thank Bob Racico for maintaining the Chemistry building enabling us to conduct research. I would like to thank Tiphani Scott, Bethanny Potter, Mary Mroz and Brenda Franklin for all their support. I would like to thank Dr. Chrisoula Vasilieou, Dr. Fangyi Shen and Dr. Ardeshir Azadnia for the opportunity they gave me as a Teaching Assistant in their respective undergraduate courses.

Most importantly, I would like to thank my family. I thank my mom (Evalyn Shikuku) and dad (Syverio Maloba) for believing in my academic ability and providing the necessary support that catapulted me to the highest level of academic achievement that I could have dreamed of. I thank my siblings; Andrew Maloba, Hildah Maloba and Simon Maloba for all their encouragement and support. Special thanks to my wife, Harriet, and my daughter, Trixie, for their unconditional love and support that helped me navigate the tough journey associated with graduate school.

I would like to thank my research collaborators for their helpful discussion and suggestions. I thank Dr. Luis Martin Mori-Quiroz who laid the foundation on Wittig rearrangement of cyclic ethers. His contribution shaped some of the work reported in this dissertation to a great extent. I thank Dr. Maria del Rosario Amado Sierra for her contributions in the silyl migration by epoxidation work. I thank Nagham Al Masraf for her contributions in the synthesis of starting materials for the silyl migration by epoxidation work. I also thank her for allowing me to be her mentor both as an ACS Project Seed high school student (East Lansing High School) and an undergraduate research fellow at Michigan State University.

I would like to thank the Maleczka group members; both former and current. This group was like my second family, and I carry with me fond memories. My special gratitude goes to Professor Jonathan Dannat for the help he accorded me when I joined the group. I thank Dr. Jose Raul Montero Bastidas (Pepe), Dr. Badru Deen Barry, Dr. Fangyi Chen, Dr. Ruwandhi Jayasundara (Ruwi) and Dr. Aditya Patil (Chemical Engineer). I thank the current Maleczka group members; Thomas, Arzoo, Sean, Chris, Pauline, Anshu, Cliff, Nehali, Junhui and Jenna. Special thanks to Darshika for taking over the Wittig rearrangement project.

I would like to thank all the friends I met while at graduate school. These include but not limited to, Dr. Kunli Liu, Dr. Jurick Lahiri, Zhen Li, Dr. Grace Hubbell, Dr. Shivangi Chugh, Aimen Al Hilfi, Dr. Timothe Melin, Katayoon Maghami, Dr. Dan Wanyama, Dr. Donald Akanga, Titus Omanga, Dare George, Dr. Katie Kwiatkowski, Bismarck Amaniampong, Nicholas Wills, Jesse Cantrell, Jiaojiao Wang, Rosemary Augustine and Morgan Mayieka. Special thanks to Ankush Chakraborty and Dr. Taylor Fiolek for helpful discussions during lunch on Fridays. I thank Dr. Amaya Mathes Hewage, Dr. Samantha Houchlei and Dr. Zhilin Hou for the fun moments and support during my time in graduate school. I thank Dr. Herbert Kavunja and Dr. Pauline Wambua for their guidance. Thanks to the Kenyan community in greater Lansing for the events that made me feel at home. Special thanks to my uncle, Ashitiva Mandale, for all the support he gave me.

I would also like to thank my mentors throughout my academic journey. I thank Mrs. Benardine Omondi, former headmistress of Endebess Centre Primary School for identifying my unique academic talent. I thank Mr. Samuel Abraham, principal of Manor House High School, and all teachers for the support they gave me in high school. I thank the academic staff of the Department of Chemistry at the University of Nairobi for guiding and teaching me throughout my undergraduate studies. Special thanks to Professor John Mmari Onyari for encouraging me to pursue a PhD.

## TABLE OF CONTENTS

LIST OF SYMBOLS AND ABBREVIATIONS ..... ix
CHAPTER 1. INTRODUCTION ..... 1
1.1. Background of Wittig rearrangements ..... 1
1.2. First examples of Wittig rearrangements ..... 2
1.3. The $[2,3]$-Wittig rearrangement ..... 2
1.4. The $[1,2]$-Wittig rearrangement ..... 7
1.5. The $[1,4]$-Wittig rearrangement ..... 14
1.6. Control of regioselectivity during Wittig rearrangements ..... 17
REFERENCES ..... 24
CHAPTER 2. THE [1,2]- AND [1,4]-WITTIG REARRANGEMENTS OF 2-SILYL-7-ARYL-2,5,6,7-TETRAHYDROOXEPINS ..... 30
2.1. Introduction ..... 30
2.2. Expected products of $[1,2]$ - and $[1,4]$-Wittig rearrangements of 2-silyl-7-aryl-2,5,6,7- tetrahydrooxepins ..... 31
2.3. Synthesis of 2-silyl-7-aryl-2,5,6,7-tetrahydrooxepins ..... 31
2.4. Wittig rearrangements of 2-trimethylsilyl-2,5,6,7-tetrahydro-7-aryl-oxepins ..... 32
2.5. Conclusion ..... 34
2.6. Experimental section ..... 34
REFERENCES ..... 66
APPENDIX ..... 68
CHAPTER 3. SILYLCYCLOPROPANES BY SELECTIVE [1,4]-WITTIG REARRANGEMENT OF 4-SILYL-5,6-DIHYDROPYRANS ..... 155
3.1. Introduction ..... 155
3.2. Synthesis of 4 -silyl-5,6-dihydro-2H-pyrans ..... 157
3.3. Optimization of reaction conditions for Wittig rearrangement ..... 158
3.4. Wittig rearrangements of 4-silyl-5,6-dihydro-2H-pyrans with alkyl substituents at the migrating carbon ..... 158
3.5. Wittig rearrangements of 4-silyl-5,6-dihydro-2H-pyrans with alkyl substituents at the migrating carbon ..... 160
3.6. Rearrangement of substrates bearing electron-deficient aryl groups and 2-naphthyl derivative ..... 161
3.7. Comparative studies on Wittig rearrangements of dihydropyrans ..... 163
3.8. Proposed mechanism of the [1,4]-Wittig rearrangement of 4-silyl-6-aryl(alkyl)-5,6-dihydroprans ..... 163
3.9. Conclusion ..... 164
3.10. Experimental section ..... 165
REFERENCES ..... 240
APPENDIX ..... 245
CHAPTER 4. A [1,2]-WITTIG / m-CPBA TRIGGERED [1,2]-CARBON-TO-CARBON SILYL MIGRATION APPROACH TO $\alpha$-SILYL- $\beta$-HYDROXY CYCLOPENTANONES AND CYCLOHEXANONES ..... 342
4.1. Introduction ..... 342
4.2. Synthesis of 2-silyl-6-aryl-5,6-dihydro-2H-pyrans ..... 343
4.3. Wittig rearrangements of 2-silyl-6-aryl-5,6-dihydro-2H-pyrans ..... 344
4.4. The [1,2]-carbon-to-carbon silyl migration in cyclic system triggered by epoxidation ..... 346
4.5. Proposed reaction mechanism for the silyl migration ..... 348
4.6. Conclusion ..... 350
4.7. Experimental section ..... 350
REFERENCES ..... 413
APPENDIX ..... 418
CHAPTER 5. SERENDIPITOUS [1,2]-CARBON-TO-CARBON SILYL MIGRATION IN $\alpha$-HYRDOXY ALLYL SILANES: ACCESS TO $\alpha$-SILYL ALKANALS ..... 601
5.1. Introduction ..... 601
5.2. Serendipitous [1,2]-carbon-to-carbon silyl migration ..... 602
5.3. Substrate scope for [1,2]-carbon-to-carbon silyl migration ..... 607
5.4. Proposed reaction mechanism of the [1,2]-carbon-to-carbon silyl migration ..... 609
5.5. Attempted mechanistic investigation of the [1,2]-carbon-to-carbon silyl migration ..... 609
5.6. Unexpected $\mathrm{S}_{\mathrm{N}} 2$-like reaction between the O -silylated
2-phenylprop-2-en-1-ol and butyl lithiums ..... 611
5.7. Conclusion ..... 612
5.8. Experimental section ..... 613
REFERENCES ..... 635
APPENDIX ..... 640
CHAPTER 6. FUTURE WORK ..... 696
6.1. Future work on Wittig rearrangements ..... 696
6.2. Future work on silyl migration ..... 697

## LIST OF SYMBOLS AND ABBREVIATIONS

| APCI | atmospheric-pressure chemical ionization |
| :---: | :---: |
| Ar | aromatic |
| $\mathrm{BF}_{3} \cdot \mathrm{OEt}_{2}$ | boron trifluoride diethyl ether |
| Bn | benzyl |
| Bz | benzoyl |
| CI | chemical ionization |
| d | doublet |
| DBU | 1,8-diazabicycloundec-7-ene |
| DCM | dichloromethane |
| DMAP | 4-diaminopyridine |
| DMF | $\mathrm{N}, \mathrm{N}$-dimethylformamide |
| DMP | Dess-Martin periodinane |
| DMSO | Dimethyl sulfoxide |
| $d r$ | Diastereomeric ratio |
| $e e$ | enantiomeric excess |
| EI | electron ionization |
| ESI | electrospray ionization |
| $\mathrm{Et}_{3} \mathrm{~N}$ | triethylamine |
| $\mathrm{Et}_{2} \mathrm{O}$ | diethyl ether |
| EtOAc | ethyl acetate |
| equiv | equivalents |
| g | gram(s) |
| GC/MS | gas chromatography / mass spectrometry |
|  | ix |


| h | hour(s) |
| :---: | :---: |
| HPLC | high pressure liquid chromatography |
| HRMS | high resolution mass spectrometry |
| Hz | hertz |
| $i-\mathrm{Pr}$ | isopropyl |
| IR | infrared |
| J | NMR coupling constant |
| m | multiplet |
| $m$-CPBA | 3-chloroperbenzoic acid |
| min | minute |
| mg | milligram |
| mL | milliliter |
| mp | melting point |
| MHz | megahertz |
| M | molar |
| Me | methyl |
| MeO | methoxy |
| $m / z$ | mass to charge ratio |
| $n-\mathrm{BuLi}$ | $n$-butyllithium |
| $n-\operatorname{Pr}$ | $n$-propyl |
| NaOH | sodium hydroxide |
| NMR | Nuclear Magnetic Resonance |
| NOE | nuclear Overhauser effect |


| Ph | phenyl |
| :--- | :--- |
| q | quartet |
| RCM | ring-closing metathesis |
| s | singlet |
| sat | saturated |
| sec-BuLi | sec-butyllithium |
| $\mathrm{SiEt}_{3}$ | triethylsilyl |
| $\mathrm{SiMe}_{2} \mathrm{Ph}$ | phenyldimethylsilyl |
| $\mathrm{SiPh}_{2} \mathrm{Me}$ | diphenylmethylsilyl |
| $\mathrm{S}_{\mathrm{N} 2}$ | bimolecular nucleophilic substitution |
| rt | room temperature |
| t | triplet |
| $t$-BuLi | tert-butyllithium |
| $t$-BuONa | sodium tert-butoxide |
| TBAF | trimethylsilyl trifluoromethane sulfonate |
| TBDPS | tetrabutylammonium fluoride |
| TBS | tert-butyldiphenylsilyl |
| THF | trimethy |
| TMS |  |
| TMSCl |  |

## CHAPTER 1.

## INTRODUCTION

### 1.1. Background of Wittig rearrangements

Wittig rearrangements entail restructuring of bonds of carbanionic ethers (usually allylic or benzylic) leading to structural isomers. The products resulting from Wittig rearrangements are of interest since they possess functional groups which can further be exploited during organic synthesis. Multiple rearrangements are possible depending on the substrate's nature (Scheme 1.1).


Scheme 1.1: Wittig rearrangements with proposed reaction mechanisms

### 1.2. First examples of Wittig rearrangements

The very first example of a [1,2]-alkyl shift was reported by Schorigin in 1924. ${ }^{1}$ Schorigin conducted further experiments on this unusual rearrangement and reported the findings a year later. ${ }^{2}$ Treatment of 1-(benzyloxy)-4-methylbenzene (1.6) with sodium at $100^{\circ} \mathrm{C}$ led to the formation of phenyl( $p$-tolyl)methanol (1.7) (Scheme 1.2a). Later on in 1942, Wittig and Löhmann reported [1,2]alkyl shifts of benzyl ethers to the corresponding alcohols triggered by deprotonation at the benzylic position. ${ }^{2}$ An example from their work was the rearrangement of (oxybis(methylene))dibenzene (1.8) to 1,2-diphenylethan-1-ol (1.9) on treatment with phenyllithium (Scheme 1.2b). This discovery led to what is currently known as Wittig rearrangements.


Scheme 1.2: First examples of [1,2]-Wittig rearrangements

### 1.3. The $[2,3]$-Wittig rearrangement

Of all the Wittig rearrangements, the [2,3]-Wittig has received the most attention from both synthetic and mechanistic standpoint. As the products of this rearrangement are homoallylic alcohols, the [2,3]-Wittig ranks highly among the known techniques for the synthesis of these alcohols in a stereoselective fashion. ${ }^{4-7}$ The first [2,3]-Wittig rearrangement was reported by Wittig in $1949^{8}$ where the rearrangement of allyl fluorenyl ether (1.10) led to the formation of 9-allyl-9H-
fluoren-9-ol (1.11) (Scheme 1.3).


Scheme 1.3: The [2,3]-Wittig rearrangement of 9-(allyloxy)-9H-fluorene
At first, the reaction mechanism of the formation of alcohol 1.11 was not clear since the above reaction could proceed by either [1,2]- or [2,3]-Wittig rearrangement. In 1960, Cast and Stevens subjected allyl fluorenyl ether $\mathbf{1 . 1 2}$ with a methyl group at carbon 1' to Wittig rearrangement conditions leading to alcohol $\mathbf{1 . 1 3}$ (Scheme 1.4). ${ }^{9}$ This was the first unambiguous scenario of [2,3]-shift that could be distinguished from the [1,2]-Wittig rearrangement.


Scheme 1.4: Wittig rearrangement of 9-(but-3-en-2-yloxy)-9H-fluorene

### 1.3.1. Mechanism and stereochemistry of the [2,3]-Wittig rearrangement

In most cases, [2,3]-Wittig rearrangements are carried out on allylic ether substrates that bear groups such as alkenes, aryl and alkynes that serve the purpose of stabilizing carbanions. Also used to stabilize the carbanion involved in this rearrangement are electron-deficient amides and esters. For example, a model substrate $\mathbf{1 . 1 3}$, proceeds by initial metalation of the substrate leading to intermediate stabilized carbanion 1.14 (Scheme 1.5). This carbanion then undergoes [2,3]-Wittig rearrangement in a concerted fashion through an envelope-like five-membered transition state $\mathbf{1 . 1 5}$
resulting in the formation of the homoallylic alkoxide 1.16. The homoallylic alcohol product $\mathbf{1 . 1 7}$ is formed after the resulting alkoxide $\mathbf{1 . 1 6}$ is protonated during aqueous workup. ${ }^{10}$


Scheme 1.5: Proposed mechanism of the [2,3]-Wittig rearrangement
Most [2,3]-Wittig rearrangements are stereoselective, that is, they result in the formation of one alkene stereoisomer. ${ }^{4}$ This is attributed to the concerted mechanism shown in Scheme 5. For example, Wittig rearrangement of model substrate $\mathbf{1 . 1 8}$ leads to the formation of $E$-alkene $\mathbf{1 . 2 0}$ through transition state $\mathbf{1 . 1 9}$ where 1,3-diaxial interactions are minimized (Scheme 1.6a). Occasionally selectivity towards the $Z$-alkenes is observed when interactions between $R$ and $R^{\prime}$ become significant in transition state 1.19. In these cases, the low energy route yields $Z$-alkene products $\mathbf{1 . 2 2}$ via the transition state 1.21 (Scheme 1.6b). The G-group in transition states $\mathbf{1 . 1 9}$ and 1.21 prefers an equatorial orientation, hence, chirality transfer is usually quite effective. As a result, optically active substrates such as $\mathbf{1 . 1 8}$ are converted to corresponding homoallylic alcohols $\mathbf{1 . 2 0}$ and $\mathbf{1 . 2 2}$ without losing optical purity.


Scheme 1.6: Chirality transfer and alkene stereochemistry in [2,3]-Wittig rearrangement
For internal alkenes, model allyl ethers $\mathbf{1 . 2 3}$ and $\mathbf{1 . 2 6}$ undergo diastereoselective [2,3]Wittig rearrangement to the corresponding homoallylic alcohols $\mathbf{1 . 2 5}$ and $\mathbf{1 . 2 8}$ respectively when $\mathrm{G}=$ aryl, alkenyl or alkynyl (Scheme $1.7 \mathrm{a}, \mathrm{b}$ ). ${ }^{10,11}$ However, switching the carbanion stabilizing group to a carbonyl-containing electron withdrawing groups such as an amide or an ester ( $\mathrm{G}=$ $\mathrm{CONR}_{2}(\mathbf{1 . 2 9})$ or $\mathrm{G}=\operatorname{COOR}(\mathbf{1 . 3 2})$ ) reverses the stereoselectivity (Scheme $1.7 \mathrm{c}, \mathrm{d}$ ). The $[2,3]-$ Wittig rearrangements of $\mathbf{1 . 2 9}$ and $\mathbf{1 . 3 2}$ proceed via transition states $\mathbf{1 . 3 0}$ and $\mathbf{1 . 3 3}$ leading to homoallylic alcohols $\mathbf{1 . 3 1}$ and $\mathbf{1 . 3 4}$ respectively. The pseudoaxial orientation of the carbonyls in transition states $\mathbf{1 . 3 0}$ and $\mathbf{1 . 3 3}$ lead to the stabilization of the developing negative charge at C3. ${ }^{10}$


Scheme 1.7: [2,3]-Wittig transition state models for reactions of internal alkene substrates

### 1.3.2. Selected application of the [2,3]-Wittig rearrangements

As mentioned earlier, the [2,3]-Wittig rearrangement is the most utilized Wittig rearrangement from a synthetic perspective. Recent examples of this rearrangement include, but are not limited to, diastereoselective rearrangement of N -allyl ammonium ylides (Scheme 1.8 a ). ${ }^{12}$ This is an example of the aza-Wittig rearrangement since the ethereal oxygen has been replaced by nitrogen. Use of organocatalysts to control stereochemical outcomes of [2,3]-Wittig rearrangement has also been reported recently. For instance, Kanger et al. developed a formal asymmetric [2,3]Wittig rearrangement using Cinchona-derived amine organocatalyst $\alpha$-branched ketones (Scheme $1.8 b) .{ }^{13}$


Scheme 1.8: Recent report on [2,3]-Wittig rearrangement
TBD = 1,5,7-triazabicyclo[4.4.0]dec-5-ene. p-NBA = para-Nitrobenzoic acid

### 1.4. The [1,2]-Wittig rearrangement

The [1,2]-Wittig rearrangement is a sigmatropic shift resulting from benzyl or allyl carbanions, leading to the formation of allyl alcohol or homoallylic alcohol depending on the site of deprotonation (Scheme 1). It was the first type of Wittig rearrangement to be reported in the literature (Scheme 1.2). ${ }^{1-3}$ The [1,2]-Wittig rearrangement's synthetic utility is usually constrained by the need for harsh reaction conditions employing strong bases that are at times not compatible with some functional groups present. Additionally, intramolecular elimination processes tend to compete with [1,2]-Wittig resulting in low yields. Despite this, construction of complex molecules in a stereoselective fashion has been achieved recently via [1,2]-Wittig rearrangements of specific classes of ethers.

### 1.4.1. Mechanism of the [1,2]-Wittig rearrangement

To carry out reactions that are expected to trigger [1,2]-Wittig rearrangements, it is important to understand the mechanism that leads to this rearrangement. This is helpful in designing
substrates that would eliminate or suppress other Wittig rearrangements in favor of the $[1,2]$ shift. After its discovery, three different mechanistic pathways were proposed for the [1,2]-Wittig rearrangement (Scheme 1.9).


Scheme 1.9: Proposed pathways for the [1,2]-Wittig rearrangement reaction mechanisms

### 1.4.1.1. The concerted mechanism (Scheme 1.9, route a)

The rearrangement of carbanion $\mathbf{1 . 4 0}$ (resulting from deprotonation of $\mathbf{1 . 3 9}$ ) to form benzylic alkoxide $\mathbf{1 . 4 3}$ was proposed to undergo a concerted mechanism by Hauser and Kantor. ${ }^{14}$ They argued for an intramolecular isomerization involving the 1,2-shift of an alkyl group presumably without its bonding pair of electrons to directly produce alkoxide $\mathbf{1 . 4 3}$ (Scheme 1.9, route a). The concerted mechanism was thought to be the pathway until Woodward and Hoffman developed orbital symmetry rules, ${ }^{15}$ which implied that such [1,2]-antarafacial migration was not possible geometrically. Furthermore, Fukui's frontier orbital theory ${ }^{16,17}$ ruled out the first mechanism since if it were to occur this way then there would be inversion of stereochemistry at the migrating carbon center (Scheme 1.10a), which would not be in agreement with the results from the experiments carried out by Schöllkopf on optically active ethers. The [1,2]-Wittig rearrangements resulting from these ethers showed stereochemical retention of high degree at the migrating carbon center (Scheme 1.10b). ${ }^{18-20}$ Recent studies by Mori and Maleczka on optically
active dihydropyrans also showed excellent retention of stereochemistry during the [1,2]-Wittig rearrangement (Scheme 1.10c, d). ${ }^{21,22}$


Scheme 1.10: Geometrically impossible [1,2]-antarafacial migration and retention of stereochemistry

### 1.4.1.2. Stepwise mechanism via carbanion and carbonyl intermediates (Scheme 9, b)

Following the above observation, a stepwise mechanism (Scheme 1.9, route b) involving elimination leading to a carbanion and carbonyl intermediate 1.41 was proposed. This carbanion then attacks the newly formed carbonyl 1.41 leading to the alkoxide $1.43 .{ }^{18,19,23}$ This agreed with the observed stereochemichal retention during the rearrangement. The isolation of $p$-nitrotoluene by Holmes when p-nitrobenzyl ethers underwent rearrangement was also considered as evidence to support this mechanism (Scheme 1.11a). ${ }^{24}$ Recently, Sanz and Faza et al. have shown evidence
supporting the ionic character of [1,2]-Wittig rearrangement. ${ }^{25}$


Scheme 1.11: Evidence for stepwise mechanism via carbanion and carbonyl intermediates
However, there was a drawback to this mechanism (Scheme 1.9, route b). It was observed that tertiary alkyl groups migrated easily followed by secondary then primary ones, hence, ruling out stepwise mechanism involving heterolytic cleavage. ${ }^{26}$ Furthermore, significant $\beta$-hydride elimination was observed during [1,2]-Wittig rearrangements of benzyl alkyl ethers with primary alkyl groups containing protons $\alpha$ to the migrating carbon (Scheme 1.12). ${ }^{3,26-28}$


Scheme 1.12: $\beta$-hydride elimination during [1,2]-Wittig rearrangements of benzyl butyl ethers

### 1.4.1.3. Stepwise mechanism via radical / radical anion intermediates (Scheme 9, route c)

The above trend suggested presence of radicals as intermediates during the migration. Therefore, Lansbury, Pattison, Sidler and Bieber proposed a stepwise mechanism where following deprotonation, homolytic cleavage of the $\mathrm{C}-\mathrm{O}$ bond occurs. The carbon radical and the carboncentered radical anion recombine. The driving force here is the transfer of the formal negative charge from carbon to the electronegative oxygen atom (Scheme 1.9, route c). ${ }^{26}$

In favor of the homolytic cleavage stepwise mechanism (Scheme 1.9, route c) and against the heterolytic cleavage stepwise mechanism (Scheme 1.9 , route $b$ ) was the following observation: 1-(Benzyloxy)adamantane (1.65) was found to undergo [1,2]-Wittig rearrangement to form adamantan-1-yl(phenyl)methanol (1.64) but 1-(benzyloxy)norbonane (1.66) did not (Scheme 1.13). ${ }^{26,29}$ This follows from the reasoning that the 1 -norbornyl radical bears more strain and is thus less stable than the 1 -adamantyl radical, ${ }^{30}$ and the opposite is true for the corresponding anions. ${ }^{31}$ This was in agreement with the stepwise mechanism involving radical / radical anion intermediates.


Scheme 1.13: Radical versus anion stability with respect to [1,2]-Wittig rearrangement

### 1.4.2. Rationale behind Retention of Stereochemistry after [1,2]-Wittig Rearrangement

The stereochemistry retention during [1,2]-Wittig rearrangement ${ }^{18-20}$ (Scheme 1.10) suggests that the radical /radical anion pair recombines at a rate that is faster than rate of
racemization of that center. This has been explained as being due to the Wittig rearrangement occurring within a "solvent cage". Radical clock experiments have been used to gather evidence for a fleeting life of the migrating radical species: The cyclopropylmethyl group in ((cyclopropylmethoxy)methyl)benzene (1.68) experienced negligible isomerization after [1,2]Wittig rearrangement (Scheme 1.14a). ${ }^{23}$ From this observation it can be inferred that the recombination between the radical and radical anion is faster than $9.4 \times 10^{7} \mathrm{~s}^{-1}$, which is the time it takes for the cyclopropyl ring to open. ${ }^{32}$ Further rationalization involves inverse approach experiments: The expected [1,2]-Wittig rearrangement product $\mathbf{1 . 7 2}$ was exclusively formed from the rearrangement of ((hex-5-en-1-yloxy)methylene)dibenzene (1.71). Isomerization of the 5hexenyl radical to form product $\mathbf{1 . 7 3}$ was not observed (Scheme 1.14b). ${ }^{33,34}$ This is in agreement with the faster rate of rearrangement and, hence, the reason behind observed retention of stereochemistry.


Scheme 1.14: Radical clock experiment during [1,2]-Wittig rearrangements

### 1.4.3. Representative applications of [1,2]-Wittig rearrangements in total synthesis

Unlike the $[2,3]$-Wittig rearrangement, the requirements to tune substrates in order to stabilize the radical of the migrating fragments has made the application of the [1,2]-Wittig
rearrangement limited. ${ }^{35}$ Furthermore, generating an $\alpha$-carbanionic ether by use of strong bases that do not always tolerate the functional groups that may be present has also limited the use of the [1,2]Wittig rearrangement in total synthesis. However, Nakai took advantage of the chirality present in carbohydrates to develop a stereocontrolled acetal systems [1,2]-Wittig rearrangement. ${ }^{36-38} \mathrm{He}$ further utilized this technology ${ }^{37}$ in one of the steps towards the total synthesis of zaragozic acid A
(1.76) (Scheme 1.15). ${ }^{38}$


Scheme 1.15: Application of [1,2]-Wittig rearrangement to total synthesis of zaragozic acid A
Garson et al. reported the isolation and total synthesis of (-)-(5R,6Z)-dendrolasin-5-acetate in five steps. ${ }^{39-42}$ The key step employed involved a [1,2]-Wittig rearrangement of geranyl 3furylmethyl ether (1.77) to produce alcohol 1.78, followed by acetylation and resolution to obtain target molecule 1.79 (Scheme 1.16).


Scheme 1.16: Application of [1,2]-Wittig rearrangement to total synthesis of dendrolasin-5-acetate
Recently, Ho and Li developed a practical method to functionalize $\gamma$-benzyloxy vinylogous urethanes into $\gamma$-benzyl butenolides through tandem [1,2]-Wittig rearrangement/lactonization. As a proof of concept, they applied this development in the total synthesis of maculalactone A, planchol C and $\gamma$-lycorane (Scheme 1.17). ${ }^{43}$


Scheme 1.17: Application of [1,2]-Wittig rearrangement to total synthesis of Maculalactone A, Planchol C and $\gamma$-Lycorane

### 1.5. The $[1,4]$-Wittig rearrangement

Transformations via [1,4]-Wittig rearrangement are rare due to the competing [1,2]-Wittig (Scheme 1.1). First reported by Falkin and Tambute in $1969,{ }^{44}$ the [1,4]-Wittig rearrangement results in the formation of enolate which is transformed into carbonyl after protonation. The enolate
can also be trapped by various electrophiles.

### 1.5.1. Mechanism of the $[1,4]$-Wittig rearrangement

There are two different proposed mechanistic pathways for the [1,4]-Wittig rearrangement:
An orbital symmetry allowed concerted pathway (Scheme 1.18, route a) or a stepwise mechanism like the [1,2]- case involving radical / radical anion intermediates (Scheme 1.18, route b).


Scheme 1.18: Proposed mechanistic pathways for [1,4]-Wittig rearrangement
Experimental evidence in support of both approaches have been reported in the literature. ${ }^{21,22,45}$ In addition, a DFT study carried out recently by Joshi et al. on dihydrofuran systems indicated that the reaction followed a stepwise mechanism during the [1,4]-Wittig rearrangement. ${ }^{22}$

### 1.5.2. Application of $[1,4]$-Wittig rearrangement

The [1,4]-Wittig rearrangement has been observed during [1,2]-Wittig rearrangements. ${ }^{37,46-}$ ${ }^{48}$ Although the bases used to initiate Wittig rearrangements play a role in the selectivity between [1,2]- and [1,4]-Wittig rearrangements, the cause of the reaction is highly substrate-dependent. Generally, low temperatures tend to favor [1,4]-Wittig. There have been recent reports on highyielding reactions that proceed via the [1,4]-Wittig rearrangement. Extensive studies on the [1,4]Wittig rearrangement have been conducted by the Maleczka group. A highly selective [1,4]-Wittig rearrangement of $\alpha$-benzyloxyallylsilane (1.89) has been reported by Onyeozili and

Maleczka (Scheme 1.19a). ${ }^{48}$ The enolate product $\mathbf{1 . 9 0}$ from this rearrangement could be trapped with various electrophiles leading to acylsilanes 1.91 substituted at the $\alpha$-position. In 2015, Maleczka and Mori were able to tune the dihydropyrans in favor of either [1,2]- or [1,4]-Wittig rearrangement. For instance, the [1,2]-Wittig rearrangement was achieved when they employed electron-deficient aromatic groups and/or small groups on silicon. The compliment was also true (Scheme 19b). ${ }^{21}$ Surprisingly, when the silyl group was moved to form 4-silyl-6aryl/alkyldihydropyrans, [1,4]-Wittig rearrangement was to the corresponding silylcyclopropane acetaldehydes predominated (Scheme 1.19c), ${ }^{49}$ (Chapter 3).


Scheme 1.19: Selective [1,4]-Wittig rearrangement examples from Maleczka group
In 2015, Xu reported that chalcone-derived allylic ethers 1.97 undergo [1,4]-Wittig rearrangement resulting in the formation of benzyl ketones $\mathbf{1 . 9 8}$, which are disubstituted at the $\beta$ position (Scheme 1.20a). ${ }^{50}$ Not surprisingly, the [1,4]-Wittig rearrangement of optically active ( $R, E$ )-(3-(benzyloxy)prop-1-ene-1,3-diyl)dibenzene (1.99) led to racemic 1,3,4-triphenylbutan-1-
one (1.100). This could be attributed to the change of hybridization to $\mathrm{sp}^{2}$ at the stereogenic carbon during the reaction (Scheme 1.20b).


Scheme 1.20: [1,4]-Wittig rearrangement of ( $E$ )-(3-(benzyloxy)prop-1-ene-1,3-diyl)dibenzenes

### 1.6. Control of regioselectivity during Wittig rearrangements

As illustrated in scheme 1.1 and throughout the previous discussion, the [1,2]-Wittig rearrangements tend to compete with [2,3]-Wittig, whereas the [1,4]-Wittig competes with the [1,2]-Wittig. For instance, Rautenstrauch observed [2,3]-, [1,2]-, [1,4]- and [3,4]-Wittig rearrangements when 3-methyl-1-((3-methylbut-2-en-1-yl)oxy)but-2-ene (1.101) was treated with a base, leading to formation of products $\mathbf{1 . 1 0 2}, \mathbf{1 . 1 0 3}, \mathbf{1 . 1 0 4}$ and $\mathbf{1 . 1 0 5}$ respectively (Scheme 1.21). ${ }^{51}$


Scheme 1.21: All possible Wittig rearrangements

### 1.6.1. Control of regioselectivity via metal/lithium exchange

For ethers that are substituted unsymmetrically, regioselective Wittig rearrangement becomes a challenge. The ability to generate the anion can be controlled by installing a displaceable group M at either the $\alpha$ or $\alpha^{\prime}$ positions. This group can then be displaced to generate an anion, which will determine the type of Wittig rearrangement the substrate will undergo, overcoming the regioselectivity challenge (Scheme 1.22).


Scheme 1.22: Control of regioselectivity via group M-directed carbanion generation
The above idea led to the discovery of what is currently known as the Wittig-Still
rearrangement (Scheme 1.23a). ${ }^{52}$ The M-group in this case is tributyl tin. The tin-lithium exchange generates unstable carbanions, which can then isomerize via Wittig rearrangements. ${ }^{53}$ The WittigStill rearrangement is a powerful synthetic tool ${ }^{54}$ that has been applied widely in the total synthesis of natural product,,${ }^{55-59}$ however, toxicity of organotin ${ }^{60-62}$ led to a search for approaches involving non-toxic materials. Mulzer and List's pioneer work on use of silicon in place of tin (Scheme $1.23 b)^{63}$ motivated Maleczka and Geng to carry out regioselective Wittig rearrangement of $\alpha$ alkoxy silanes via silicon/lithium exchange (Scheme 1.23c). ${ }^{64}$
(a)

Scheme 1.23: Examples involving silicon-lithium exchange / Wittig rearrangements
The absence of groups that could stabilize the resulting anion in Mulzer's examples enabled efficient $\mathrm{Si} / \mathrm{Li}$ exchange and subsequent Wittig rearrangement. In prior work by the Maleczka group, however, the trimethylsilyl group's carbanion-stabilizing effect together with phenyl and olefin groups present in these molecules caused competition between deprotonation / rearrangement
and $\mathrm{Si} / \mathrm{Li}$ exchange.
Replacement of tin with silicon solved the toxicity problem. However, use of strong bases limits the substrate scope due to incompatibility of these bases with many functional groups. To solve this problem, the generation of carbanions by use of fluoride anions has been studied by Nakai, ${ }^{35}$ Reetz ${ }^{65}$ and Maleczka. ${ }^{66}$ The resulting carbanions then undergo Wittig rearrangements. In this case silyl groups are thought to be carbanion masks. Displacement of these groups generates carbanions, which then undergo Wittig rearrangements (Scheme 1.24).


Scheme 1.24: Fluoride-promoted Wittig rearrangements

### 1.6.2. Control of regioselectivity via EWG-directed deprotonation

Another way of controlling the regioselectivity of unsymmetrically substituted ethers is by modifying the $p \mathrm{Ka}$ of the protons at the $\alpha$ and $\alpha^{\prime}$ position. This in turn determines the deprotonation
site and therefore the possible Wittig rearrangements. For instance, placing an EWG group at the $\alpha$ (or $\alpha^{\prime}$ ) position can act as an anion stabilizing group and therefore make deprotonation and Wittig rearrangement regioselective (Scheme 25). Groups such as phenyl, alkynes, carbonyls (esters, amides, ketones, aldehydes), sulfonyl, silyl, and cyano can act to stabilize the anion and are therefore good substituents (EWG) at promoting Wittig rearrangements. ${ }^{67,68}$


Scheme 1.25: Control of regioselectivity via group EWG-directed deprotonation
Of particular interest are the silyl groups since they have the ability to delocalize the negative charge of adjacent carbanions through silicon's d orbitals and, thereby stabilizing the charge. ${ }^{69,70}$ Even though this ability has been attributed to hyperconjugation, ${ }^{71,72}$ the silyl group makes the $\alpha$-proton more acidic by reducing the conjugated acid's $p \mathrm{Ka}$. Therefore, the presence of silicon in a molecule can change the regioselective outcome of the Wittig rearrangement. For example, silicon-free bisallylic ethers were found to undergo selective deprotonation at the less substituted $\alpha^{\prime}$ position (Scheme 1.26a), ${ }^{31}$ however, selective deprotonation at the $\alpha$ position occurred after introduction of the silyl group at the $\gamma$ position. This was followed by Wittig
rearrangement (most substituted position) (Scheme 1.26b). ${ }^{52}$ In this case, vinyl trialkylsilyl group was used as the G-group. The Wittig rearrangements of 6-aryl-4-silyl-5,6-dihydro-( $2 H$ )-pyrans has also been studied. The vinyl silyl group led to exclusive deprotonation at the allylic position. The resulting carbanions underwent [1,2]- and [1,4]-Wittig rearrangements. Surprisingly, the major products isolated were mostly the result of [1,4]-Wittig rearrangement (Scheme 1.26c). ${ }^{49}$


Scheme 1.26: Silicon-directed deprotonation at the allylic position
With regard to the earlier observation on competitive deprotonation vs $\mathrm{Li} / \mathrm{Si}$ exchange (Scheme 1.19a), ${ }^{62}$ placing the silane group at the $\alpha^{\prime}$ position resulted in exclusive deprotonation and avoided the $\mathrm{Si} / \mathrm{Li}$ exchange as reported by Maleczka and Onyeozili. ${ }^{48}$ The $\alpha$ benzyloxyallylsilanes were able to undergo [1,4]-Wittig rearrangement in a very efficient manner. The resulting enolate intermediates were trapped with various electrophiles. This provided a new
synthetic approach to substituted acylsilanes (Scheme 19a). ${ }^{48}$ Cyclic versions of the above have also been studied by the Maleczka group. ${ }^{21,49}$ Wittig rearrangements of diastereomeric 2-silyl-5,6-dihydro-6-aryl- $(2 H)$-pyrans resulted in regiodivergent ring contractions to the corresponding $\alpha$ silylcyclopentenols and/or ( $\alpha$-cyclopropyl)acylsilanes. The [1,4]-Wittig was found to predominate when the aryl substituent on starting pyrans contained electron donating and/or by sterics emanating from the substituents on silicon. The [1,2]-Wittig was achieved with electron withdrawing groups and smaller silyl groups. Substituting the olefin part of the dihydropyran proximal to the silyl group led to exclusive [1,2]-Wittig rearrangements. Furthermore, the cis and trans diastereomers rearranged in a convergent manner leading to the corresponding $\alpha$-silylcyclopentenols and cyclopropyl acyl silanes as a single diastereomer (Scheme 19b).

Regarding the relative stereochemistry of 2-silyl-6-aryl dihydropyrans, trans diastereomers are more reactive than the cis isomers. Mori hypothesized that these reactivity differences were "presumably because an optimal conformation suitable for allylic deprotonation is easily attainable" in the trans isomer and prevented by sterics in the cis. Mori's hypothesis is consistent with conformational analysis and the fact that increased sterics about the silyl groups or aryl rings did not change the reactivity of the trans isomers, while the cis cyclic ethers become less reactive. Computational studies recently done by Joshi et al. confirmed this. ${ }^{22}$

## REFERENCES

(1) Schorigin, P. Über die Carbinol-Umlagerung von Benzyläthern. Ber. Dtsch. Chem. Ges. 1924, 57, 1634.
(2) Schorigin, P. Über die Umlagerungen von Benzyläthern. Ber. Dtsch. Chem. Ges. 1925, 58, 2028.
(3) Wittig, G.; Löhmann, L. Über die kationotrope Isomerisation gewisser Benzyläther bei Einwirkung von Phenyl-lithium. Liebigs Ann. Chem. 1942, 550, 260.
(4) Marshall, J. A. The Wittig Rearrangement. Comprehensive Organic Synthesis 1991, 975.
(5) Mikami, K.; Nakai, T. Acyclic stereocontrol via [2,3]-Wittig sigmatropic rearrangement. Synthesis 1991, 594.
(6) Nakai, T.; Mikami, K. The [2,3]-Wittig rearrangement. Org. React. 1994, 46, 105.
(7) Ahmad, N.M. Rearrangements: [2,3]-Wittig rearrangement. Name Reactions for Homologations, Part II 2009, 241.
(8) Wittig, G.; Doser, H.; Lorenz, I. Über die Isomerisierbarkeit metallierter Fluorenyläther. Liebigs Ann. Chem. 1949, 562, 192.
(9) Cast, J.; Stevens, T. S.; Holmes, J. Molecular rearrangement and fission of ethers by alkaline reagents. J. Chem. Soc. 1960, 3521.
(10) Wolfe, J. 3.21 The Wittig Rearrangement. Comprehensive Organic Synthesis (Second Edition), 2014, 1038.
(11) Wu, Y. D.; Houk, K. N.; Marshall, J. A.Transition structure for the [2,3]-Wittig rearrangement and analysis of stereoselectivities. J. Org. Chem. 1990, 55, 1421.
(12) Murre, A.; Erkman, K.; Kaabel, S.; Järving, I.; Kanger, T. Diastereoselective [2,3]sigmatropic rearrangement of $N$-allyl ammonium ylides. Synthesis 2019, 51, 4183.
(13) Kimm, M.; Ošeka, M.; Kaabel, S.; Metsala, A.; Järving, I.; Kanger, T. [2,3]-Wittig rearrangement as a formal asymmetric alkylation of $\alpha$-branched ketones. Org. Lett. 2019, 21, 4976.
(14) Kantor, S.W.; Hauser, C.R. Rearrangements of benzyltrimethylammonium ion and related quaternary ammonium ions by sodium amide involving migration into the ring. J. Am. Chem. Soc. 1951, 73, 1437.
(15) Woodward, R. B.; Hoffmann, R. The Conservation of Orbital Symmetry, Academic Press, New York, N.Y., 1969.
(16) Fukui, K. An MO-theoretical illumination for the principle of stereoselection. Bull. Chem. Soc. Jp. 1966, 39, 498.
(17) Fukui, K. Recognition of stereochemical paths by orbital interaction. Acc. Chem. Res. 1971, 4, 57.
(18) Schöllkopf, U.; Fabian, W. Umlagerungen organischer anionen, I. stereochemischer Ablauf der umlagerung des metallierten Benzyl-sek.-butyl-äthers; ein beitrag zum problem der SNi-substitution. Liebigs Ann Chem. 1961, 642, 1.
(19) Schöllkopf, U.; Schäfer, H. Umlagerungen organischer anionen, III. Stereochemischer ablauf der umlagerung des lithium-2-benzyloxy-2-phenyl-butans. Liebigs Ann Chem. 1963, 663, 22.
(20) Tomooka, K.; Igarashi, T. Nakai, T. [1, 2]-Wittig rearrangement of enantio-defined $\alpha$ alkoxyalkyllithiums: Structural requirement and steric course at the Li-bearing terminus. Tetrahedron 1994, 50, 5927.
(21) Mori-Quiroz, L. M.; Maleczka, R. E., Jr. Stereoconvergent [1,2]- and [1,4]-Wittig rearrangements of 2-silyl-6-aryl-5,6-dihydropyrans: A tale of steric vs electronic regiocontrol of divergent pathways. J. Org. Chem. 2015, 80, 1163
(22) Nath, S. R.; Joshi, K. A. Mechanistic investigation in the [1,4] and [1,2] Wittig rearrangement reactions: a DFT study. Phys. Chem. Chem. Phys. 2018, 20, 21457
(23) Lansbury, P. T.; Pattison, V. A. The Wittig rearrangement of the benzyl ethers of cyclobutanol and cyclopropylcarbinol. J. Am. Chem. Soc. 1962, 84, 4295.
(24) Cast, J.; Stevens, T. S.; Holmes, J. Molecular rearrangement and fission of ethers by alkaline reagents. J. Chem. Soc. 1960, 3521.
(25) Velasco, R.; Silva López, C.; Nieto Faza, O.; Sanz, R. Exploring the reactivity of $\alpha$-lithiated aryl benzyl ethers: Inhibition of the [1,2]-Wittig rearrangement and the mechanistic proposal revisited. Chem. - Eur. J. 2016, 22, 15058.
(26) Lansbury, P. T.; Pattison, V. A.; Sidler, J. D.; Bieber, J. B. Mechanistic aspects of the rearrangement and elimination reactions of $\alpha$-metalated benzyl alkyl ethers. J. Am. Chem. Soc. 1966, 88, 78.
(27) Letsinger, R. L.; Pollart, D. F. $\alpha$-versus $\beta$-Elimination in the cleavage of ethers by organoalkali metal compounds. J. Am. Chem. Soc. 1956, 78, 6079.
(28) d'Orchymont, H.; Goeldner, M. P.; Biellmann, J.-F. $\alpha^{\prime} \beta$-Elimination and Wittig rearrangement of the carbanion from benzyl cyclooctyl ether. Tetrahedron Lett. 1982, 23, 1727.
(29) Lansbury, P. T.; Pattison, V. A. Some reactions of $\alpha$-metalated ethers. J. Org. Chem. 1962, 27, 1933.
(30) Humphrey, L. B.; Hodgson, B.; Pincock, R. E. Bridgehead free radical stability. The decomposition of t-butylperoxyesters of the 1-adamantyl, 1-bicyclo [2•2•2] octyl, and 1norbornyl systems. Can. J. Chem. 1968, 46, 3099.
(31) Lansbury, P. T.; Sidler, J. D. Relative stabilities of 1-(bicyclo(2.2.1)heptyllithium and 1bicyclo(2.2.2)octyllithium. Tetrahedron Lett. 1965, 6, 691.
(32) Newcomb, M.; Glenn, A. G. A convenient method for kinetic studies of fast radical rearrangements. Rate constants and Arrhenius function for the cyclopropylcarbinyl radical ring opening. J. Am. Chem. Soc. 1989, 111, 275.
(33) Garst, J. F.; Smith, C. D. Mechanisms of Wittig rearrangements and ketyl-alkyl iodide reactions. J. Am. Chem. Soc. 1973, 95, 6870.
(34) Garst, J. F.; Smith, C. D. Wittig rearrangements of aralkyl alkyl ethers. J. Am. Chem. Soc. 1976, 98, 1526.
(35) Takahashi, O.; Maeda, T.; Mikami, K.; Nakai, T. [3,3]-Claisen vs. [2,3]-Wittig shift in thermal and fluolide ion-promoted rearrangements of the O- and C-silylated forms of $\alpha$ allyloxy esters. Chem. Lett. 1986, 1355.
(36) Tomooka, K.; Yamamoto, H.; Nakai, T. [1, 2]-Wittig rearrangement of acetal systems: A highly stereocontrolled conversion of O-glycosides to C-glycosides. J. Am. Chem. Soc. 1996, 118, 3317.
(37) Tomooka. K.; Yamamoto, H.; Nakai, T. Stereoselective synthesis of highly functionalized C-glycosides based on acetal [1, 2] and [1, 4] Wittig rearrangements. Angew. Chem. Int. Ed. 2000, 39, 4500.
(38) Tomooka. K.; Kikuchi, M.; Igawa, K.; Suzuki, M.; Keong, P. -H.; Nakai, T. Stereoselective total synthesis of zaragozic acid A based on an acetal [1, 2] Wittig rearrangement. Angew. Chem. Int. Ed. 2000, 39, 4502.
(39) Kakou, Y.; Crews, P.; Bakus, G. J. Dendrolasin and latrunculin A from the Fijian sponge Spongia mycofijiensis and an associated nudibranch Chromodoris lochi. J. Nat. Prod. 1987, 50, 482.
(40) Hochlowski, J. E.; Walker, R. P.; Ireland, C.; Faulkner, D. J. Metabolites of four nudibranchs of the genus Hypselodoris. J. Org. Chem. 1982, 47, 88.
(41) Tran, N. H.; Hooper, J. N. A.; Capon, R. J. New oxygenated sesquiterpenes from a southern Australian marine sponge, Dictyodendrilla sp. Aust. J. Chem. 1995, 48, 1757.
(42) Mudianta, W; Challinor, V. L; Winters, A. E; Cheney, K. L.; De Voss, J. J.; Garson, M. J. Synthesis and determination of the absolute configuration of (-)-(5R, 6Z)-dendrolasin-5acetate from the nudibranch Hypselodoris jacksoni. Beilstein J. Org. Chem. 2013, 9, 2925.
(50) Gao, P.-S.; Ye, F.; Dong, X.-Y.; Chen, Y.; Gao, Z.-W.; Zhang, W.-Q.; Xu, L.-W. Basepromoted [1,4]-Wittig rearrangement of chalcone-derived allylic ethers leading to aromatic $\beta$-benzyl ketones. RSC Adv. 2015, 5, 33818.
(51) Rautenstrauch, V. The Wittig rearrangement of some allyl ethers. Chem. Comm. 1970, 4.
(52) Still, W. C.; Mitra, A. A highly stereoselective synthesis of Z-trisubstituted olefins via [2, 3]-sigmatropic rearrangement. Preference for a pseudoaxially substituted transition state. J. Am. Chem. Soc. 1978, 100, 1927.
(53) Tomooka, K.; Nakai, T. [1,2]-Wittig rearrangement stereochemical features and synthetic utilities. J. Synth. Org. Chem., Jpn. 1997, 54, 1000.

Rycek, L.; Hudlicky, T. Applications of the Wittig-Still rearrangement in organic synthesis. Angew. Chem. Int. Ed. 2017, 56, 6022.
(55) Ogura, A.; Yamada, K.; Yokoshima, S.; Fukuyama, T. Total synthesis of (-)-Anisatin. Org. Lett. 2012, 14, 1632.
(56) Kikuchi, T.; Mineta, M.; Ohtaka, J.; Matsumoto, N.; Katoh, T. Enantioselective total synthesis of ( - )-subglutinols A and B: potential immunosuppressive agents isolated from a microorganism. Eur. J. Org. Chem. 2011, 5020.
(57) Blackburn, T. J.; Helliwell, M.; Kilner, M. J.; Lee, A. T. L.; Thomas, E. J. Further studies of an approach to a total synthesis of phomactins. Tetrahedron Lett. 2009, 50, 3550.
(58) Millar, J. G.; Moreira, J. A.; McElfresh, J. S.; Daane, K. M.; Freund, A. S. Sex pheromone of the longtailed mealybug: a new class of monoterpene structure. Org. Lett. 2009, 11, 2683.
(59) Ohkubo, M.; Hirai, G.; Sodeoka, M. Synthesis of the DFGH ring system of type B physalins: highly oxygenated, cage-shaped molecules. Angew. Chem. Int. Ed. 2009, 48, 3862.
(60) Smith, P. J. Chemistry of Tin; Blackie Academic \& Professional: New York, 1998.
(61) Davies, A. G. In Organotin Chemistry; VCH: New York, 1997.
(62) Pereyre, M.; Quintard, J.-P.; Rahm, A. In Tin in Organic Synthesis; Butterworth: Toronto, 1987.
(63) Mulzer, J.; List, B. [2,3]-Wittig rearrangements of (trimethylsilyl) methyl allyl ethers. Tetrahedron Lett. 1996, 37, 2403.
(64) Maleczka, R. E.; Geng, F. Methyllithium-promoted Wittig rearrangements of $\alpha-$ alkoxysilanes. Org. Lett. 1999, 1, 1115.
(65) Reetz, M. T.; Greif, N. Dyotrope Umlagerungen, XIII. Fluoridionen-katalysierte umlagerungen von allyl- und benzyl-(silylmethyl)-ethern. Chem. Ber. 1977, 110, 2958.
(66) Maleczka, R. E.; Geng. F. Synthesis and Fluoride-Promoted Wittig Rearrangements of $\alpha-$ Alkoxysilanes. Org. Lett. 1999, 1, 1111.
(67) Damrauer, R.; Crowell, A. J.; Craig, C. F. Electron, hydride, and fluoride affinities of silicon-containing species: Computational studies. J. Am. Chem. Soc. 2003, 125, 10759.
(68) Chan, T. H.; Wang, D. Silylallyl anions in organic synthesis: A study in regio-and stereoselectivity. Chem. Rev. 1995, 95, 1279.
(69) Mikami, K.; Kishi, N.; Nakai, T. Silicon-directed regiocontrol in wittig rearrangements of bis-allyl ethers and allyl propargyl ethers. Chem. Lett. 1989, 18, 1683.
(70) Kishi, N.; Maeda, T.; Mikami, K.; Nakai, T. [2,3] Wittig rearrangement-Peterson olefination sequence: a stereocontrolled entry to terminal conjugated trienes. Tetrahedron, 1992, 48, 4087.
(71) Schleyer, P. V. R.; Clark, T.; Kos, A. J.; Spitznagel, G. W.; Rohde, C.; Arad, D.; Houk, K. N. Rondan, N. G. Structures and stabilities of $\alpha$-hetero-substituted organolithium and organosodium compounds. Energetic unimportance of d-orbital effects. J. Am. Chem. Soc. 1984, 106, 6467.
(72) Wetzel, D. M.; Brauman, J. I. Quantitative measure of. alpha.-silyl carbanion stabilization. The electron affinity of (trimethylsilyl) methyl radical. J. Am. Chem. Soc. 1988, 110, 8333.

## CHAPTER 2.

## THE [1,2]- AND [1,4]-WITTIG REARRANGEMENTS OF 2-SILYL-7-ARYL-2,5,6,7-TETRAHYDROOXEPINS

### 2.1. Introduction

The first reported Wittig rearrangement reaction was the [1,2]-Wittig. ${ }^{1}$ Despite this fact, the [1,2]-Wittig rearrangement has fewer synthetic applications compared to its [2,3]- counterpart. This is partly attributed to competition observed between the [1,2]- and the [2,3]-Wittig rearrangement. Furthermore, the [1,2]-Wittig rearrangement experiences narrower substrate scope and relatively lower yields restricting its synthetic applicability. ${ }^{2-11}$ In addition, the [1,4]-Wittig rearrangement, which was first observed by Felkin and Tambute in 1969 also competes with the [1,2]-Wittig rearrangement. ${ }^{7,12-14}$

In 2015, Mori and Maleczka reported that diastereomeric 2-silyl-5,6-dihydro-6-aryl-(2H)pyrans undergo stereoconvergent [1,2]- and/or [1,4]-Wittig rearrangements to afford $\alpha$ silylcyclopentenols and/or ( $\alpha$-cyclopropyl)acylsilanes respectively (Scheme 1.1). ${ }^{15}$ Although there is usually a competition between the [1,2]- and [1,4]-Wittig pathways, they have shown that the isomerization can be selective towards the [1,2]- or [1,4]-Wittig rearrangements. For instance, the [1,2]-Wittig pathway predominated with electron-deficient aryl groups and smaller alkyl/aryl substituents on silicon. The opposite was true for the [1,4]-Wittig rearrangement. Furthermore, cis and trans dihydropyrans exhibited different reactivities, but converged to single diastereomeric Wittig products (Scheme 2.1). The [1,2]- and [1,4]-Wittig rearrangements are proposed to follow stepwise mechanism. This proposal has been supported by recent computational studies. ${ }^{16}$


Scheme 2.1: [1,2]- and [1,4]-Wittig rearrangements of 2-silyl-6-aryl-5,6-dihydropyrans ${ }^{15}$

### 2.2. Expected products of $[1,2]$ - and $[1,4]$-Wittig rearrangements of 2-silyl-7-aryl-2,5,6,7-tetrahydrooxepins

From the above results, we hypothesized that increasing the ring size of the starting material from 2-silyl-6-aryl-5,6-dihydropyrans to 2-silyl-7-aryl-2,5,6,7-tetrahydrooxepins would enable access to 1-silyl-6-arylcyclohex-2-en-1-ols and/or ( $\alpha$-cyclobutyl)acylsilanes via [1,2]- and/or [1,4]Wittig rearrangement (Scheme 2.2). To the best of our knowledge, the ring contraction of tetrahydrooxepins by Wittig rearrangements has not been previously reported.


Scheme 2.2: Proposed [1,2]- and [1,4]-Wittig rearrangements of 2-silyl-7-aryl-2,5,6,7-tetrahydrooxepins

### 2.3. Synthesis of 2-silyl-7-aryl-2,5,6,7-tetrahydrooxepins

To test the above hypothesis, we first synthesized the starting material by homoallylation of benzaldehydes followed by conversion of the resulting alcohols $\mathbf{2 . 1}$ to trichloroacetimidates $\mathbf{2 . 2}$. The trichloroacetimidates were then coupled with 1-(trimethylsilyl)prop-2-en-1-ol in the presence of catalytic amount of a Lewis acid $\left(\mathrm{BF}_{3} \cdot \mathrm{OEt}_{2}\right.$ or TMSOTf) to form diastereomeric dienes 2.3. The assignment of relative stereochemistry of these dienes was done after the ring closing metathesis by Grubbs $2^{\text {nd }}$ generation catalyst. In some instances, the diastereomeric dienes were separable by column chromatography. In these cases, the dienes were subjected to ring closing metathesis as
single diastereomers. In cases where the diastereomeric dienes $\mathbf{2 . 3}$ were subjected to ring closing metathesis as a mixture of diastereomers (syn:anti), the resulting diastereomers of 2-silyl-2,5,6,7-tetrahydro-7-aryl-oxepins (2.4) were separable by column chromatography and their relative stereochemistry assigned by ${ }^{1} \mathrm{H}$ NMR NOESY experiments (Scheme 2.3). Typically, the anti dienes underwent ring closing metathesis to the corresponding cis oxepins whereas the syn dienes led to trans oxepins.


Scheme 2.3: Synthesis of cis/trans -silyl-2,5,6,7-tetrahydro-7-aryl-oxepins

### 2.4. Wittig rearrangements of 2-trimethylsilyl-2,5,6,7-tetrahydro-7-aryl-oxepins

With the starting materials at hand, we subjected them to Wittig rearrangement conditions. Exposure of (trans)-2-trimethylsilyl-2,5,6,7-tetrahydro-7-aryl-oxepins (2.4) to $n$-butyllithium led to formation of compounds $\mathbf{2 . 5}, \mathbf{2} \mathbf{6}$, and 2.7 in varying yields (Table 1).

Table 2.1 ${ }^{\text {a }}$ : Wittig rearrangement of 2-trimethylsilyl-2,5,6,7-tetrahydro-7-aryl-oxepins

|  |  <br> 2.4 | $\xrightarrow[\substack{\text { THF, } \\ 30 \text { min }}]{\substack{\text { nBuLi } \\(1.2 \text { equiv })}} \text { Ar }$ |  <br> 2.5 <br> [1,2]-Wittig |  |  <br> ittig |  <br> 2.7 |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Entry | Substrate | Ar | (2.5) | $d r$ <br> (2.5) | $\%$ <br> (2.6) | (2.6) | (2.7) |
| 1 | 2.4a | $\mathrm{C}_{6} \mathrm{H}_{5}$ | 25 | 5:1 | n.d | - | 51 |
| 2 | 2.4b | 4-Cl-C6 $\mathrm{H}_{4}$ | 72 | 5:1 | n.d | - | 22 |
| 3 | trans-2.4c | $4-\mathrm{OMe}-\mathrm{C}_{6} \mathrm{H}_{4}$ | 13 | 1.5:1 | 3 | 2:1 | 58 |
| 4 | cis-2.4c ${ }^{\text {b }}$ | $4-\mathrm{OMe}-\mathrm{C}_{6} \mathrm{H}_{4}$ | 24 | 12:1 | 16 | 1:1 | $n . \mathrm{d}^{\text {c }}$ |
| 5 | 2.4d | 2-Naph | 48 | $4: 1$ | 26 | 1:1 | $n . d^{\text {d }}$ |

${ }^{\text {a }}$ Diastereomeric ratio determined by ${ }^{1} \mathrm{H}$ NMR spectroscopy of the crude reaction mixture. The relative stereochemistry was determined by 1D and 2D NMR NOESY analysis. ${ }^{b}$ Reaction performed with sec-butyllithium ( 3.0 equiv.) at $-10^{\circ} \mathrm{C}$ for 5 hours then quenched with deuterium oxide. n.d $=$ not detected. ${ }^{\text {c }} 48 \%$ of ortho deuterated starting material recovered. ${ }^{\text {d }}$ Approximately $10 \%$ of desilylated cyclohexanone $\mathbf{2 . 8 c}$ was formed instead (see experimental section)

Unlike the dihydropyrans, the cis diastereomers were resistant to rearrangement even after employing the conditions successfully used earlier on their cis dihydropyrans counterpart, that is, sec-butyllithium at $-78{ }^{\circ} \mathrm{C}$ for 3 hours. ${ }^{1}$ Increasing the temperature to $-10{ }^{\circ} \mathrm{C}$ led to Wittig rearrangement in case of cis-2.4c (see experimental section). Compound $\mathbf{2 . 7}$ was formed as a resultant of deprotonated trans-tetrahydrooxepins reluctance to rearrange. Subjecting compound
2.7 to butyllithium results in the formation of 2-arylcyclohexanone 2.8 with a concomitant loss of the silyl group (See experimental section).

### 2.5. Conclusion

In summary, we report the first ring contraction of 2-trimethylsilyl-2,5,6,7-tetrahydro-7-aryl-oxepins by Wittig rearrangements, which proceeds with modest diastereoselectivities via [1,2]and [1,4]-Wittig pathways to produce silyl cyclohexenols and cyclobutyl acyl silanes, respectively. An unexpected product resulting from alkene migration was also observed. Treatment of this unexpected product with butyllithium did not yield Wittig rearrangement products but resulted in the formation of 2-arylcyclohexan-1-one with a loss of the silyl group. Finally, unlike the cis dihydropyrans, the cis oxepins could only undergo rearrangement after increasing the reaction temperature.

### 2.6. Experimental section

### 2.6.1. General Information

Unless otherwise noted, all reactions were run under a positive atmosphere of nitrogen in oven-dried or flame-dried round-bottomed flasks or conical vials or disposable drum vials capped with rubber septa. Solvents were removed by rotary evaporation under reduced pressure at temperatures lower than $45^{\circ} \mathrm{C}$. Column chromatography was run on 230-400 mesh silica gel. Tetrahydrofuran and diethyl ether were distilled from sodium-benzophenone ketyl; dichloromethane, benzene, trimethylsilyl chloride were distilled from calcium hydride. Trimethylsilyltrifluoromethane sulfonate (TMSOTf) was redistilled and stored under nitrogen at $10{ }^{\circ} \mathrm{C}$ before the reaction. tert-butyllithium ( 1.7 M in pentane) and $\mathrm{BF}_{3} . \mathrm{OEt}_{2}$ were used as received. $n$-Butyllithum ( 2.5 M in hexanes) and sec-butyllithium (1.4 M in cyclohexane) were purchased from Aldrich and their concentration calculated by titration with diphenylacetic acid (average of
three runs). ${ }^{1} \mathrm{H}$ NMR spectra was collected in 500 MHz Varian instruments using $\mathrm{CDCl}_{3}$ as solvent, which was referenced at 7.26 ppm (residual chloroform proton) and ${ }^{13} \mathrm{C}$ NMR spectra was collected in $\mathrm{CDCl}_{3}$ at 126 MHz and referenced at 77.0 ppm . High-resolution mass spectrometric analysis was run in TOF instruments.

### 2.6.2. Preparation of 1-arylpent-4-en-1-ols (2.1a-2.1c) - general procedure $A$



To a 250 mL 3-neck round-bottomed flask fitted with a magnetic stir bar was weighed 2.2 g of magnesium powder ( $90 \mathrm{mmol}, 3.0$ equiv.) and 2 crystals of iodine. The side necks of the flask were sealed by two rubber septa and a reflux condenser attached to the middle neck then purged with nitrogen. An oil bath was placed underneath the flask. This was followed by addition of 60 mL dry THF and the resulting brown suspension was vigorously stirred. Homoallylic bromide (7.6 $\mathrm{mL}, 10.13 \mathrm{~g}, 75 \mathrm{mmol}, 2.5$ equiv.) was then added slowly. After complete addition of the bromide, the temperature of the oil bath had risen to $40^{\circ} \mathrm{C}$. The mixture was then further heated on an oil bath to $80^{\circ} \mathrm{C}$ (reflux) for 1 hour. The oil bath was removed, and the mixture was allowed to cool down to room temperature. The mixture was cooled down further to $0^{\circ} \mathrm{C}$ by placing an ice bath underneath the flask. This was followed by dropwise addition of appropriate aryl aldehyde (30 mmol, 1.0 equiv.) as a solution in 20 mL dry THF. The resulting mixture was stirred at $0^{\circ} \mathrm{C}$ to room temperature over a period of 2 hours. The mixture was then cooled to $0^{\circ} \mathrm{C}$ and quenched by slow addition of 20 mL saturated aqueous ammonium chloride solution. The mixture was diluted with 40 mL of diethyl ether and 20 mL of saturated aqueous ammonium chloride solution. The
resulting mixture was transferred into a 1000 mL separating funnel and the layers were separated. The aqueous layer was extracted with diethyl ether ( $50 \mathrm{~mL} \times 3$ ). The combined organic layers were washed with saturated aqueous ammonium chloride solution ( 50 mL ), water ( $50 \mathrm{~mL} \times 2$ ) and brine ( 50 mL ) respectively and then dried over anhydrous magnesium sulfate. This was followed by filtration and the filtrate was concentrated on the rotorvap under reduced pressure affording 1-arylpent-4-en-1-ol 2.1 which was used in the next step without need for further purification.

### 2.6.3. Preparation of trichloroacetimidates 2.2 - general procedure $\mathbf{B}$



Following our reported procedure with a slight modification, ${ }^{1}$ to a dry 250 mL roundbottomed flask fitted with a magnetic stir bar and sealed with a rubber septum was added 60 mL of dry dichloromethane under nitrogen. The desired 1-arylpent-4-en-1-ol $\mathbf{2 . 1}$ ( $20 \mathrm{mmol}, 1.00$ equiv) in dichloromethane ( 20 mL ) was then transferred into the flask. This was followed by addition of 0.54 mL of 1,8-Diazabicyclo(5.4.0)undec-7-ene (DBU) ( $0.55 \mathrm{~g}, 3.6 \mathrm{mmol}, 0.18$ equiv.). After stirring for 5 minutes, the solution was cooled to $0^{\circ} \mathrm{C}$ on an ice bath. This was followed by dropwise addition of 2.8 mL of tricchloroacetonitrile ( $4.04 \mathrm{~g}, 28 \mathrm{mmol}, 1.40$ equiv.). After 12 hours, the resulting dark brown mixture was filtered through a plug of silica ( 5 cm thick) to remove the dark residue. The filtrate was concentrated, and the crude mixture subjected to column chromatography (EtOAc/hexanes) to afford the desired trichloroacetimidate 2.2.

### 2.6.4. Alternative preparation of trichloroacetimidates 2.2 - general procedure $\mathbf{C}^{\mathbf{1 5}}$



Following our reported procedure, ${ }^{15} 240 \mathrm{mg}$ of sodium hydride $60 \% \mathrm{w} / \mathrm{w}$ dispersion in mineral oil ( $6 \mathrm{mmol}, 0.18$ equiv) was weighed into a dry 100 mL round-bottomed flask fitted with a magnetic stir bar and 20 mL of freshly distilled diethyl ether was added into the flask. The flask was sealed with a rubber septum and purged with nitrogen. The resulting grey suspension was cooled on an ice bath and the desired 1-arylpent-4-en-1-ol $\mathbf{2 . 1}$ ( $30 \mathrm{mmol}, 1.00$ equiv) in dry diethyl ether ( 20 mL ) was then transferred into the flask slowly resulting in a fizzy reaction. The mixture was stirred at $0{ }^{\circ} \mathrm{C}$ for 10 minutes. This was followed by dropwise addition of 4.2 mL of trichloroacetonitrile ( $6.06 \mathrm{~g}, 42 \mathrm{mmol}, 1.40$ equiv.). The mixture turned dark brown after complete addition of the trichloroacetonitrile. The mixture was stirred at $0^{\circ} \mathrm{C}$ for 20 minutes and then the ice bath was removed, and the mixture stirred at room temperature for 1 hour. The diethyl ether was then removed by rotorvap and methanol ( $0.25 \mathrm{~mL}, 6.0 \mathrm{mmol}, 0.18$ equiv.) in 15 mL pentane was added to the crude mixture. The mixture was further diluted with 40 mL pentane and filtered through a plug of celite ( 5 cm thick). The filtrate was concentrated, and the crude mixture subjected to column chromatography (EtOAc/hexanes) to afford the desired trichloroacetimidate 2.2.

### 2.6.5. Preparation of diastereomeric dienes 2.3 - general procedure $D^{15}$



A dry 250 mL round-bottomed flask with a magnetic stir bar was sealed with a rubber septum and purged with nitrogen. A solution of 1-(trimethylsilyl)prop-2-en-1-ol ( $10 \mathrm{mmol}, 1$ equiv) in 20 mL hexanes was transferred into the flask followed by a solution of the corresponding trichloroacetimidate 2.2 ( $15 \mathrm{mmol}, 1.5$ equiv.) in 20 mL hexanes. Additional 40 mL of hexanes was then added into the flask and the resulting mixture was cooled on ice bath to $0^{\circ} \mathrm{C}$ while stirring. To the cold solution was added appropriate Lewis acid: $\mathrm{BF}_{3} \cdot \mathrm{OEt}_{2}(0.12 \mathrm{~mL}, 1 \mathrm{mmol}, 0.1$ equiv.) or TMSOTf ( $0.18 \mathrm{~mL}, 1 \mathrm{mmol}, 0.1$ equiv.). After complete addition, a thick precipitate was formed. The mixture was stirred at $0^{\circ} \mathrm{C}$ to room temperature for 6 hours and filtered through a plug of celite ( 5 cm thick) and the filtrate was transferred into a separating funnel. The filtrate was then washed with saturated solution of aqueous sodium bicarbonate ( $50 \mathrm{~mL} \times 3$ ), water ( $50 \mathrm{~mL} \times 2$ ) and brine $(50 \mathrm{~mL})$ respectively. The organic layer was dried over anhydrous sodium sulfate and then filtered. The filtrate was concentrated under reduced pressure to afford diene $\mathbf{2 . 3}$ as a mixture of diastereomers. The resulting crude reaction mixture was purified by column chromatography (dichloromethane/hexanes). It is important to note that some diastereomers were separable by column chromatography and hence only one of them (syn) was taken to the next step (RCM). The diastereomers that could not be separated by column chromatography were taken to the next step as a mixture. The stereochemistry of these diastereomers were determined after the RCM reaction: syn diastereomers underwent RCM to form cis dihydropyrans or tetrahyrooxepins and vice versa.

It is also worth noting that for most of the compounds reported herein, the syn diastereomer exhibited lower $R_{f}$ value than its anti counterpart (dichloromethane/hexanes).

### 2.6.6. Preparation of tetrahydrooxepins 2.4 via ring closing metathesis (RCM) - general procedure $\mathbf{E}^{15}$



To a dry 250 mL round-bottomed flask with a magnetic stir bar was weighed 170 mg of Grubbs catalyst $2^{\text {nd }}$ generation ( $0.2 \mathrm{mmol}, 0.04$ equiv.) and the flask was sealed with a rubber septum and purged with nitrogen. This was followed by addition of 80 mL of dry dichloromethane and corresponding diene $\mathbf{2 . 3}$ ( 5 mmol , 1.0 equiv) as a solution in 20 mL dry dichloromethane as a single diastereomer (syn) or as a mixture of diastereomers (syn:anti=1:1). The resulting mixture was stirred at room temperature for 12 hours. The mixture was concentrated under reduced pressure to afford tetrahydrooxepins 2.4. The resulting crude reaction mixture was purified by column chromatography (dichloromethane/hexanes). The cis and trans diastereomers were separable by column chromatography. The stereochemistry of these diastereomers were determined by 1D NOESY experiment and confirmed with X-ray crystallography (trans-2.4d). It is also worth noting that for most of the compounds reported here the trans diastereomer has lower $R_{f}$ value than its cis counterpart (dichloromethane/hexanes).

## Synthesis of 1-phenylpent-4-en-1-ol (2.1a)



Applying general procedure A to magnesium powder ( $2.9 \mathrm{~g}, 120 \mathrm{mmol}, 1.2$ equiv), homoallylic bromide ( $12.2 \mathrm{~mL}, 120 \mathrm{mmol}, 1.2$ equiv), benzaldehyde $(9.65 \mathrm{~mL}, 100 \mathrm{mmol}, 1.0$ equiv), and THF ( 200 mL ) afforded $15.8 \mathrm{~g}, 97.4 \mathrm{mmol}(97 \%$ isolated yield) of compound 2.1a as a yellow oil after column chromatography, $R_{f}=0.5$ ( $20 \%$ ethyl acetate in hexanes): ${ }^{1} \mathrm{H}$ NMR ( 500 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.39-7.32(\mathrm{~m}, 4 \mathrm{H}), 7.29(\mathrm{dtd}, J=7.0,5.7,2.5 \mathrm{~Hz}, 1 \mathrm{H}), 5.85(\mathrm{ddt}, J=16.9,10.2$, $6.6 \mathrm{~Hz}, 1 \mathrm{H}), 5.05(\mathrm{dq}, J=17.1,1.7 \mathrm{~Hz}, 1 \mathrm{H}), 5.00(\mathrm{ddt}, J=10.2,2.2,1.3 \mathrm{~Hz}, 1 \mathrm{H}), 4.68(\mathrm{dd}, J=7.7$, $5.5 \mathrm{~Hz}, 1 \mathrm{H}), 2.24-2.01(\mathrm{~m}, 3 \mathrm{H}), 1.90(\mathrm{dddd}, J=13.7,8.8,7.7,6.0 \mathrm{~Hz}, 1 \mathrm{H}), 1.80(\mathrm{dddd}, J=13.6$, 9.2, 6.4, $5.5 \mathrm{~Hz}, 1 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 144.6,138.1,128.4,127.5,125.9,114.9,73.9$, 38.0, 30.0. IR (FTIR, film, $\mathrm{cm}^{-1}$ ) 3389, 2935, 1640, 1451, 909, 800. 2.1a is a known compound and spectroscopic data are in agreement with those reported in literature. ${ }^{17}$

## Synthesis of 1-(4-chlorophenyl)pent-4-en-1-ol (2.1b)



Following general procedure A, magnesium powder ( $2.9 \mathrm{~g}, 120 \mathrm{mmol}, 1.2$ equiv), 4-bromobut-1-ene ( $12.2 \mathrm{~mL}, 120 \mathrm{mmol}, 1.2$ equiv), 4-chlorobenzaldehyde ( $14.6 \mathrm{~g}, 100 \mathrm{mmol}, 1.0$
equiv.) and THF ( 220 mL ) afforded $18.81 \mathrm{~g}, 95.6 \mathrm{mmol}(96 \%$ isolated yield) of 1-(4-chlorophenyl)pent-4-en-1-ol 2.1b as a yellow oil after column chromatography, $R_{f}=0.4$ ( $20 \%$ ethyl acetate in hexanes). ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta=7.31(\mathrm{~d}, J=8.5 \mathrm{~Hz}, 2 \mathrm{H}), 7.26(\mathrm{~d}, J=8.5 \mathrm{~Hz}$, $2 \mathrm{H}), 5.82(\mathrm{ddt}, J=16.9,10.2,6.6 \mathrm{~Hz}, 1 \mathrm{H}), 5.03(\mathrm{dq}, J=17.1,1.7 \mathrm{~Hz}, 1 \mathrm{H}), 4.99(\mathrm{dq}, J=10.2,1.4$ $\mathrm{Hz}, 1 \mathrm{H}), 4.66(\mathrm{dd}, J=7.7,5.5 \mathrm{~Hz}, 1 \mathrm{H}), 2.19-2.01(\mathrm{~m}, 3 \mathrm{H}), 1.90-1.81(\mathrm{~m}, 1 \mathrm{H}), 1.75(\mathrm{dddd}, J=$ 13.9, 9.0, 6.5, $5.5 \mathrm{~Hz}, 1 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta=143.0,137.9,133.1,128.5,127.2$, 115.2, 73.3, 38.0, 29.9. IR (FTIR, film, $\mathrm{cm}^{-1}$ ) $\tilde{\mathrm{v}}=3326,2935,1640,1491,1090,1012,911,828$. 2.1b is a known compound and spectroscopic data are in agreement with those reported in literature. ${ }^{17}$

## Synthesis of 1-(4-methoxyphenyl)pent-4-en-1-ol (2.1c)



Following general procedure A, magnesium powder ( $2.2 \mathrm{~g}, 90 \mathrm{mmol}, 3.0$ equiv), 4-bromobut-1-ene ( $7.6 \mathrm{~mL}, 75 \mathrm{mmol}, 2.5$ equiv), 3.65 mL of $p$-anisaldehyde ( $4.08 \mathrm{~g}, 30 \mathrm{mmol}, 1.0$ equiv.) and THF ( 120 mL ) afforded $7.35 \mathrm{~g}, 38 \mathrm{mmol}$ (quantitative crude yield) of 1-(4-methoxyphenyl)pent-4-en-1-ol 2.1c as a yellow oil in THF. The product was taken to the next step without further purification. ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}, \mathrm{ppm}\right) \delta=7.26(\mathrm{~d}, J=8.5 \mathrm{~Hz}, 2 \mathrm{H}), 6.88$ $(\mathrm{d}, J=8.7 \mathrm{~Hz}, 2 \mathrm{H}), 5.84(\mathrm{ddt}, J=16.9,10.2,6.6 \mathrm{~Hz}, 1 \mathrm{H}), 5.03(\mathrm{dq}, J=17.1,1.7 \mathrm{~Hz}, 1 \mathrm{H}), 4.98$ (ddt, $J=10.2,2.2,1.3 \mathrm{~Hz}, 1 \mathrm{H}), 4.63(\mathrm{dd}, J=7.5,5.8 \mathrm{~Hz}, 1 \mathrm{H}), 3.80(\mathrm{~s}, 3 \mathrm{H}), 2.19-2.01(\mathrm{~m}, 3 \mathrm{H}), 1.89$ (dddd, $J=13.7,8.9,7.5,6.1 \mathrm{~Hz}, 1 \mathrm{H}), 1.77(\mathrm{ddt}, J=13.6,9.3,6.1 \mathrm{~Hz}, 1 \mathrm{H}) .{ }^{13} \mathrm{C}\left\{{ }^{1} \mathrm{H}\right\}-\mathrm{NMR}(126$
$\left.\mathrm{MHz}, \mathrm{CDCl}_{3}, \mathrm{ppm}\right) \delta=159.0,138.2,136.7,127.1,114.8,113.8,73.5,55.2,37.9,30.1 .2 .1 \mathrm{c}$ is a known compound and spectroscopic data are in agreement with those reported in literature. ${ }^{18}$

## Synthesis of 1-(naphthalen-2-yl)pent-4-en-1-ol (2.1d)



Following general procedure A, magnesium powder ( $2.33 \mathrm{~g}, 96 \mathrm{mmol}, 1.2$ equiv), 4-bromobut-1-ene ( $9.8 \mathrm{~mL}, 96 \mathrm{mmol}, 1.2$ equiv), 2-naphthaldehyde ( $12.96 \mathrm{~g}, 80 \mathrm{mmol}, 1.0$ equiv.) and THF ( 150 mL ) afforded $16.8 \mathrm{~g}, 79 \mathrm{mmol}$ ( $99 \%$ crude yield) of 1-(naphthalen-2-yl)pent-4-en-1-ol 2.2d as a yellow oil in THF. The product was taken to the next step without further purification. ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta=7.88-7.81(\mathrm{~m}, 3 \mathrm{H}), 7.78(\mathrm{~s}, 1 \mathrm{H}), 7.48(\mathrm{qd}, J=5.2,4.2,2.6 \mathrm{~Hz}$, $3 \mathrm{H}), 5.87(\mathrm{ddt}, J=16.9,10.1,6.6 \mathrm{~Hz}, 1 \mathrm{H}), 5.06(\mathrm{dq}, J=17.1,1.7 \mathrm{~Hz}, 1 \mathrm{H}), 5.01(\mathrm{dq}, J=10.2,1.5$ $\mathrm{Hz}, 1 \mathrm{H}), 4.91-4.82(\mathrm{~m}, 1 \mathrm{H}), 2.26-2.04(\mathrm{~m}, 3 \mathrm{H}), 2.04-1.95(\mathrm{~m}, 1 \mathrm{H}), 1.90(\mathrm{ddt}, J=13.5,9.1,6.0$ $\mathrm{Hz}, 1 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR $\left(126 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta=141.9,138.1,133.2,132.9,128.3,127.9,127.7,126.1$, 125.8, 124.6, 124.0, 115.0, 74.1, 37.9, 30.0. IR (FTIR, film, $\mathrm{cm}^{-1}$ ) $\tilde{\mathrm{v}}=3324,3055,2919,1638$, 1507, 1270, 1017, 817, 745. 2.2d is a known compound and spectroscopic data are in agreement with those reported in literature. ${ }^{18}$

## Synthesis of 1-phenylpent-4-en-1-yl 2,2,2-trichloroacetimidate (2.2a)



Applying general procedure B to 1-phenylpent-4-en-1-ol 2.1a (15.74 g, $97 \mathrm{mmol}, 1$ equiv), DBU ( $2.6 \mathrm{~mL}, 17.5 \mathrm{mmol}, 0.18$ equiv), and trichloroacetonitrile ( $13.6 \mathrm{~mL}, 135.8 \mathrm{mmol}, 1.4$ equiv), in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(180 \mathrm{~mL})$ afforded $27.5 \mathrm{~g}, 89.7 \mathrm{mmol}$ ( $92 \%$ crude yield) of compound 2.2a as a yellow liquid. ${ }^{1} \mathrm{H}$ NMR $\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 8.26(\mathrm{~s}, 1 \mathrm{H}), 7.43-7.39(\mathrm{~m}, 2 \mathrm{H}), 7.39-7.33(\mathrm{~m}, 2 \mathrm{H}), 7.33$ - $7.28(\mathrm{~m}, 1 \mathrm{H}), 5.90-5.78(\mathrm{~m}, 2 \mathrm{H}), 5.06(\mathrm{dq}, J=17.2,1.6 \mathrm{~Hz}, 1 \mathrm{H}), 5.02(\mathrm{dt}, J=10.1,1.6 \mathrm{~Hz}$, 1H), $2.30-2.12(\mathrm{~m}, 3 \mathrm{H}), 2.00-1.89(\mathrm{~m}, 1 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 161.5,140.2,137.4$, 128.4, 127.9, 126.1, 115.5, 91.7, 80.1, 36.1, 29.6. IR (FTIR, film, $\mathrm{cm}^{-1}$ ) 3341, 3065, 2943, 1661, 1284, 1071, 792.

## Synthesis of 1-(4-chlorophenyl)pent-4-en-1-yl 2,2,2-trichloroacetimidate (2.2b)



Following general procedure B, 1-(4-chlorophenyl)pent-4-en-1-ol 2.1b (15.74 g, 80 mmol , 1.0 equiv), DBU ( $2.2 \mathrm{~mL}, 14.4 \mathrm{mmol}, 0.18$ equiv), trichloroacetonitrile ( $11.2 \mathrm{~mL}, 112 \mathrm{mmol}, 1.4$ equiv.) and dichloromethane ( 200 mL ), $27.6 \mathrm{~g}, 80.9 \mathrm{mmol}$ ( $>99 \%$ isolated yield) of 1-(4-chlorophenyl)pent-4-en-1-yl 2,2,2-trichloroacetimidate $\mathbf{2 . 2 b}$ was obtained as a yellow oil after column chromatography, $R_{f}=0.4\left(10 \% \mathrm{EtOAc}\right.$ in hexanes). ${ }^{1} \mathrm{H} \mathrm{NMR}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta=8.27$
$(\mathrm{s}, 1 \mathrm{H}), 7.33(\mathrm{~s}, 4 \mathrm{H}), 5.88-5.76(\mathrm{~m}, 2 \mathrm{H}), 5.09-4.98(\mathrm{~m}, 2 \mathrm{H}), 2.28-2.08(\mathrm{~m}, 3 \mathrm{H}), 1.96-1.85(\mathrm{~m}$, 1H). ${ }^{13} \mathrm{C}$ NMR ( $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta=161.4,138.7,137.1,133.7,128.6,127.6,115.7,91.5,79.3$, 36.0, 29.5. IR (FTIR, film, $\mathrm{cm}^{-1}$ ) $\tilde{\mathrm{v}}=3340,2943,1662,1282,1069,791 . \mathrm{HRMS}(\mathrm{ESI}), \mathrm{m} / \mathrm{z}[\mathrm{M}-$ $\left.\mathrm{Cl}_{3} \mathrm{CONH}\right]^{+}$calcd for $\mathrm{C}_{11} \mathrm{H}_{12} \mathrm{Cl}$ : 179.0628; found: 179.0603.

Synthesis of 1-(4-methoxyphenyl)pent-4-en-1-yl 2,2,2-trichloroacetimidate (2.2c)


Following general procedure C, 1-(4-methoxyphenyl)pent-4-en-1-ol 2.1c ( $5.77 \mathrm{~g}, 30 \mathrm{mmol}$, 1.0 equiv), $\mathrm{NaH} 60 \% \mathrm{w} / \mathrm{w}$ dispersion in mineral oil ( $240 \mathrm{mg}, 6 \mathrm{mmol}, 0.20$ equiv), trichloroacetonitrile ( $4.2 \mathrm{~mL}, 42 \mathrm{mmol}, 1.4$ equiv.) and diethyl ether $(15 \mathrm{~mL}), 10.67 \mathrm{~g}, 31.7 \mathrm{mmol}$ (quantitative crude yield) of 1-(4-methoxyphenyl)pent-4-en-1-yl 2,2,2-trichloroacetimidate 2.2c which was used in the next step without further purification. ${ }^{1} \mathrm{H}$ NMR $\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta=8.25$ $(\mathrm{s}, 1 \mathrm{H}), 7.34(\mathrm{~d}, J=8.6 \mathrm{~Hz}, 2 \mathrm{H}), 6.90(\mathrm{~d}, J=8.8 \mathrm{~Hz}, 2 \mathrm{H}), 5.91-5.76(\mathrm{~m}, 2 \mathrm{H}), 5.09-4.99(\mathrm{~m}, 2 \mathrm{H})$, $3.81(\mathrm{~s}, 3 \mathrm{H}), 2.28-2.12(\mathrm{~m}, 3 \mathrm{H}), 1.97-1.86(\mathrm{~m}, 1 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $\left.126 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta=161.5$, $159.3,137.5,127.6,115.4,113.8,91.8,80.0,55.2,36.0,29.7 . \operatorname{IR}\left(\right.$ FTIR, $\left.\mathrm{cm}^{-1}\right) \tilde{\mathrm{v}}=3362,3239$, 3177, 1690, 1609, 1381, 1108, 829 MS (GC/MS): m/z (\%) = 175 (100) [ $\left.\mathrm{M}-\mathrm{Cl}_{3} \mathrm{CCONH}\right]^{+}, 121$ (80). HRMS (ESI), $m / z[\mathrm{M}+\mathrm{H}]^{+}$calcd for $\mathrm{C}_{14} \mathrm{H}_{17} \mathrm{Cl}_{3} \mathrm{NO}_{2}$ : 336.0325; found: 336.0333.

## Synthesis of 1-(naphthalen-2-yl)pent-4-en-1-yl 2,2,2-trichloroacetimidate (2.2d)



Following general procedure B, 1-(naphthalen-2-yl)pent-4-en-1-ol 2.1d (14.86 g, 70 mmol , 1.0 equiv), DBU ( $1.89 \mathrm{~mL}, 12.6 \mathrm{mmol}, 0.18$ equiv), trichloroacetonitrile ( $9.83 \mathrm{~mL}, 98 \mathrm{mmol}, 1.4$ equiv.) and dichloromethane ( 200 mL ), $25.89 \mathrm{~g}, 72.58 \mathrm{mmol}$ (quantitative crude yield) of 1-(naphthalen-2-yl)pent-4-en-1-yl 2,2,2-trichloroacetimidate 2.2d was obtained as a dark brown oil in dichloromethane. The product was used in the next step without further purification. ${ }^{1} \mathrm{H}$ NMR $\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}, \mathrm{ppm}\right) \delta=8.30(\mathrm{~s}, 1 \mathrm{H}), 7.96-7.80(\mathrm{~m}, 4 \mathrm{H}), 7.56(\mathrm{dd}, J=8.5,1.8 \mathrm{~Hz}, 1 \mathrm{H}), 7.50$ (qt, $J=7.2,3.9 \mathrm{~Hz}, 2 \mathrm{H}), 6.03(\mathrm{dd}, J=7.9,5.0 \mathrm{~Hz}, 1 \mathrm{H}), 5.89(\mathrm{ddt}, J=16.4,10.0,6.2 \mathrm{~Hz}, 1 \mathrm{H}), 5.14$ $-5.00(\mathrm{~m}, 2 \mathrm{H}), 2.36-2.19(\mathrm{~m}, 3 \mathrm{H}), 2.05(\mathrm{tdd}, J=10.9,7.4,3.9 \mathrm{~Hz}, 1 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( 126 MHz , $\left.\mathrm{CDCl}_{3}, \mathrm{ppm}\right) \delta=161.5,137.5,137.3,133.1,128.3,128.0,127.7,126.2,126.0,125.4,123.9,115.5$, 91.7, 80.2, 36.0, 29.7. IR (FTIR, $\left.\mathrm{cm}^{-1}\right) ~ \tilde{v}=3337,3057,2941,1661,1284,1071,986,791 \mathrm{MS}$ (GC/MS): m/z (\%) = 355 (0.1) [ $\left.{ }^{+}\right], 301$ (15), 184 (70), 156 (45), 141 (100), 115 (48). HRMS (ESI), $m / z[\mathrm{M}+\mathrm{H}]^{+}$calcd for $\mathrm{C}_{17} \mathrm{H}_{17} \mathrm{Cl}_{3} \mathrm{NO}: 356.0376$; found: 356.0370.

## Synthesis of syn/anti-(1-((1-phenylpent-4-en-1-yl)oxy)allyl)trimethylsilane (syn/anti-2.3a)



Applying general procedure D to 1-(trimethylsilyl)prop-2-en-1-ol (5.2 g, $40 \mathrm{mmol}, 1.0$ equiv), trichloroacetimidate $\mathbf{2 . 2 a}(17.7 \mathrm{~g}, 60 \mathrm{mmol}, 1.5$ equiv), and boron trifluoro diethyl etherate ( $0.5 \mathrm{~mL}, 4.0 \mathrm{mmol}, 0.1$ equiv) in hexane ( 200 mL ) afforded after column chromatography ( $5 \%$ $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ in hexanes) a total of $8.04 \mathrm{~g}, 29.3 \mathrm{mmol}(73 \%$ isolated yield) of syn/anti-2.3a (1:1) as a colorless oil. Compounds syn-2.3a and anti-2.3a were partially separable by column chromatography.

Spectroscopic data for syn-2.3a: ${ }^{1} \mathrm{H}$ NMR $\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.36-7.30(\mathrm{~m}, 4 \mathrm{H}), 7.25$ (ddd, $J=8.7,5.0,3.8 \mathrm{~Hz}, 1 \mathrm{H}), 5.82$ (ddt, $J=16.9,10.2,6.6 \mathrm{~Hz}, 1 \mathrm{H}), 5.67(\mathrm{ddd}, J=17.4,10.6,7.1$ $\mathrm{Hz}, 1 \mathrm{H}), 5.00(\mathrm{dq}, J=17.2,1.7 \mathrm{~Hz}, 1 \mathrm{H}), 4.98-4.89(\mathrm{~m}, 2 \mathrm{H}), 4.83(\mathrm{dt}, J=10.6,1.8 \mathrm{~Hz}, 1 \mathrm{H}), 4.36$ $(\mathrm{t}, J=6.0 \mathrm{~Hz}, 1 \mathrm{H}), 3.78(\mathrm{dt}, J=7.2,1.5 \mathrm{~Hz}, 1 \mathrm{H}), 2.09-1.97(\mathrm{~m}, 2 \mathrm{H}), 1.88(\mathrm{ddt}, J=12.8,9.3,6.2$ $\mathrm{Hz}, 1 \mathrm{H}), 1.80-1.71(\mathrm{~m}, 1 \mathrm{H}), 0.09(\mathrm{~s}, 9 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $126 \mathrm{MHz}, \mathrm{CDCl}_{3}, \mathrm{ppm}$ ) $\delta$ 143.9, 138.7, $138.0,127.9,126.8,126.6,114.4,111.7,80.9,75.7,36.2,29.3,-3.7$.

Spectroscopic data for anti-2.3a: ${ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.37$ - $7.32(\mathrm{~m}, 3 \mathrm{H}), 7.30$ $-7.26(\mathrm{~m}, 2 \mathrm{H}), 5.85(\mathrm{ddt}, J=16.9,10.3,6.6 \mathrm{~Hz}, 1 \mathrm{H}), 5.77(\mathrm{ddd}, J=17.4,10.6,7.7 \mathrm{~Hz}, 1 \mathrm{H}), 5.11$ $-5.03(\mathrm{~m}, 2 \mathrm{H}), 5.01-4.94(\mathrm{~m}, 2 \mathrm{H}), 4.43(\mathrm{dd}, J=8.2,5.3 \mathrm{~Hz}, 1 \mathrm{H}), 3.43(\mathrm{dt}, J=7.7,1.3 \mathrm{~Hz}, 1 \mathrm{H})$, 2.21 (dddt, $J=10.2,5.1,2.2,1.2 \mathrm{~Hz}, 1 \mathrm{H}), 2.08(\mathrm{dddd}, J=16.4,8.3,4.9,2.0 \mathrm{~Hz}, 1 \mathrm{H}), 1.93-1.85$
$(\mathrm{m}, 1 \mathrm{H}), 1.68(\mathrm{dddd}, J=13.6,9.6,6.1,5.3 \mathrm{~Hz}, 1 \mathrm{H}), 0.00(\mathrm{~s}, 9 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$, ppm) $\delta 143.0,138.8,137.8,128.4,128.1,127.3,114.3,113.0,78.8,72.9,37.7,30.3,-4.0$.

Synthesis of syn/anti-(1-((1-(4-chlorophenyl)pent-4-en-1-yl)oxy)allyl)trimethylsilane (syn/anti-2.3b)


Compound 2.3b was prepared following general procedure D , a solution of 2-(trimethylsilyl)prop-2-en-1-ol ( $3.91 \mathrm{~g}, 30 \mathrm{mmol}, 1$ equiv.) and 1-(4-chlorophenyl)pent-4-en-1-yl 2,2,2-trichloroacetimidate $\mathbf{2 . 2 b}$ ( $15.35 \mathrm{~g}, 45 \mathrm{mmol}, 1.5$ equiv. $), \mathrm{BF}_{3} \cdot \mathrm{OEt}_{2}(0.37 \mathrm{~mL}, 3.0 \mathrm{mmol}, 0.1$ equiv.) and hexanes ( 165 mL ) for 12 hours followed by workup, concentration and column chromatography, $R_{f}$ for $a n t i-\mathbf{2 . 3 b}=0.6$ and $R_{f}$ for syn-2.3b $=0.4(2 \%$ DCM in hexanes $)$ afforded a total of $5.40 \mathrm{~g}, 17.5 \mathrm{mmol}$ ( $58 \%$ isolated yield) partially separable mixture of diastereomers of compound 2.3 as colorless liquid.

Spectroscopic data for syn-2.3b: ${ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta=7.27(\mathrm{~d}, J=8.5 \mathrm{~Hz}, 2 \mathrm{H})$, $7.22(\mathrm{~d}, J=8.5 \mathrm{~Hz}, 2 \mathrm{H}), 5.78(\mathrm{ddt}, J=16.9,10.2,6.6 \mathrm{~Hz}, 1 \mathrm{H}), 5.62(\mathrm{ddd}, J=17.6,10.5,7.2 \mathrm{~Hz}$, $1 \mathrm{H}), 5.01-4.92(\mathrm{~m}, 2 \mathrm{H}), 4.88(\mathrm{dt}, J=17.2,1.8 \mathrm{~Hz}, 1 \mathrm{H}), 4.82(\mathrm{dt}, J=10.5,1.6 \mathrm{~Hz}, 1 \mathrm{H}), 4.31(\mathrm{t}, J$ $=6.1 \mathrm{~Hz}, 1 \mathrm{H}), 3.74(\mathrm{dt}, J=7.3,1.5 \mathrm{~Hz}, 1 \mathrm{H}), 2.06-1.94(\mathrm{~m}, 2 \mathrm{H}), 1.83(\mathrm{ddt}, J=13.6,9.6,6.0 \mathrm{~Hz}$, $1 \mathrm{H}), 1.70(\mathrm{ddt}, J=13.7,9.6,6.1 \mathrm{~Hz}, 1 \mathrm{H}), 0.06(\mathrm{~s}, 9 \mathrm{H}) .{ }^{13} \mathrm{C} \mathrm{NMR}\left(126 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta=142.5$, $138.4,137.8,132.4,128.1,127.9,114.6,111.9,80.3,76.1,36.1,29.2,-3.8 . \operatorname{IR}\left(\right.$ FTIR, $\left.\mathrm{cm}^{-1}\right) \tilde{v}=$

3078, 2954, 2926, 1640, 1627, 1490, 1246, 839. MS (GC/MS): m/z (\%) = 179 (17.5) [M $\mathrm{C}_{6} \mathrm{H}_{13} \mathrm{OSi}^{+}, 125$ (100).

Spectroscopic data for anti-2.3b: ${ }^{1} \mathrm{H} \operatorname{NMR}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}, \mathrm{ppm}\right) \delta=7.30(\mathrm{~d}, J=8.4 \mathrm{~Hz}$, $2 \mathrm{H}), 7.18(\mathrm{~d}, J=8.3 \mathrm{~Hz}, 2 \mathrm{H}), 5.85-5.78(\mathrm{~m}, 1 \mathrm{H}), 5.78-5.70(\mathrm{~m}, 1 \mathrm{H}), 5.06-4.98(\mathrm{~m}, 2 \mathrm{H}), 4.98$ - $4.92(\mathrm{~m}, 2 \mathrm{H}), 4.39(\mathrm{dd}, J=8.1,5.4 \mathrm{~Hz}, 1 \mathrm{H}), 3.37(\mathrm{dt}, J=7.8,1.3 \mathrm{~Hz}, 1 \mathrm{H}), 2.22-2.12(\mathrm{~m}, 1 \mathrm{H})$, $2.09-2.00(\mathrm{~m}, 1 \mathrm{H}), 1.90-1.79(\mathrm{~m}, 2 \mathrm{H}), 1.63(\mathrm{ddt}, J=13.5,9.5,5.8 \mathrm{~Hz}, 1 \mathrm{H}),-0.01(\mathrm{~s}, 9 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR (126 MHz, $\left.\mathrm{CDCl}_{3}, \mathrm{ppm}\right) \delta=141.5,138.5,137.5,128.6,128.3,127.2,114.5,113.2,78.1$, 73.1, 37.6, 30.1, -4.0. IR (FTIR, $\left.\mathrm{cm}^{-1}\right) \tilde{\mathrm{v}}=3105,2949,1641,1592,1490,1089,1013,822$. HRMS (APCI), $m / z[\mathrm{M}+\mathrm{H}]^{+}$calcd for $\mathrm{C}_{17} \mathrm{H}_{26} \mathrm{ClOSi}$ : 309.1441; found: 309.1428.

Synthesis of syn/anti-(1-((1-(4-methoxyphenyl)pent-4-en-1-yl)oxy)allyl)trimethylsilane (syn/anti-2.3c)


Compound 2.3c was prepared following general procedure D with slight modification to minimize formation of the side product as a result of elimination. A solution of 1-(trimethylsilyl)prop-2-en-1-ol (1.95 g, $15 \mathrm{mmol}, 1$ equiv.) and 1-(4-methoxyphenyl)pent-4-en-1-yl 2,2,2-trichloroacetimidate $2.2 \mathrm{c}(5.05 \mathrm{~g}, 15 \mathrm{mmol}, 1$ equiv.) in dichloromethane ( 120 mL ) was cooled to $-78^{\circ} \mathrm{C}$. TMSOTf ( $0.27 \mathrm{~mL}, 1.5 \mathrm{mmol}, 0.1$ equiv.) was added dropwise and the mixture stirred at $-78^{\circ} \mathrm{C}$ for 6 hours. The rubber septum was removed, and 7 g of sodium bicarbonate was
poured into the flask. The dry ice-acetone bath was removed, and the mixture was allowed to warm up to room temperature. The mixture was filtered and concentrated under reduced pressure to remove dichloromethane. Hexanes was then added to the resulting mixture resulting in the formation of white precipitate. Subsequent filtration and concentration furnished a residue which was purified by column chromatography, $R_{f}$ for anti-2.3c $=0.5$ and $R_{f}$ for $\operatorname{syn} \mathbf{- 2 . 3} \mathbf{c}=0.3(10 \% \mathrm{DCM}$ in hexanes) to afford a total of $3.02 \mathrm{~g}, 9.9 \mathrm{mmol}$ ( $66 \%$ isolated yield) partially separable mixture of diastereomers of compound 2.3c (syn:anti =1:1) as colorless liquid.

Spectroscopic data for syn-2.3c: ${ }^{1} \mathrm{H}$ NMR $\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}, \mathrm{ppm}\right) \delta=7.21(\mathrm{~d}, J=8.5 \mathrm{~Hz}$, $2 \mathrm{H}), 6.84(\mathrm{~d}, J=8.6 \mathrm{~Hz}, 2 \mathrm{H}), 5.80(\mathrm{ddt}, J=16.8,10.2,6.5 \mathrm{~Hz}, 1 \mathrm{H}), 5.65(\mathrm{ddd}, J=17.4,10.5,7.0$ $\mathrm{Hz}, 1 \mathrm{H}), 4.98(\mathrm{dq}, J=17.2,1.7 \mathrm{~Hz}, 1 \mathrm{H}), 4.96-4.87(\mathrm{~m}, 2 \mathrm{H}), 4.81(\mathrm{ddd}, J=10.6,2.1,1.5 \mathrm{~Hz}, 1 \mathrm{H})$, $4.28(\mathrm{t}, J=6.2 \mathrm{~Hz}, 1 \mathrm{H}), 3.80(\mathrm{~s}, 3 \mathrm{H}), 3.74(\mathrm{dt}, J=7.1,1.6 \mathrm{~Hz}, 1 \mathrm{H}), 2.05-1.96(\mathrm{~m}, 2 \mathrm{H}), 1.91-$ $1.81(\mathrm{~m}, 1 \mathrm{H}), 1.76-1.66(\mathrm{~m}, 1 \mathrm{H}), 0.06(\mathrm{~s}, 9 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR $\left(126 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta=158.5,138.7$, 138.1, 136.0, 127.7, 114.3, 113.2, 111.4, 80.6, 75.5, 55.2, 36.2, 29.4, -3.7. HRMS (ESI), $m / z[\mathrm{M}-$ $\mathrm{H}]^{-}$calcd for $\mathrm{C}_{18} \mathrm{H}_{27} \mathrm{O}_{2} \mathrm{Si}$ : 303.1780; found: 303.1789.

Spectroscopic data for anti-2.3c: ${ }^{1} \mathrm{H} \operatorname{NMR}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta=7.17(\mathrm{~d}, J=8.7 \mathrm{~Hz}, 2 \mathrm{H})$, $6.87(\mathrm{~d}, J=8.6 \mathrm{~Hz}, 2 \mathrm{H}), 5.84(\mathrm{ddt}, J=16.9,10.2,6.5 \mathrm{~Hz}, 1 \mathrm{H}), 5.75(\mathrm{ddd}, J=17.2,10.5,7.6 \mathrm{~Hz}$, $1 \mathrm{H}), 5.06-4.91(\mathrm{~m}, 4 \mathrm{H}), 4.36(\mathrm{dd}, J=8.0,5.5 \mathrm{~Hz}, 1 \mathrm{H}), 3.82(\mathrm{~s}, 3 \mathrm{H}), 3.41(\mathrm{dt}, J=7.6,1.4 \mathrm{~Hz}, 1 \mathrm{H})$, $2.23-2.13(\mathrm{~m}, 1 \mathrm{H}), 2.10-2.00(\mathrm{~m}, 1 \mathrm{H}), 1.88(\mathrm{dddd}, J=13.5,9.3,8.0,5.6 \mathrm{~Hz}, 1 \mathrm{H}), 1.65(\mathrm{ddt}, J=$ $13.5,9.6,5.8 \mathrm{~Hz}, 1 \mathrm{H}),-0.02(\mathrm{~s}, 9 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta=158.8,138.8,137.9,135.0$, $128.4,114.3,113.5,112.8,78.3,72.5,55.2,37.7,30.3,-4.0 . \operatorname{HRMS}(\mathrm{APCI}), m / z[\mathrm{M}+\mathrm{H}]^{+}$calcd for $\mathrm{C}_{18} \mathrm{H}_{29} \mathrm{O}_{2} \mathrm{Si}: 305.1937$; found: 305.1928 .

Synthesis of syn/anti-trimethyl(1-((1-(naphthalen-2-yl)pent-4-en-1-yl)oxy)allyl)silane (syn/anti-2.3d)


Compound 2.3d was prepared following general procedure D , a solution of 1-(trimethylsilyl)prop-2-en-1-ol (1.30 g, $10 \mathrm{mmol}, 1$ equiv.) and 1-(naphthalen-2-yl)pent-4-en-1-yl 2,2,2-trichloroacetimidate $\mathbf{2 . 2 d}\left(5.36 \mathrm{~g}, 15 \mathrm{mmol}, 1.5\right.$ equiv.), $\mathrm{BF}_{3} \cdot \mathrm{OEt}_{2}(0.13 \mathrm{~mL}, 1.0 \mathrm{mmol}, 0.1$ equiv.) and hexanes ( 80 mL ) for 12 hour followed by workup, concentration and column chromatography, $R_{f}$ for anti/syn-2.3d $=0.5(10 \% \mathrm{DCM}$ in hexanes $)$ afforded a total of $2.19 \mathrm{~g}, 6.8$ mmol ( $68 \%$ isolated yield) partially separable mixture of diastereomers of compound $\mathbf{2 . 3 d}$ as colorless liquid.

Spectroscopic data for syn/anti-2.3d: ${ }^{1} \mathrm{H} \operatorname{NMR}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta=7.89-7.80(\mathrm{~m}, 6 \mathrm{H})$, $7.75(\mathrm{~s}, 1 \mathrm{H}), 7.70(\mathrm{~s}, 1 \mathrm{H}), 7.54-7.45(\mathrm{~m}, 6 \mathrm{H}), 5.93-5.78(\mathrm{~m}, 3 \mathrm{H}), 5.69(\mathrm{ddd}, J=17.5,10.5,7.1$ $\mathrm{Hz}, 1 \mathrm{H}), 5.10(\mathrm{ddd}, J=10.6,2.0,1.2 \mathrm{~Hz}, 1 \mathrm{H}), 5.08-4.93(\mathrm{~m}, 6 \mathrm{H}), 4.82(\mathrm{ddd}, J=10.6,2.1,1.5 \mathrm{~Hz}$, $1 \mathrm{H}), 4.62(\mathrm{dd}, J=8.0,5.5 \mathrm{~Hz}, 1 \mathrm{H}), 4.54(\mathrm{t}, J=6.1 \mathrm{~Hz}, 1 \mathrm{H}), 3.86(\mathrm{dt}, J=7.1,1.5 \mathrm{~Hz}, 1 \mathrm{H}), 3.49(\mathrm{dt}$, $J=7.6,1.4 \mathrm{~Hz}, 1 \mathrm{H}), 2.30-2.20(\mathrm{~m}, 1 \mathrm{H}), 2.17-2.06(\mathrm{~m}, 3 \mathrm{H}), 2.06-1.95(\mathrm{~m}, 2 \mathrm{H}), 1.90-1.82(\mathrm{~m}$, $1 \mathrm{H}), 1.79(\mathrm{ddt}, J=11.6,9.6,3.8 \mathrm{~Hz}, 1 \mathrm{H}), 0.13(\mathrm{~s}, 9 \mathrm{H}), 0.04(\mathrm{~s}, 9 \mathrm{H}) .{ }^{13} \mathrm{C} \mathrm{NMR}\left(126 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ $=141.4,140.4,138.7,138.0,137.8,133.1,133.0,132.8,128.1,127.9,127.8,127.7,127.6,126.4$, $125.9,125.8,125.6,125.4,125.3,125.1,124.9,114.5,114.4,113.0,111.7,81.2,79.0,76.0,72.9$,
37.5, 36.2, 30.3, 29.4, -3.7, -4.0.

Spectroscopic data for syn-2.3d only: ${ }^{1} \mathrm{H}$ NMR $\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta=7.86-7.78(\mathrm{~m}, 3 \mathrm{H})$, 7.73 (s, 1H), $7.51-7.43(\mathrm{~m}, 3 \mathrm{H}), 5.83(\mathrm{ddt}, J=16.8,10.1,6.5 \mathrm{~Hz}, 1 \mathrm{H}), 5.67(\mathrm{ddd}, J=17.4,10.5$, $7.1 \mathrm{~Hz}, 1 \mathrm{H}), 5.00(\mathrm{dq}, J=17.1,1.7 \mathrm{~Hz}, 1 \mathrm{H}), 4.98-4.90(\mathrm{~m}, 2 \mathrm{H}), 4.80(\mathrm{dt}, J=10.6,1.8 \mathrm{~Hz}, 1 \mathrm{H})$, $4.52(\mathrm{t}, J=6.1 \mathrm{~Hz}, 1 \mathrm{H}), 3.83(\mathrm{dt}, J=7.1,1.5 \mathrm{~Hz}, 1 \mathrm{H}), 2.11-2.02(\mathrm{~m}, 2 \mathrm{H}), 1.96(\mathrm{ddt}, J=12.7,9.1$, $6.2 \mathrm{~Hz}, 1 \mathrm{H}), 1.83(\mathrm{ddt}, J=13.4,9.2,6.3 \mathrm{~Hz}, 1 \mathrm{H}), 0.11(\mathrm{~s}, 9 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR $\left(126 \mathrm{MHz}, \mathrm{CDCl}_{3}, \mathrm{ppm}\right) \delta$ $=141.4,138.6,138.0,133.1,132.8,127.9,127.63,127.62,125.8,125.4,125.3,124.9,114.5,111.7$, 81.2, 76.0, 36.2, 29.4, -3.7. HRMS (ESI), $m / z[\mathrm{M}+\mathrm{H}]^{+}$calcd for $\mathrm{C}_{21} \mathrm{H}_{29} \mathrm{OSi}$ : 325.1988; found: 325.2025.

## Synthesis of trans-(7-phenyl-2,5,6,7-tetrahydrooxepin-2-yl)trimethylsilane (trans-2.4a)



Applying general procedure E to syn-2.3a ( $4.1 \mathrm{~g}, 15 \mathrm{mmol}, 1$ equiv) and second-generation Grubbs catalyst ( $127 \mathrm{mg}, 0.15 \mathrm{mmol}, 0.01$ equiv) in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ ( 200 mL ) afforded after column chromatography ( $40 \% \mathrm{CH}_{2} \mathrm{Cl}_{2}$ in hexanes) $3.41 \mathrm{~g}, 13.8 \mathrm{mmol}$ ( $92 \%$ isolated yield) of trans-2.4a as a colorless oil. ${ }^{1} \mathrm{H}$ NMR $\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.46-7.40(\mathrm{~m}, 2 \mathrm{H}), 7.37-7.32(\mathrm{~m}, 2 \mathrm{H}), 7.30-$ $7.25(\mathrm{~m}, 1 \mathrm{H}), 5.78-5.72(\mathrm{~m}, 1 \mathrm{H}), 5.60(\mathrm{dt}, J=11.3,2.8 \mathrm{~Hz}, 1 \mathrm{H}), 4.87(\mathrm{dd}, J=10.9,4.3 \mathrm{~Hz}, 1 \mathrm{H})$, $4.04(\mathrm{dtd}, J=4.5,3.1,1.8 \mathrm{~Hz}, 1 \mathrm{H}), 2.58-2.48(\mathrm{~m}, 1 \mathrm{H}), 2.41-2.28(\mathrm{~m}, 2 \mathrm{H}), 2.18-2.10(\mathrm{~m}, 1 \mathrm{H})$, $-0.00(\mathrm{~s}, 9 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR (126 MHz, $\left.\mathrm{CDCl}_{3}\right) \delta 142.6,130.8,128.6,128.0,127.2,127.0,80.3,68.6$, 33.6, 26.5, -3.1.

Synthesis of cis/trans- (7-(4-chlorophenyl)-2,5,6,7-tetrahydrooxepin-2-yl)trimethylsilane (cis/trans-2.4b)


Compound 2.4b was prepared following general procedure $E$ : Grubbs catalyst $2^{\text {nd }}$ generation (191 mg, $0.225 \mathrm{mmol}, 0.015$ equiv.) and syn/anti-(1-((1-(4-chlorophenyl)pent-4-en-1yl)oxy)allyl)trimethylsilane, syn:anti $=1: 1, \mathbf{2 . 4 b}$ ( $4.64 \mathrm{~g}, 15 \mathrm{mmol}$, 1 equiv.) and dichloromethane $(120 \mathrm{~mL})$ at $40^{\circ} \mathrm{C}$ for 12 hours followed by concentration and column chromatography, $R_{f}$ for cis$\mathbf{2 . 4 b}=0.7$ and $R_{f}$ for trans-2.4b $=0.4(20 \%$ DCM in hexanes) afforded a total of $3.09 \mathrm{~g}, 11 \mathrm{mmol}$ (73\% isolated yield) fully separable mixture of diastereomers of compound $\mathbf{2 . 4 b}$ as colorless liquid.

Spectroscopic data for cis-2.4b: ${ }^{1} \mathrm{H}$ NMR ( $\left.500 \mathrm{MHz}, \mathrm{CDCl}_{3}, \mathrm{ppm}\right) \delta=7.28(\mathrm{~s}, 4 \mathrm{H}), 5.69$ (dddd, $J=11.9,7.3,4.7,2.5 \mathrm{~Hz}, 1 \mathrm{H}), 5.54(\mathrm{dt}, J=11.2,2.6 \mathrm{~Hz}, 1 \mathrm{H}), 4.73(\mathrm{dd}, J=8.4,5.4 \mathrm{~Hz}, 1 \mathrm{H})$, $4.05(\mathrm{dt}, J=4.4,3.3 \mathrm{~Hz}, 1 \mathrm{H}), 2.62(\mathrm{dddd}, J=16.2,8.1,3.7,1.4 \mathrm{~Hz}, 1 \mathrm{H}), 2.32(\mathrm{ddt}, J=13.7,10.5$, $5.2 \mathrm{~Hz}, 1 \mathrm{H}), 2.02-1.93(\mathrm{~m}, 1 \mathrm{H}), 1.83(\mathrm{ddt}, J=13.6,8.4,5.1 \mathrm{~Hz}, 1 \mathrm{H}), 0.11(\mathrm{~s}, 9 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR (126 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta=143.7,132.0,130.6,128.1,126.6,126.1,80.8,77.3,37.4,23.5,-3.7$. IR (FTIR, film, $\left.\mathrm{cm}^{-1}\right) \tilde{\mathrm{v}}=3014,2932,2780,1719,1490,1246,1089,937 . \operatorname{MS}(\mathrm{GC} / \mathrm{MS}): m / \mathrm{z}(\%)=280(0.2)$ $[\mathrm{M}]^{+}, 245$ (0.5), 142 (95), 127 (30), 73 (100). HRMS (ESI), $m / z\left[\mathrm{M}-\mathrm{H}^{-}\right]^{+}$calcd for $\mathrm{C}_{15} \mathrm{H}_{20} \mathrm{ClOSi}$ : 279.0972; found: 279.0959.

Spectroscopic data for trans-2.4b: ${ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta=7.35(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 2 \mathrm{H})$, $7.31(\mathrm{~d}, J=8.6 \mathrm{~Hz}, 2 \mathrm{H}), 5.73(\mathrm{ddt}, J=10.1,7.3,2.4 \mathrm{~Hz}, 1 \mathrm{H}), 5.58(\mathrm{dt}, J=11.2,2.8 \mathrm{~Hz}, 1 \mathrm{H}), 4.82$ (dd, $J=11.0,4.4 \mathrm{~Hz}, 1 \mathrm{H}), 3.98(\mathrm{dtd}, J=4.4,2.8,1.6 \mathrm{~Hz}, 1 \mathrm{H}), 2.50(\mathrm{dddt}, J=16.2,9.3,4.0,2.5 \mathrm{~Hz}$,
$1 \mathrm{H}), 2.35-2.24(\mathrm{~m}, 2 \mathrm{H}), 2.10(\mathrm{ddt}, J=13.2,6.5,2.9 \mathrm{~Hz}, 1 \mathrm{H}),-0.01(\mathrm{~s}, 9 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( 126 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta=141.2,132.9,130.7,128.5,128.4,128.2,79.6,68.7,33.7,26.4,-3.1 . \operatorname{HRMS}(\mathrm{ESI}), m / z$ $[\mathrm{M}+\mathrm{H}]^{+}$calcd for $\mathrm{C}_{15} \mathrm{H}_{22} \mathrm{ClOSi}$ : 281.1129; found: 281.1107.

## Synthesis of cis-(7-(4-methoxyphenyl)-2,5,6,7-tetrahydrooxepin-2-yl)trimethylsilane (cis-2.4c)



Compound cis-2.4c was prepared following general procedure E : Grubbs catalyst $2^{\text {nd }}$ generation ( $119 \mathrm{mg}, 0.14 \mathrm{mmol}, 0.02$ equiv.) and anti-(1-((1-(4-methoxyphenyl)pent-4-en-1yl)oxy)allyl)trimethylsilane anti-2.3c ( $2132 \mathrm{mg}, 7.0 \mathrm{mmol}$, 1 equiv.) and benzene ( 100 mL ) at 80 ${ }^{\circ} \mathrm{C}$ for 4 hours followed by concentration and column chromatography, $R_{f}$ for cis-2.4c $=0.6(40 \%$ DCM in hexanes) afforded a total of $1786 \mathrm{mg}, 6.5 \mathrm{mmol}(92 \%$ isolated yield) of compound cis$\mathbf{2 . 4} \mathbf{c}$ as a colorless liquid.

Spectroscopic data for cis-2.4c: ${ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta=7.29(\mathrm{~d}, J=8.8 \mathrm{~Hz}, 2 \mathrm{H})$, $6.88(\mathrm{~d}, J=8.8 \mathrm{~Hz}, 2 \mathrm{H}), 5.71(\mathrm{dddd}, J=11.3,7.2,4.7,2.5 \mathrm{~Hz}, 1 \mathrm{H}), 5.56(\mathrm{ddd}, J=11.2,3.2,2.1$ $\mathrm{Hz}, 1 \mathrm{H}), 4.74(\mathrm{dd}, J=8.4,5.3 \mathrm{~Hz}, 1 \mathrm{H}), 4.07(\mathrm{qd}, J=3.3,1.2 \mathrm{~Hz}, 1 \mathrm{H}), 3.81(\mathrm{~s}, 3 \mathrm{H}), 2.72-2.60(\mathrm{~m}$, $1 \mathrm{H}), 2.33$ (ddt, $J=13.7,10.4,5.1 \mathrm{~Hz}, 1 \mathrm{H}), 2.00(\mathrm{ddtd}, J=15.3,7.4,4.9,1.2 \mathrm{~Hz}, 1 \mathrm{H}), 1.89$ (ddt, $J$ $=13.6,8.4,5.1 \mathrm{~Hz}, 1 \mathrm{H}), 0.13(\mathrm{~s}, 9 \mathrm{H}) .{ }^{13} \mathrm{C} \operatorname{NMR}\left(126 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta=158.2,137.5,130.7,126.3$, 126.2, 113.4, 81.2, 77.1, 55.2, 37.4, 23.6, -3.7. HRMS (ESI), $m / z[\mathrm{M}-\mathrm{OH}]^{+}$calcd for $\mathrm{C}_{16} \mathrm{H}_{23} \mathrm{OSi}$ : 259.1518; found: 259.1513 .

Synthesis of trans-(7-(4-methoxyphenyl)-2,5,6,7-tetrahydrooxepin-2-yl)trimethylsilane (trans-2.4c)


Compound trans-2.4c was prepared following general procedure E: Grubbs catalyst $2^{\text {nd }}$ generation ( $81 \mathrm{mg}, 0.095 \mathrm{mmol}, 0.02$ equiv.) and syn-(1-((1-(4-methoxyphenyl)pent-4-en-1yl)oxy)allyl)trimethylsilane $\operatorname{syn}-\mathbf{2 . 4 c}\left(1.45 \mathrm{~g}, 4.75 \mathrm{mmol}\right.$, 1 equiv.) and benzene ( 80 mL ) at $80^{\circ} \mathrm{C}$ for 2 hours followed by concentration and column chromatography, $R_{f}$ for trans-2.4c $=0.4(40 \%$ DCM in hexanes) afforded $943 \mathrm{mg}, 3.4 \mathrm{mmol}$ ( $72 \%$ isolated yield) of compound trans-2.4c as a colorless liquid. ${ }^{1} \mathrm{H}$ NMR $\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta=7.35(\mathrm{~d}, J=8.3 \mathrm{~Hz}, 2 \mathrm{H}), 6.88(\mathrm{~d}, J=8.8 \mathrm{~Hz}, 2 \mathrm{H})$, $5.74(\mathrm{ddt}, J=13.6,6.0,3.8 \mathrm{~Hz}, 1 \mathrm{H}), 5.57(\mathrm{dt}, J=11.1,2.8 \mathrm{~Hz}, 1 \mathrm{H}), 4.84(\mathrm{dd}, J=11.1,4.5 \mathrm{~Hz}, 1 \mathrm{H})$, $3.97(\mathrm{dtd}, J=4.1,2.8,1.4 \mathrm{~Hz}, 1 \mathrm{H}), 3.81(\mathrm{~s}, 3 \mathrm{H}), 2.52(\mathrm{dddq}, J=16.7,9.3,5.2,2.5 \mathrm{~Hz}, 1 \mathrm{H}), 2.40-$ $2.24(\mathrm{~m}, 2 \mathrm{H}), 2.13-2.05(\mathrm{~m}, 1 \mathrm{H}),-0.02(\mathrm{~s}, 9 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $\left.126 \mathrm{MHz}, \mathrm{CDCl}_{3}, \mathrm{ppm}\right) \delta=158.8$, $134.7,130.9,128.7,128.4,113.3,79.8,67.9,55.2,33.4,26.4,-3.2$. IR (FTIR, $\left.\mathrm{cm}^{-1}\right) \tilde{\mathrm{v}}=3010,2952$, 1511, 1242, 1032, 825. HRMS (ESI), $m / z[\mathrm{M}+\mathrm{H}]^{+}$calcd for $\mathrm{C}_{16} \mathrm{H}_{25} \mathrm{O}_{2} \mathrm{Si}: 277.1624$; found: 277.1624.

Synthesis of trans-trimethyl(7-(naphthalen-2-yl)-2,5,6,7-tetrahydrooxepin-2-yl)silane (trans-2.4d)


Compound trans-2.4d was prepared following general procedure E: Grubbs catalyst $2^{\text {nd }}$ generation ( $76 \mathrm{mg}, 0.1 \mathrm{mmol}, 0.015$ equiv.) and syn-(1-((1-(4-chlorophenyl)pent-4-en-1yl)oxy)allyl)trimethylsilane, 2.3 d ( $2.12 \mathrm{~g}, 6 \mathrm{mmol}$, 1 equiv.) and benzene ( 80 mL ) at $80^{\circ} \mathrm{C}$ for 2 hours followed by concentration and column chromatography, $R_{f}$ for trans-2.4d $=0.5(30 \% \mathrm{DCM}$ in hexanes) afforded a total of $1.493 \mathrm{~g}, 5 \mathrm{mmol}$ ( $84 \%$ isolated yield) of trans-2.4d as a light yellow crystalline solid (mp 30-32 ${ }^{\circ} \mathrm{C}$ ). The crystal structure of compound trans-2.4d was solved by Xray crystallography and the results deposited to the Cambridge Crystallographic Data Centre and assigned CCDC 1902771. ${ }^{1} \mathrm{H}$ NMR $\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta=7.91-7.78(\mathrm{~m}, 4 \mathrm{H}), 7.57(\mathrm{dd}, J=8.5$, $1.7 \mathrm{~Hz}, 1 \mathrm{H}), 7.52-7.44(\mathrm{~m}, 2 \mathrm{H}), 5.78(\mathrm{ddt}, J=10.3,6.7,2.7 \mathrm{~Hz}, 1 \mathrm{H}), 5.61(\mathrm{dt}, J=11.2,2.8 \mathrm{~Hz}$, $1 \mathrm{H}), 5.04(\mathrm{dd}, J=10.9,4.4 \mathrm{~Hz}, 1 \mathrm{H}), 4.05(\mathrm{td}, J=4.3,2.7 \mathrm{~Hz}, 1 \mathrm{H}), 2.59(\mathrm{dddq}, J=16.4,11.7,4.6$, $2.5 \mathrm{~Hz}, 1 \mathrm{H}), 2.48(\mathrm{dtd}, J=13.6,11.3,1.8 \mathrm{~Hz}, 1 \mathrm{H}), 2.41-2.32(\mathrm{~m}, 1 \mathrm{H}), 2.26(\mathrm{dt}, J=13.8,4.4 \mathrm{~Hz}$, $1 \mathrm{H}), 0.00(\mathrm{~s}, 9 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR (126 MHz, $\left.\mathrm{CDCl}_{3}, \mathrm{ppm}\right) \delta=140.0,133.0,132.8,130.9,128.6,128.0$, $127.7,127.6,126.3,125.9,125.7,124.9,80.2,68.6,33.6,26.4,-3.1 . \operatorname{IR}\left(\right.$ FTIR, film, $\left.\mathrm{cm}^{-1}\right) \tilde{v}=3052$, 2948, 2773, 1600, 1245, 1097, 829. HRMS (ESI), $m / z[\mathrm{M}+\mathrm{H}]^{+}$calcd for $\mathrm{C}_{19} \mathrm{H}_{25} \mathrm{OSi}$ 297.1675; found: 297.1663.

### 2.6.7. General procedure $F$ : Wittig rearrangement of trans-2-trimethylsilyl-2,5,6,7-tetrahydro-7-aryl-oxepins



Following a reported procedure, ${ }^{15}$ freshly prepared and purified trans-2-trimethylsilyl-2,5,6,7-tetrahydro-7-aryl-oxepin was dissolved in THF under nitrogen (concentration 0.08 M , unless otherwise noted) and the solution cooled at $-78{ }^{\circ} \mathrm{C}$ (dry ice/acetone bath), $n$-butyllithium (1.2 equiv, 1.6 M or 2.5 M in hexanes) was added dropwise ( $1 \mathrm{drop} / \mathrm{s}$ ) to give a colored solution. The reaction was quenched after the indicated time ( $10-30 \mathrm{~min}$ ) by adding saturated $\mathrm{NH}_{4} \mathrm{Cl}_{(\mathrm{aq})}$ and diluted with $\mathrm{H}_{2} \mathrm{O}$ and diethyl ether. The aqueous phase was extracted with diethyl ether three times. Combined organic extracts were washed with saturated $\mathrm{NH}_{4} \mathrm{Cl}_{(\mathrm{aq})}, \mathrm{H}_{2} \mathrm{O}$, and brine. The solution was dried over magnesium sulfate, filtered, quickly concentrated in a rotovap at temperatures lower than $45^{\circ} \mathrm{C}$. Column chromatography with EtOAc in hexanes afforded cyclohexenols. Other products including the ones resulting from [1,4]-Wittig rearrangement were also observed (see individual substrate).

Synthesis of 2-(trimethylsilyl)-1,2,5,6-tetrahydro-[1,1'-biphenyl]-2-ol (2.5a) and 7-phenyl-4,5,6,7-tetrahydrooxepin-2-yl)trimethylsilane (2.7a)


Applying general procedure F to trans-trimethyl(-7- phenyl-2,5,6,7-tetrahydrooxepin-2yl)silane 2.4a ( $1.6 \mathrm{~g}, 6.5 \mathrm{mmol}$, 1.0 equiv), $n$-butyllithium ( 2.3 M in hexanes, $3.4 \mathrm{~mL}, 7.8 \mathrm{mmol}$, 1.2 equiv), and THF ( 70 mL ) afforded after column chromatography ( $5-10 \%$ EtOAc in hexanes) $404 \mathrm{mg}, 1.64 \mathrm{mmol}(25 \%)$ of compound $\mathbf{2 . 5 a}$ and $819 \mathrm{mg}, 3.3 \mathrm{mmol}(51 \%$ isolated yield) of $\mathbf{2 . 7 a}$ as colorless oils.

Spectroscopic data for 2.5a: ${ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.40-7.35(\mathrm{~m}, 2 \mathrm{H}), 7.32-$ $7.27(\mathrm{~m}, 2 \mathrm{H}), 7.25-7.21(\mathrm{~m}, 1 \mathrm{H}), 5.96(\mathrm{ddd}, J=10.0,4.8,2.6 \mathrm{~Hz}, 1 \mathrm{H}), 5.88(\mathrm{dt}, J=10.0,1.9 \mathrm{~Hz}$, $1 \mathrm{H}), 2.91(\mathrm{dd}, J=10.9,3.0 \mathrm{~Hz}, 1 \mathrm{H}), 2.20(\mathrm{dtdd}, J=18.2,5.1,3.5,1.6 \mathrm{~Hz}, 1 \mathrm{H}), 2.16-2.06(\mathrm{~m}$, $1 \mathrm{H}), 2.00(\mathrm{dddd}, J=13.0,10.9,9.6,5.4 \mathrm{~Hz}, 1 \mathrm{H}), 1.68(\mathrm{ddt}, J=12.8,6.3,3.4 \mathrm{~Hz}, 1 \mathrm{H}), 1.30(\mathrm{~s}$, 1H), -0.11 (s, 9H). ${ }^{13} \mathrm{C}$ NMR (126 MHz, $\mathrm{CDCl}_{3}$ ) $\delta 143.8,130.8,130.4,129.3,128.2,126.5,66.2$, 47.3, 25.7, 25.3, -3.4.

## Synthesis of 4'-chloro-2-(trimethylsilyl)-1,2,5,6-tetrahydro-[1,1'-biphenyl]-2-ol (2.5b) and (7-(4-chlorophenyl)-4,5,6,7-tetrahydrooxepin-2-yl)trimethylsilane (2.7b)



Applying general procedure F to trans-(7-(4-chlorophenyl)-2,5,6,7-tetrahydrooxepin-2yl)trimethylsilane trans-2.4b ( $2.4 \mathrm{~g}, 8.5 \mathrm{mmol}, 1.0$ equiv), $n$-butyllithium ( 2.5 M in hexanes, 5.6 $\mathrm{mL}, 6.0 \mathrm{mmol}, 1.2$ equiv $)$, and THF $(100 \mathrm{~mL})$ afforded after column chromatography, $R_{f}$ for $\mathbf{2 . 7 b}=$ 0.7 and $R_{f}$ for $\mathbf{2 . 5 b}=0.3$ ( $2 \%$ EtOAc in hexanes) $667 \mathrm{mg}, 2.4 \mathrm{mmol}(28 \%$ isolated yield $)$ of compound 2.5b and $515.8 \mathrm{mg}, 1.84 \mathrm{mmol}$ ( $21 \%$ isolated yield) of 2.7b as colorless oils.

Spectroscopic data for 2.5b. ${ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}, \mathrm{ppm}$ ): $\delta=7.34(\mathrm{~d}, J=8.5 \mathrm{~Hz}, 2 \mathrm{H}), 7.27$
(d, $J=8.4 \mathrm{~Hz}, 2 \mathrm{H}$ ), 5.97 (ddd, $J=10.0,4.9,2.6 \mathrm{~Hz}, 1 \mathrm{H}), 5.89(\mathrm{dt}, J=10.0,2.0 \mathrm{~Hz}, 1 \mathrm{H}), 2.87$ (dd, $J=11.2,2.9 \mathrm{~Hz}, 1 \mathrm{H}), 2.25-2.07(\mathrm{~m}, 2 \mathrm{H}), 1.96(\mathrm{dddd}, J=13.0,11.1,9.8,5.6 \mathrm{~Hz}, 1 \mathrm{H}), 1.64$ (ddt, $J=12.6,6.6,3.1 \mathrm{~Hz}, 1 \mathrm{H}), 1.24(\mathrm{~s}, 1 \mathrm{H}),-0.10(\mathrm{~s}, 9 \mathrm{H}) .{ }^{13} \mathrm{C} \mathrm{NMR}\left(126 \mathrm{MHz}, \mathrm{CDCl}_{3}, \mathrm{ppm}\right): \delta=142.6$, 132.2, 130.8, 130.7, 130.5, 128.2, 66.0, 46.7, 25.7, 25.3, -3.4. HRMS (ESI): $m / z[M]^{+}$calcd for $\mathrm{C}_{15} \mathrm{H}_{21} \mathrm{ClOSi}$ : 280.1050 ; found: 280.1015 .

Spectroscopic data for 2.7b. ${ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}, \mathrm{ppm}$ ): $\delta=7.20(\mathrm{~d}, J=8.6 \mathrm{~Hz}$, $2 \mathrm{H}), 7.10(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 2 \mathrm{H}), 5.03(\mathrm{td}, J=4.0,1.1 \mathrm{~Hz}, 1 \mathrm{H}), 3.28(\mathrm{tq}, J=6.0,1.7 \mathrm{~Hz}, 1 \mathrm{H}), 2.16-$ $2.01(\mathrm{~m}, 2 \mathrm{H}), 1.95(\mathrm{dddd}, J=12.9,9.5,6.1,3.3 \mathrm{~Hz}, 1 \mathrm{H}), 1.62(\mathrm{dddd}, J=13.3,10.9,5.9,3.0 \mathrm{~Hz}$, $1 \mathrm{H}), 1.54-1.40(\mathrm{~m}, 2 \mathrm{H}), 0.01(\mathrm{~s}, 9 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $\left.126 \mathrm{MHz}, \mathrm{CDCl}_{3}, \mathrm{ppm}\right): \delta=150.2,142.9,131.5$, 129.7, 128.1, 106.1, 45.4, 32.7, 24.0, 19.2, 0.2. HRMS (ESI): $m / z[\mathrm{M}+\mathrm{H}]^{+}$calcd for $\mathrm{C}_{15} \mathrm{H}_{22} \mathrm{ClOSi}$ :
281.1129; found: 281.1147.

Synthesis of 4'-methoxy-2-(trimethylsilyl)-1,2,5,6-tetrahydro-[1,1'-biphenyl]-2-ol (2.5c), 2-(2-(4-methoxyphenyl)cyclobutyl)-1-(trimethylsilyl)ethan-1-one (2.6c) and (7-(4-methoxyphenyl)-4,5,6,7-tetrahydrooxepin-2-yl)trimethylsilane (2.7c)


Applying general procedure F to trans-(7-(4-methoxyphenyl)-2,5,6,7-tetrahydrooxepin-2yl)trimethylsilane trans-2.4c ( $885 \mathrm{mg}, 3.2 \mathrm{mmol}, 1.0$ equiv), $n$-butyllithium ( 2.5 M in hexanes, 5.6 $\mathrm{mL}, 6.0 \mathrm{mmol}$, 1.2 equiv $)$, and $\operatorname{THF}(100 \mathrm{~mL})$ afforded after column chromatography, $R_{f}$ for $\mathbf{2 . 6} \mathbf{c}=$ 0.7, $R_{f}$ for $\mathbf{2 . 7} \mathbf{c}=0.5$ and $R_{f}$ for $\mathbf{2 . 5 c}=0.3(5 \% \mathrm{EtOAc}$ in hexanes) $117 \mathrm{mg}, 0.42 \mathrm{mmol}(13 \%$ isolated yield) of compound $\mathbf{2 . 5 c}, 28 \mathrm{mg}, 0.1 \mathrm{mmol}(3 \%$ isolated yield) of $\mathbf{2 . 6 c}$ and $562 \mathrm{mg}, 2 \mathrm{mmol}$, ( $64 \%$ isolated yield) of $\mathbf{2 . 7} \mathbf{c}$ as colorless oils.

Spectroscopic data for 2.5c: ${ }^{1} \mathrm{H}$ NMR ( $\left.500 \mathrm{MHz}, \mathrm{CDCl}_{3}, \mathrm{ppm}\right): \delta=7.30(\mathrm{~d}, J=8.7 \mathrm{~Hz}$, $2 \mathrm{H}), 7.27(\mathrm{~d}, J=7.8 \mathrm{~Hz}, 2 \mathrm{H}), 6.88(\mathrm{~d}, J=8.7 \mathrm{~Hz}, 1 \mathrm{H}), 6.85(\mathrm{~d}, J=8.7 \mathrm{~Hz}, 2 \mathrm{H}), 5.95(\mathrm{dddd}, J=$ $10.0,4.8,2.6,0.7 \mathrm{~Hz}, 1 \mathrm{H}), 5.87(\mathrm{ddd}, J=10.0,2.3,1.6 \mathrm{~Hz}, 1 \mathrm{H}), 5.84-5.79(\mathrm{~m}, 1 \mathrm{H}), 5.69$ (ddd, $J$ $=9.9,2.5,1.7 \mathrm{~Hz}, 1 \mathrm{H}), 3.82(\mathrm{~s}, 2 \mathrm{H}), 3.81(\mathrm{~s}, 3 \mathrm{H}), 2.92-2.85(\mathrm{~m}, 2 \mathrm{H}), 2.31-2.05(\mathrm{~m}, 5 \mathrm{H}), 2.01-$ $1.91(\mathrm{~m}, 2 \mathrm{H}), 1.69-1.62(\mathrm{~m}, 2 \mathrm{H}), 1.52(\mathrm{~s}, 1 \mathrm{H}), 1.28(\mathrm{~s}, 1 \mathrm{H}),-0.10(\mathrm{~s}, 9 \mathrm{H}),-0.21(\mathrm{~s}, 9 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR (126 MHz, $\left.\mathrm{CDCl}_{3}, \mathrm{ppm}\right): \delta=158.4,158.2,135.9,135.1,132.8,130.9,130.3,130.2,129.4,127.1$, $113.54,113.47,72.0,66.3,55.3,55.2,50.3,46.4,25.7,25.52,250.5,25.45,-2.4,-3.4$. HRMS (ESI): $m / z[\mathrm{M}+\mathrm{H}]^{+}$calcd for $\mathrm{C}_{16} \mathrm{H}_{25} \mathrm{O}_{2} \mathrm{Si}$ : 277.1624; found: 277.1623.

Spectroscopic data for 2.6c (cis:trans $=1: 1)$ : ${ }^{1} \mathrm{H}$ NMR $\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}, \mathrm{ppm}\right): \delta=7.14$ $(\mathrm{d}, J=8.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.04(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 2 \mathrm{H}), 6.86-6.80(\mathrm{~m}, 4 \mathrm{H}), 3.82-3.77(\mathrm{~m}, 7 \mathrm{H}), 3.70(\mathrm{q}, J=$ $8.1 \mathrm{~Hz}, 1 \mathrm{H}), 3.20(\mathrm{dtd}, J=14.6,7.9,7.3,1.4 \mathrm{~Hz}, 1 \mathrm{H}), 3.01(\mathrm{q}, J=8.8 \mathrm{~Hz}, 1 \mathrm{H}), 2.87-2.71(\mathrm{~m}, 2 \mathrm{H})$, $2.51(\mathrm{dd}, J=17.6,7.6 \mathrm{~Hz}, 1 \mathrm{H}), 2.32-2.23(\mathrm{~m}, 3 \mathrm{H}), 2.23-2.16(\mathrm{~m}, 2 \mathrm{H}), 2.16-2.05(\mathrm{~m}, 1 \mathrm{H}), 1.98$ $(\mathrm{qd}, J=10.3,8.6 \mathrm{~Hz}, 1 \mathrm{H}), 1.66-1.59(\mathrm{~m}, 2 \mathrm{H}), 1.53(\mathrm{tt}, J=10.3,8.6 \mathrm{~Hz}, 1 \mathrm{H}), 0.13(\mathrm{~s}, 9 \mathrm{H}), 0.00(\mathrm{~s}$, 9H). ${ }^{13} \mathrm{C}$ NMR ( $126 \mathrm{MHz}, \mathrm{CDCl}_{3}, \mathrm{ppm}$ ): $\delta=248.1,248.0,158.0,157.8,136.4,134.0,128.8,127.7$, $113.7,113.5,55.3,54.8,50.1,46.5,41.6,38.4,33.5,27.4,25.6,24.5,23.5,-3.3,-3.4$. IR (FTIR, $\left.\mathrm{cm}^{-1}\right): \tilde{\mathrm{v}}=2951,2938,2865,2835,1639,1610,1511,1244,1175,1034,827$. HRMS (ESI): $\mathrm{m} / \mathrm{z}$ $[\mathrm{M}+\mathrm{H}]^{+}$calcd for $\mathrm{C}_{16} \mathrm{H}_{25} \mathrm{O}_{2}$ Si: 277.1624; found: 277.1625.

Spectroscopic data for 2.7c: ${ }^{1} \mathrm{H}$ NMR ( $\left.500 \mathrm{MHz}, \mathrm{CDCl}_{3}, \mathrm{ppm}\right): \delta=7.31(\mathrm{~d}, J=8.4 \mathrm{~Hz}$, $2 \mathrm{H}), 6.88(\mathrm{~d}, J=8.8 \mathrm{~Hz}, 2 \mathrm{H}), 5.40(\mathrm{dd}, J=6.2,5.1 \mathrm{~Hz}, 1 \mathrm{H}), 4.50(\mathrm{dd}, J=10.9,2.4 \mathrm{~Hz}, 1 \mathrm{H}), 3.81$ (s, 3H), 2.36-2.19 (m, 2H), 2.17-2.08(m, 1H), 1.96 (dddt, $J=15.5,11.4,7.1,4.2 \mathrm{~Hz}, 2 \mathrm{H}), 1.52$ (dtt, $J=11.6,6.8,2.2 \mathrm{~Hz}, 1 \mathrm{H}), 0.06(\mathrm{~s}, 9 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $126 \mathrm{MHz}, \mathrm{CDCl}_{3}, \mathrm{ppm}$ ): $\delta=166.2,158.4$, $136.4,126.8,122.9,113.4,83.2,55.2,39.4,27.3,25.5,-2.3$. IR (FTIR, $\left.\mathrm{cm}^{-1}\right): \tilde{\mathrm{v}}=2952,2927,2835$, 1613, 1512, 1244, 1100, 837. HRMS (ESI): $m / z[\mathrm{M}+\mathrm{H}]^{+}$calcd for $\mathrm{C}_{16} \mathrm{H}_{25} \mathrm{O}_{2} \mathrm{Si}: 277.1624$; found: 277.1624.

Synthesis of 4'-methoxy-2-(trimethylsilyl)-1,2,5,6-tetrahydro-[1,1'-biphenyl]-3'-d-2-ol (2.5c- $d_{1}$ ), 2-(2-(4-methoxyphenyl-3- $d$ )cyclobutyl)-1-(trimethylsilyl)ethan-1-one-2-d (2.6c- $d_{2}$ ) and (7-(4-methoxyphenyl-3-d)-2,5,6,7-tetrahydrooxepin-2-yl)trimethylsilane (cis-2.4c- $d_{1}$ )


Compounds $\mathbf{2 . 5} \mathbf{c}-\boldsymbol{d}_{\boldsymbol{1}}, \mathbf{2 . 6 c} \mathbf{-} \boldsymbol{d}_{\mathbf{2}}$, and $c i s-\mathbf{2 . 4 c} \mathbf{-} \boldsymbol{d}_{\boldsymbol{I}}$ were prepared as follows: $1104 \mathrm{mg}(4.0 \mathrm{mmol}$, 1.0 equiv) of freshly prepared and purified cis-(7-(4-methoxyphenyl)-2,5,6,7-tetrahydrooxepin-2yl)trimethylsilane (cis-2.4c) was dissolved in 50 mL dry THF under nitrogen, and the resulting solution was cooled at $-78{ }^{\circ} \mathrm{C}$ (dry ice/acetone bath), sec-butyllithium, 1.4 M in cyclohexane ( 8.6
$\mathrm{mL}, 12.0 \mathrm{mmol}, 3.0$ equiv) was added dropwise ( 1 drop/s) to give a dark brown solution. The reaction mixture was allowed to warm up slowly without removing the cooling bath to $-10{ }^{\circ} \mathrm{C}$. The mixture was stirred at this temperature for 5 hours then cooled back to $-78{ }^{\circ} \mathrm{C}$. The reaction was quenched at $-78{ }^{\circ} \mathrm{C}$ with $\mathrm{D}_{2} \mathrm{O}(5 \mathrm{~mL})$, and the cooling bath was removed. After warming up to room temperature, the reaction mixture was diluted with 20 mL ether and loaded into a separating funnel. The layers were separated, and the aqueous phase was extracted with diethyl ether ( 20 mL $x$ 3). Combined organic extracts were washed with brine ( 15 mL ), and dried over anhydrous magnesium sulfate, filtered, quickly concentrated in a rotovap at temperatures lower than $45{ }^{\circ} \mathrm{C}$. The crude material was purified by column chromatography, $R_{f}$ for cis-2.4c- $\boldsymbol{d}_{\boldsymbol{I}}=0.6, R_{f}$ for $\mathbf{2 . 6 c} \mathbf{-} \boldsymbol{d}_{\mathbf{2}}$ $=0.4$ and $R_{f}$ for $\mathbf{2 . 5 c}-\boldsymbol{d}_{\boldsymbol{l}}=0.3$ ( $10 \% \mathrm{EtOAc}$ in hexanes) $259 \mathrm{mg}, 0.93 \mathrm{mmol}(23 \%$ isolated yield $)$ of compound 2.5c- $\boldsymbol{d}_{\mathbf{1}}, 172 \mathrm{mg}, 0.62 \mathrm{mmol}\left(16 \%\right.$ isolated yield) of $\mathbf{2 . 6 c}-\boldsymbol{d}_{\mathbf{2}}$ and $503 \mathrm{mg}, 1.8 \mathrm{mmol}(45 \%)$ of recovered starting material cis-2.4c- $\boldsymbol{d}_{\boldsymbol{I}}$ with deuterium incorporation in the aromatic ring as colorless oils.

Spectroscopic data for $\mathbf{2 . 5 c}-\boldsymbol{d}_{\boldsymbol{l}}:{ }^{1} \mathrm{H}$ NMR $\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}, \mathrm{ppm}\right): \delta=7.29(\mathrm{dq}, J=4.6,2.3$ $\mathrm{Hz}, 2 \mathrm{H}), 6.84(\mathrm{~d}, J=9.0 \mathrm{~Hz}, 1 \mathrm{H}), 5.94(\mathrm{dddd}, J=10.0,4.9,2.6,0.7 \mathrm{~Hz}, 1 \mathrm{H}), 5.86(\mathrm{ddd}, J=10.0$, $2.4,1.7 \mathrm{~Hz}, 1 \mathrm{H}), 3.79(\mathrm{~s}, 3 \mathrm{H}), 2.87(\mathrm{dd}, J=10.9,3.0 \mathrm{~Hz}, 1 \mathrm{H}), 2.18(\mathrm{dtdd}, J=18.3,5.3,3.6,1.6 \mathrm{~Hz}$, $1 \mathrm{H}), 2.14-2.04(\mathrm{~m}, 1 \mathrm{H}), 1.95(\mathrm{dddd}, J=13.0,10.9,9.6,5.5 \mathrm{~Hz}, 1 \mathrm{H}), 1.64(\mathrm{ddt}, J=12.7,6.3,3.2$ $\mathrm{Hz}, 1 \mathrm{H}), 1.28(\mathrm{~s}, 1 \mathrm{H}),-0.11(\mathrm{~s}, 9 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $\left.126 \mathrm{MHz}, \mathrm{CDCl}_{3}, \mathrm{ppm}\right): \delta=158.2,135.9,130.9$, $130.22,130.20,130.1,113.6,66.3,55.1,46.4,25.7,25.5,-3.4 .{ }^{29} \mathrm{Si} \operatorname{NMR}\left(99 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta=$ 5.03. HRMS (ESI): $m / z\left[\mathrm{M}-\mathrm{H}^{-}\right]^{+}$calcd for $\mathrm{C}_{16} \mathrm{H}_{22} \mathrm{DO}_{2} \mathrm{Si}: 276.1536$; found: 276.1523.

Spectroscopic data for 2.6c- $\boldsymbol{d}_{2}$ (cis:trans $=1: 1$ ): ${ }^{1} \mathrm{H} \operatorname{NMR}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}, \mathrm{ppm}\right): \delta=7.14$ $(\mathrm{dq}, J=4.1,2.2 \mathrm{~Hz}, 2 \mathrm{H}), 7.04(\mathrm{dq}, J=3.8,2.2 \mathrm{~Hz}, 2 \mathrm{H}), 6.84(\mathrm{~d}, J=5.1 \mathrm{~Hz}, 1 \mathrm{H}), 6.82(\mathrm{~d}, J=5.1$ $\mathrm{Hz}, 1 \mathrm{H}), 3.78(\mathrm{~s}, 3 \mathrm{H}), 3.78(\mathrm{~s}, 3 \mathrm{H}), 3.70(\mathrm{q}, J=8.1 \mathrm{~Hz}, 1 \mathrm{H}), 3.19(\mathrm{p}, J=7.9 \mathrm{~Hz}, 1 \mathrm{H}), 3.01(\mathrm{q}, J=$
$9.1 \mathrm{~Hz}, 1 \mathrm{H}), 2.85-2.67(\mathrm{~m}, 2 \mathrm{H}), 2.47(\mathrm{dt}, J=7.5,2.5 \mathrm{~Hz}, 1 \mathrm{H}), 2.30-2.23(\mathrm{~m}, 2 \mathrm{H}), 2.22-2.14$ $(\mathrm{m}, 2 \mathrm{H}), 2.13-2.05(\mathrm{~m}, 1 \mathrm{H}), 1.98(\mathrm{qd}, J=10.3,8.5 \mathrm{~Hz}, 1 \mathrm{H}), 1.66-1.58(\mathrm{~m}, 1 \mathrm{H}), 1.58-1.48(\mathrm{~m}$, $1 \mathrm{H}), 0.13(\mathrm{~s}, 9 \mathrm{H}), 0.00(\mathrm{~s}, 9 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $\left.126 \mathrm{MHz}, \mathrm{CDCl}_{3}, \mathrm{ppm}\right): \delta=248.2,248.1,157.9,157.7$, $136.3,133.9,128.8,128.7,127.7,127.6,113.7,113.5,55.2,55.2,54.4,49.7,46.4,41.6,38.3,33.4$, 27.4, 25.5, 24.4, 23.5, -3.3, -3.4. ${ }^{29}$ Si NMR ( $99 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta=-10.71,-10.89$. HRMS (ESI): $\mathrm{m} / \mathrm{z}$ $[\mathrm{M}+\mathrm{H}]^{+}$calcd for $\mathrm{C}_{16} \mathrm{H}_{22} \mathrm{D}_{2} \mathrm{O}_{2}$ Si: 279.1744; found: 279.1738.

Spectroscopic data for cis-2.4c- $\boldsymbol{d}_{\mathbf{l}}:{ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}, \mathrm{ppm}$ ): $\delta=7.34-7.27(\mathrm{~m}$, $2 \mathrm{H}), 6.89(\mathrm{~d}, J=9.1 \mathrm{~Hz}, 1 \mathrm{H}), 5.72$ (dddd, $J=11.1,7.2,4.7,2.5 \mathrm{~Hz}, 1 \mathrm{H}), 5.57(\mathrm{ddd}, J=11.2,3.1$, $2.1 \mathrm{~Hz}, 1 \mathrm{H}), 4.74(\mathrm{dd}, J=8.4,5.3 \mathrm{~Hz}, 1 \mathrm{H}), 4.08(\mathrm{qd}, J=3.3,1.2 \mathrm{~Hz}, 1 \mathrm{H}), 3.81(\mathrm{~s}, 3 \mathrm{H}), 2.71-2.61$ $(\mathrm{m}, 1 \mathrm{H}), 2.33(\mathrm{ddt}, J=13.6,10.4,5.0 \mathrm{~Hz}, 1 \mathrm{H}), 2.05-1.96(\mathrm{~m}, 1 \mathrm{H}), 1.89(\mathrm{ddt}, J=13.6,8.4,5.2 \mathrm{~Hz}$, $1 \mathrm{H}), 0.13(\mathrm{~s}, 9 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $126 \mathrm{MHz}, \mathrm{CDCl}_{3}, \mathrm{ppm}$ ): $\delta=158.1,137.4,130.7,126.3,126.23$, $126.18,113.4,81.2,77.1,55.2,37.4,23.6,-3.7 .{ }^{29}$ Si NMR ( $99 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta=-1.06$. HRMS (ESI): $m / z[\mathrm{M}-\mathrm{OH}]^{+}$calcd for $\mathrm{C}_{16} \mathrm{H}_{22}$ DOSi: 260.1575; found: 260.1572.

Synthesis of 6-(naphthalen-2-yl)-1-(trimethylsilyl)cyclohex-2-en-1-ol (2.5d), 2-(2-(naphthalen-2-yl)cyclobutyl)-1-(trimethylsilyl)ethan-1-one (2.6d) and 2-(naphthalen-2-yl)cyclohexan-1-one (2.8d)


Applying general procedure F to trans-trimethyl(7-(naphthalen-2-yl)-2,5,6,7-tetrahydrooxepin-2-yl)silane trans-2.4d ( $593 \mathrm{mg}, 2.0 \mathrm{mmol}, 1.0$ equiv), $n$-butyllithium ( 2.5 M in hexanes, $5.6 \mathrm{~mL}, 6.0 \mathrm{mmol}, 1.2$ equiv), and THF ( 100 mL ) afforded after column chromatography, $R_{f}$ for $\mathbf{2 . 6 d}=0.7, R_{f}$ for $\mathbf{2 . 5 d}=0.5, R_{f}$ for $\mathbf{2 . 5 d} \mathbf{d}^{\prime}=0.4$ and $R_{f}$ for $\mathbf{2 . 8 d}=0.3(10 \%$ EtOAc in hexanes $)$ $283 \mathrm{mg}, 0.95 \mathrm{mmol}\left(48 \%\right.$ isolated yield) of compound $\mathbf{2 . 5 d}$ and $\mathbf{2 . 5 d} \mathbf{d}^{\prime}, 156 \mathrm{mg}, 0.53 \mathrm{mmol}(26 \%$ isolated yield) of $\mathbf{2 . 6 d}$ and $59 \mathrm{mg}, 0.26 \mathrm{mmol}$ ( $13 \%$ isolated yield) of $\mathbf{2 . 8 d}$ as colorless oils.

Spectroscopic data for $\mathbf{2 . 5 d}$ (silyl and naphthalenyl groups cis to one another): ${ }^{1} \mathrm{H}$ NMR ( $\left.500 \mathrm{MHz}, \mathrm{CDCl}_{3}, \mathrm{ppm}\right): \delta=7.85-7.77(\mathrm{~m}, 4 \mathrm{H}), 7.57(\mathrm{dd}, J=8.4,1.7 \mathrm{~Hz}, 1 \mathrm{H}), 7.49-7.41(\mathrm{~m}$, $2 \mathrm{H}), 6.01(\mathrm{ddd}, J=10.1,5.0,2.3 \mathrm{~Hz}, 1 \mathrm{H}), 5.93(\mathrm{dt}, J=10.0,1.9 \mathrm{~Hz}, 1 \mathrm{H}), 3.10(\mathrm{dd}, J=10.6,3.1$ $\mathrm{Hz}, 1 \mathrm{H}), 2.28-2.05(\mathrm{~m}, 3 \mathrm{H}), 1.78-1.69(\mathrm{~m}, 1 \mathrm{H}), 1.38(\mathrm{~s}, 1 \mathrm{H}),-0.10(\mathrm{~s}, 9 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( 126 MHz , $\left.\mathrm{CDCl}_{3}, \mathrm{ppm}\right): \delta=141.5,133.4,132.3,131.0,130.4,128.1,127.73,127.68,127.6,127.5,125.8$,
125.3, 66.4, 47.5, 25.7, 25.5, -3.3. HRMS (ESI): $m / z[\mathrm{M}]^{+}$calcd for $\mathrm{C}_{19} \mathrm{H}_{24} \mathrm{OSi}$ 296.1596; found: 296.1592.

Spectroscopic data for $\mathbf{2 . 5 d}$ ' (silyl and naphthalenyl groups trans to one another): ${ }^{1} \mathrm{H}$ NMR (500 MHz, $\left.\mathrm{CDCl}_{3}, \mathrm{ppm}\right): \delta=7.85-7.78(\mathrm{~m}, 3 \mathrm{H}), 7.75(\mathrm{~s}, 1 \mathrm{H}), 7.55(\mathrm{dd}, J=8.4,1.8 \mathrm{~Hz}, 1 \mathrm{H}), 7.50$ $-7.41(\mathrm{~m}, 2 \mathrm{H}), 5.87(\mathrm{dd}, J=10.0,3.4 \mathrm{~Hz}, 1 \mathrm{H}), 5.74(\mathrm{~d}, J=10.3 \mathrm{~Hz}, 1 \mathrm{H}), 3.10(\mathrm{dt}, J=12.8,3.5$ $\mathrm{Hz}, 1 \mathrm{H}), 2.32-2.26(\mathrm{~m}, 2 \mathrm{H}), 2.10(\mathrm{dp}, J=6.7,5.1 \mathrm{~Hz}, 1 \mathrm{H}), 1.59(\mathrm{~s}, 1 \mathrm{H}),-0.26(\mathrm{~s}, 9 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR (126 MHz, $\left.\mathrm{CDCl}_{3}, \mathrm{ppm}\right): \delta=140.7,133.3,132.9,132.3,127.78,127.76,127.6,127.5,127.3,126.7$, $125.9,125.4,72.0,51.3,25.48,25.46,-2.3 . \mathrm{MS}(\mathrm{GC} / \mathrm{MS}): m / \mathrm{z}(\%)=296(5)[\mathrm{M}]^{+}, 279(38), 165$ (14), 73 (100). HRMS (ESI): $m / z\left[\mathrm{M}-\mathrm{H}^{-}\right]^{+}$calcd for $\mathrm{C}_{19} \mathrm{H}_{23} \mathrm{OSi}$ : 295.1518; found: 295.1505.

Spectroscopic data for 2.6d: ${ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}, \mathrm{ppm}$ ) for trans diastereomer: $\delta$ $=7.82-7.78(\mathrm{~m}, 2 \mathrm{H}), 7.75(\mathrm{~d}, \mathrm{~J}=8.5 \mathrm{~Hz}, 1 \mathrm{H}), 7.58(\mathrm{~s}, 1 \mathrm{H}), 7.48-7.41(\mathrm{~m}, 2 \mathrm{H}), 7.27-7.22(\mathrm{~m}$, $1 \mathrm{H}), 3.93(\mathrm{q}, \mathrm{J}=8.2 \mathrm{~Hz}, 1 \mathrm{H}), 3.40-3.29(\mathrm{~m}, 1 \mathrm{H}), 2.54(\mathrm{dd}, \mathrm{J}=17.7,7.5 \mathrm{~Hz}, 1 \mathrm{H}), 2.51-2.44(\mathrm{~m}$, $1 \mathrm{H}), 2.42-2.36(\mathrm{~m}, 1 \mathrm{H}), 2.36-2.25(\mathrm{~m}, 2 \mathrm{H}), 1.72(\mathrm{ddtd}, \mathrm{J}=11.5,9.4,5.8,0.9 \mathrm{~Hz}, 1 \mathrm{H}),-0.10(\mathrm{~s}$, 9H). ${ }^{13} \mathrm{C}$ NMR ( $126 \mathrm{MHz}, \mathrm{CDCl}_{3}, \mathrm{ppm}$ ) trans diastereomer: $\delta=248.0,139.4,133.4,132.0,127.54$, $127.48,127.46,127.0,125.8,125.7,125.2,50.0,42.4,33.6,24.5,23.1,-3.6 .{ }^{13} \mathrm{C}$ NMR ( 126 MHz , $\left.\mathrm{CDCl}_{3}, \mathrm{ppm}\right)$ cis/trans isomers: $\delta=249.2,248.0,139.5,135.7,134.4,133.2,133.4,132.0,129.8$, $127.54,127.48,127.47,127.3,127.04,127.00,126.4,126.1,125.8,125.7,125.5,125.2,125.0$, 50.0, 45.7, 42.4, 41.4, 33.6, 27.1, 26.2, 24.6, 23.1, 21.7, -3.3, -3.5. HRMS (ESI): $m / z[\mathrm{M}+\mathrm{H}]^{+}$calcd for $\mathrm{C}_{19} \mathrm{H}_{25} \mathrm{OSi}$ : 297.1675; found: 297.1661.

Spectroscopic data for 2.8d: ${ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}, \mathrm{ppm}$ ): $\delta=7.63(\mathrm{~s}, 1 \mathrm{H}), 7.53-$ $7.41(\mathrm{~m}, 2 \mathrm{H}), 7.31(\mathrm{dd}, J=8.5,1.8 \mathrm{~Hz}, 1 \mathrm{H}), 3.80(\mathrm{dd}, J=12.2,5.5 \mathrm{~Hz}, 1 \mathrm{H}), 2.59(\mathrm{dtd}, J=13.7$, $4.0,1.3 \mathrm{~Hz}, 1 \mathrm{H}), 2.51$ (dddd, $J=13.7,12.4,5.9,1.2 \mathrm{~Hz}, 1 \mathrm{H}), 2.35(\mathrm{dddd}, J=15.2,7.3,3.4,2.3 \mathrm{~Hz}$, $1 \mathrm{H}), 2.24-2.12(\mathrm{~m}, 2 \mathrm{H}), 2.05$ (dddq, $J=10.5,6.8,4.5,2.4,2.0 \mathrm{~Hz}, 1 \mathrm{H}), 1.95-1.82(\mathrm{~m}, 2 \mathrm{H}) .{ }^{13} \mathrm{C}$

NMR (126 MHz, $\left.\mathrm{CDCl}_{3}, \mathrm{ppm}\right): \delta=210.4,136.3,133.4,132.5,127.72,127.65,127.6,126.92$, $126.90,125.8,125.5,57.4,42.2,35.0,27.7,25.2$. 2.8d is a known compound, and the spectroscopic data are in agreement with those reported in the literature. ${ }^{19}$

## REFERENCES

(1) Wittig, G.; Lohmann, L. Übe die kationtrope isomerisation gewisser benzyläther bei einwirkung von phenyl-lithium. Liebigs Ann. Chem. 1942, 550, 260.
(2) Schöllkopf, U. Recent results in carbanion chemistry. Angew. Chem., Int. Ed. Engl. 1970, 9, 763.
(3) Schäfer, H.; Schöllkopf, U.; Walter, D. Migration tendency of alkyl groups in the Wittig rearrangement. Tetrahedron Lett. 1968, 9, 2809.
(4) Sheldon, J. C.; Taylor, M. S.; Bowie, J. H.; Dua, Brian-Chia S. C.; Eichinger, P. C. H. The gas phase 1,2 -Wittig rearrangement is an anion reaction. A joint experimental and theoretical study. J. Chem. Soc., Perkin Trans. 2 1999, 333.
(5) Strunk, S.; Schlosser, M. Wittig rearrangement of lithiated allyl aryl ethers: A mechanistic study. Eur. J. Org. Chem. 2006, 4393.
(6) Antoniotti, P.; Tonachini, G. J. Mechanism of the anionic Wittig rearrangement. An ab initio theoretical study. J. Org. Chem. 1998, 63 ,9756.
(7) Schreiber, S. L.; Goulet, M. T. Stereochemistry of the 1,2-Wittig rearrangement: A synthesis of syn-1,3-diol monoethers. Tetrahedron Lett. 1987, 28, 1043.
(8) Tomooka, K.; Nakai, T. [1,2]-Wittig rearrangement stereochemical features and synthetic utilities. J. Synth. Org. Chem., Jpn. 1997, 54, 1000.
(9) Tomooka, K.; Yamamoto, H.; Nakai, T. Recent developments in the [1,2]-Wittig rearrangement. Liebigs Ann./Recl. 1997, 1275.
(10) Yadav, J. S.; Ravishankar, R. A novel approach towards the synthesis of functionalized taxane skeleton employing Wittig rearrangement. Tetrahedron Lett. 1991, 32, 2629.
(11) Maleczka, R. E., Jr.; Geng, F. Synthesis and fluoride-promoted Wittig rearrangements of $\alpha$-alkoxysilanes. Org. Lett. 1999, 1, 1111.
(12) Felkin, H. Tambute, A. 1,4-Alkyl shifts in the Wittig rearrangement of alkyl allyl ethers. Tetrahedron Lett. 1969, 821.
(13) Courtois, G.; Miginiac, L. Transposition of allyl ethers under the action of allyllithium. Tetrahedron Lett. 1972, 24, 2411.
(14) Rautenstrauch, V.; Büchi, G.; Wüst, H. Vinyl migration in Wittig rearrangements. J. Am. Chem. Soc. 1974, 96, 2576.
(15) Mori-Quiroz, L. M.; Maleczka, R, E., Jr. Stereoconvergent [1,2]- and [1,4]-Wittig rearrangements of 2-silyl-6-aryl-5,6-dihydropyrans: A tale of steric vs electronic regiocontrol of divergent pathways. J. Org. Chem. 2015, 80, 1163.
(16) Nath, S. R.; Joshi, K. A. Mechanistic investigation in the [1,4] and [1,2] Wittig rearrangement reactions: a DFT study. Phys. Chem. Chem. Phys. 2018, 20, 21457.
(17) Tomizuka, A.; Moriyama, K. Bromoetherification of alkenyl alcohols by aerobic oxidation of bromide: Asymmetric synthesis of 2-bromomethyl 5-substituted tetrahydrofurans. Adv. Synth. Catal. 2019, 361, 1447.
(18) Chan, C-K; Huang, Y-H; Chang, M-Y. Sodium amalgam mediated desulfonylative reduction of $\alpha$-functionalized $\beta$-ketosulfones. Tetrahedron, 2016, 72, 5521.
(19) Cheon, C. H.; Kanno, O.; Toste, F. D. Chiral Brønsted acid from a cationic gold(I) complex: Catalytic enantioselective protonation of silyl enol ethers of ketones. J. Am. Chem. Soc. 2011, 133, 13248.

## APPENDIX

## X-ray data for compound 2.4d

Crystal structure confirms relative stereo chemistry, both enantiomers are present.

## Crystal data and experimental



Figure 2.1: Crystal structure of compound 2.4d
Experimental. Single yellow needle-shaped crystals of 2.4d were used as received. A suitable crystal $0.43 \times 0.12 \times 0.05 \mathrm{~mm}^{3}$ was selected and mounted on a nylon loop with paratone oil on a Bruker APEX-II CCD diffractometer. The crystal was kept at a steady $T=173$ (2) K during data collection. The structure was solved with the ShelXT (Sheldrick, G.M. (2015). Acta Cryst. A71, 3-8) structure solution program using the Intrinsic Phasing solution method and by using Olex2 (Dolomanov et al., 2009) as the graphical interface. The model was refined with version 2018/3 of ShelXL (Sheldrick, Acta Cryst. A64 2008, 112-122) using Least Squares minimization.

Crystal data. $\mathrm{C}_{19} \mathrm{H}_{24} \mathrm{OSi}, M_{r}=296.47$, monoclinic, $C 2 / c$ (No. 15), $\mathrm{a}=32.3150(4) \AA, \mathrm{b}=$ $6.20960(10) \AA, \mathrm{c}=22.7493(3) \AA, \beta=132.1830(10)^{\circ}, \alpha=\gamma=90^{\circ}, V=3382.64(9) \AA^{3}, T=173(2) \mathrm{K}$, $Z=8, Z^{\prime}=1, \mu\left(\mathrm{CuK}_{\alpha}\right)=1.182,25668$ reflections measured, 3340 unique $\left(R_{\text {int }}=0.0417\right)$ which were used in all calculations. The final $w R_{2}$ was 0.1021 (all data) and $R_{l}$ was 0.0367 (I > 2(I)).

Table 2.2: Crystal data

| Compound | 2.4d |
| :---: | :---: |
| CCDC | 1902771 |
| Formula | $\mathrm{C}_{19} \mathrm{H}_{24} \mathrm{OSi}$ |
| $D_{\text {calc. }} / \mathrm{g} \mathrm{cm}^{-3}$ | 1.164 |
| $\mu / \mathrm{mm}^{-1}$ | 1.182 |
| Formula Weight | 296.47 |
| Color | yellow |
| Shape | needle |
| Size/mm ${ }^{3}$ | $0.43 \times 0.12 \times 0.05$ |
| T/K | 173(2) |
| Crystal System | monoclinic |
| Space Group | C2/c |
| $a /$ Å | 32.3150(4) |
| b/Å | 6.20960 (10) |
| c/Å | 22.7493(3) |
| $\alpha 1^{\circ}$ | 90 |
| $\beta 1^{\circ}$ | 132.1830(10) |
| $\gamma 1^{\circ}$ | 90 |
| V/A ${ }^{3}$ | 3382.64(9) |
| Z | 8 |
| Z' | 1 |
| Wavelength/Å | 1.541838 |
| Radiation type | $\mathrm{CuK}_{\alpha}$ |
| $\Theta_{\text {min }}{ }^{\circ}$ | 3.692 |
| $\Theta_{\max }{ }^{\circ}$ | 72.149 |
| Measured Refl. | 25668 |
| Independent Refl. | 3340 |
| Reflections with I $>2(\mathrm{I})$ | 2924 |
| $R_{\text {int }}$ | 0.0417 |
| Parameters | 193 |
| Restraints | 0 |
| Largest Peak | 0.293 |
| Deepest Hole | -0.226 |
| GooF | 1.044 |
| $w R_{2}$ (all data) | 0.1021 |
| $w R_{2}$ | 0.0975 |
| $R_{l}$ (all data) | 0.0425 |
| $R_{1}$ | 0.0367 |


| Reflections: | $\min (\mathrm{Cu}) \quad 0.81$ |  | 40.6 |  | 4.17\% | ete | 100\% |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Refinement: | Shift 0.001 | Max Peak | 0.3 | Min P | -0.2 | God | 1.044 |

Figure 2.2: Structure quality indicators
A yellow needle-shaped crystal with dimensions $0.43 \times 0.12 \times 0.05 \mathrm{~mm}^{3}$ was mounted on a nylon loop with paratone oil. Data were collected using a Bruker APEX-II CCD diffractometer equipped with an Oxford Cryosystems low-temperature device, operating at $T=173(2) \mathrm{K}$.

Data were measured using $\omega$ and $\phi$ of $1.00^{\circ}$ per frame for 100.00 s using $\mathrm{CuK}_{\alpha}$ radiation (sealed tube, $40 \mathrm{kV}, 30 \mathrm{~mA}$ ). The total number of runs and images was based on the strategy calculation from the program COSMO (BRUKER, V1.61, 2009). The actually achieved resolution was $\Theta=$ 72.149 .

Cell parameters were retrieved using the SAINT (Bruker, V8.38A, after 2013) software and refined using SAINT (Bruker, V8.38A, after 2013) on 9911 reflections, $39 \%$ of the observed reflections. Data reduction was performed using the SAINT (Bruker, V8.38A, after 2013) software which corrects for Lorentz polarization. The final completeness is 100.00 out to 72.149 in $\Theta$ SADABS-2016/2 (Bruker,2016/2) was used for absorption correction. $w R_{2}$ (int) was 0.0735 before and 0.0532 after correction. The Ratio of minimum to maximum transmission is 0.8774 . The $\lambda / 2$ correction factor is Not present.

The structure was solved in the space group C2/c (\# 15) by Intrinsic Phasing using the ShelXT (Sheldrick, G.M. (2015). Acta Cryst. A71, 3-8) structure solution program. The structure was refined by Least Squares using version 2014/6 of XL (Sheldrick, 2008) incorporated in Olex2 (Dolomanov et al., 2009). All non-hydrogen atoms were refined anisotropically. Hydrogen atom positions were calculated geometrically and refined using the riding model, except for the hydrogen atom on the non-carbon atom(s) which was found by difference Fourier methods and refined isotropically.

CCDC 1902771 contains the supplementary crystallographic data for this paper. The data can be obtained free of charge from The Cambridge Crystallographic Data Centre via www.ccdc.cam.ac.uk/structures.

There is a single molecule in the asymmetric unit, which is represented by the reported sum formula. In other words: Z is 8 and $\mathrm{Z}^{\prime}$ is 1 .


Figure 2.3: Model has Chirality at C1 (Centro SPGR) R Verify: Model has Chirality at C6 (Centro SPGR) S Verify. Being in a centrosymmetric space group means both enantiomers are present in the crystal


Figure 2.4: Packing diagram of 2.4d

## Data Plots: Diffraction data



Figure 2.5: Data plots: Diffraction data

Figure 2.5 (cont'd)



## Data Plots: Refinement and data




Figure 2.6: Data plots: Refinement and data
Table 2.3: Reflection statistics

| Total reflections (after <br> filtering) | 26650 |
| :--- | :--- |
| Completeness | 0.998 |
| hkl $_{\text {max }}$ collected | $(39,7,25)$ |
| hkl $l_{\text {max }}$ used | $(29,7,28)$ |
| Lim d $\mathrm{d}_{\text {max }}$ collected | 100.0 |
| $\mathrm{~d}_{\text {max }}$ used | 16.86 |
| Friedel pairs | 3742 |
| Inconsistent equivalents | 0 |


| Unique reflections | 3340 |
| :--- | :--- |
|  |  |
| Mean $\mathrm{I} / \sigma$ | 25.28 |
| $\mathrm{hk} 1_{\min }$ collected | $(-39,-7,-28)$ |
| $\mathrm{hk} 1_{\min }$ used | $(-39,0,0)$ |
| Lim d $_{\text {min }}$ collected | 0.77 |
| $\mathrm{~d}_{\text {min }}$ used | 0.81 |
| Friedel pairs merged | 1 |
| $\mathrm{R}_{\text {int }}$ | 0.0417 |

Table 2.3 (cont'd)

| $\mathrm{R}_{\text {sigma }}$ <br> Omitted reflections | 0.0246 | Intensity transformed <br> Omitted by user <br> (OMIT hkl) | 0 |
| :--- | :--- | :--- | :--- |
| Multiplicity |  |  |  |

Images of the crystal on the diffractometer


Figure 2.7: Images of the crystal on the diffractometer

Table 2.4: Fractional atomic coordinates $\left(\times 10^{4}\right)$ and equivalent isotropic displacement parameters $\left(\AA^{2} \times 10^{3}\right)$ for 2.4d. $U_{e q}$ is defined as $1 / 3$ of the trace of the orthogonalised $U_{i j}$

| Atom | $\mathbf{x}$ | $\mathbf{Y}$ | $\mathbf{z}$ | $\boldsymbol{U}_{\boldsymbol{e q}}$ |
| :--- | :--- | :--- | :--- | :--- |
| Si1 | $6251.6(2)$ | $3547.1(6)$ | $3967.8(2)$ | $31.32(12)$ |
| O1 | $5536.0(4)$ | $4241.9(14)$ | $4154.0(5)$ | $31.5(2)$ |
| C1 | $5847.6(6)$ | $2421(2)$ | $4236.2(8)$ | $30.7(3)$ |
| C2 | $5474.3(7)$ | $562(2)$ | $3721.4(9)$ | $41.6(3)$ |
| C3 | $4970.0(7)$ | $142(3)$ | $3455.3(9)$ | $47.8(4)$ |
| C4 | $4691.5(7)$ | $1294(3)$ | $3685.7(9)$ | $47.5(4)$ |
| C5 | $5109.8(6)$ | $2106(2)$ | $4539.5(8)$ | $37.6(3)$ |
| C6 | $5385.3(6)$ | $4203(2)$ | $4622.3(8)$ | $33.2(3)$ |
| C7 | $5877.6(6)$ | $4880(2)$ | $5471.9(8)$ | $32.3(3)$ |
| C8 | $6123.5(7)$ | $6915(2)$ | $5592.9(10)$ | $39.8(3)$ |
| C9 | $6551.7(7)$ | $7674(2)$ | $6334.7(10)$ | $43.3(4)$ |
| C10 | $6773.1(6)$ | $6447(2)$ | $7018.7(9)$ | $39.0(3)$ |
| C11 | $7224.5(7)$ | $7161(3)$ | $7804.0(11)$ | $52.2(4)$ |
| C12 | $7420.5(7)$ | $5904(3)$ | $8436.9(11)$ | $58.4(5)$ |
| C13 | $7180.1(8)$ | $3889(3)$ | $8325.1(10)$ | $56.1(5)$ |
| C14 | $6746.8(7)$ | $3149(3)$ | $7578.3(9)$ | $45.7(4)$ |
| C15 | $6533.2(6)$ | $4401(2)$ | $6907.3(9)$ | $36.0(3)$ |
| C16 | $6084.0(6)$ | $3673(2)$ | $6122.1(8)$ | $34.0(3)$ |
| C17 | $6598.9(9)$ | $1286(3)$ | $3899.5(14)$ | $57.4(5)$ |
| C18 | $5762.6(6)$ | $5013(2)$ | $3011.0(9)$ | $39.0(3)$ |
| C19 | $6787.4(6)$ | $5447(3)$ | $4770.3(9)$ | $45.3(4)$ |

Table 2.5: Anisotropic displacement parameters $\left(\times 10^{4}\right)$ 2.4d. The anisotropic displacement factor exponent takes the form: $-2 \pi^{2}\left[h^{2} a^{* 2} \times U_{11}+\ldots+2 h k a^{*} \times b^{*} \times U_{12}\right]$

| Atom | $\boldsymbol{U}_{\boldsymbol{1 1}}$ | $\boldsymbol{U}_{\mathbf{2 2}}$ | $\boldsymbol{U}_{\mathbf{3 3}}$ | $\boldsymbol{U}_{\mathbf{2 3}}$ | $\boldsymbol{U}_{\mathbf{1 3}}$ | $\boldsymbol{U}_{\mathbf{1 2}}$ |
| :--- | :--- | :--- | :--- | ---: | :--- | ---: |
| Si1 | $32.8(2)$ | $28.5(2)$ | $37.4(2)$ | $3.82(14)$ | $25.53(18)$ | $1.96(14)$ |
| O1 | $38.8(5)$ | $27.6(4)$ | $36.8(5)$ | $5.9(4)$ | $28.9(4)$ | $4.1(4)$ |
| C1 | $35.8(7)$ | $26.1(6)$ | $34.8(7)$ | $1.9(5)$ | $25.6(6)$ | $0.8(5)$ |
| C2 | $59.4(9)$ | $31.0(7)$ | $47.4(8)$ | $-6.3(6)$ | $41.2(8)$ | $-8.4(7)$ |
| C3 | $58.3(10)$ | $47.3(9)$ | $39.0(8)$ | $-11.6(7)$ | $33.2(8)$ | $-22.8(8)$ |
| C4 | $38.5(8)$ | $61.0(10)$ | $41.9(8)$ | $-2.4(7)$ | $26.5(7)$ | $-16.8(7)$ |
| C5 | $38.6(7)$ | $43.9(8)$ | $38.8(7)$ | $-0.2(6)$ | $29.4(7)$ | $-5.3(6)$ |
| C6 | $36.5(7)$ | $33.8(7)$ | $38.6(7)$ | $5.0(6)$ | $29.0(6)$ | $4.4(6)$ |
| C7 | $37.1(7)$ | $30.3(6)$ | $41.0(7)$ | $0.2(5)$ | $30.9(6)$ | $1.8(6)$ |
| C8 | $48.8(8)$ | $32.7(7)$ | $52.1(9)$ | $2.7(6)$ | $39.7(8)$ | $0.7(6)$ |
| C9 | $47.9(9)$ | $33.2(7)$ | $63.4(10)$ | $-8.0(7)$ | $43.3(8)$ | $-7.3(6)$ |
| C10 | $34.9(7)$ | $39.8(8)$ | $49.9(8)$ | $-10.1(6)$ | $31.7(7)$ | $-0.3(6)$ |
| C11 | $37.3(8)$ | $55.2(10)$ | $62.7(11)$ | $-22.6(9)$ | $33.0(8)$ | $-4.6(7)$ |
| C12 | $37.0(8)$ | $76.2(12)$ | $44.8(9)$ | $-18.6(9)$ | $20.4(8)$ | $7.7(9)$ |
| C13 | $48.8(9)$ | $71.1(12)$ | $41.1(9)$ | $2.1(8)$ | $27.1(8)$ | $17.5(9)$ |
| C14 | $45.1(9)$ | $51.1(9)$ | $41.5(8)$ | $3.9(7)$ | $29.2(7)$ | $8.1(7)$ |
| C15 | $34.5(7)$ | $39.2(7)$ | $41.7(8)$ | $-1.6(6)$ | $28.6(7)$ | $3.9(6)$ |
| C16 | $36.8(7)$ | $30.6(7)$ | $40.6(7)$ | $-0.9(5)$ | $28.5(6)$ | $-2.1(6)$ |
| C17 | $67.0(11)$ | $42.6(9)$ | $94.9(14)$ | $10.9(9)$ | $67.5(12)$ | $12.4(8)$ |
| C18 | $43.2(8)$ | $40.1(7)$ | $36.1(7)$ | $2.3(6)$ | $27.6(7)$ | $-1.4(6)$ |
| C19 | $35.8(8)$ | $48.4(9)$ | $40.8(8)$ | $2.5(7)$ | $21.2(7)$ | $-7.7(7)$ |

Table 2.6: Bond Lengths in Å for 2.4d

| Atom | Atom | Length/Å |
| :--- | :--- | :--- |
| Si1 | C1 | $1.9072(13)$ |
| Si1 | C17 | $1.8680(16)$ |
| Si1 | C18 | $1.8559(15)$ |
| Si1 | C19 | $1.8649(16)$ |
| O1 | C1 | $1.4402(15)$ |
| O1 | C6 | $1.4382(15)$ |
| C1 | C2 | $1.5044(19)$ |
| C2 | C3 | $1.328(2)$ |
| C3 | C4 | $1.493(2)$ |
| C4 | C5 | $1.526(2)$ |
| C5 | C6 | $1.5164(19)$ |
| C6 | C7 | $1.521(2)$ |
| C7 | C8 | $1.4180(19)$ |
| C7 | C16 | $1.3674(19)$ |
| C8 | C9 | $1.360(2)$ |
| C9 | C10 | $1.421(2)$ |
| C10 | C11 | $1.420(2)$ |
| C10 | C15 | $1.420(2)$ |
| C11 | C12 | $1.363(3)$ |
| C12 | C13 | $1.403(3)$ |
| C13 | C14 | $1.365(2)$ |
| C14 | C15 | $1.413(2)$ |
| C15 | C16 | $1.421(2)$ |

Table 2.7: Bond Angles in ${ }^{\circ}$ for 2.4d

| Atom | Atom | Atom | Angle $^{\circ}$ |
| :--- | :--- | :--- | :--- |
| C17 | Si1 | C1 | $109.25(7)$ |
| C18 | Si1 | C1 | $109.17(6)$ |
| C18 | Si1 | C17 | $110.94(8)$ |
| C18 | Si1 | C19 | $109.72(7)$ |
| C19 | Si1 | C1 | $107.68(7)$ |
| C19 | Si1 | C17 | $110.02(9)$ |
| C6 | O1 | C1 | $116.45(9)$ |
| O1 | C1 | Si1 | $103.89(8)$ |
| O1 | C1 | C2 | $112.10(11)$ |
| C2 | C1 | Si1 | $113.30(9)$ |
| C3 | C2 | C1 | $127.04(14)$ |
| C2 | C3 | C4 | $126.15(14)$ |
| C3 | C4 | C5 | $112.19(13)$ |
| C6 | C5 | C4 | $112.71(12)$ |
| O1 | C6 | C5 | $112.66(11)$ |
| O1 | C6 | C7 | $110.56(11)$ |
| C5 | C6 | C7 | $114.90(11)$ |
| C8 | C7 | C6 | $117.69(12)$ |
| C16 | C7 | C6 | $123.87(12)$ |
| C16 | C7 | C8 | $118.41(13)$ |
| C9 | C8 | C7 | $121.37(14)$ |
| C8 | C9 | C10 | $121.18(14)$ |
| C11 | C10 | C9 | $123.17(15)$ |
| C11 | C10 | C15 | $118.70(16)$ |
| C15 | C10 | C9 | $118.13(14)$ |
| C12 | C11 | C10 | $120.44(17)$ |
| C11 | C12 | C13 | $120.73(16)$ |
| C14 | C13 | C12 | $120.41(18)$ |
| C13 | C14 | C15 | $120.54(17)$ |
| C10 | C15 | C16 | $118.91(13)$ |
| C14 | C15 | C10 | $119.18(14)$ |
| C14 | C15 | C16 | $121.91(14)$ |
| C7 | C16 | C15 | $122.00(13)$ |
|  |  |  |  |

Table 2.8: Torsion angles in ${ }^{\circ}$ for 2.4d

| Atom | Atom | Atom | Atom | Angle ${ }^{\circ}$ |
| :--- | :--- | :--- | :--- | :---: |
| Si1 | C1 | C2 | C3 | $-142.51(15)$ |
| O1 | C1 | C2 | C3 | $-25.3(2)$ |
| O1 | C6 | C7 | C8 | $56.42(16)$ |
| O1 | C6 | C7 | C16 | $-125.63(13)$ |
| C1 | O1 | C6 | C5 | $-50.23(15)$ |
| C1 | O1 | C6 | C7 | $79.84(14)$ |
| C1 | C2 | C3 | C4 | $-7.4(3)$ |
| C2 | C3 | C4 | C5 | $-31.9(2)$ |
| C3 | C4 | C5 | C6 | $81.04(17)$ |
| C4 | C5 | C6 | O1 | $-41.76(17)$ |
| C4 | C5 | C6 | C7 | $-169.59(12)$ |
| C5 | C6 | C7 | C8 | $-174.70(12)$ |
| C5 | C6 | C7 | C16 | $3.25(19)$ |
| C6 | O1 | C1 | Si1 | $-154.03(9)$ |
| C6 | O1 | C1 | C2 | $83.27(14)$ |
| C6 | C7 | C8 | C9 | $177.60(13)$ |
| C6 | C7 | C16 | C15 | $-177.84(12)$ |
| C7 | C8 | C9 | C10 | $0.5(2)$ |
| C8 | C7 | C16 | C15 | $0.1(2)$ |
| C8 | C9 | C10 | C11 | $179.14(14)$ |
| C8 | C9 | C10 | C15 | $-0.2(2)$ |
| C9 | C10 | C11 | C12 | $-179.65(15)$ |
| C9 | C10 | C15 | C14 | $179.89(13)$ |
| C9 | C10 | C15 | C16 | $-0.16(19)$ |
| C10 | C11 | C12 | C13 | $-0.1(2)$ |
| C10 | C15 | C16 | C7 | $0.2(2)$ |
| C11 | C10 | C15 | C14 | $0.5(2)$ |
| C11 | C10 | C15 | C16 | $-179.54(12)$ |
| C11 | C12 | C13 | C14 | $0.4(3)$ |
| C12 | C13 | C14 | C15 | $-0.1(2)$ |
| C13 | C14 | C15 | C10 | $-0.3(2)$ |
| C13 | C14 | C15 | C16 | $179.75(14)$ |
| C15 | C15 | C16 | C7 | $-179.84(13)$ |
| C10 | C1 | C8 | C12 | $-0.3(2)$ |
| C9 | $-0.5(2)$ |  |  |  |
| C15 |  |  |  |  |

Table 2.9: Hydrogen fractional atomic coordinates $\left(\times 10^{4}\right)$ and equivalent isotropic displacement parameters $\left(\AA^{2} \times 10^{3}\right)$ for 2.4d. $U_{e q}$ is defined as $1 / 3$ of the trace of the orthogonalised $U_{i j}$

| Atom |  | $\mathbf{Y}$ | $\mathbf{z}$ |  | $\boldsymbol{U}_{\boldsymbol{e q}}$ |
| :--- | :--- | :--- | :--- | :--- | :--- |
| H1 | 6117.96 | 1947.53 | 4803.17 | 37 |  |
| H2 | 5614.28 | -418.73 | 3570.24 | 50 |  |
| H3 | 4767.49 | -999.26 | 3083.29 | 57 |  |
| H4A | 4423.78 | 306.51 | 3623.63 | 57 |  |
| H4B | 4478.04 | 2530.58 | 3324.32 | 57 |  |
| H5A | 4915.08 | 2317.72 | 4733.37 | 45 |  |
| H5B | 5401.68 | 997.83 | 4877.39 | 45 |  |
| H6 | 5095.22 | 5345.81 | 4396.99 | 40 |  |
| H8 | 5986.08 | 7766.51 | 5146.42 | 48 |  |
| H9 | 6705.7 | 9049.64 | 6396.69 | 52 |  |
| H11 | 7391.04 | 8522.55 | 7889 | 63 |  |
| H12 | 7723.27 | 6398 | 8959.48 | 70 |  |
| H13 | 7319.7 | 3034.48 | 8771.89 | 67 |  |
| H14 | 6587.52 | 1781.35 | 7509.17 | 55 |  |
| H16 | 5922.26 | 2306.59 | 6046.79 | 41 |  |
| H17A | 6836.27 | 479.74 | 4399.32 | 86 |  |
| H17B | 6828.38 | 1867.19 | 3799.03 | 86 |  |
| H17C | 6314.09 | 323.48 | 3464.26 | 86 |  |
| H18A | 5471.27 | 4025.42 | 2595.97 | 58 |  |
| H18B | 5967.82 | 5584.8 | 2868.34 | 58 |  |
| H18C | 5590.34 | 6204.57 | 3063.59 | 58 |  |
| H19A | 6602.6 | 6597.83 | 4815.47 | 68 |  |
| H19B | 6997.22 | 6073.94 | 4642.44 | 68 |  |
| H19C | 7044.08 | 4668.02 | 5274.65 | 68 |  |

## Citations

COSMO-V1.61 - Software for the CCD Detector Systems for Determining Data Collection Parameters, Bruker axs, Madison, WI (2000).
O.V. Dolomanov and L.J. Bourhis and R.J. Gildea and J.A.K. Howard and H. Puschmann, Olex2: A complete structure solution, refinement and analysis program, J. Appl. Cryst., (2009), 42, 339-341.

Sheldrick, G.M., A short history of ShelX, Acta Cryst., (2008), A64, 339-341.
Software for the Integration of CCD Detector System Bruker Analytical X-ray Systems, Bruker axs, Madison, WI (after 2013).

## Copies of NMR Spectra





2.1a
${ }^{1} \mathrm{H}$ NMR, $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$



${ }^{13} \mathrm{C}\left\{{ }^{1} \mathrm{H}\right\}$ NMR, $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$










2.2a
${ }^{1} \mathrm{H}$ NMR, $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$








2.2c
${ }^{1} \mathrm{H}$ NMR, $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$







syn-2.3a $d r=8: 1$
${ }^{1} \mathrm{H}$ NMR, $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$



anti-2.3a $d r=8: 1$
${ }^{1} \mathrm{H}$ NMR, $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$



syn-2.3b dr = 9:1
${ }^{1} \mathrm{H}$ NMR, $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$











syn/anti-2.3d dr = 1:1
${ }^{1} \mathrm{H}$ NMR, $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$


syn／anti－2．3d dr＝1：1
${ }^{13} \mathrm{C}\left\{{ }^{1} \mathrm{H}\right\}$ NMR， $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$




药


trans-2.4a
${ }^{1} \mathrm{H}$ NMR, $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$





trans-2.4b
${ }^{1} \mathrm{H}$ NMR, $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$





cis-2.4c
${ }^{13} \mathrm{C}\left\{{ }^{1} \mathrm{H}\right\}$ NMR, $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$


trans-2.4c
${ }^{1} \mathrm{H}$ NMR, $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$









| 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 210 | 200 | 190 | 180 | 170 | 160 | 150 | 140 | 130 | 120 | 110 | 100 | 90 | 80 | 70 | 60 | 50 | 40 | 30 | 20 | 10 | 0 | -10 |

$$
\begin{aligned}
& \text { 芘 }
\end{aligned}
$$







$$
\frac{\mathrm{m}}{8}
$$








2.7c
${ }^{1} \mathrm{H}$ NMR, $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$



咢
 -

2.5c- $d_{1} d r=12: 1$
${ }^{1} \mathrm{H}$ NMR, $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$



2.5c- $d_{1} d r=12: 1$
${ }^{29}$ Si NMR, $99 \mathrm{MHz}, \mathrm{CDCl}_{3}$


т


蒈




2.6c- $d_{2} d r=1: 1$ ${ }^{1} \mathrm{H}$ NMR, $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$



| 1 | 1 | 1 |  | 1 | 1 | 1 | 1 | 1 |  | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 |  | 1 |  | 1 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 250 | 240 | 230 | 220 | 210 | 200 | 190 | 180 | 170 | 160 | 150 | 140 | 130 | 120 | 110 | 100 | 90 | 80 | 70 | 60 | 50 | 40 | 30 | 20 | 10 | 0 | -10 |



$$
11111111
$$


cis-2.4c-d ${ }_{1}$
${ }^{13} \mathrm{C}\left\{{ }^{1} \mathrm{H}\right\}$ NMR, $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$



cis-2.4c-d ${ }_{1}$
${ }^{29} \mathrm{Si}$ NMR, $99 \mathrm{MHz}, \mathrm{CDCl}_{3}$


| 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 90 | 80 | 70 | 60 | 50 | 40 | 30 | 20 | 10 | $\begin{gathered} 0 \\ \text { f1 (ppm) } \end{gathered}$ | -10 | -20 | -30 | -40 | -50 | -60 | -70 | -80 | -90 |






|  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 210 | 200 | 190 | 180 | 170 | 160 | 150 | 140 | 130 | 120 | 10 | $\begin{gathered} 100 \\ \mathrm{f} 1(\mathrm{ppm}) \end{gathered}$ | 90 | 80 | 70 | 60 | 50 | 40 | 30 | 20 | 10 | 0 | -1c |


trans-2.6d
${ }^{1} \mathrm{H}$ NMR, $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$


trans-2.6d
${ }^{13} \mathrm{C}\left\{{ }^{1} \mathrm{H}\right\} \mathrm{NMR}, 126 \mathrm{MHz}, \mathrm{CDCl}_{3}$



$\stackrel{\sim}{n}$


$\begin{array}{lllllllllllllllllllllllllllllll}250 & 240 & 230 & 220 & 210 & 200 & 190 & 180 & 170 & 160 & 150 & 140 & 130 & 120 & 110 & 100 & 90 & 80 & 70 & 60 & 50 & 40 & 30 & 20 & 10 & 0 & -10\end{array}$ f1 (ppm)


| T | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | T |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| :50 | 240 | 230 | 220 | 210 | 200 | 190 | 180 | 170 | 160 | 150 | 140 | 130 | $\begin{gathered} 120 \\ \mathrm{f} 1(\mathrm{ppm}) \end{gathered}$ | 110 | 100 | 90 | 80 | 70 | 60 | 50 | 40 | 30 | 20 | 10 | 0 | -1 |





## CHAPTER 3.

## SILYLCYCLOPROPANES BY SELECTIVE [1,4]-WITTIG REARRANGEMENT OF 4-SILYL-5,6-DIHYDROPYRANS

This work has been adapted with permission from: Mori-Quiroz, L. M.; Maloba, E. W.; Maleczka, R. E., Jr. Silylcyclopropanes by selective [1,4]-Wittig rearrangement of 4-silyl-5,6dihydropyrans. Org. Lett. 2021, 23, 5724 - 5728. Copyright 2021 American Chemical Scoiety. This work was done in collaboration with Dr. Luis Martin Mori-Quiroz. Dr. Mori worked on reaction design and optimization. I worked on additional substrate scope, comparative studies, and derivatization of compound $\mathbf{3 . 2} \mathbf{c}$.

### 3.1. Introduction

[1,4]-Wittig rearrangements of allyl ethers generate enolates, whereas the more common [2,3]- and [1,2]-pathways produce alkoxides. ${ }^{1-6}$ In addition, the [1,4]-Wittig pathway is inherently interwoven with the [1,2]-manifold, which typically predominates. Despite its synthetic potential, the [1,4]-migration path is largely underdeveloped with selective and efficient [1,4]-Wittig pathways being relatively rare and of limited scope. ${ }^{7-9}$

In 2006, our research group found that (1-trimethylsilyl)allylbenzyl ether rearranged selectively through the [1,4]-pathway, forming the acylsilane product. ${ }^{10}$ The apparent ability of the silyl group to allow (1) selective allylic deprotonation and (2) selective [1,4]-migration of the benzyl group led us to explore more complex acyclic analogues. These studies were hampered by lower reactivity of such higher analogues, which we speculated was due to sterics hindering access to conformations necessary for deprotonation. ${ }^{11}$ In contrast, we found that related cyclic ethers rearrange efficiently to give $\alpha$-cyclopropyl acylsilanes or $\alpha$-silylcyclopentenols by [1,4]- and [1,2]Wittig migrations, respectively. ${ }^{12} \mathrm{We}$ also learned that cis/trans diastereomers of these cyclic ethers exhibited very different rates of deprotonation, again presumably reflecting their different ability to
achieve the optimal conformation for deprotonation. Once deprotonated, [1,4]-migration and the competing [1,2]-pathway proceed in a stereoconvergent fashion, with [1,4]-/[1,2]- selectivity being highly sensitive to steric and electronic factors (Scheme 3.1). ${ }^{12}$


Scheme 3.1: Wittig rearrangements of 2-silyl-6-aryl-5,6-dihydropyrans
A question that arose from these studies was whether relocation of the silyl group to the 4 position of the dihydropyran scaffold would favor the [1,4]- or [1,2]-pathway. Herein, we report that 4 -silyl-5,6-dihydropyrans undergo highly selective [1,4]-Wittig rearrangement to afford silylcyclopropyl acetaldehydes.

Silylcyclopropanes are versatile building blocks in organic synthesis. ${ }^{13}$ For instance, they engage in reactions with both nucleophilic and electrophilic partners. Traditional synthetic approaches (Scheme 3.2) involve the cyclopropanation of vinylsilanes ${ }^{14-24}$ and the addition of silyl carbenoids to olefins. ${ }^{25-32}$ Other metal-catalyzed processes have been developed, such as the addition of silyl reagents to cyclopropenes, ${ }^{33-37}$ intramolecular $\mathrm{C}-\mathrm{H}$ silylation of cyclopropanes, ${ }^{38}$ and annulation reactions. ${ }^{39-43}$ To the best of our knowledge, the synthesis of silylcyclopropanes by means of ring contraction had not been reported.


Scheme 3.2: General approaches to silylcyclopropanes

### 3.2. Synthesis of 4-silyl-5,6-dihydro-2H-pyrans

For our purpose, the 4-silyl-5,6-dihydropyrans were prepared from readily available homopropargylic alcohols in three steps involving regioselective alkyne hydrosilylation using Trost catalyst ${ }^{44,45}$ or Tomooka's Pt-catalyzed method, ${ }^{46}$ followed by O-allylation, and ring-closing metathesis (RCM) of the diene precursor using Grubbs' second-generation catalyst (Scheme 3.3). ${ }^{47,48} \mathrm{~A}$ variety of substrates bearing different silyl groups were thus accessed.


Scheme 3.3: Synthetic route to dihydropyrans 3.1

### 3.3. Optimization of reaction conditions for Wittig rearrangement

We started this study by evaluating dihydropyran 3.1a under Wittig conditions used in our previous reports (Scheme 3.4). Treatment of 3.1a with $n$-butyllithium in THF at $-78{ }^{\circ} \mathrm{C}$ for 3.5 h (conditions A) afforded exclusively [1,4]-Wittig product 3.2a in $80 \%$ yield with modest diastereoselectivity (3.3:1), together with a small amount of unreacted 3.1a (7\%). The use of the stronger sec-butyllithium (conditions B) resulted in complete deprotonation followed by rearrangement to afford 3.2a in $91 \%$ yield after only 20 min . Slightly higher diastereoselectivity (4.7:1) was also realized. Under both reaction conditions, we were unable to detect any [1,2]-Wittig product by ${ }^{1} \mathrm{H}$ NMR analysis of the crude reaction mixtures.


Scheme 3.4: $[1,4]$-Wittig rearrangement of model substrate 3.1a

### 3.4. Wittig rearrangements of 4-silyl-5,6-dihydro-2H-pyrans with aryl substituents at the migrating carbon

We next evaluated a variety of substrates bearing different silyl groups at the 4-position and aryl substituents at the migrating carbon (Scheme 3.5). The smaller $\mathrm{EtMe}_{2} \mathrm{Si}$ group afforded silylcyclopropylacetaldehyde 3.2b in $85 \%$ yield and 3.3:1 diastereoselectivity, whereas the more sterically demanding $\mathrm{Et}_{3} \mathrm{Si}$ group led to silylcyclopropane 3.2c in a slightly lower yield (70\%) but higher diastereoselectivity (11:1).


Scheme 3.5: Substrate scope of aryl-substituted dihydropyrans bearing different silyl groups ${ }^{\text {a }}$
${ }^{a}$ Diastereoselectivity determined by ${ }^{1} \mathrm{H}$ NMR of the crude reaction mixture
${ }^{b}$ Reaction run on a 2 mmol scale
${ }^{c}$ A small amount ( $<5 \%$ ) of the presumed [1,2]-Wittig product within a complex mixture was observed but not fully characterized
${ }^{d} 15 \%$ of unreacted dihydropyran $\mathbf{3 . 1 h}$ was recovered
${ }^{e} 2.2$ equiv of $\sec$-BuLi was used
Consistent with prior observations, electron-donating groups such as a 4-methyl on the phenyl group afforded exclusively silylcyclopropanes 3.2d and 3.2e bearing $\mathrm{PhMe}_{2} \mathrm{Si}$ and $\mathrm{BnMe}_{2} \mathrm{Si}$ groups in good yields and low diastereoselectivities. o-Methyl substitution at the aryl group was tolerated, leading to silylcyclopropane $\mathbf{3 . 2 f}$ in $91 \%$ yield and 8.3:1 diastereoselectivity. $m$-Methoxy substitution of the aryl ring, which confers an electron-deficient character to the migrating (benzylic) carbon, afforded predominantly the [1,4]-Wittig product $\mathbf{3 . 2 g}$ in $61 \%$ yield. This was in contrast with the observation in our previous work on 2-
silyl-6-aryl-5,6-dihydropyrans, where a near equal mixture of [1,2], and [1,4] products was observed. ${ }^{12}$ Other (hetero)aromatic substituents at the migrating center such as ferrocenyl and 2-thiophene-yl were tolerated, providing access to silylcyclopropyl acetaldehydes $\mathbf{3 . 2 h}$ and $\mathbf{3 . 2} \mathbf{i n}$ $69 \%$ and $71 \%$ yield, respectively. However, in contrast to all previous examples, the major diastereomer in $\mathbf{3 . 2} \mathbf{i}$ was trans. This outcome is best explained by the fact that 2.2 equiv of $\sec -\mathrm{BuLi}$ was used to ensure complete allylic deprotonation of 3.2i. Such conditions were used because the 2-thiophenyl group undergoes competitive deprotonation at the 5 position, as previously observed. ${ }^{12}$ Therefore, the actual species that undergoes rearrangement is likely the dianion $\mathbf{L i}_{\mathbf{2}} \mathbf{-} \mathbf{3 . 1 i}$ (Scheme 3.5, inset), whose unique electronic characteristics might be responsible for the observed stereochemistry of 3.2i.

We determined the relative stereochemistry of the major diastereomer in 3.2a by NOESY studies and assigned the relative stereochemistry of compounds $\mathbf{3 . 2 b} \mathbf{- 3 . 2}$ by comparison. Specifically, protons corresponding to the alkyl groups attached to silicon (Me, Et) appeared upfield in the NMR spectrum relative to those in the minor diastereomer, presumably due to shielding effects by the cis-oriented aromatic group. In addition, protons corresponding to methyl groups in dimethylsilyl products (i.e., 3.2b, 3.2d-3.2h) became inequivalent due to the expected slow rotation induced by the bulky aryl groups. We further confirmed the structure of compound 3.2c by X-ray crystallographic analysis of its 2,4-dinitrophenylhydrazine derivative (see the experimental section).

### 3.5. Wittig rearrangements of 4-silyl-5,6-dihydro-2H-pyrans with alkyl substituents at the migrating carbon

We next evaluated dihydropyrans bearing alkyl substituents at the migrating carbon (Scheme 3.6). These substrates underwent slow deprotonation under conditions A and B (see Scheme 3.4). However, addition of sec-butyllithium at $-78{ }^{\circ} \mathrm{C}$ and warming to $-10{ }^{\circ} \mathrm{C}$
(conditions C ) allowed deprotonation and rearrangement with excellent [1,4]-selectivity. Dihydropyrans bearing $\mathrm{PhMe}_{2} \mathrm{Si}$ groups on the 4-position and n-propyl and cyclohexyl substituents at the migrating carbon led to the corresponding silylcyclopropanes $\mathbf{3 . 2 j}$ and $\mathbf{3 . 2 k}$ in $83 \%$ and $76 \%$ yields, respectively. The $n$-propyl-substituted dihydropyran ( $\mathbf{3 . 2} \mathbf{j}$ ) rearranged with higher diastereoselectivity compared to the dihydropyrans bearing cycloalkyl groups (Scheme 3.6). Interestingly, cyclopropyl-substituted dihydropyrans $\mathbf{3 . 1 1}$ and $\mathbf{3 . 1} \mathbf{m}$ underwent rearrangement without observable formation of the ring-opened products.
(Conditions C)

Scheme 3.6: Selective [1,4]-Wittig rearrangement of dihydropyrans $\mathbf{3 . 1}$ bearing alkyl groups at the migrating center ${ }^{\text {a }}$
${ }^{\text {a }}$ Diastereoselectivity determined by ${ }^{1} \mathrm{HNMR}$ of the crude reaction mixture

### 3.6. Rearrangement of substrates bearing electron-deficient aryl groups and 2-naphthyl derivative

Dihydropyrans with electron-deficient aryl groups such as 3.1n underwent Wittig rearrangements with flipped [1,4]-/[1,2]-selectivity. Here, the predominant product was the [1,2]Wittig alcohol 3.3n (54\%), followed by the [1,4]-silylcyclopropane 3.2n (17\%) and a small amount of an isomeric [1,2]-Wittig product $\mathbf{3 . 4 n}(6 \%)$. Formation of $\mathbf{4 n}$ indicates that benzylic deprotonation becomes competitive when electron-deficient aryl groups are present. Similarly, 2-pyridyl-substituted dihydropyran (3.1o) predominantly afforded diastereomeric [1,2]-Wittig
products $\mathbf{3 . 3 o}$ and $\mathbf{3 . 3 o}^{\prime}$ (2:1 ratio), resulting from allylic deprotonation. Unreacted $\mathbf{3 . 1 0}$ could not be isolated and instead underwent oxidation during workup and purification to give lactone 3.5o. ${ }^{49}$ Attempts to access the 4-pyridyl analogue using our established route (Scheme 3.3) were unsuccessful due to reluctance of the diene precursor to undergo ring-closing metathesis (see the experimental section). 2-Naphthyl-substituted dihydropyran (3.1p) failed to undergo Wittig rearrangement, and instead, ring-opened products $\mathbf{3 . 6 p}$ and $\mathbf{3 . 7} \mathbf{p}$ were observed (Scheme 3.7).


Scheme 3.7: Rearrangement of substrates bearing electron-deficient aryl groups and 2-naphthyl derivative

### 3.7. Comparative studies on Wittig rearrangements of dihydropyrans

On a last note, it is worth comparing the ability of silyldihydropyrans $\mathbf{3 . 1}$ and isomeric 3.9a/b ${ }^{12}$ to undergo clean rearrangements relative to the unsubstituted analogue 3.8 (Figure 3.1). While $\mathbf{3 . 1}$ and $\mathbf{3 . 9 a} / \mathbf{b}$ undergo Wittig rearrangements in good yields, dihydropyran 3.8 reacts sluggishly to give a low yield of $[1,4]$-Wittig product together with a complex mixture of undetermined byproducts. On the other hand, the exclusive [1,4]-selectivity of $\mathbf{3 . 1}$ is independent of the nature of the silyl groups, while those of 3.9a or 3.9b are very sensitive to the sterics of the silyl group.
yield rearrangement

Figure 3.1: Comparison of yields and [1,4]-/[1,2]-selectivities of $\mathbf{3 . 1}$ vs 2-silyl analogues $\mathbf{3 . 9 a} / \mathbf{3 . 9 b}$ and desilylated analogue 3.8

### 3.8. Proposed mechanism of the $[1,4]$-Wittig rearrangement of 4-silyl-6-aryl(alkyl)-5,6-dihydroprans

In line with our previously proposed mechanistic hypothesis, we maintain that the [1,4]Wittig rearrangement of silyl dihydropyrans proceeds primarily by a stepwise process involving a homolytic $\mathrm{C}-\mathrm{O}$ bond cleavage and intramolecular radical/radical anion recombination (Scheme 3.8), ${ }^{12}$ a process that must be faster than $\sim 7 \times 10^{7} \mathrm{~s}^{-1}$ given that cyclopropyl-bearing substrates did not lead to ring opened products. ${ }^{50}$ As previously reported, ${ }^{12}$ the product distributions from 3.9a or 3.9b suggest that increasing the steric demand of the silyl group prevents [1,2]recombination due to steric clash with the phenyl group. These observations, together with the exclusive [1,4]-selectivity displayed by $\mathbf{3 . 1}$ suggest that the [1,4]-/[1,2]-selectivity is determined
by the ability of the silyl group to transiently and locally stabilize the allylic radical, ${ }^{51}$ guiding recombination toward the Si-bearing carbon.


Scheme 3.8: Proposed mechanism of the [1,4]-Wittig rearrangement of 4-silyl-6-aryl(alkyl)-5,6-dihydroprans

However, there remains the question as to why varying diastereoselectivities are observed with different silyl or aryl groups (Scheme 3.5). For instance, the diastereoselectivity increases nearly 3-fold from the relatively small $\mathrm{SiMe}_{2} \mathrm{Et}$ group (3.2b, $d r=3.3$ :1) to the more sterically demanding $\mathrm{SiEt}_{3}$ group (3.2c, $d r=11: 1$ ). Similarly, the bulkier aryl group 2-methyl phenyl in $\mathbf{3 . 2 f}$ affords a higher diastereoselectivity (8.3:1) relative to the phenyl analogue $\mathbf{3 . 2 b}$ (Scheme 3.5). At this point, we conjecture that a concerted mechanism is operative to a certain extent and leads to the minor diastereomer (trans). In this scenario, bulkier silyl or aryl groups preclude such a competitive mechanism, indirectly leading to higher diastereoselectivity by the dominant, stepwise mechanism.

### 3.9. Conclusion

In conclusion, silylcyclopropane acetaldehydes with a variety of silyl groups can be accessed efficiently by selective [1,4]-Wittig rearrangement of 4-silyl-5,6-dihyropyrans. High selectivity is achievable with substrates whose migrating group has an electron-neutral or electronrich character. In general, the diastereoselectivity of the [1,4]-migration is such that the bulkier groups (silyl and aryl/alkyl) end up in a cis relationship.

The rearrangement proceeds even when the substituent at the 6-position of the dihydropyran is alkyl. The [1,4]-Wittig selectivity is independent of the substituents on silicon,
but it is influenced by the electronic character of the migrating center.

### 3.10. Experimental section

### 3.10.1. General Information

Unless otherwise noticed all reactions were run under a positive atmosphere of nitrogen in oven- dried (at least 4 hours) or flame-dried round bottom flasks or disposable drum vials capped with rubber septa. Solvents were removed by rotary evaporation at temperatures lower than $45^{\circ} \mathrm{C}$. Column chromatography was run on 230-400 mesh silica gel. Tetrahydrofuran and diethyl ether were distilled from sodium-benzophenone ketyl; dichloromethane, benzene, trimethylsilyl chloride were distilled from calcium hydride. Acetone was distilled from drierite and used immediately. Triethylsilane, dimethylbenzylsilane, dimethylphenysilane, dimethylethylsilane, and vinyldimethylchlorosilane were used as received. $n$-Butyllithum ( 1.6 M in hexanes) and secbutyllithium (1.4 M in cyclohexane) were purchased from Aldrich and their concentration calculated by titration with diphenylacetic acid (average of three runs). ${ }^{1} \mathrm{H}$ NMR spectra was collected in 500 MHz and 600 MHz Varian instruments using $\mathrm{CDCl}_{3}$ as solvent, which was referenced at 7.24 ppm (residual chloroform proton) and ${ }^{13} \mathrm{C}$ NMR spectra was collected in $\mathrm{CDCl}_{3}$ at 126 MHz or 151 MHz and referenced at 77.0 ppm . Other deuterated solvents used for NMR analysis were dimethyl sulfoxide (referenced at 2.50 for ${ }^{1} \mathrm{HNMR}$ and 39.51 for ${ }^{13} \mathrm{CNMR}$ ) and benzene (referenced at 7.16 for ${ }^{1} \mathrm{HNMR}$ and 128.39 for ${ }^{13} \mathrm{CNMR}$ ). High resolution mass spectrometric (HRMS) analysis was run in TOF instruments.

### 3.10.2. Synthesis of 4 -silyl-5,6-dihydropyrans and precursors

3.10.2.1. Preparation of aryl homopropargylic alcohols 3.10 - general procedure $A$


Following a reported procedure, ${ }^{52}$ to a vigorously stirred suspension of Zinz dust ( $5.3 \mathrm{~g}, 81$ mmol, 3 equiv) in THF ( 200 mL ) at $0^{\circ} \mathrm{C}$ was added propargyl bromide $(80 \% \mathrm{w} / \mathrm{w}$ in toluene, 12 $\mathrm{g}, 81 \mathrm{mmol}, 3$ equiv) followed by $\mathrm{TiCl}_{4}\left(1 \mathrm{M}\right.$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}, 1.35 \mathrm{~mL}, 1.35 \mathrm{mmol}, 0.05$ equiv). After 10 minutes, the desired aryl aldehyde ( $27 \mathrm{mmol}, 1$ equiv) in THF ( 60 mL ) was adde via syringe slowly. The reaction was followed by TLC and was typically complete in 3-4 hours. The reaction was quenched by adding $\mathrm{NH}_{4} \mathrm{Cl}_{\text {(sat) }}$ ( $\sim 150 \mathrm{~mL}$ ) and slightly acidified with 1 M HCl to remove the emulsion. The mixture was extracted with $\mathrm{Et}_{2} \mathrm{O}(3 \times 150 \mathrm{~mL})$. Combined organic extracts were dried over $\mathrm{MgSO}_{4}$ and concentrated. The product was purified by column chromatography.
3.10.2.2. Preparation of alkyl homopropargylic alcohols 3.10 - general procedure B


Following a reported procedure slightly modified, ${ }^{53}$ to a solution of 1,2-dibromoethane ( $0.15 \mathrm{~mL}, 1.783 \mathrm{mmol}, 0.2$ equiv) in THF ( 20 mL ) was added Zn dust ( $1.17 \mathrm{~g}, 17.83 \mathrm{mmol}, 2$ equiv) with vigorous stirring. The mixture was heated at $65^{\circ} \mathrm{C}$ for 10 minutes and then cooled
down at room temperature. After 20 minutes trimethylsilyl chloride ( $23 \mu \mathrm{~L}, 0.178 \mathrm{mmol}, 0.02$ equiv) was added dropwise and 20 minutes later the reaction was cooled down at $0^{\circ} \mathrm{C}$. Propargyl bromide ( $80 \% \mathrm{w} / \mathrm{w}$ in toluene, $2.65 \mathrm{~g}, 17.83 \mathrm{mmol}, 2$ equiv) was added slowly with vigorous stirring. After 1 hour the mixture was cooled down at $-78^{\circ} \mathrm{C}$ and the desired alkyl aldehyde (8.915 mmol, 1 equiv) was slowly added as a solution in THF ( 10 mL ). The temperature was slowly raised to $0{ }^{\circ} \mathrm{C}$. The reaction was monitored by TLC until completion. The reaction was quenched by adding $\mathrm{NH}_{4} \mathrm{Cl}_{\text {(sat) }}(10 \mathrm{~mL})$ and extracted with $\mathrm{Et}_{2} \mathrm{O}(3 \times 15 \mathrm{~mL})$. Combined organic extracts were dried over $\mathrm{MgSO}_{4}$ and concentrated. The product was purified by column chromatography.

### 3.10.2.3. Preparation of 3-silyl homoallylic alcohols 3.11 - general procedure $\mathbf{C}$



Following a reported procedure, ${ }^{46}$ to a solution of the desired homopropargylic alcohol ( $1.286 \mathrm{mmol}, 1$ equiv) and dimethylvinylsilyl chloride ( $1.929 \mathrm{mmol}, 1.5$ equiv) in THF ( 5.6 mL ) at room temperature was added imidazole ( $1.929 \mathrm{mmol}, 1.5$ equiv) in one portion and the mixture was stirred for 2-12 hours under nitrogen. The reaction was monitored by TLC until completion. The mixture was then filtered through a plug of celite, rinsing with hexanes, and the filtrate was concentrated and suspended in hexanes. Filtration through a plug of celite and concentration afforded the crude dimethylvinylsiloxy that was used in the next step without further purification. In some cases, it was necessary to repeat the treatment with hexanes to remove all insoluble material.

Alternatively, to an 8 mL conical vial fitted with a vane magnetic stir bar and sealed with a rubber septum, the desired homopropargylic alcohol ( 5.0 mmol , 1 equiv) and dimethylvinylsilyl chloride ( $6.0 \mathrm{mmol}, 1.2$ equiv) were dissolved in 5 mL of dry DMF (dried over molecular sieves) under nitrogen. Triethyl amine ( $1.7 \mathrm{~mL}, 12 \mathrm{mmol}, 2.4$ equiv) was added dropwise and the mixture stirred at room temperature for 5 minutes then on a pre-heated oil bath at $80^{\circ} \mathrm{C}$ for 4 hours. The mixture was allowed to cool to room temperature filtered, and the filtrate was transferred to a separating funnel and a cold solution of saturated aqueous sodium hydrogen carbonate ( 5 mL ) was added resulting in an exothermic reaction. The mixture was extracted with pentane ( $30 \mathrm{~mL} \times 3$ ) and the organic portion was washed with brine and dried over anhydrous magnesium sulfate. Filtration and concentration yielded the desired product which was used in the next step without further purification.

To a mixture of the above $O$-dimethylvinylsilyl homopropargylic alcohol ( $1.813 \mathrm{mmol}, 1$ equiv) and the corresponding silane ( $1.813 \mathrm{mmol}, 1$ equiv) was added Karstedt catalyst as a solution in xylenes ( $2 \% \mathrm{w} / \mathrm{w}$ in xylenes, $80.7 \mu \mathrm{~L}, 0.002$ equiv) and the mixture was heated under nitrogen at $80^{\circ} \mathrm{C}$ for 1-1.5 hours. The reaction mixture was cooled down at room temperature and diluted with THF ( 18 mL ) and TBAF ( 1 M in THF, $2.18 \mathrm{mmol}, 2.18 \mathrm{~mL}, 1.2$ equiv) was added slowly. After 20 minutes the solution was concentrated and the residue subjected to column chromatography (EtOAc/hexanes) to afford the desired 3-silyl homoallylic alcohol 3.11.

### 3.10.2.4. Preparation of 3-silyl homoallylic alcohols 3.11 - general procedure $D$



Following a literature procedure, ${ }^{45}$ a solution of the desired homopropargylic alcohol ( $3.625 \mathrm{mmol}, 1$ equiv) and silane ( $4.35 \mathrm{mmol}, 1.2$ equiv) in dry dichloromethane was cooled down at $0{ }^{\circ} \mathrm{C}$ and $\left[\mathrm{Cp} * \mathrm{Ru}(\mathrm{MeCN})_{3}\right] \mathrm{PF}_{6}(36.6 \mathrm{mg}, 0.072 \mathrm{mmol}, 0.02$ equiv) was added quickly, the reaction was kept under nitrogen and the cold bath was removed. After about 1 hour the reaction mixture was concentrated and the product purified by column chromatography (EtOAc/hexanes).

### 3.10.2.5. Etherification of 3-silyl homoallylic alcohols 3.11 to RCM precursors 3.12 - general procedure $\mathbf{E}$



To a solution of 3-silyl homoallylic alcohol 3.11 ( 0.918 mmol , 1 equiv) and allyl bromide (194 $\mu \mathrm{L}, 2.296 \mathrm{mmol}, 2.5$ equiv) in THF $(2 \mathrm{~mL})$ at $0^{\circ} \mathrm{C}$ was added $t$-BuONa ( $265 \mathrm{mg}, 2.75 \mathrm{mmol}$, 3 equiv) and the mixture was vigorously stirred at room temperature. After 4 hours the reaction was quenched with water $(3 \mathrm{~mL})$ and diluted with $\operatorname{EtOAc}(5 \mathrm{~mL})$. The aqueous phase was extracted with EtOAc $(3 \times 5 \mathrm{~mL})$. Combined organic extracts were washed with water ( 3 mL ), brine, dried over $\mathrm{MgSO}_{4}$ and concentrated. The diene product $\mathbf{3 . 1 2}$ was purified by short column chromatography.

Alternatively, sodium hydride, $60 \% \mathrm{w} / \mathrm{w}$ dispersion in mineral oil ( $345 \mathrm{mg}, 9.0 \mathrm{mmol}, 3.0$ equiv) was weighed into a 100 mL dry round bottomed flask fitted with a magnetic stir bar. The flask was sealed and purged with nitrogen, then 20 mL THF was added followed by 0.52 mL of allyl bromide ( $726 \mathrm{mg}, 6.0 \mathrm{mmol}, 2.0$ equiv) and the resulting suspension was cooled on an icebath at $0^{\circ} \mathrm{C}$. To the cold suspension, 3-silyl homoallylic alcohol $\mathbf{3 . 1 1}$ ( $3.0 \mathrm{mmol}, 1$ equiv) in THF
$(10 \mathrm{~mL})$ was added in a dropwise manner and the resulting mixture stirred at $0{ }^{\circ} \mathrm{C}$ to room temperature. After 4 hours, the reaction was quenched with saturated aqueous ammonium chloride $(10 \mathrm{~mL})$ and diluted with ether $(20 \mathrm{~mL})$ and water $(10 \mathrm{~mL})$. The layers were separated and the aqueous layer was extracted with ether ( 20 mL x 3 ). The combined organic layer was washed with brine and dried over anhydrous magnesium sulfate. This was followed by filtration and the filtrate was concentrated under reduced pressure on a rotorvap. The diene product $\mathbf{3 . 1 2}$ was purified by short column chromatography.

### 3.10.2.6. Preparation of 4 -silyl dihydropyrans 3.1 via ring-closing metathesis of dienes 3.12 - general procedure $F$



A round-bottom flask was charged with a magnetic stirred, diene $\mathbf{3 . 1 2}$ ( $100 \mathrm{mg}, 0.31 \mathrm{mmol}$, 1 equiv) and benzene ( 6.2 mL ), and then second-generation Grubbs catalyst ( $10.5 \mathrm{mg}, 0.0124$ mmol, 0.04 equiv) was added in one portion. A condenser was attached, and the system flushed with nitrogen. The reaction was heated in an oil bath at $80^{\circ} \mathrm{C}$ for $1-1.5$ hours. The reaction was then cooled down at room temperature and concentrated. The residue was subjected to column chromatography $\left(\mathrm{CH}_{2} \mathrm{Cl}_{2}\right.$ in hexanes) to afford the desired product $\mathbf{3 . 1}$ as a colorless oil.

## Preparation of 1-phenyl-3-butyn-1-ol (3.10a)



Following general procedure A, 1-Phenyl-3-butyn-1-ol (3.10a) was prepared in $\sim 100 \%$ yield. ${ }^{1} \mathrm{H}$ NMR $\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.41-7.34(\mathrm{~m}, 4 \mathrm{H}), 7.34-7.28(\mathrm{~m}, 1 \mathrm{H}), 4.86(\mathrm{t}, J=6.4 \mathrm{~Hz}$, $1 \mathrm{H}), 2.64(\mathrm{dd}, J=6.4,2.7 \mathrm{~Hz}, 2 \mathrm{H}), 2.56(\mathrm{~s}, 1 \mathrm{H}), 2.08(\mathrm{t}, J=2.7 \mathrm{~Hz}, 1 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( 126 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta 142.4,128.4,127.9,125.7,80.6,72.2,70.9,29.3$. Spectral data are in accord with reported literature values. ${ }^{54}$

## Preparation of 1-(4-methylphenyl)-3-butyn-1-ol (3.10d)



Following general procedure A, 1-(4-methylphenyl)-3-butyn-1-ol (3.10d) was prepared in $94 \%$ yield. ${ }^{1} \mathrm{H} \operatorname{NMR}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.28(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 2 \mathrm{H}), 7.18(\mathrm{~d}, J=7.8 \mathrm{~Hz}, 2 \mathrm{H}), 2.64$ $(\mathrm{dd}, J=6.4,2.6 \mathrm{~Hz}, 2 \mathrm{H}), 2.35(\mathrm{~s}, 3 \mathrm{H}), 2.31(\mathrm{~s}, 1 \mathrm{H}), 2.07(\mathrm{t}, J=2.6 \mathrm{~Hz}, 1 \mathrm{H})$. Spectral data are in accord with reported literature values. ${ }^{55}$

Preparation of 1-(2-methylphenyl)-3-butyn-1-ol (3.10f)


Following General Procedure A, 1-(2-methylphenyl)-3-butyn-1-ol (3.10f) was prepared in $90 \%$ yield. Spectral data are in accord with reported literature values. ${ }^{56}$

Preparation of 1-(3-methoxyphenyl)-3-butyn-1-ol (3.10g)


Following General Procedure A, 1-(3-methoxyphenyl)-3-butyn-1-ol (3.10g) was prepared in $\sim 100 \%$ yield. Spectral data are in accord with reported literature values. ${ }^{56}$

## Preparation of 1-(ferrocenyl)-3-butyn-1-ol (3.10h)



Following General Procedure A, 1-(ferrocenyl)-3-butyn-1-ol (3.10h) was prepared in 87\% yield. ${ }^{1} \mathrm{H}$ NMR $\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 4.53(\mathrm{td}, J=6.2,4.0 \mathrm{~Hz}, 1 \mathrm{H}), 4.33-4.27(\mathrm{~m}, 1 \mathrm{H}), 4.27-$ $4.23(\mathrm{~m}, 1 \mathrm{H}), 4.22-4.18(\mathrm{~m}, 5 \mathrm{H}), 4.17(\mathrm{dt}, J=2.4,0.8 \mathrm{~Hz}, 2 \mathrm{H}), 2.69-2.49(\mathrm{~m}, 2 \mathrm{H}), 2.24(\mathrm{dd}, J$ $=4.1,0.5 \mathrm{~Hz}, 1 \mathrm{H}), 2.06(\mathrm{td}, J=2.7,0.5 \mathrm{~Hz}, 1 \mathrm{H}) .{ }^{13} \mathrm{C} \operatorname{NMR}\left(126 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 91.82,81.1$, $70.61,68.5,68.2,68.1,68.0,67.0,65.8,28.2$. (This compound was used in the next step without further purification. See preparation of compound 3.11h)

Preparation of 1-(2-thiophenyl)-3-butyn-1-ol (3.10i)


Following General Procedure A, 1-(2-thiophenyl)-3-butyn-1-ol (3.10i) was prepared in $91 \%$ yield. Spectral data are in accord with reported literature values. ${ }^{57}$

## Preparation of 1-heptyn-4-ol (3.10j)



Following General Procedure B, 1-heptyn-4-ol ( $\mathbf{( 3 . 1 0 j}$ ) was prepared in $84 \%$ yield. Spectral data are in accord with reported literature values. ${ }^{58}$

## Preparation of 1-cyclohexylbut-3-yn-1-ol (3.10k)



Following general procedure B, 1-cyclohexylbut-3-yn-1-ol (3.10k) was prepared in $84 \%$ yield. ${ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 3.49(\mathrm{dp}, J=7.2,3.8 \mathrm{~Hz}, 1 \mathrm{H}), 2.45(\mathrm{ddd}, J=16.7,4.1,2.7$ $\mathrm{Hz}, 1 \mathrm{H}), 2.35(\mathrm{ddd}, J=16.7,7.5,2.6 \mathrm{~Hz}, 1 \mathrm{H}), 2.05(\mathrm{td}, J=2.7,0.6 \mathrm{~Hz}, 1 \mathrm{H}), 1.97-1.86(\mathrm{~m}, 2 \mathrm{H})$, 1.75 (ddddd, $J=13.2,6.4,5.0,3.3,1.8 \mathrm{~Hz}, 2 \mathrm{H}), 1.66(\mathrm{dddd}, J=13.6,6.9,3.3,1.7 \mathrm{~Hz}, 2 \mathrm{H}), 1.46$ (tdt, $J=11.7,6.7,3.4 \mathrm{~Hz}, 1 \mathrm{H}), 1.24(\mathrm{dddd}, J=16.2,9.0,3.4,1.9 \mathrm{~Hz}, 2 \mathrm{H}), 1.20-1.11(\mathrm{~m}, 1 \mathrm{H})$, $1.07-0.96(\mathrm{~m}, 2 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR (126 MHz, $\left.\mathrm{CDCl}_{3}\right) \delta 81.3,74.0,70.7,42.5,29.0,28.1,26.3,26.1$, 25.9, 24.6. Spectral data are in accord with reported literature values. ${ }^{57}$

## Preparation of 1-cyclopropylbut-3-yn-1-ol (3.101)



Following General Procedure B, 1-cyclopropylbut-3-yn-1-ol (3.10I) was prepared in $73 \%$ yield. ${ }^{1} \mathrm{H}$ NMR $\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 3.06(\mathrm{ddd}, J=8.6,6.8,4.5 \mathrm{~Hz}, 1 \mathrm{H}), 2.56(\mathrm{dddd}, J=16.7,4.6$, $2.6,1.1 \mathrm{~Hz}, 1 \mathrm{H}), 2.47$ (dddd, $J=16.9,6.8,2.7,1.1 \mathrm{~Hz}, 1 \mathrm{H}), 2.06(\mathrm{td}, J=2.6,1.1 \mathrm{~Hz}, 1 \mathrm{H}), 2.02(\mathrm{~s}$, $1 \mathrm{H}), 1.09-0.99(\mathrm{~m}, 1 \mathrm{H}), 0.59-0.51(\mathrm{~m}, 2 \mathrm{H}), 0.37$ (dddt, $J=8.8,5.0,2.5,1.0 \mathrm{~Hz}, 1 \mathrm{H}), 0.26$ (dddt, $J=10.5,4.5,2.3,1.1 \mathrm{~Hz}, 1 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR (126 MHz, $\left.\mathrm{CDCl}_{3}\right) \delta 81.0,74.6,70.5,27.1,16.8,2.9$,
2.5.

## Preparation of 1-(3-chlorophenyl)-3-butyn-1-ol (3.10n)



Following General Procedure A, 1-(3-chlorophenyl)-3-butyn-1-ol was prepared in $\sim 100 \%$ yield. Spectral data are in accord with reported literature values. ${ }^{59}$

Preparation of 1-(pyridin-2-yl)but-3-yn-1-ol (3.100)


Following General Procedure A, 1-(pyridin-2-yl)but-3-yn-1-ol (3.10o) was prepared in $\sim 100 \%$ yield. ${ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 8.54(\mathrm{~d}, J=4.9 \mathrm{~Hz}, 1 \mathrm{H}), 7.70(\mathrm{td}, J=7.7,1.8 \mathrm{~Hz}$, $1 \mathrm{H}), 7.41(\mathrm{~d}, J=7.9 \mathrm{~Hz}, 1 \mathrm{H}), 7.23(\mathrm{ddd}, J=7.6,4.9,1.2 \mathrm{~Hz}, 1 \mathrm{H}), 4.89(\mathrm{t}, J=6.1 \mathrm{~Hz}, 1 \mathrm{H}), 4.49$ $(\mathrm{s}, 1 \mathrm{H}), 2.70(\mathrm{ddd}, J=6.1,2.7,1.0 \mathrm{~Hz}, 2 \mathrm{H}), 2.01(\mathrm{t}, J=2.7 \mathrm{~Hz}, 1 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 159.9,148.3,136.7,122.8,120.7,80.4,71.0,70.8,28.3$. Spectral data are in accord with reported literature values. ${ }^{57}$

## Preparation of 1-(naphthalen-2-yl)but-3-yn-1-ol (3.10p)



Following General Procedure A, 1-(naphthalen-2-yl)but-3-yn-1-ol (3.10p) was prepared in $\sim 100 \%$ yield. ${ }^{1} \mathrm{H}$ NMR $\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.87-7.82(\mathrm{~m}, 4 \mathrm{H}), 7.52-7.48(\mathrm{~m}, 3 \mathrm{H}), 5.03(\mathrm{t}, J=$ $6.4 \mathrm{~Hz}, 1 \mathrm{H}), 2.74(\mathrm{dd}, J=6.4,2.6 \mathrm{~Hz}, 2 \mathrm{H}), 2.70(\mathrm{~s}, 1 \mathrm{H}), 2.10(\mathrm{t}, J=2.6 \mathrm{~Hz}, 1 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR (126 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 139.7,133.1,133.0,128.2,128.0,127.6,126.2,126.0,124.6,123.6,80.6,72.4$, 71.0, 29.3. Spectral data are in accord with reported literature values. ${ }^{57}$

Preparation of 1-(pyridin-4-yl)but-3-yn-1-ol (3.10q)


Following General Procedure A, 1-(pyridin-4-yl)but-3-yn-1-ol (3.10q) was prepared in $\sim 100 \%$ yield as white solid that was soluble in DMF and DMSO. The solid was insoluble in THF, diethyl ether, pentane, hexanes, cyclohexane, benzene, toluene, dichloromethane, and chloroform. Spectroscopic and melting point data for this compound: mp 136 to $137{ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}$ NMR ( 500 MHz , DMSO- $d_{6}$ ) $\delta 8.52(\mathrm{~d}, J=6.2 \mathrm{~Hz}, 2 \mathrm{H}), 7.50(\mathrm{~d}, J=6.2 \mathrm{~Hz}, 2 \mathrm{H}), 5.84(\mathrm{~d}, J=4.5 \mathrm{~Hz}, 1 \mathrm{H}(\mathrm{OH})), 4.75$ $(\mathrm{q}, J=6.0 \mathrm{~Hz}, 1 \mathrm{H}), 2.75(\mathrm{t}, J=2.6 \mathrm{~Hz}, 1 \mathrm{H}), 2.61-2.51(\mathrm{~m}, 2 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( 126 MHz , DMSO-
$\left.d_{6}\right) \delta 155.1,149.1,122.5,81.4,73.5,69.9,28.6$. IR (neat) $3492,3270,3097,1618,1559,1424$, $1392,1334,1065,1028,845,817 \mathrm{~cm}^{-1}$.

Preparation of 3-(dimethyl(phenyl)silyl)-1-phenylbut-3-en-1-ol (3.11a)


Following general procedure C, 1-phenylbut-3-yn-1-ol (3.10a) (1.68 g, $11.5 \mathrm{mmol}, 1$ equiv), dimethylvinylsilyl chloride ( $1.19 \mathrm{~g}, 17.5 \mathrm{mmol}, 1.5$ equiv), imidazole ( $2.08,17.5 \mathrm{mmol}$, 1.5 equiv), in THF ( 50 mL ) afforded the O-dimethylvinylsilyl derivative in nearly quantitative yield, which was used in the next step without further purification. The O-dimethylvinylsilyl intermediate ( $1.22 \mathrm{~g}, 5.3 \mathrm{mmol}, 1$ equiv) was then treated with dimethylphenylsilane $(0.72 \mathrm{~g}, 5.3$ mmol, 1 equiv), and Karstedt catalyst solution ( $236 \mu \mathrm{~L}, 0.002$ equiv) at $80^{\circ} \mathrm{C}$ for 1.5 h and cooled down to room temperature. Treatment with TBAF solution ( $6.3 \mathrm{~mL}, 6.3 \mathrm{mmol}, 1.2$ equiv), workup and purification by silica gel chromatography ( $12 \%$ EtOAc in hexanes) afforded 3.11a as a colorless oil ( $507 \mathrm{mg}, 1.8 \mathrm{mmol}$ ) and a mixture its regioisomer ( $278 \mathrm{mg}, 1.0 \mathrm{mmol}$ ) in $53 \%$ overall yield. 3.11a is a known compound and its spectral data are in accord with reported literature values. ${ }^{60}{ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.53(\mathrm{~m}, 2 \mathrm{H}), 7.36(\mathrm{~m}, 3 \mathrm{H}), 7.28(\mathrm{~m}, 2 \mathrm{H}), 7.21(\mathrm{~m}, 3$ H), $5.84(\mathrm{~m}, 1 \mathrm{H}), 5.62(\mathrm{~m}, 1 \mathrm{H}), 4.53(\mathrm{dt}, J=3.0,9.5 \mathrm{~Hz}, 1 \mathrm{H}), 2.58(\mathrm{~A}$ of ABX system, m, 1 H ), 2.43 (B of ABX system, dd, $J=10.0,14.0 \mathrm{~Hz}, 1 \mathrm{H}), 1.94(\mathrm{~d}, J=2.5 \mathrm{~Hz}, 1 \mathrm{H}), 0.43(\mathrm{~s}, 6 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR (126 MHz, $\mathrm{CDCl}_{3}$ ) $\delta 147.4,144.1,137.6,133.9$ (2 C), 130.0, 129.3, 128.3 (2 C), 128.0 (2
C), 127.3, 125.7 (2 C), $72.2,47.0,-2.90,-2.99$. IR (neat) $3420,3067,2957,1427,1250,1111,833$ $\mathrm{cm}^{-1}$.

Preparation of 3-(ethyldimethylsilyl)-1-phenylbut-3-en-1-ol (3.11b)


Compound 3.11b was prepared by a slight modification of General Procedure C. 1-phenylbut-3-yn-1-ol ( $0.77 \mathrm{~g}, 5.27 \mathrm{mmol}, 1$ equiv), dimethylvinylsilyl chloride ( $0.95 \mathrm{~g}, 7.9 \mathrm{mmol}$, 1.5 equiv), imidazole ( $538 \mathrm{mg}, 7.9 \mathrm{mmol}, 1.5$ equiv), in THF ( 20 mL ) afforded the Odimethylvinylsilyl derivative in nearly quantitative yield. The O-dimethylvinylsilyl intermediate $(0.5 \mathrm{~g}, 2.17 \mathrm{mmol}, 1$ equiv) was then treated with ethyldimethylsilane $(0.21 \mathrm{~g}, 2.39 \mathrm{mmol}, 1.1$ equiv), and Karstedt catalyst solution ( $97 \mu \mathrm{~L}, 0.002$ equiv) at $45^{\circ} \mathrm{C}$ for 1.5 h and cooled down to room temperature. Treatment with TBAF solution ( $2.4 \mathrm{~mL}, 2.4 \mathrm{mmol}, 1.1$ equiv), workup and purification by silica gel chromatography ( $10 \%$ EtOAc in hexanes) afforded 3.11b as a colorless oil ( $193 \mathrm{mg}, 0.82 \mathrm{mmol}$ ) and a mixture of $\mathbf{3 . 1 1 b}$ and its regioisomer ( $200 \mathrm{mg}, 0.85 \mathrm{mmol}$ ) in $77 \%$ overall yield. ${ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.35(\mathrm{~m}, 4 \mathrm{H}), 7.26(\mathrm{~m}, 1 \mathrm{H}), 5.76(\mathrm{~m}, 1 \mathrm{H}), 5.53(\mathrm{dt}$, $J=0.5,8.0, \mathrm{~Hz}, 1 \mathrm{H}), 4.73(\mathrm{ddd}, J=2.0,3.5,10.0 \mathrm{~Hz}, 1 \mathrm{H}), 2.62(\mathrm{dddd}, J=0.5,1.5,3.5,14.0 \mathrm{~Hz}$, $1 \mathrm{H}), 2.43(\mathrm{dd}, J=9.5,14.5 \mathrm{~Hz}, 1 \mathrm{H}), 2.10(\mathrm{~m}, 1 \mathrm{H}), 0.93(\mathrm{t}, J=8.0 \mathrm{~Hz}, 3 \mathrm{H}), 0.61(\mathrm{q}, J=8.0 \mathrm{~Hz}$, $2 \mathrm{H}), 0.11(\mathrm{~s}, 6 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 148.2,128.7$, 128.4 (2 C), 127.4, 125.8 (2 C), 72.2, 47.1, 7.3, 6.9, -3.6, -3.7. IR (film) 3389, 3031, 2955, 1248, 1049, $833 \mathrm{~cm}^{-1}$. HRMS (EI) $\mathrm{m} / \mathrm{z}$
$217.1401\left[(\mathrm{M}-\mathrm{HO})^{+}\right.$; calcd for $\left.\mathrm{C}_{14} \mathrm{H}_{21} \mathrm{Si}, 217.1413\right]$.

## Preparation of 1-phenyl-3-(triethylsilyl)but-3-en-1-ol (3.11c)



Following general procedure C, 1-phenylbut-3-yn-1-ol (3.10a) ( $1.5 \mathrm{~g}, 10.26 \mathrm{mmol}, 1$ equiv), dimethylvinylsilyl chloride ( $1.86 \mathrm{~g}, 15.39 \mathrm{mmol}, 1.5$ equiv), imidazole ( $1.05,15.39 \mathrm{mmol}$, 1.5 equiv), in THF ( 45 mL ) afforded the O-dimethylvinylsilyl derivative in nearly quantitative yield, which was used in the next step without further purification. The O-dimethylvinylsilyl intermediate ( $1.4 \mathrm{~g}, 6.07 \mathrm{mmol}, 1$ equiv) was then treated with triethylsilane $(0.7 \mathrm{~g}, 6.07 \mathrm{mmol}, 1$ equiv), and Karstedt catalyst solution ( $230 \mu \mathrm{~L}, 0.002$ equiv) at $80^{\circ} \mathrm{C}$ for 1.5 h and cooled down to room temperature. Treatment with TBAF solution ( $6.1 \mathrm{~mL}, 6.1 \mathrm{mmol}, 1$ equiv), workup and purification by silica gel chromatography ( $9 \%$ EtOAc in hexanes) afforded $\mathbf{3 . 1 1 c}$ as a colorless oil
 overall yield. ${ }^{1} \mathrm{H}$ NMR ( $600 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.37-7.32(\mathrm{~m}, 4 \mathrm{H}), 7.26(\mathrm{~m}, 1 \mathrm{H}), 5.82(\mathrm{~m}, 1 \mathrm{H})$, $5.52(\mathrm{~d}, J=2.5 \mathrm{~Hz}, 1 \mathrm{H}), 4.73(\mathrm{dd}, J=2.5,8.0 \mathrm{~Hz}, 1 \mathrm{H}), 2.60(\mathrm{~m}, 1 \mathrm{H}), 2.40(\mathrm{dd}, J=8.0,11.5 \mathrm{~Hz}$, $1 \mathrm{H}), 2.13(\mathrm{~s}, 1 \mathrm{H}), 0.95(\mathrm{t}, J=6.5 \mathrm{~Hz}, 9 \mathrm{H}), 0.65(\mathrm{dq}, J=1.5,6.0 \mathrm{~Hz}, 6 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( 126 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta$ 146.2, 144.1, 129.5, 128.4 (2 C), 127.4, 125.8 (2 C), 72.0, 47.2, 7.3, 2.9. IR (film) 3406, 3032, 2953, 2876, 1456, 1008, $721 \mathrm{~cm}^{-1}$. HRMS (EI) $m / z 245.1712\left[(\mathrm{M}-\mathrm{OH})^{+}\right.$; calcd for $\mathrm{C}_{16} \mathrm{H}_{25} \mathrm{Si}$, 245.1726].

## Preparation of 3-(dimethyl(phenyl)silyl)-1-(p-tolyl)but-3-en-1-ol (3.11d)



Following general procedure C , 1 -(p-tolyl)but-3-yn-1-ol (3.10d) ( $1.1 \mathrm{~g}, 6.9 \mathrm{mmol}, 1$ equiv), dimethylvinylsilyl chloride ( $1.25 \mathrm{~g}, 10.35 \mathrm{mmol}, 1.5$ equiv), imidazole ( $0.7,10.35 \mathrm{mmol}$, 1.5 equiv), in THF ( 30 mL ) afforded the O-dimethylvinylsilyl derivative in nearly quantitative yield, which was used in the next step without further purification. The O-dimethylvinylsilyl intermediate ( $957 \mathrm{mg}, 3.92 \mathrm{mmol}$, 1 equiv) was then treated with phenyldimethylsilane ( 534 mg , 3.92 mmol , 1 equiv), and Karstedt catalyst solution ( $175 \mu \mathrm{~L}, 0.002$ equiv) at $80^{\circ} \mathrm{C}$ for 1.1 h and cooled down to room temperature. Treatment with TBAF solution ( $4.7 \mathrm{~mL}, 4.7 \mathrm{mmol}, 1.2$ equiv), workup and purification by silica gel chromatography ( $12 \% \mathrm{EtOAc}$ in hexanes) afforded 3.11d as a colorless oil ( $233 \mathrm{mg}, 0.79 \mathrm{mmol}$ ), and a mixture of $\mathbf{3 . 1 1 d}$ and its regioisomer ( $506 \mathrm{mg}, 1.71$ mmol ) in $77 \%$ overall yield. ${ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.54(\mathrm{~m}, 2 \mathrm{H}), 7.37(\mathrm{~m}, 3 \mathrm{H}), 7.11(\mathrm{~s}$, $4 \mathrm{H}), 5.85(\mathrm{~m}, 1 \mathrm{H}), 5.62(\mathrm{~d}, J=2.5 \mathrm{~Hz}, 1 \mathrm{H}), 4.53(\mathrm{ddd}, J=2.5,3.5,9.5 \mathrm{~Hz}, 1 \mathrm{H}), 2.57(\mathrm{~A}$ of ABX system, m, 1 H ), 2.44 (B of ABX system, dd, $J=10.0,14.0 \mathrm{~Hz}, 1 \mathrm{H}$ ), $2.32(\mathrm{~s}, 3 \mathrm{H}), 1.92$ (d, $J=2.0 \mathrm{~Hz}, 1 \mathrm{H}), 0.438(\mathrm{~s}, 3 \mathrm{H}), 0.429(\mathrm{~s}, 3 \mathrm{H}) .{ }^{13} \mathrm{C} \mathrm{NMR}\left(126 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 147.5,141.1,137.7$, 136.9, 133.9 (2 C), 129.8, 129.2, 128.9 (2 C), 127.9 (2 C), 125.6 (2 C), 72.0, 46.9, 21.1, -2.896, 2.977. IR (neat) $3422,3047,2955,1427,1248,1111,1047,815 \mathrm{~cm}^{-1}$. HRMS (EI) $\mathrm{m} / \mathrm{z} 279.1568$ [(M-OH) ${ }^{+}$; calcd for $\left.\mathrm{C}_{19} \mathrm{H}_{23} \mathrm{Si}, 279.1569\right]$.

## Preparation of 3-(ethyldimethylsilyl)-1-(o-tolyl)but-3-en-1-ol (3.11f)



Following General Procedure D, 1-(o-tolyl)but-3-yn-1-ol (3.10f) ( $400 \mathrm{mg}, 2.5 \mathrm{mmol}, 1$ equiv), ethyldimethylsilane ( $265 \mathrm{mg}, 3 \mathrm{mmol}, 1.2$ equiv), dichloromethane ( 5 mL ) and $\left[\mathrm{Cp} * \mathrm{Ru}(\mathrm{MeCN})_{3}\right] \mathrm{PF}_{6}(25.2 \mathrm{mg}, 0.05 \mathrm{mmol}, 0.02$ equiv) at room temperature for 1 h , followed by silica gel chromatography ( $10 \%$ EtOAc in hexanes) afforded $\mathbf{3 . 1 1 f}$ as colorless oil ( $435 \mathrm{mg}, 1.75$ mmol ) and a mixture of $\mathbf{3 . 1 1 f}$ and its regioisomer ( $110 \mathrm{mg}, 0.44 \mathrm{mmol}$ ) in $88 \%$ overall yield. ${ }^{1} \mathrm{H}$ NMR ( $600 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.52(\mathrm{dd}, J=1.2,7.2 \mathrm{~Hz}, 1 \mathrm{H}), 7.23(\mathrm{~m}, 1 \mathrm{H}), 7.16(\mathrm{dt}, J=1.8,7.2 \mathrm{~Hz}$, $1 \mathrm{H}), 7.12(\mathrm{dd}, J=0.6,7.8 \mathrm{~Hz}, 1 \mathrm{H}), 5.82(\mathrm{~m}, 1 \mathrm{H}), 5.58(\mathrm{~d}, J=3.0 \mathrm{~Hz}, 1 \mathrm{H}), 4.96(\mathrm{dd}, J=3.0,9.6$ Hz, 1 H ), 2.59 (m, 1 H ), 2.36 (dd, $J=10.2,13.8 \mathrm{~Hz}, 1 \mathrm{H}), 2.35(\mathrm{~s}, 3 \mathrm{H}), 2.05(\mathrm{~s}, 1 \mathrm{H}), 0.94$ (t, $J=$ $8.4 \mathrm{~Hz}, 3 \mathrm{H}), 0.62(\mathrm{q}, J=7.8 \mathrm{~Hz}, 2 \mathrm{H}), 0.12(\mathrm{~s}, 6 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 148.4,142.3$, $134.2,130.3,128.8,127.1,126.3,125.3,68.4,45.4,19.2,7.3,6.9,-3.6,-3.7$. IR (film) 3408,3052 , 2953, 1458, 1248, 1049, $819 \mathrm{~cm}^{-1}$. HRMS (EI) $\mathrm{m} / \mathrm{z} 231.1555\left[(\mathrm{M}-\mathrm{OH})^{+}\right.$; calcd for $\mathrm{C}_{15} \mathrm{H}_{23} \mathrm{Si}$, 231.1569].

Preparation of 3-(ethyldimethylsilyl)-1-(3-methoxyphenyl)but-3-en-1-ol (3.11g)


Following General Procedure D, 1-(3-methoxyphenyl)but-3-yn-1-ol (3.10g) (400 mg, 2.27
mmol, 1 equiv), ethyldimethylsilane ( $240 \mathrm{mg}, 2.72 \mathrm{mmol}, 1.2$ equiv), dichloromethane ( 4.5 mL ) and $\left[\mathrm{Cp} * \mathrm{Ru}(\mathrm{MeCN})_{3}\right] \mathrm{PF}_{6}(22.9 \mathrm{mg}, 0.045 \mathrm{mmol}, 0.02$ equiv) at room temperature for 1.5 h , followed by silica gel chromatography (12-15\% EtOAc in hexanes) afforded $\mathbf{3 . 1 1 g}$ as colorless
 $53 \%$ overall yield. ${ }^{1} \mathrm{H}$ NMR $\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.24(\mathrm{t}, J=7.0 \mathrm{~Hz}, 1 \mathrm{H}), 6.93(\mathrm{~m}, 2 \mathrm{H}), 6.80(\mathrm{~m}$, $1 \mathrm{H}), 5.76(\mathrm{~m}, 1 \mathrm{H}), 5.53(\mathrm{~d}, J=2.5 \mathrm{~Hz}, 1 \mathrm{H}), 4.70(\mathrm{dt}, J=3.0,4.5 \mathrm{~Hz}, 1 \mathrm{H}), 3.81(\mathrm{~s}, 3 \mathrm{H}), 2.62$ $(\mathrm{m}, 1 \mathrm{H}), 2.41(\mathrm{dd}, J=10.0,14.0 \mathrm{~Hz}, 1 \mathrm{H}), 2.08(\mathrm{~d}, J=2.0 \mathrm{~Hz}, 1 \mathrm{H}), 0.93(\mathrm{t}, J=8.0 \mathrm{~Hz}, 3 \mathrm{H}), 0.60$ $(\mathrm{q}, J=8.0 \mathrm{~Hz}, 2 \mathrm{H}), 0.10(\mathrm{~s}, 6 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta$ 159.7, 148.2, 145.9, 129.4, 128.7, 118.1, 112.8, 111.3, 72.1, 55.2, 47.0, 7.3, 6.9, -3.6, -3.7. IR (film) 3049, 2955, 1603, 1255, 1045, $777 \mathrm{~cm}^{-1}$. HRMS (EI) $m / z 246.1441$ [(M-H2O) $)^{+}$; calcd for $\mathrm{C}_{15} \mathrm{H}_{22} \mathrm{OSi}$, 246.1440].

Preparation of 3-(dimethyl(phenyl)silyl)-1-(ferrocenyl)but-3-en-1-ol (3.11h)


Following general procedure C, 1-(ferrocenyl)but-3-yn-1-ol (3.10h) (1.6 g, $6.3 \mathrm{mmol}, 1$ equiv), dimethylvinylsilyl chloride ( $1.14 \mathrm{~g}, 9.4 \mathrm{mmol}, 1.5$ equiv), imidazole ( $642 \mathrm{mg}, 9.4 \mathrm{mmol}$, 1.5 equiv), in THF ( 27 mL ) afforded the O-dimethylvinylsilyl derivative in nearly quantitative yield, which was used in the next step without further purification. The O-dimethylvinylsilyl intermediate ( $6.3 \mathrm{mmol}, 1$ equiv) was then treated with phenyldimethylsilane ( $858 \mathrm{mg}, 6.3 \mathrm{mmol}$, 1 equiv), and Karstedt catalyst solution ( $281 \mu \mathrm{~L}, 0.013 \mathrm{mmol}, 0.002$ equiv) at $75^{\circ} \mathrm{C}$ for 2 h and cooled down to room temperature. Treatment with TBAF solution ( $7.6 \mathrm{~mL}, 7.6 \mathrm{mmol}, 1.2$ equiv),
workup and purification by silica gel chromatography ( $10 \% \mathrm{EtOAc}$ in hexanes) afforded a mixture of 3.11h and its regioisomer (1.0:0.2 ratio) as an orange oil ( $1.72 \mathrm{~g}, 4.47 \mathrm{mmol}, 71 \%$ isolated yield). ${ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.53(\mathrm{~m}, 2 \mathrm{H}), 7.35(\mathrm{~m}, 3 \mathrm{H}), 5.79(\mathrm{~m}, 1 \mathrm{H}), 5.57(\mathrm{~d}, J=3.0$ $\mathrm{Hz}, 1 \mathrm{H}), 4.28(\mathrm{dq}, J=4.0,9.5 \mathrm{~Hz}, 1 \mathrm{H}), 4.15(\mathrm{q}, J=1.5 \mathrm{~Hz}, 1 \mathrm{H}), 4.08(\mathrm{t}, J=3.0 \mathrm{~Hz}, 2 \mathrm{H}), 4.07$ $(\mathrm{s}, 5 \mathrm{H}), 4.01(\mathrm{q}, J=3.0 \mathrm{~Hz}, 1 \mathrm{H}), 2.51(\mathrm{~m}, 1 \mathrm{H}), 2.42(\mathrm{dd}, J=9.0,14.0 \mathrm{~Hz}, 1 \mathrm{H}), 1.86(\mathrm{~d}, J=3.5$ $\mathrm{Hz}, 1 \mathrm{H}), 0.40(\mathrm{~s}, 3 \mathrm{H}), 0.39(\mathrm{~s}, 3 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 147.3$, 138.0, $134.0(2 \mathrm{C})$, 129.2, 129.1, 127.9 (2 C), 93.3, 68.3 (5 C), 68.1, 67.7, 67.6, 66.8, 65.7, 45.1, -2.8, -2.9. IR (film) 3412, 3097, 2955, 1427, 1248, 1107, $817 \mathrm{~cm}^{-1}$. HRMS (ESI) $\mathrm{m} / \mathrm{z} 373.1069\left[(\mathrm{M}-\mathrm{OH})^{+}\right.$; calcd for $\left.\mathrm{C}_{22} \mathrm{H}_{25} \mathrm{FeSi}, 373.1075\right]$.

## Preparation of 3-(ethyldimethylsilyl)-1-(thiophen-2-yl)but-3-en-1-ol (3.11i)



Following General Procedure D, 1-(thiophen-2-yl)but-3-yn-1-ol (3.10i) ( $400 \mathrm{mg}, 2.63$ mmol, 1 equiv), ethyldimethylsilane ( $278 \mathrm{mg}, 3.15 \mathrm{mmol}, 1.2$ equiv), dichloromethane ( 5.3 mL ) and $\left[\mathrm{Cp} * \mathrm{Ru}(\mathrm{MeCN})_{3}\right] \mathrm{PF}_{6}(26.5 \mathrm{mg}, 0.053 \mathrm{mmol}, 0.02$ equiv) at room temperature for 1.5 h , followed by silica gel chromatography ( $15 \%$ EtOAc in hexanes) afforded 3.11i as colorless oil (178 mg, $0.61 \mathrm{mmol}, 23 \%$ isolated yield). ${ }^{1} \mathrm{H} \operatorname{NMR}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.22(\mathrm{dd}, J=1.5,5.0 \mathrm{~Hz}$, $1 \mathrm{H}), 6.96(\mathrm{~m}, 2 \mathrm{H}), 5.75(\mathrm{~m}, 1 \mathrm{H}), 5.53(\mathrm{~d}, J=2.5 \mathrm{~Hz}, 1 \mathrm{H}), 4.99(\mathrm{dt}, J=3.0,9.5 \mathrm{~Hz}, 1 \mathrm{H}), 2.73$ (m, 1 H), $2.58(\mathrm{dd}, J=9.5,14.5 \mathrm{~Hz}, 1 \mathrm{H}), 2.17(\mathrm{~d}, J=3.0 \mathrm{~Hz}, 1 \mathrm{H}), 0.93(\mathrm{t}, J=8.0 \mathrm{~Hz}, 3 \mathrm{H}), 0.59$ $(\mathrm{q}, J=8.0 \mathrm{~Hz}, 2 \mathrm{H}), 0.09(\mathrm{~s}, 6 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 147.9,147.6,128.9$ 126.5, 124.4, $123.5,68.5,46.8,7.3,6.8,-3.7,-3.8$. IR (film) 3402, 2955, 1248, 1116, $833 \mathrm{~cm}^{-1}$. HRMS (EI) $\mathrm{m} / \mathrm{z}$
$223.0975\left[(\mathrm{M}-\mathrm{OH})^{+}\right.$; calcd for $\mathrm{C}_{12} \mathrm{H}_{19} \mathrm{SSi}$, 223.0977].

## Preparation of 2-(dimethyl(phenyl)silyl)hept-1-en-4-ol (3.11j)



Following general procedure $C$, hept-1-yn-4-ol ( $\mathbf{3 . 1 0 j}$ ) ( $1.1 \mathrm{~g}, 9.8 \mathrm{mmol}, 1$ equiv), dimethylvinylsilyl chloride ( $1.77 \mathrm{~g}, 14.7 \mathrm{mmol}, 1.5$ equiv), imidazole ( $1 \mathrm{~g}, 14.7 \mathrm{mmol}, 1.5$ equiv), in THF (43 mL) afforded the O-dimethylvinylsilyl derivative in nearly quantitative yield, which was used in the next step without further purification. The O-dimethylvinylsilyl intermediate (1.9 $\mathrm{g}, 9.8 \mathrm{mmol}, 1$ equiv) was then treated with phenyldimethylsilane ( $1.3 \mathrm{~g}, 9.8 \mathrm{mmol}, 1$ equiv), and Karstedt catalyst solution ( $440 \mu \mathrm{~L}, 0.02 \mathrm{mmol}, 0.002$ equiv) at $80^{\circ} \mathrm{C}$ for 1.5 h and cooled down to room temperature. Treatment with TBAF solution ( $10.5 \mathrm{~mL}, 10.5 \mathrm{mmol}, 1.2$ equiv), in THF (100 mL ), workup and purification by silica gel chromatography ( $15 \%$ EtOAc in hexanes) afforded $\mathbf{3 . 1 1} \mathbf{j}$ as a colorless oil ( $845 \mathrm{mg}, 3.40 \mathrm{mmol}$ ) and a mixture of $\mathbf{3 . 1 1 \mathbf { j }}$ and its regioisomer $(929 \mathrm{mg}$, $3.74 \mathrm{mmol})$ in $73 \%$ overall yield. ${ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.49(\mathrm{~m}, 2 \mathrm{H}), 7.34(\mathrm{~m}, 3 \mathrm{H}), 5.78$ $(\mathrm{m}, 1 \mathrm{H}), 5.57(\mathrm{~d}, J=3.0 \mathrm{~Hz}, 1 \mathrm{H}), 3.48(\mathrm{~m}, 1 \mathrm{H}), 2.37(\mathrm{~m}, 1 \mathrm{H}), 2.13(\mathrm{dd}, J=9.0,13.5 \mathrm{~Hz}, 1 \mathrm{H})$, $1.50(\mathrm{~s}, 1 \mathrm{H}), 1.35(\mathrm{~m}, 3 \mathrm{H}), 1.23(\mathrm{~m}, 1 \mathrm{H}), 0.84(\mathrm{t}, J=6.5 \mathrm{~Hz}, 3 \mathrm{H}), 0.39(\mathrm{~s}, 3 \mathrm{H}), 0.38(\mathrm{~s}, 3 \mathrm{H})$. ${ }^{13}{ }^{13}$ NMR (126 MHz, $\mathrm{CDCl}_{3}$ ) $\delta 147.8,137.8,133.8$ (2 C), 129.4, 129.2, 127.9 (2 C), 69.3, 45.0, 39.1, 18.8, 14.0, -2.87, -2.93. IR (film) 3379, 3049, 2957, 2872, 1427, 1250, 1111, $817 \mathrm{~cm}^{-1}$. HRMS (EI) $\mathrm{m} / \mathrm{z} 230.1503$ [(M-H2O) ${ }^{+}$; calcd for $\mathrm{C}_{15} \mathrm{H}_{22} \mathrm{Si}$, 230.1491].

## Preparation of 1-cyclohexyl-3-(dimethyl(phenyl)silyl)but-3-en-1-ol (3.11k)



Following general procedure C, 1-cyclohexylbut-3-yn-1-ol (3.10k) (1.05 g, $6.9 \mathrm{mmol}, 1$ equiv), dimethylvinylsilyl chloride ( $1.25 \mathrm{~g}, 10.39 \mathrm{mmol}, 1.5$ equiv), imidazole ( $0.7 \mathrm{~g}, 10.39 \mathrm{mmol}$, 1.5 equiv), in THF ( 30 mL ) afforded the O-dimethylvinylsilyl derivative in nearly quantitative yield, which was used in the next step without further purification. The O-dimethylvinylsilyl intermediate ( $1.06 \mathrm{~g}, 4.46 \mathrm{mmol}, 1$ equiv) was then treated with phenyldimethylsilane ( 608 mg , 4.46 mmol, 1 equiv), and Karstedt catalyst solution (199 $\mu \mathrm{L}, 0.009 \mathrm{mmol}, 0.002$ equiv) at $75^{\circ} \mathrm{C}$ for 1.5 h and cooled down to room temperature. Treatment with TBAF solution ( $5.35 \mathrm{~mL}, 5.35$ mmol, 1.2 equiv), in THF ( 50 mL ), workup and purification by silica gel chromatography ( $10 \%$ EtOAc in hexanes) afforded a mixture of $\mathbf{3 . 1 1 \mathbf { k }}$ and its regioisomer (1.0:0.14 ratio) as a colorless oil ( $856 \mathrm{mg}, 2.99 \mathrm{mmol}, 67 \%$ isolated yield). ${ }^{1} \mathrm{H} \mathrm{NMR}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.50(\mathrm{~m}, 2 \mathrm{H}), 7.34$ (m, 3 H), 5.79 (quintet, 1 H ), $5.57(\mathrm{~d}, J=3.0 \mathrm{~Hz}, 1 \mathrm{H}), 3.24(\mathrm{~m}, 1 \mathrm{H}), 2.45$ (A of ABX system, m, 1 H ), 2.07 (B of ABX system, dd, $J=10.0,13.5 \mathrm{~Hz}, 1 \mathrm{H}$ ), $1.75-1.67(\mathrm{~m}, 3 \mathrm{H}), 1.61(\mathrm{~m}, 1 \mathrm{H}), 1.55$ $(\mathrm{m}, 1 \mathrm{H}), 1.30-1.23(\mathrm{~m}, 2 \mathrm{H}), 1.22-1.08(\mathrm{~m}, 3 \mathrm{H}), 0.94(\mathrm{~m}, 2 \mathrm{H}), 0.39(\mathrm{~s}, 3 \mathrm{H}), 0.38(\mathrm{~s}, 3 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR (126 MHz, $\mathrm{CDCl}_{3}$ ) $\delta 148.3,137.8,133.8$ (2 C), 129.5, 129.1, 127.9 (2 C), 73.4, 43.3, 41.7, 28.9, 28.1, 26.5, 26.3, 26.1, -2.77, -2.86. IR (neat) 3472, 3049, 2926, 2853, 1427, 1250, 1113, 817 $\mathrm{cm}^{-1}$. HRMS (EI) $m / z 288.1894$ [(M) ${ }^{+}$; calcd for $\mathrm{C}_{18} \mathrm{H}_{28} \mathrm{OSi}$, 288.1909].

## Preparation of 1-cyclopropyl-3-(dimethyl(phenyl)silyl)but-3-en-1-ol (3.111)



Following general procedure C, 1-cyclopropylbut-3-yn-1-ol (3.101) ( $928 \mathrm{mg}, 8.42 \mathrm{mmol}$, 1 equiv), dimethylvinylsilyl chloride ( $1.53 \mathrm{~g}, 12.64 \mathrm{mmol}, 1.5$ equiv), imidazole ( $861 \mathrm{mg}, 12.64$ mmol, 1.5 equiv), in THF ( 34 mL ) afforded the O-dimethylvinylsilyl derivative in nearly quantitative yield, which was used in the next step without further purification. The Odimethylvinylsilyl intermediate ( $833 \mathrm{mg}, 4.2 \mathrm{mmol}, 1$ equiv) was then treated with phenyldimethylsilane ( $572 \mathrm{mg}, 4.2 \mathrm{mmol}, 1$ equiv), and Karstedt catalyst solution ( $187 \mu \mathrm{~L}, 0.008$ mmol, 0.002 equiv) at $80^{\circ} \mathrm{C}$ for 1.5 h and cooled down to room temperature. Treatment with TBAF solution ( $4.2 \mathrm{~mL}, 4.2 \mathrm{mmol}$, 1.2 equiv), in THF ( 70 mL ), workup and purification by silica gel chromatography ( $15 \%$ EtOAc in hexanes) afforded $\mathbf{3 . 1 1 1}$ as a colorless oil ( $527 \mathrm{mg}, 2.14 \mathrm{mmol}$, $51 \%$ isolated yield). ${ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.49(\mathrm{~m}, 2 \mathrm{H}), 7.34(\mathrm{~m}, 3 \mathrm{H}), 5.81(\mathrm{~m}, 1 \mathrm{H})$, $5.56(\mathrm{dd}, J=1.0,3.0 \mathrm{~Hz}, 1 \mathrm{H}), 2.76(\mathrm{dt}, J=3.0,9.0 \mathrm{~Hz}, 1 \mathrm{H}), 2.52(\mathrm{~m}, 1 \mathrm{H}), 2.29(\mathrm{dd}, J=4.5,13.5$ Hz, 1 H$), 1.58(\mathrm{~s}, 1 \mathrm{H}), 0.80(\mathrm{~m}, 1 \mathrm{H}), 0.50-0.36(\mathrm{~m}, 2 \mathrm{H}), 0.38(\mathrm{~s}, 6 \mathrm{H}), 0.21(\mathrm{~m}, 1 \mathrm{H}), 0.00(\mathrm{~m}, 1$ H). ${ }^{13} \mathrm{C}$ NMR ( $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 147.5,137.8,133.8$ (2 C), 129.3, 129.1, 127.9 (2 C), 74.5, 44.3, 17.3, 2.9, 2.2, -2.8, -2.9. IR (film) 3408, 3069, 2957, 2909, 1427, 1250, 1111, $817 \mathrm{~cm}^{-1}$. HRMS (EI) $m / z 228.1331\left[\left(\mathrm{M}-\mathrm{H}_{2} \mathrm{O}\right)^{+}\right.$; calcd for $\mathrm{C}_{15} \mathrm{H}_{20} \mathrm{Si}$, 228.1334].

## Preparation of 1-cyclopropyl-3-(triethylsilyl)but-3-en-1-ol (3.11m)



Following general procedure C, 1-cyclopropylbut-3-yn-1-ol (3.101) (928 mg, 8.42 mmol , 1 equiv), dimethylvinylsilyl chloride ( $1.53 \mathrm{~g}, 12.64 \mathrm{mmol}, 1.5$ equiv), imidazole ( $861 \mathrm{mg}, 12.64$ mmol, 1.5 equiv), in THF ( 34 mL ) afforded the O-dimethylvinylsilyl derivative in nearly quantitative yield, which was used in the next step without further purification. The Odimethylvinylsilyl intermediate ( $833 \mathrm{mg}, 4.2 \mathrm{mmol}$, 1 equiv) was then treated with triethylsilane ( $488 \mathrm{mg}, 4.2 \mathrm{mmol}, 1$ equiv), and Karstedt catalyst solution ( $187 \mu \mathrm{~L}, 0.008 \mathrm{mmol}, 0.002$ equiv) at $80^{\circ} \mathrm{C}$ for 1.5 h and cooled down to room temperature. Treatment with TBAF solution (4.2 mL, 4.2 mmol , 1.0 equiv), in THF ( 70 mL ), workup and purification by silica gel chromatography ( $10 \%$ EtOAc in hexanes) afforded $\mathbf{3 . 1 1 m}$ as a colorless oil ( $570 \mathrm{mg}, 2.52 \mathrm{mmol}, 60 \%$ isolated yield). ${ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 5.76(\mathrm{ddd}, J=2.9,1.6,1.1 \mathrm{~Hz}, 1 \mathrm{H}), 5.46(\mathrm{dt}, J=3.0,0.7 \mathrm{~Hz}, 1 \mathrm{H}), 2.93$ (ddd, $J=9.7,8.1,3.1 \mathrm{~Hz}, 1 \mathrm{H}), 2.53$ (dddd, $J=14.1,3.1,1.6,0.7 \mathrm{~Hz}, 1 \mathrm{H}), 2.26(\mathrm{ddt}, J=13.9,9.6$, $0.8 \mathrm{~Hz}, 1 \mathrm{H}), 0.95-0.84(\mathrm{~m}, 10 \mathrm{H}), 0.59(\mathrm{qd}, J=7.9,3.1 \mathrm{~Hz}, 6 \mathrm{H}), 0.55-0.45(\mathrm{~m}, 2 \mathrm{H}), 0.38-0.30$ $(\mathrm{m}, 1 \mathrm{H}), 0.22-0.14(\mathrm{~m}, 1 \mathrm{H}) .{ }^{13} \mathrm{C} \operatorname{NMR}\left(126 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 146.2,128.8,74.2,44.6,17.4,7.3$, 3.0, 2.2. HRMS (EI) $m / z 206.1484$ [(M-H2O) ${ }^{+}$; calcd for $\mathrm{C}_{13} \mathrm{H}_{24} \mathrm{Si}$, 206.1491].

## Preparation of 1-(3-chlorophenyl)-3-(ethyldimethylsilyl)but-3-en-1-ol (3.11n)



Following general procedure D, 1-(3-chlorophenyl)but-3-yn-1-ol (3.10n) (400 mg, 2.21 mmol, 1 equiv), ethyldimethylsilane ( $234 \mathrm{mg}, 2.66 \mathrm{mmol}$, 1.2 equiv), dichloromethane ( 4.4 mL ) and $\left[\mathrm{Cp} * \mathrm{Ru}(\mathrm{MeCN})_{3}\right] \mathrm{PF}_{6}(22.3 \mathrm{mg}, 0.044 \mathrm{mmol}, 0.02$ equiv) at room temperature for 1.5 h , followed by silica gel chromatography ( $10 \%$ EtOAc in hexanes) afforded 3.11n as colorless oil $(438 \mathrm{mg}, 1.63 \mathrm{mmol})$ and a mixture of its regioisomer ( $47 \mathrm{mg}, 0.17 \mathrm{mmol}$ ) in $81 \%$ overall yield. ${ }^{1} \mathrm{H}$ NMR ( $\left.500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.36(\mathrm{~m}, 2 \mathrm{H}), 7.25-7.21(\mathrm{~m}, 3 \mathrm{H}), 5.76(\mathrm{~m}, 1 \mathrm{H}), 5.55(\mathrm{dt}, J=1.0$, $3.0 \mathrm{~Hz}, 1 \mathrm{H}), 4.69(\mathrm{dd}, J=3.5,10.5 \mathrm{~Hz}, 1 \mathrm{H}), 2.59(\mathrm{~m}, 1 \mathrm{H}), 2.36(\mathrm{ddd}, J=0.5,10.0,14.0 \mathrm{~Hz}, 1$ H), $2.14(\mathrm{~s}, 1 \mathrm{H}), 0.93(\mathrm{t}, J=7.5 \mathrm{~Hz}, 3 \mathrm{H}), 0.60(\mathrm{q}, J=8.0 \mathrm{~Hz}, 2 \mathrm{H}), 0.10(\mathrm{~s}, 6 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR (126 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 147.9,146.2,134.3,129.6,129.1,127.5,126.0,123.9,71.5,47.1,7.3,6.9,-3.6,-$ 3.7. IR (film) $3402,3051,2955,1431,1248,1055,817 \mathrm{~cm}^{-1}$. HRMS (EI) $m / z 251.1014\left[(\mathrm{M}-\mathrm{OH})^{+}\right.$; calcd for $\left.\mathrm{C}_{14} \mathrm{H}_{20} \mathrm{SiCl}, 251.1023\right]$.

Preparation of 1-(pyridin-2-yl)-3-(triethylsilyl)but-3-en-1-ol (3.110)


Following general procedure C, 1-(pyridin-2-yl)but-3-yn-1-ol (3.10o) (2.21 g, 15.0 mmol , 1.0 equiv), dimethylvinylsilyl chloride ( $2.5 \mathrm{~mL}, 18.0 \mathrm{mmol}, 1.2$ equiv), triethyl amine ( 5.1 mL , 36.0 mmol , 2.4 equiv), in DMF ( 15 mL ) afforded $3.08 \mathrm{~g}, 89 \%$ of O-dimethylvinylsilyl derivative, which was used in the next step without further purification. The O-dimethylvinylsilyl intermediate ( $1.39 \mathrm{~g}, 6.0 \mathrm{mmol}, 1.0$ equiv) was then treated with triethylsilane ( $0.96 \mathrm{~mL}, 6.0 \mathrm{mmol}$, 1.0 equiv), and Karstedt catalyst solution ( $270 \mu \mathrm{~L}, 0.012 \mathrm{mmol}, 0.002$ equiv) at $80^{\circ} \mathrm{C}$ for 1.5 h and cooled down to room temperature. Treatment with TBAF solution 1.0 M in THF (7.2 mL, 7.2 mmol, 1.2 equiv), in THF ( 40 mL ), workup and purification by silica gel chromatography ( $50 \%$ EtOAc in hexanes) afforded $350 \mathrm{mg}, 1.32 \mathrm{mmol}$ ( $22 \%$ isolated yield) of homoallylic alcohol 3.11o as a colorless oil and about $75 \%$ of unreacted homopropargyl alcohol. Spectroscopic data for compound 3.110: ${ }^{1} \mathrm{H}$ NMR $\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 8.54(\mathrm{ddd}, J=4.9,1.7,1.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.66(\mathrm{td}, J=$ $7.7,1.8 \mathrm{~Hz}, 1 \mathrm{H}), 7.34-7.28(\mathrm{~m}, 1 \mathrm{H}), 7.18(\mathrm{ddd}, J=7.5,4.8,1.2 \mathrm{~Hz}, 1 \mathrm{H}), 5.80(\mathrm{dt}, J=2.9,1.4$ $\mathrm{Hz}, 1 \mathrm{H}), 5.50(\mathrm{dt}, J=2.8,0.8 \mathrm{~Hz}, 1 \mathrm{H}), 4.83(\mathrm{dd}, J=9.1,4.1 \mathrm{~Hz}, 1 \mathrm{H}), 3.69(\mathrm{~s}, 1 \mathrm{H}), 2.67(\mathrm{dddd}, J$ $=14.4,4.2,1.6,0.8 \mathrm{~Hz}, 1 \mathrm{H}), 2.41(\mathrm{ddt}, J=14.4,9.0,1.1 \mathrm{~Hz}, 1 \mathrm{H}), 0.94(\mathrm{t}, J=7.9 \mathrm{~Hz}, 9 \mathrm{H}), 0.65$ $(\mathrm{q}, J=7.9 \mathrm{~Hz}, 6 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR $\left(126 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 162.2,148.4,145.5,136.5,128.9,122.2,120.5$, 71.9, 45.6, 7.3, 2.9. HRMS (ESI) $m / z 264.1783\left[(\mathrm{M}+\mathrm{H})^{+}\right.$; calcd for $\mathrm{C}_{15} \mathrm{H}_{26} \mathrm{NOSi}$, 264.1784].

Preparation of 3-(dimethyl(phenyl)silyl)-1-(naphthalen-2-yl)but-3-en-1-ol (3.11p)


Following general procedure C, 1-(naphthalen-2-yl)but-3-yn-1-ol (3.10p) (2.65 g, 13.5
mmol, 1 equiv), dimethylvinylsilyl chloride ( $2.8 \mathrm{~mL}, 20.25 \mathrm{mmol}, 1.5$ equiv), imidazole ( 1.38 g , 20.25 mmol , 1.5 equiv), in THF ( 60 mL ) afforded the O-dimethylvinylsilyl derivative in nearly quantitative yield, which was used in the next step without further purification. The O dimethylvinylsilyl intermediate $(1.4 \mathrm{~g}, 5.0 \mathrm{mmol}, 1$ equiv) was then treated with dimethylphenylsilane ( $0.77 \mathrm{~mL}, 5.0 \mathrm{mmol}$, 1 equiv), and Karstedt catalyst solution ( $220 \mu \mathrm{~L}, 0.01$ mmol, 0.002 equiv) at $80^{\circ} \mathrm{C}$ for 1.5 h and cooled down to room temperature. Treatment with TBAF solution ( $6.0 \mathrm{~mL}, 5.0 \mathrm{mmol}, 1.2$ equiv), in THF ( 70 mL ), workup and purification by silica gel chromatography (15\% EtOAc in hexanes) afforded 3.11p as a colorless oil ( $1.26 \mathrm{~g}, 3.8 \mathrm{mmol}, 76 \%$ isolated yield). ${ }^{1} \mathrm{H}$ NMR ( 500 MHz , Chloroform-d) $\delta 7.85-7.80(\mathrm{~m}, 3 \mathrm{H}), 7.66(\mathrm{~s}, 1 \mathrm{H}), 7.63-$ $7.58(\mathrm{~m}, 2 \mathrm{H}), 7.52-7.45(\mathrm{~m}, 2 \mathrm{H}), 7.45-7.40(\mathrm{~m}, 3 \mathrm{H}), 7.37(\mathrm{dd}, J=8.5,1.7 \mathrm{~Hz}, 1 \mathrm{H}), 5.91(\mathrm{dt}, J$ $=2.8,1.3 \mathrm{~Hz}, 1 \mathrm{H}), 5.68(\mathrm{~d}, J=2.8 \mathrm{~Hz}, 1 \mathrm{H}), 4.74(\mathrm{dd}, J=9.9,3.7 \mathrm{~Hz}, 1 \mathrm{H}), 2.72(\mathrm{dddd}, J=14.1$, $3.8,1.5,0.6 \mathrm{~Hz}, 1 \mathrm{H}), 2.56(\mathrm{ddt}, J=14.1,9.6,1.0 \mathrm{~Hz}, 1 \mathrm{H}), 2.15(\mathrm{~s}, 1 \mathrm{H}), 0.51(\mathrm{~s}, 3 \mathrm{H}), 0.49(\mathrm{~s}, 3 \mathrm{H})$. ${ }^{13} \mathrm{C}$ NMR (126 MHz, Chloroform-d) $\delta 147.3$, 141.4, 137.6, 133.9, 133.2, 132.8, 130.0, 129.3, $127.99,128.97,127.87,127.6,126.0,125.6,124.3,123.9,72.3,46.9,-2.90,-2.94$. IR (neat) 3424 , 3051, 2955, 1427, 1248, 1111, $815 \mathrm{~cm}^{-1}$. HRMS (EI) m/z $315.1553\left[(\mathrm{M}-\mathrm{OH})^{+}\right.$; calcd for $\mathrm{C}_{22} \mathrm{H}_{23} \mathrm{Si}$, 315.1569].

Preparation of 1-(pyridin-4-yl)-3-(triethylsilyl)but-3-en-1-ol (3.11q)


Following general procedure C , 1-(pyridin-4-yl)but-3-yn-1-ol (3.10q) $(2.21 \mathrm{~g}, 15.0 \mathrm{mmol}$, 1.0 equiv), dimethylvinylsilyl chloride ( $2.5 \mathrm{~mL}, 18.0 \mathrm{mmol}, 1.2$ equiv), triethyl amine ( 5.1 mL ,
36.0 mmol , 2.4 equiv), in DMF ( 15 mL ) afforded $2.85 \mathrm{~g}, 82 \%$ of O-dimethylvinylsilyl derivative, which was used in the next step without further purification. The O-dimethylvinylsilyl intermediate ( $926 \mathrm{mg}, 4.0 \mathrm{mmol}, 1.0$ equiv) was then treated with triethylsilane ( $0.64 \mathrm{~mL}, 4.0$ mmol, 1.0 equiv), and Karstedt catalyst solution ( $180 \mu \mathrm{~L}, 0.008 \mathrm{mmol}, 0.002$ equiv) at $80^{\circ} \mathrm{C}$ for 1.5 h and cooled down to room temperature. Treatment with TBAF solution 1.0 M in THF (4.8 $\mathrm{mL}, 4.8 \mathrm{mmol}$, 1.2 equiv), in THF ( 40 mL ), workup and purification by silica gel chromatography ( $60 \% \mathrm{EtOAc}$ in hexanes) afforded $706 \mathrm{mg}, 2.68 \mathrm{mmol}$ ( $67 \%$ isolated yield) of homoallylic alcohol 3.11q as a colorless oil. ${ }^{1} \mathrm{H} \operatorname{NMR}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 8.48(\mathrm{~d}, J=6.2 \mathrm{~Hz}, 2 \mathrm{H}), 7.28(\mathrm{~d}, J=6.1$ $\mathrm{Hz}, 2 \mathrm{H}), 5.79(\mathrm{dt}, J=2.7,1.3 \mathrm{~Hz}, 1 \mathrm{H}), 5.53(\mathrm{~d}, J=2.7 \mathrm{~Hz}, 1 \mathrm{H}), 4.71(\mathrm{dd}, J=9.8,3.6 \mathrm{~Hz}, 1 \mathrm{H})$, $3.21(\mathrm{~s}, 1 \mathrm{H}), 2.56(\mathrm{dd}, J=14.2,3.6 \mathrm{~Hz}, 1 \mathrm{H}), 2.34(\mathrm{dd}, J=14.2,9.8 \mathrm{~Hz}, 1 \mathrm{H}), 0.93(\mathrm{t}, J=7.9 \mathrm{~Hz}$, 9H), 0.64 (qd, $J=7.8,2.1 \mathrm{~Hz}, 6 \mathrm{H}) .{ }^{13} \mathrm{C} \operatorname{NMR}\left(126 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 153.4,149.5,145.3,129.9$, $120.8,70.6,46.7,7.3,2.9$ HRMS (ESI) $m / z 264.1788\left[(\mathrm{M}+\mathrm{H})^{+}\right.$; calcd for $\mathrm{C}_{15} \mathrm{H}_{26} \mathrm{NOSi}$, 264.1784].

Preparation of 3-(dimethyl(phenyl)silyl)-1-(pyridin-4-yl)but-3-en-1-ol (3.11r)


The O-dimethylvinylsilyl of 1-(pyridin-4-yl)but-3-yn-1-ol ( $\mathbf{3 . 1 0 q}$ ) intermediate ( 1273 mg , 5.5 mmol , 1.0 equiv) was treated with dimethylphenyl silane ( $0.84 \mathrm{~mL}, 5.5 \mathrm{mmol}, 1.0$ equiv), and Karstedt catalyst solution ( $250 \mu \mathrm{~L}, 0.008 \mathrm{mmol}, 0.002$ equiv) at $80^{\circ} \mathrm{C}$ for 1.5 h and cooled down to room temperature. Treatment with TBAF solution 1.0 M in THF ( $6.6 \mathrm{~mL}, 6.6 \mathrm{mmol}, 1.2$ equiv), in THF ( 40 mL ), workup and purification by silica gel chromatography ( $20 \% \mathrm{EtOAc}$ in hexanes)
afforded $1175 \mathrm{mg}, 4.13 \mathrm{mmol}$ ( $75 \%$ isolated yield) of homoallylic alcohol $\mathbf{3 . 1 1 r}$ as a colorless oil. ${ }^{1} \mathrm{H} \operatorname{NMR}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 8.41(\mathrm{~d}, J=6.1 \mathrm{~Hz}, 2 \mathrm{H}), 7.57-7.50(\mathrm{~m}, 2 \mathrm{H}), 7.41-7.34(\mathrm{~m}, 3 \mathrm{H})$, $7.08(\mathrm{~d}, J=6.1 \mathrm{~Hz}, 2 \mathrm{H}), 5.83(\mathrm{dt}, J=2.6,1.3 \mathrm{~Hz}, 1 \mathrm{H}), 5.65(\mathrm{~d}, J=2.7 \mathrm{~Hz}, 1 \mathrm{H}), 4.48(\mathrm{dd}, J=9.7$, $3.7 \mathrm{~Hz}, 1 \mathrm{H}), 3.01(\mathrm{~s}, 1 \mathrm{H}), 2.55(\mathrm{ddd}, J=13.9,3.7,1.4 \mathrm{~Hz}, 1 \mathrm{H}), 2.37(\mathrm{dd}, J=14.1,9.7 \mathrm{~Hz}, 1 \mathrm{H})$, $0.44(\mathrm{~s}, 3 \mathrm{H}), 0.43(\mathrm{~s}, 3 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 153.2,149.4,146.7,137.4,133.8,130.4$, 129.4, 128.0, 120.7, 70.7, 46.6, -3.0, -3.1. HRMS (ESI) $m / z 284.1475\left[(\mathrm{M}+\mathrm{H})^{+}\right.$; calcd for $\mathrm{C}_{17} \mathrm{H}_{22} \mathrm{NOSi}$, 284.1471].

Preparation of (4-(allyloxy)-4-phenylbut-1-en-2-yl)dimethyl(phenyl)silane (3.12a)


Following General Procedure E, alcohol 3.11a ( $486 \mathrm{mg}, 1.72 \mathrm{mmol}$, 1 equiv), allyl bromide ( $0.36 \mathrm{~mL}, 4.3 \mathrm{mmol}, 2.5$ equiv), THF ( 3.6 mL ), and $t-\mathrm{BuONa}(496 \mathrm{mg}, 5.16 \mathrm{mmol}, 3$ equiv) afforded, after silica gel chromatography $\left(25 \% \mathrm{CH}_{2} \mathrm{Cl}_{2}\right.$ in hexanes $)$, compound 3.12a as a colorless oil (499 mg, $1.55 \mathrm{mmol}, 90 \%$ isolated yield). ${ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.51(\mathrm{~m}, 2 \mathrm{H}), 7.33$ (m, 3 H ), $7.27(\mathrm{t}, J=6.5 \mathrm{~Hz}, 2 \mathrm{H}), 7.22(\mathrm{~m}, 1 \mathrm{H}), 7.13(\mathrm{~m}, 2 \mathrm{H}), 5.77(\mathrm{~m}, 1 \mathrm{H}), 5.69(\mathrm{~m}, 1 \mathrm{H}), 5.47$ $(\mathrm{d}, J=2.5 \mathrm{~Hz}, 1 \mathrm{H}), 5.14(\mathrm{~m}, 1 \mathrm{H}), 5.07(\mathrm{~m}, 1 \mathrm{H}), 4.19(\mathrm{dd}, J=5.5,8.0 \mathrm{~Hz}, 1 \mathrm{H}), 3.74(\mathrm{~A}$ of ABX system, ddt, $J=1.5,5.5,13.0 \mathrm{~Hz}, 1 \mathrm{H}$ ), 3.52 (B of ABX system, ddt, $J=1.5,6.0,13.0 \mathrm{~Hz}, 1 \mathrm{H}$ ), 2.64 (A of ABX system, dd, $J=8.0,14.5 \mathrm{~Hz}, 1 \mathrm{H}$ ), 2.39 (B of ABX system, $\mathrm{dd}, J=5.5,14.5 \mathrm{~Hz}$, $1 \mathrm{H}), 0.36(\mathrm{~s}, 6 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR (126 MHz, $\mathrm{CDCl}_{3}$ ) $\delta 146.3,142.4,138.3,135.0,134.0(2 \mathrm{C}), 129.2$, 128.9, 128.2 (2 C), 127.7 (2 C), 127.4, 126.8 (2 C), 116.5, 80.7, 69.4, 44.7, -2.924, -2.917. IR (neat) $3067,2955,1427,1248,1111,815 \mathrm{~cm}^{-1}$. HRMS (EI) $\mathrm{m} / \mathrm{z} 307.1504\left[\left(\mathrm{M}-\mathrm{CH}_{3}\right)^{+}\right.$; calcd for $\left.\mathrm{C}_{20} \mathrm{H}_{23} \mathrm{OSi}, 307.1518\right]$.

## Preparation of (4-(allyloxy)-4-phenylbut-1-en-2-yl)(ethyl)dimethylsilane (3.12b)



Following general procedure E, alcohol 3.11b ( $181 \mathrm{mg}, 0.77 \mathrm{mmol}$, 1 equiv), allyl bromide ( $0.16 \mathrm{~mL}, 1.93 \mathrm{mmol}, 2.5$ equiv), THF ( 1.6 mL ), and $t$-BuONa ( $223 \mathrm{mg}, 2.32 \mathrm{mmol}, 3$ equiv) afforded, after silica gel chromatography $\left(30 \% \mathrm{CH}_{2} \mathrm{Cl}_{2}\right.$ in hexanes $)$, compound $\mathbf{3 . 1 2 b}$ as a colorless oil ( $165.2 \mathrm{mg}, 0.60 \mathrm{mmol}, 78 \%$ isolated yield). ${ }^{1} \mathrm{H} \mathrm{NMR}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.32(\mathrm{~m}, 2 \mathrm{H}), 7.26$ $(\mathrm{m}, 2 \mathrm{H}), 7.24(\mathrm{~m}, 1 \mathrm{H}), 5.86(\mathrm{~m}, 1 \mathrm{H}), 5.58(\mathrm{~m}, 1 \mathrm{H}), 5.37(\mathrm{dt}, J=0.5,3.0 \mathrm{~Hz}, 1 \mathrm{H}), 5.19(\mathrm{dq}, J=$ $1.5,17.0 \mathrm{~Hz}, 1 \mathrm{H}), 5.11(\mathrm{~m}, 1 \mathrm{H}), 4.36(\mathrm{dd}, J=5.5,8.0 \mathrm{~Hz}, 1 \mathrm{H}), 3.85(\mathrm{ddt}, \mathrm{A}$ of ABX system, $J=$ $1.5,5.0,12.5, \mathrm{~Hz}, 1 \mathrm{H}$ ), 3.72 (ddt, B of ABX system, $J=1.5,6.0,12.5 \mathrm{~Hz}, 1 \mathrm{H}$ ), $2.64(\mathrm{~m}, 1 \mathrm{H})$, $2.39(\mathrm{~m}, 1 \mathrm{H}), 0.89(\mathrm{t}, J=8.0 \mathrm{~Hz}, 3 \mathrm{H}), 0.54(\mathrm{q}, J=8.0 \mathrm{~Hz}, 2 \mathrm{H}), 0.04(\mathrm{~s}, 3 \mathrm{H}), 0.02(\mathrm{~s}, 3 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR (126 MHz, $\mathrm{CDCl}_{3}$ ) $\delta$ 147.2, 142.5, 135.0, 128.3 (2 C), 127.6, 127.4, 126.9 (2 C), 116.6, 81.1, 69.5, 44.5, 7.4, 6.8, -3.7, -3.8. IR (film) 3030, 2955, 2874, 1248, 1087, $819 \mathrm{~cm}^{-1}$. HRMS (EI) $m / z 274.1748\left[(\mathrm{M})^{+}\right.$; calcd for $\mathrm{C}_{17} \mathrm{H}_{26} \mathrm{OSi}$, 274.1753].

Preparation of (4-(allyloxy)-4-phenylbut-1-en-2-yl)triethylsilane (3.12c)


Following general procedure E, alcohol 3.11c ( $402 \mathrm{mg}, 1.53 \mathrm{mmol}$, 1 equiv), allyl bromide
( $0.32 \mathrm{~mL}, 3.83 \mathrm{mmol}, 2.5$ equiv), THF ( 3.1 mL ), and $t$-BuONa ( $442 \mathrm{mg}, 4.59 \mathrm{mmol}, 3$ equiv) afforded, after silica gel chromatography $\left(20 \% \mathrm{CH}_{2} \mathrm{Cl}_{2}\right.$ in hexanes $)$, compound $\mathbf{3 . 1 2} \mathbf{c}$ as a colorless oil ( $407 \mathrm{mg}, 1.35 \mathrm{mmol}, 88 \%$ isolated yield). ${ }^{1} \mathrm{H} \operatorname{NMR}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.33-7.23(\mathrm{~m}, 5 \mathrm{H})$, $5.86(\mathrm{~m}, 1 \mathrm{H}), 5.64(\mathrm{~m}, 1 \mathrm{H}), 5.36(\mathrm{~d}, J=3.5 \mathrm{~Hz}, 1 \mathrm{H}), 5.19(\mathrm{dq}, J=2.0,17.5 \mathrm{~Hz}, 1 \mathrm{H}), 5.10(\mathrm{~m}$, $1 \mathrm{H}), 4.37(\mathrm{dd}, J=5.5,8.0 \mathrm{~Hz}, 1 \mathrm{H}), 3.85(\mathrm{~m}, 1 \mathrm{H}), 3.71(\mathrm{~m}, 1 \mathrm{H}), 2.62(\mathrm{dd}, \mathrm{A}$ of ABX system, $J$ $=8.0,14.5 \mathrm{~Hz}, 1 \mathrm{H}), 2.37(\mathrm{dd}, \mathrm{B}$ of ABX system, $J=5.5,14.5 \mathrm{~Hz}, 1 \mathrm{H}), 0.90(\mathrm{t}, J=8.0 \mathrm{~Hz}, 9 \mathrm{H})$, 0.58 (dq, $J=2.0,8.0 \mathrm{~Hz}, 6 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 144.9,142.6,135.0,128.38$, 128., $127.4,126.9,116.6,81.0,69.5,44.5,7.3,2.8$. IR (film) 3030, 2953, 1454, 1086, 924, $700 \mathrm{~cm}^{-1}$. HRMS (EI) $m / z 302.2051\left[(\mathrm{M})^{+} ;\right.$calcd for $\left.\mathrm{C}_{19} \mathrm{H}_{30} \mathrm{OSi}, 302.2066\right]$.

Preparation of (4-(allyloxy)-4-(p-tolyl)but-1-en-2-yl)dimethyl(phenyl)silane (3.12d)


Following general procedure E, alcohol 3.11d ( $714 \mathrm{mg}, 2.41 \mathrm{mmol}, 1$ equiv), allyl bromide ( $0.51 \mathrm{~mL}, 6.02 \mathrm{mmol}, 2.5$ equiv), THF ( 4.9 mL ), and $t$ - BuONa ( $694 \mathrm{mg}, 7.2 \mathrm{mmol}, 3$ equiv) afforded, after silica gel chromatography $\left(25 \% \mathrm{CH}_{2} \mathrm{Cl}_{2}\right.$ in hexanes), compound 3.12d as a colorless oil ( $698 \mathrm{mg}, 2.07 \mathrm{mmol}, 86 \%$ isolated yield). ${ }^{1} \mathrm{H} \mathrm{NMR}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.51(\mathrm{~m}, 2 \mathrm{H}), 7.34$ (m, 3 H$), 7.09(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 2 \mathrm{H}), 7.03(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 2 \mathrm{H}), 5.78(\mathrm{~m}, 1 \mathrm{H}), 5.70(\mathrm{~m}, 1 \mathrm{H}), 5.47$ (m, 1 H ), $5.14(\mathrm{~m}, 1 \mathrm{H}), 5.07(\mathrm{~m}, 1 \mathrm{H}), 4.17(\mathrm{dd}, J=5.0,8.0 \mathrm{~Hz}, 1 \mathrm{H}), 3.74(\mathrm{~A}$ of ABX system, m, $1 \mathrm{H}), 3.52(\mathrm{~B}$ of ABX system, m, 1 H), 2.64(A of ABX system, ddd, $J=0.5,8.0,14.5 \mathrm{~Hz}, 1 \mathrm{H}$ ), 2.38 (B, of ABX system, ddd, $J=0.5,5.0,14.0 \mathrm{~Hz}, 1 \mathrm{H}), 2.32(\mathrm{~s}, 3 \mathrm{H}), 0.364(\mathrm{~s}, 3 \mathrm{H}), 0.361(\mathrm{~s}, 3$ H). ${ }^{13} \mathrm{C}$ NMR ( $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 146.4,139.3,138.3,137.0,135.1,134.0$ (2 C), 129.1, 128.9,
128.8 (2 C), 127.7 (2 C), 126.8 (2 C), 116.4, 80.4, 69.3, 44.7, 21.1, -2.89, -2.91. IR (neat) 3049, 2955, 2858, 1427, 1248, 1111, $815 \mathrm{~cm}^{-1}$. HRMS (EI) $m / z 321.1667\left[\left(\mathrm{M}-\mathrm{CH}_{3}\right)^{+}\right.$; calcd for $\left.\mathrm{C}_{21} \mathrm{H}_{25} \mathrm{OSi}, 321.1675\right]$.

## Preparation of (4-(allyloxy)-4-(p-tolyl)but-1-en-2-yl)(benzyl)dimethylsilane (3.12e)



Compound 3.12e was prepared by a modification of General Procedure D. A 25 mL round bottom flask was charged with a magnetic stirrer, 1-(1-(allyloxy)but-3-yn-1-yl)-4-methylbenzene ( $500 \mathrm{mg}, 2.5 \mathrm{mmol}$, 1 equiv), dichloromethane ( 5 mL ), and benzydimethylsilane ( $451 \mathrm{mg}, 3 \mathrm{mmol}$, 1.2 equiv). The flask was flushed with nitrogen and $\left[\mathrm{Cp} * \mathrm{Ru}(\mathrm{MeCN})_{3}\right] \mathrm{PF}_{6}(25.2 \mathrm{mg}, 0.05 \mathrm{mmol}$, 0.02 equiv) was added quickly, the reaction was kept under nitrogen. The reaction was run overnight at room temperature. The reaction mixture was concentrated and the crude product purified by silica gel chromatography $\left(30 \% \mathrm{CH}_{2} \mathrm{Cl}_{2}\right.$ in hexanes) to afford 3.12e as a colorless oil ( $297 \mathrm{mg}, 1.48 \mathrm{mmol}, 59 \%$ isolated yield). ${ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.16(\mathrm{~m}, 5 \mathrm{H}), 7.05(\mathrm{t}, J$ $=7.0 \mathrm{~Hz}, 1 \mathrm{H}), 6.96(\mathrm{~d}, J=7.5 \mathrm{~Hz}, 2 \mathrm{H}), 5.86(\mathrm{~m}, 1 \mathrm{H}), 5.64(\mathrm{~s}, 1 \mathrm{H}), 5.38(\mathrm{~s}, 1 \mathrm{H}), 5.21(\mathrm{~d}, J=$ $17.5 \mathrm{~Hz}, 1 \mathrm{H}), 5.11(\mathrm{~d}, J=10.0 \mathrm{~Hz}, 1 \mathrm{H}), 4.35(\mathrm{~m}, 1 \mathrm{H}), 3.86(\mathrm{dd}, \mathrm{A}$ of ABX system, $J=4.5,12.5$ $\mathrm{Hz}, 1 \mathrm{H}), 3.71$ (dd, B of ABX system, $J=5.5,12.5 \mathrm{~Hz}, 1 \mathrm{H}), 2.65(\mathrm{dd}, \mathrm{C}$ of CDX system, $J=8.0$, $14.5 \mathrm{~Hz}, 1 \mathrm{H}), 2.38(\mathrm{dd}, J=5.0,14.5 \mathrm{~Hz}, 1 \mathrm{H}), 2.34(\mathrm{~s}, 3 \mathrm{H}), 2.13(\mathrm{~s}, 2 \mathrm{H}), 0.03(\mathrm{~s}, 3 \mathrm{H}), 0.01(\mathrm{~s}$, $3 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 146.8,140.0,139.3,137.1,135.0,129.0(2 \mathrm{C}), 128.3$ (2 C), 128.2, 128.0 (2 C), 126.9 (2 C), 123.9, 116.6, 81.0, 69.4, 44.4, 25.4, 21.1, -3.5. IR (film) 3025, 2922, 1493, 1248, 1084, $815 \mathrm{~cm}^{-1}$. HRMS (EI) $\mathrm{m} / \mathrm{z} 350.2060\left[\left(\mathrm{M}^{+}\right)\right.$; calcd for $\mathrm{C}_{23} \mathrm{H}_{30} \mathrm{OSi}$,
350.2066].

## Preparation of (4-(allyloxy)-4-(o-tolyl)but-1-en-2-yl)(ethyl)dimethylsilane (3.12f)



Following general procedure E, alcohol $\mathbf{3 . 1 1 f}(412 \mathrm{mg}, 1.66 \mathrm{mmol}$, 1 equiv), allyl bromide ( $0.35 \mathrm{~mL}, 4.15 \mathrm{mmol}, 2.5$ equiv), THF ( 3.3 mL ), and $t$ - BuONa ( $478 \mathrm{mg}, 4.98 \mathrm{mmol}, 3$ equiv) afforded, after silica gel chromatography ( $30 \% \mathrm{CH}_{2} \mathrm{Cl}_{2}$ in hexanes), compound $\mathbf{3 . 1 2 f}$ as a colorless oil ( $439 \mathrm{mg}, 1.53 \mathrm{mmol}, 92 \%$ isolated yield). ${ }^{1} \mathrm{H} \operatorname{NMR}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.40(\mathrm{dd}, J=1.0,7.5$ $\mathrm{Hz}, 1 \mathrm{H}), 7.20(\mathrm{t}, J=7.5 \mathrm{~Hz}, 1 \mathrm{H}), 7.15(\mathrm{dt}, J=1.5,7.5 \mathrm{~Hz}, 1 \mathrm{H}), 7.10(\mathrm{~m}, 1 \mathrm{H}), 5.87(\mathrm{~m}, 1 \mathrm{H})$, 5.67 (quintet, $J=1.5 \mathrm{~Hz}, 1 \mathrm{H}), 5.41(\mathrm{dt}, J=1.0,3.0 \mathrm{~Hz}, 1 \mathrm{H}), 5.19(\mathrm{dq}, J=1.0,17.5 \mathrm{~Hz}, 1 \mathrm{H})$, $5.11(\mathrm{dq}, J=1.5,10.5 \mathrm{~Hz}, 1 \mathrm{H}), 4.64(\mathrm{dd}, J=4.0,9.0 \mathrm{~Hz}, 1 \mathrm{H}), 3.85(\mathrm{ddt}, \mathrm{A}$ of ABX system, $J=$ $1.5,5.0,12.5 \mathrm{~Hz}, 1 \mathrm{H}$ ), 3.69 (ddt, B of ABX system, $J=1.5,6.0,13.0 \mathrm{~Hz}, 1 \mathrm{H}$ ), $2.54(\mathrm{~m}, 1 \mathrm{H}), 2.34$ $(\mathrm{m}, 1 \mathrm{H}), 2.30(\mathrm{~s}, 3 \mathrm{H}), 0.90(\mathrm{t}, J=8.0 \mathrm{~Hz}, 3 \mathrm{H}), 0.56(\mathrm{q}, J=8.0 \mathrm{~Hz}, 2 \mathrm{H}), 0.05(\mathrm{~s}, 3 \mathrm{H}), 0.04(\mathrm{~s} 3$ H). ${ }^{13} \mathrm{C}$ NMR (126 MHz, $\mathrm{CDCl}_{3}$ ) $\delta 147.8,140.8,135.20,135.18,130.3,127.1,127.0,126.22$, $126.18,116.5,77.6,69.4,43.2,19.3,7.4,6.8,-3.75,-3.78$. IR (film) 3049, 2957, 1426, 1248, 1109, $815 \mathrm{~cm}^{-1}$. HRMS (EI) $\mathrm{m} / \mathrm{z} 230.1484\left[\left(\mathrm{M}-\mathrm{OCH}_{2} \mathrm{CHCH}_{2}\right)^{+}\right.$; calcd for $\left.\mathrm{C}_{15} \mathrm{H}_{22} \mathrm{Si}, 230.1491\right]$.

## Preparation of (4-(allyloxy)-4-(3-methoxyphenyl)but-1-en-2-yl)(ethyl)dimethylsilane (3.12g)



Following general procedure E, alcohol 3.11g ( $289 \mathrm{mg}, 1.09 \mathrm{mmol}$, 1 equiv), allyl bromide ( $0.23 \mathrm{~mL}, 2.73 \mathrm{mmol}, 2.5$ equiv), THF ( 2.2 mL ), and $t$-BuONa ( $315 \mathrm{mg}, 3.28 \mathrm{mmol}, 3$ equiv) afforded, after silica gel chromatography ( $30 \% \mathrm{CH}_{2} \mathrm{Cl}_{2}$ in hexanes), compound $\mathbf{3 . 1 2 g}$ as a colorless oil ( $283 \mathrm{mg}, 0.93 \mathrm{mmol}, 85 \%$ isolated yield). ${ }^{1} \mathrm{H} \mathrm{NMR}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.22(\mathrm{t}, J=8.0 \mathrm{~Hz}, 1$ H), $6.85(\mathrm{~m}, 2 \mathrm{H}), 6.79(\mathrm{~m}, 1 \mathrm{H}), 5.86(\mathrm{~m}, 1 \mathrm{H}), 5.60(\mathrm{~m}, 1 \mathrm{H}), 5.38(\mathrm{~d}, J=3.0 \mathrm{~Hz}, 1 \mathrm{H}), 5.20(\mathrm{dq}$, $J=1.5,17.5 \mathrm{~Hz}, 1 \mathrm{H}), 5.11(\mathrm{dq}, J=1.5,10.5 \mathrm{~Hz}, 1 \mathrm{H}), 4.34(\mathrm{dd}, J=5.0,8.0 \mathrm{~Hz}, 1 \mathrm{H}), 3.88$ (ddt, A of ABX system, $J=1.5,5.0,12.5 \mathrm{~Hz}, 1 \mathrm{H}), 3.80(\mathrm{~s}, 3 \mathrm{H}), 3.71$ (ddt, B of ABX system, $J=1.5$, $6.0,12.5 \mathrm{~Hz}, 1 \mathrm{H}), 2.62$ (dd, C of CDX system, $J=8.0,14.5 \mathrm{~Hz}, 1 \mathrm{H}$ ), 2.38 (dd, D of CDX system, $J=5.0,14.5 \mathrm{~Hz}, 1 \mathrm{H}), 0.89(\mathrm{t}, J=8.0 \mathrm{~Hz}, 3 \mathrm{H}), 0.54(\mathrm{q}, J=8.0 \mathrm{~Hz}, 2 \mathrm{H}), 0.04(\mathrm{~s}, 3 \mathrm{H}), 0.03(\mathrm{~s}, 3$ H). ${ }^{13} \mathrm{C}$ NMR ( $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 159.7,147.2,144.3,135.0,129.2,127.5,119.4,116.7,112.9$, $112.2,81.0,69.6,55.2,44.5,7.4,6.8,-3.71,-3.73$. IR (film) 3049, 2953, 1601, 1257, 1045, 819, $779 \mathrm{~cm}^{-1}$. HRMS (EI) $m / z 246.1441$ [(M-C $\left.\mathrm{C}_{3} \mathrm{H}_{6} \mathrm{O}\right)^{+}$; calcd for $\mathrm{C}_{15} \mathrm{H}_{22} \mathrm{OSi}$, 246.1440].

Preparation of (4-(allyloxy)-4-(ferrocenyl)but-1-en-2-yl)dimethyl(phenyl)silane (3.12h)


Following general procedure E, alcohol 3.11h ( $1.72 \mathrm{~g}, 4.4 \mathrm{mmol}$, 1 equiv), allyl bromide ( $0.93 \mathrm{~mL}, 11 \mathrm{mmol}, 2.5$ equiv), THF ( 8.8 mL ), and $t$ - $\mathrm{BuONa}(1.27 \mathrm{~g}, 13.2 \mathrm{mmol}, 3$ equiv) afforded, after silica gel chromatography ( $25-35 \% \mathrm{CH}_{2} \mathrm{Cl}_{2}$ in hexanes), compound $\mathbf{3 . 1 2 h}$ as an orange oil $\left(1.49 \mathrm{~g}, 3.43 \mathrm{mmol}, 78 \%\right.$ isolated yield). ${ }^{1} \mathrm{H}$ NMR $\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.57(\mathrm{~m}, 2 \mathrm{H}), 7.36(\mathrm{~m}, 3$ H), $5.85(\mathrm{~m}, 1 \mathrm{H}), 5.78(\mathrm{ddt}, J=5.5,10.5,17.5 \mathrm{~Hz}, 1 \mathrm{H}), 5.56(\mathrm{~d}, J=2.5 \mathrm{~Hz}, 1 \mathrm{H}), 5.18(\mathrm{dq}, J=$ $1.2,17.0 \mathrm{~Hz}, 1 \mathrm{H}), 5.06(\mathrm{dq}, J=1.5,10.5 \mathrm{~Hz}, 1 \mathrm{H}), 4.12(\mathrm{dd}, J=3.5,9.0 \mathrm{~Hz}, 1 \mathrm{H}), 4.08(\mathrm{~m}, 4 \mathrm{H})$, $3.99(\mathrm{~s}, 5 \mathrm{H}), 3.89(\mathrm{ddt}, J=1.5,5.0,12.5 \mathrm{~Hz}, 1 \mathrm{H}), 3.69(\mathrm{ddt}, J=1.5,5.5,12.5 \mathrm{~Hz}, 1 \mathrm{H}), 2.72(\mathrm{~A}$ of ABX system, m, 1 H), 2.65 (B of ABX system, dd, $J=4.5,15.0 \mathrm{~Hz}, 1 \mathrm{H}), 0.429(\mathrm{~s}, 3 \mathrm{H}), 0.421$ $(\mathrm{s}, 3 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 147.1,138.3,135.4$, 134.0 (2 C), 129.1, 128.4, 127.8 (2 C), 89.3, 76.1, 69.2, 68.6 (5 C), 68.4, 67.8, 67.1, 66.1, 42.1, -2.69, -2.75. IR (neat) 3070, 2955, 1427, 1248, 1107, $815 \mathrm{~cm}^{-1}$. HRMS (EI) $m / z 430.1402\left[(\mathrm{M})^{+}\right.$; calcd for $\left.\mathrm{C}_{25} \mathrm{H}_{30} \mathrm{SiOFe}, 430.1415\right]$.

## Preparation of (4-(allyloxy)-4-(thiophen-2-yl)but-1-en-2-yl)(ethyl)dimethylsilane (3.12i)



Following general procedure E, alcohol 3.11i ( $169 \mathrm{mg}, 0.7 \mathrm{mmol}$, 1 equiv), allyl bromide ( $0.15 \mathrm{~mL}, 1.76 \mathrm{mmol}, 2.5$ equiv), THF ( 1.5 mL ), and $t$ - BuONa ( $203 \mathrm{mg}, 2.11 \mathrm{mmol}, 3$ equiv) afforded, after silica gel chromatography $\left(20 \% \mathrm{CH}_{2} \mathrm{Cl}_{2}\right.$ in hexanes), compound $\mathbf{3 . 1 2}$ i as a colorless oil $\left(166 \mathrm{mg}, 0.59 \mathrm{mmol}, 84 \%\right.$ isolated yield). ${ }^{1} \mathrm{H} \operatorname{NMR}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.23(\mathrm{dd}, J=0.5,5.0$ $\mathrm{Hz}, 1 \mathrm{H}), 6.92$ (dd, $J=3.5,5.0 \mathrm{~Hz}, 1 \mathrm{H}), 6.89$ (dd, $J=1.0,3.0 \mathrm{~Hz}, 1 \mathrm{H}), 5.86(\mathrm{~m}, 1 \mathrm{H}), 5.59$ (dt, $J$ $=1.5,2.5 \mathrm{~Hz}, 1 \mathrm{H}), 5.37(\mathrm{~d}, J=3.0 \mathrm{~Hz}, 1 \mathrm{H}), 5.21(\mathrm{dq}, J=1.5,17.0 \mathrm{~Hz}, 1 \mathrm{H}), 5.13(\mathrm{dq}, J=1.5$, $10.0 \mathrm{~Hz}, 1 \mathrm{H}), 6.64(\mathrm{t}, J=7.0 \mathrm{~Hz}, 1 \mathrm{H}), 3.95(\mathrm{ddt}, \mathrm{A}$ of ABX system, $J=1.5,5.5,12.5 \mathrm{~Hz}, 1 \mathrm{H})$,
3.79 (ddt, B of ABX system, $J=1.5,6.5,12.5 \mathrm{~Hz}, 1 \mathrm{H}), 2.76(\mathrm{~m}, 1 \mathrm{H}), 2.51(\mathrm{~m}, 1 \mathrm{H}), 0.89(\mathrm{t}, J=$ $8.0 \mathrm{~Hz}, 3 \mathrm{H}), 0.54(\mathrm{q}, J=8.0 \mathrm{~Hz}, 2 \mathrm{H}), 0.04(\mathrm{~s}, 3 \mathrm{H}), 0.03(\mathrm{~s}, 3 \mathrm{H}) .{ }^{13} \mathrm{C} \mathrm{NMR}\left(126 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$ $\delta 146.7,146.4,134.8,127.7,126.2,125.3,124.7,117.0,76.3,69.4,44.8,7.3,6.8,-3.7,-3.8$. HRMS (EI) $m / z 280.1310\left[(\mathrm{M})^{+}\right.$; calcd for $\left.\mathrm{C}_{15} \mathrm{H}_{24} \mathrm{SiOS}, 280.1317\right]$.

Preparation of (4-(allyloxy)hept-1-en-2-yl)dimethyl(phenyl)silane (3.12j)


Following general procedure E, alcohol $\mathbf{3 . 1 1 j}$ ( $828 \mathrm{mg}, 3.33 \mathrm{mmol}$, 1 equiv), allyl bromide ( $0.7 \mathrm{~mL}, 8.33 \mathrm{mmol}, 2.5$ equiv), THF ( 6.7 mL ), and $t$ - BuONa ( $961 \mathrm{mg}, 10 \mathrm{mmol}, 3$ equiv) afforded, after silica gel chromatography ( $20 \% \mathrm{CH}_{2} \mathrm{Cl}_{2}$ in hexanes), compound $\mathbf{3 . 1 2} \mathbf{j}$ as a colorless oil (864 $\mathrm{mg}, 3.0 \mathrm{mmol}, 90 \%$ isolated yield). ${ }^{1} \mathrm{H} \operatorname{NMR}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.49(\mathrm{~m}, 2 \mathrm{H}), 7.33(\mathrm{~m}, 3 \mathrm{H})$, $5.81(\mathrm{~m}, 1 \mathrm{H}), 5.74(\mathrm{~m}, 1 \mathrm{H}), 5.49(\mathrm{~d}, J=3.0 \mathrm{~Hz}, 1 \mathrm{H}), 5.16(\mathrm{dq}, J=1.5,17.0 \mathrm{~Hz}, 1 \mathrm{H}), 5.08(\mathrm{~m}$, 1 H ), 3.82 (ddt, A of ABX system, $J=1.5,6.0,13.0 \mathrm{~Hz}, 1 \mathrm{H}$ ), 3.76 (ddt, B of ABX system, $J=$ $1.5,6.0,12.5 \mathrm{~Hz}, 1 \mathrm{H}), 3.22(\mathrm{~m}, 1 \mathrm{H}), 2.44(\mathrm{ddt}, \mathrm{C}$ of CDX system, $J=1.0,5.5,14.0 \mathrm{~Hz}, 1 \mathrm{H})$, 2.15 (dd, D of CDX system, $J=7.5,14.0 \mathrm{~Hz}, 1 \mathrm{H}), 1.35-1.25(\mathrm{~m}, 3 \mathrm{H}), 1.13(\mathrm{~m}, 1 \mathrm{H}), 0.79(\mathrm{t}, J=$ $7.0 \mathrm{~Hz}, 3 \mathrm{H}), 0.38(\mathrm{~s}, 3 \mathrm{H}), 0.37(\mathrm{~s}, 3 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 147.2,138.1,135.5,133.9$ (2 C), 129.0, 128.9, 127.7 (2 C), 116.3, 77.9, 69.9, 41.4, 36.0, 18.5, 14.1, -2.7, -2.9. IR (neat) 3062, 2924, 2851, 1248, 1111, $816 \mathrm{~cm}^{-1}$. HRMS (EI) $\mathrm{m} / \mathrm{z} 273.1660\left[\left(\mathrm{M}-\mathrm{CH}_{3}\right)^{+}\right.$; calcd for $\mathrm{C}_{17} \mathrm{H}_{25} \mathrm{OSi}$, 273.1675].

## Preparation of (4-(allyloxy)-4-cyclohexylbut-1-en-2-yl)dimethyl(phenyl)silane (3.12k)



Following general procedure E, alcohol 3.11k ( $827 \mathrm{mg}, 2.87 \mathrm{mmol}$, 1 equiv), allyl bromide ( $0.61 \mathrm{~mL}, 7.18 \mathrm{mmol}, 2.5$ equiv), THF ( 6 mL ), and $t$ - BuONa ( $827 \mathrm{mg}, 8.6 \mathrm{mmol}, 3$ equiv) afforded, after silica gel chromatography ( $20-25 \% \mathrm{CH}_{2} \mathrm{Cl}_{2}$ in hexanes), compound $\mathbf{3 . 1 2} \mathbf{k}$ as a colorless oil $\left(647 \mathrm{mg}, 1.98 \mathrm{mmol}, 69 \%\right.$ isolated yield). ${ }^{1} \mathrm{H}$ NMR ( $\left.500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.50(\mathrm{~m}, 2 \mathrm{H}), 7.33(\mathrm{~m}$, $3 \mathrm{H}), 5.81(\mathrm{~m}, 1 \mathrm{H}), 5.77(\mathrm{~m}, 1 \mathrm{H}), 5.49(\mathrm{~d}, J=3.0 \mathrm{~Hz}, 1 \mathrm{H}), 5.15(\mathrm{dq}, J=2.0,17.5 \mathrm{~Hz}, 1 \mathrm{H}), 5.05$ (dq, $J=1.5,10.5 \mathrm{~Hz}, 1 \mathrm{H}$ ), 3.79 (A of ABX system, ddt, $J=1.5,5.5,12.5 \mathrm{~Hz}, 1 \mathrm{H}$ ), 3.74 (B of ABX system, ddt, $J=1.5,6.0,13.0 \mathrm{~Hz}, 1 \mathrm{H}), 3.02(\mathrm{~m}, 1 \mathrm{H}), 2.28(\mathrm{~d}, J=6.5 \mathrm{~Hz}, 1 \mathrm{H}), 1.67(\mathrm{~m}, 2$ H), 1.59 (m, 2 H ), $1.39-1.29(\mathrm{~m}, 2 \mathrm{H}), 1.15-1.01(\mathrm{~m}, 4 \mathrm{H}), 0.97(\mathrm{~m}, 1 \mathrm{H}), 0.39(\mathrm{~s}, 3 \mathrm{H}), 0.37(\mathrm{~s}, 3$ H). ${ }^{13} \mathrm{C}$ NMR ( $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 147.5,138.3,135.7,133.9$ (2 C), 128.9, 128.8, 127.7 (2 C), $116.0,82.3,71.1,40.8,38.0,29.2,27.4,26.6,26.5,26.4,-2.71,-2.86$. IR (neat) $3069,2926,2853$, 1450, 1248, 1111, $817 \mathrm{~cm}^{-1}$. HRMS (EI) $m / z 328.2233\left[(\mathrm{M})^{+}\right.$; calcd for $\left.\mathrm{C}_{21} \mathrm{H}_{32} \mathrm{OSi}, 328.2222\right]$.

## Preparation of (4-(allyloxy)-4-cyclopropylbut-1-en-2-yl)dimethyl(phenyl)silane (3.121)



Following general procedure E, alcohol $\mathbf{3 . 1 1 1}$ ( $312 \mathrm{mg}, 1.27 \mathrm{mmol}$, 1 equiv), allyl bromide ( $0.27 \mathrm{~mL}, 3.17 \mathrm{mmol}, 2.5$ equiv), THF ( 2.5 mL ), and $t-\mathrm{BuONa}(365 \mathrm{mg}, 3.8 \mathrm{mmol}, 3$ equiv)
afforded, after silica gel chromatography ( $20-25 \% \mathrm{CH}_{2} \mathrm{Cl}_{2}$ in hexanes), compound $\mathbf{3 . 1 2 1}$ as a colorless oil ( $290 \mathrm{mg}, 1.02 \mathrm{mmol}, 80 \%$ isolated yield). ${ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.50(\mathrm{~m}, 2$ H), 7.32 (m, 3 H ), 5.81 (m, 2 H$), 5.48$ (d, $J=2.5 \mathrm{~Hz}, 1 \mathrm{H}), 5.18$ (dq, $J=1.5,14.5 \mathrm{~Hz}, 1 \mathrm{H}), 5.07$ (dq, $J=1.5,9.0 \mathrm{~Hz}, 1 \mathrm{H}$ ), 4.04 (ddt, A of ABX system, $J=1.0,4.5,10.5 \mathrm{~Hz}, 1 \mathrm{H}$ ), 3.75 (ddt, B of ABX system, $J=1.5,4.5,10.5 \mathrm{~Hz}, 1 \mathrm{H}), 2.60(\mathrm{dt}, J=3.5,6.5 \mathrm{~Hz}, 1 \mathrm{H}), 2.45(\mathrm{dd}, \mathrm{C}$ of CDX system, $J=6.5,12.0 \mathrm{~Hz}, 1 \mathrm{H}), 2.39(\mathrm{dd}, \mathrm{D}$ of CDX system, $J=3.5,12.0 \mathrm{~Hz}, 1 \mathrm{H}), 0.73(\mathrm{~m}, 1 \mathrm{H})$, $0.49(\mathrm{~m}, 1 \mathrm{H}), 0.36(\mathrm{~s}, 6 \mathrm{H}), 0.34(\mathrm{~m}, 1 \mathrm{H}), 0.25(\mathrm{~m}, 1 \mathrm{H}),-0.10(\mathrm{~m}, 1 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( 126 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta 147.0,138.4,135.7,134.0(2 \mathrm{C}), 128.9,128.8,127.7$ (2 C), 116.0, 82.2, 69.7, 42.0, 15.1, 4.4, 1.3, -2.7, -2.8. IR (film) 3069, 3005, 2957, 2858, 1427, 1111, $815 \mathrm{~cm}^{-1}$. HRMS (EI) $\mathrm{m} / \mathrm{z}$ 286.1741 [(M) ${ }^{+}$; calcd for $\mathrm{C}_{18} \mathrm{H}_{26} \mathrm{OSi}$, 286.1753].

## Preparation of (4-(allyloxy)-4-cyclopropylbut-1-en-2-yl)triethylsilane (3.12m)



Following general procedure E , alcohol $\mathbf{3 . 1 1 \mathrm { m }}$ ( $265 \mathrm{mg}, 1.17 \mathrm{mmol}$, 1 equiv), allyl bromide ( $0.25 \mathrm{~mL}, 2.93 \mathrm{mmol}, 2.5$ equiv), THF ( 2.4 mL ), and $t$-BuONa ( $337 \mathrm{mg}, 3.5 \mathrm{mmol}, 3$ equiv) afforded, after silica gel chromatography ( $20-25 \% \mathrm{CH}_{2} \mathrm{Cl}_{2}$ in hexanes), compound $\mathbf{3 . 1 2 m}$ as a colorless oil ( $261 \mathrm{mg}, 0.98 \mathrm{mmol}, 84 \%$ isolated yield). ${ }^{1} \mathrm{H}$ NMR ( $600 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 5.88(\mathrm{~m}, 1$ H), $5.48(\mathrm{dt}, J=1.8,3.0 \mathrm{~Hz}, 1 \mathrm{H}), 5.37(\mathrm{~d}, J=3.0 \mathrm{~Hz}, 1 \mathrm{H}), 5.22(\mathrm{dq}, J=1.8,17.4 \mathrm{~Hz}, 1 \mathrm{H}), 5.09$ (dq, $J=1.8,12.0 \mathrm{~Hz}, 1 \mathrm{H}$ ), 4.16 (ddt, A of ABX system, $J=1.8,6.0,13.2 \mathrm{~Hz}, 1 \mathrm{H}$ ), 3.93 (ddt, B of ABX system, $J=1.2,5.4,12.6 \mathrm{~Hz}, 1 \mathrm{H}), 2.79(\mathrm{dt}, J=4.8,7.8 \mathrm{~Hz}, 1 \mathrm{H}), 2.42(\mathrm{dd}, \mathrm{C}$ of CDX system, $J=7.2,14.4 \mathrm{~Hz}, 1 \mathrm{H}), 2.35(\mathrm{dd}, \mathrm{D}$ of CDX system, $J=4.8,14.4 \mathrm{~Hz}, 1 \mathrm{H}), 0.91(\mathrm{t}, J=7.8$
$\mathrm{Hz}, 9 \mathrm{H}), 0.83(\mathrm{~m}, 1 \mathrm{H}), 0.59(\mathrm{q}, J=7.8 \mathrm{~Hz}, 6 \mathrm{H}), 0.55(\mathrm{~m}, 1 \mathrm{H}), 0.43(\mathrm{~m}, 1 \mathrm{H}), 0.36(\mathrm{~m}, 1 \mathrm{H})$, $0.08(\mathrm{~m}, 1 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR (151 MHz, $\left.\mathrm{CDCl}_{3}\right) \delta 145.6,135.7,128.0,116.1,82.3,69.8,41.9,15.2$, 7.4, 4.4, 3.0, 1.4. IR (film) 3070, 2957, 1427, 1113, $815 \mathrm{~cm}^{-1}$. HRMS (EI) $m / z 266.2062\left[(\mathrm{M})^{+}\right.$; calcd for $\mathrm{C}_{16} \mathrm{H}_{30} \mathrm{OSi}$, 266.2066].

Preparation of (4-(allyloxy)-4-(3-chlorophenyl)but-1-en-2-yl)(ethyl)dimethylsilane (3.12n)


Following general procedure E, alcohol 3.11n ( $423 \mathrm{mg}, 1.57 \mathrm{mmol}$, 1 equiv), allyl bromide ( $0.33 \mathrm{~mL}, 3.93 \mathrm{mmol}, 2.5$ equiv), THF ( 3.2 mL ), and $t$-BuONa ( $454 \mathrm{mg}, 4.72 \mathrm{mmol}, 3$ equiv) afforded, after silica gel chromatography ( $30 \% \mathrm{CH}_{2} \mathrm{Cl}_{2}$ in hexanes), compound $\mathbf{3 . 1 2 n}$ as a colorless oil (442 mg, $1.43 \mathrm{mmol}, 91 \%$ isolated yield). ${ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.26(\mathrm{~s}, 1 \mathrm{H}), 7.25-$ $7.21(\mathrm{~m}, 2 \mathrm{H}), 7.15(\mathrm{dt}, J=1.5,6.5 \mathrm{~Hz}, 1 \mathrm{H}), 5.85(\mathrm{~m}, 1 \mathrm{H}), 5.56(\mathrm{~m}, 1 \mathrm{H}), 5.38(\mathrm{~d}, J=2.5 \mathrm{~Hz}, 1$ H), $5.20(\mathrm{dq}, J=1.5,17.0 \mathrm{~Hz}, 1 \mathrm{H}), 5.13(\mathrm{dq}, J=1.5 .10 .5 \mathrm{~Hz}, 1 \mathrm{H}), 4.33(\mathrm{dd}, J=5.5,8.0 \mathrm{~Hz}, 1$ H), 3.86 (ddt, A of ABX system, $J=1.5,5.5,13.0 \mathrm{~Hz}, 1 \mathrm{H}$ ), 3.72 (ddt, B of ABX system, $J=1.5$, $6.0,13.0 \mathrm{~Hz}, 1 \mathrm{H}$ ), 2.61 (dd, C of CDX system, $J=8.0,14.5 \mathrm{~Hz}, 1 \mathrm{H}$ ), 2.35 (dd, D of CDX system, $J=5.5,15.0 \mathrm{~Hz}, 1 \mathrm{H}), 0.89(\mathrm{t}, J=7.5 \mathrm{~Hz}, 3 \mathrm{H}), 0.53(\mathrm{q}, J=7.5 \mathrm{~Hz}, 2 \mathrm{H}), 0.04(\mathrm{~s}, 3 \mathrm{H}), 0.02(\mathrm{~s}, 3$ H). ${ }^{13} \mathrm{C}$ NMR ( $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 146.8,144.8,134.7,134.2,129.6,127.9,127.6,127.0,125.1$, $116.9,80.5,69.7,44.4,7.3,6.8,-3.7,-3.8$. IR (film) $3060,2955,1427,1248,1092,815 \mathrm{~cm}^{-1}$. HRMS (EI) $\mathrm{m} / \mathrm{z} 308.1372$ [( $\mathrm{M}^{+}$); calcd for $\left.\mathrm{C}_{17} \mathrm{H}_{25} \mathrm{OSiCl}, 308.1363\right]$.

## Preparation of 2-(1-(allyloxy)-3-(triethylsilyl)but-3-en-1-yl)pyridine (3.12o)



Following alternative general procedure E, alcohol $\mathbf{3 . 1 1 0}$ ( $356 \mathrm{mg}, 1.35 \mathrm{mmol}, 1$ equiv), allyl bromide ( $0.24 \mathrm{~mL}, 2.70 \mathrm{mmol}, 2.0$ equiv), THF ( 30 mL ), and sodium hydride, $60 \% \mathrm{w} / \mathrm{w}$ dispersion in mineral oil ( $155 \mathrm{mg}, 4.05 \mathrm{mmol}, 3$ equiv) afforded, after silica gel chromatography (30\% EtOAc in hexanes), compound $\mathbf{3 . 1 2 \mathrm { o }}$ as a colorless oil ( $391 \mathrm{mg}, 1.28 \mathrm{mmol}, 95 \%$ isolated yield). ${ }^{1} \mathrm{H} \operatorname{NMR}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 8.55(\mathrm{ddd}, J=4.9,1.8,0.9 \mathrm{~Hz}, 1 \mathrm{H}), 7.66(\mathrm{td}, J=7.7,1.8$ $\mathrm{Hz}, 1 \mathrm{H}), 7.39(\mathrm{dt}, J=7.8,1.1 \mathrm{~Hz}, 1 \mathrm{H}), 7.16(\mathrm{ddd}, J=7.5,4.9,1.2 \mathrm{~Hz}, 1 \mathrm{H}), 5.87(\mathrm{ddt}, J=17.2$, $10.5,5.6 \mathrm{~Hz}, 1 \mathrm{H}), 5.72(\mathrm{dt}, J=2.8,1.5 \mathrm{~Hz}, 1 \mathrm{H}), 5.37(\mathrm{dd}, J=2.8,1.0 \mathrm{~Hz}, 1 \mathrm{H}), 5.22(\mathrm{dq}, J=17.2$, $1.7 \mathrm{~Hz}, 1 \mathrm{H}), 5.11(\mathrm{dq}, J=10.4,1.4 \mathrm{~Hz}, 1 \mathrm{H}), 4.56(\mathrm{dd}, J=7.4,5.7 \mathrm{~Hz}, 1 \mathrm{H}), 3.90(\mathrm{ddt}, J=12.7$, $5.3,1.5 \mathrm{~Hz}, 1 \mathrm{H}), 3.83(\mathrm{ddt}, J=12.7,5.8,1.4 \mathrm{~Hz}, 1 \mathrm{H}), 2.55(\mathrm{dq}, J=6.9,1.2 \mathrm{~Hz}, 2 \mathrm{H}), 0.90(\mathrm{t}, J=$ $7.9 \mathrm{~Hz}, 9 \mathrm{H}), 0.60(\mathrm{qd}, J=7.9,2.1 \mathrm{~Hz}, 6 \mathrm{H}) .{ }^{13} \mathrm{C} \operatorname{NMR}\left(126 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 162.4,149.0,144.7$, 136.4, 134.7, 128.2, 122.3, 120.9, 116.7, 82.2, 70.1, 42.8, 7.3, 2.8. HRMS (ESI) m/z 304.2100 [(M $+\mathrm{H})^{+}$; calcd for $\left.\mathrm{C}_{18} \mathrm{H}_{30} \mathrm{NOSi}, 304.2097\right]$.

Preparation of (4-(allyloxy)-4-(naphthalen-2-yl)but-1-en-2-yl)dimethyl(phenyl)silane (3.12p)


Following alternative general procedure $E$, alcohol $\mathbf{3 . 1 1 p}(998 \mathrm{mg}, 3.0 \mathrm{mmol}, 1$ equiv),
allyl bromide ( $0.52 \mathrm{~mL}, 6.0 \mathrm{mmol}, 2.0$ equiv), THF ( 30 mL ), and sodium hydride, $60 \% \mathrm{w} / \mathrm{w}$ dispersion in mineral oil ( $345 \mathrm{mg}, 9.0 \mathrm{mmol}, 3$ equiv) afforded, after silica gel chromatography ( $30 \% \mathrm{CH}_{2} \mathrm{Cl}_{2}$ in hexanes), compound 3.12 p as a colorless oil ( $1.07 \mathrm{~g}, 2.88 \mathrm{mmol}, 96 \%$ isolated yield). ${ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.85-7.77(\mathrm{~m}, 3 \mathrm{H}), 7.59-7.54(\mathrm{~m}, 2 \mathrm{H}), 7.52-7.47(\mathrm{~m}$, $3 \mathrm{H}), 7.44-7.36(\mathrm{~m}, 4 \mathrm{H}), 5.84(\mathrm{dddd}, J=17.2,10.4,6.0,5.2 \mathrm{~Hz}, 1 \mathrm{H}), 5.74(\mathrm{dt}, J=2.7,1.3 \mathrm{~Hz}$, $1 \mathrm{H}), 5.55-5.49(\mathrm{~m}, 1 \mathrm{H}), 5.20(\mathrm{dq}, J=17.2,1.7 \mathrm{~Hz}, 1 \mathrm{H}), 5.14(\mathrm{dq}, J=10.3,1.4 \mathrm{~Hz}, 1 \mathrm{H}), 4.40$ (dd, $J=7.8,5.5 \mathrm{~Hz}, 1 \mathrm{H}), 3.82(\mathrm{ddt}, J=12.7,5.2,1.5 \mathrm{~Hz}, 1 \mathrm{H}), 3.62(\mathrm{ddt}, J=12.7,6.1,1.4 \mathrm{~Hz}$, $1 \mathrm{H}), 2.80(\mathrm{ddt}, J=14.3,7.8,1.1 \mathrm{~Hz}, 1 \mathrm{H}), 2.55(\mathrm{ddt}, J=14.4,5.5,1.0 \mathrm{~Hz}, 1 \mathrm{H}), 0.43(\mathrm{~s}, 3 \mathrm{H}), 0.42$ (s, 3H). ${ }^{13} \mathrm{C}$ NMR ( $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta$ 146.2, 139.7, 138.3, 134.9, 134.0, 133.1, 133.0, 129.3, $129.0,128.1,127.81,127.76,127.7,126.0,125.9,125.7,124.6,116.7,80.8,69.4,44.6,-2.9,-3.0$. IR (neat) 3051, 2957, 2856, 1427, 1248, 1111, 1082, $817 \mathrm{~cm}^{-1}$. HRMS (EI) m/z $372.1906\left[(\mathrm{M})^{+}\right.$; calcd for $\mathrm{C}_{25} \mathrm{H}_{28} \mathrm{OSi}$, 372.1909].

## Preparation of 4-(1-(allyloxy)-3-(triethylsilyl)but-3-en-1-yl)pyridine (3.12q)



Following alternative general procedure E, alcohol 3.11q ( $606 \mathrm{mg}, 2.30 \mathrm{mmol}, 1$ equiv), allyl bromide ( $0.4 \mathrm{~mL}, 4.60 \mathrm{mmol}$, 2.0 equiv), THF ( 30 mL ), and sodium hydride, $60 \% \mathrm{w} / \mathrm{w}$ dispersion in mineral oil ( $264.5 \mathrm{mg}, 6.90 \mathrm{mmol}, 3$ equiv) afforded, after silica gel chromatography (25\% EtOAc in hexanes), compound $\mathbf{3 . 1 2 q}$ as a colorless oil ( $542 \mathrm{mg}, 1.8 \mathrm{mmol}, 78 \%$ isolated yield). ${ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 8.55(\mathrm{~d}, J=6.0 \mathrm{~Hz}, 2 \mathrm{H}), 7.20(\mathrm{~d}, J=6.1 \mathrm{~Hz}, 2 \mathrm{H}), 5.85$ (dddd, $J=17.2,10.3,6.0,5.2 \mathrm{~Hz}, 1 \mathrm{H}), 5.63(\mathrm{dt}, J=2.7,1.4 \mathrm{~Hz}, 1 \mathrm{H}), 5.38(\mathrm{~d}, J=2.7 \mathrm{~Hz}, 1 \mathrm{H})$,
$5.21(\mathrm{dq}, J=17.2,1.7 \mathrm{~Hz}, 1 \mathrm{H}), 5.14(\mathrm{dq}, J=10.4,1.4 \mathrm{~Hz}, 1 \mathrm{H}), 4.36(\mathrm{dd}, J=7.8,5.4 \mathrm{~Hz}, 1 \mathrm{H}), 3.87$ (ddt, $J=12.6,5.2,1.5 \mathrm{~Hz}, 1 \mathrm{H}), 3.75(\mathrm{ddt}, J=12.6,6.0,1.4 \mathrm{~Hz}, 1 \mathrm{H}), 2.58(\mathrm{ddt}, J=14.4,7.9,1.1$ $\mathrm{Hz}, 1 \mathrm{H}), 2.33(\mathrm{ddt}, J=14.5,5.4,1.2 \mathrm{~Hz}, 1 \mathrm{H}), 0.90(\mathrm{t}, J=7.9 \mathrm{~Hz}, 9 \mathrm{H}), 0.58(\mathrm{qd}, J=7.9,1.7 \mathrm{~Hz}$, $6 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( 126 MHz , Chloroform- $d$ ) $\delta 151.5,149.8,144.1,134.3,129.0,121.9,117.1,79.8$, 70.0, 44.1, 7.3, 2.8. HRMS (ESI) $m / z 304.2102\left[(\mathrm{M}+\mathrm{H})^{+}\right.$; calcd for $\mathrm{C}_{18} \mathrm{H}_{30} \mathrm{NOSi}$, 304.2097].

## Preparation of 4-(1-(allyloxy)-3-(dimethyl(phenyl)silyl)but-3-en-1-yl)pyridine (3.12r)



Following alternative general procedure E, alcohol $\mathbf{3 . 1 1 \mathbf { r }}$ ( $1077 \mathrm{mg}, 3.8 \mathrm{mmol}, 1$ equiv), allyl bromide ( $0.66 \mathrm{~mL}, 7.60 \mathrm{mmol}$, 2.0 equiv), THF ( 30 mL ), and sodium hydride, $60 \% \mathrm{w} / \mathrm{w}$ dispersion in mineral oil ( $437 \mathrm{mg}, 11.40 \mathrm{mmol}, 3$ equiv) afforded, after silica gel chromatography (30\% EtOAc in hexanes), compound 3.12r as a colorless oil (1010 mg, $3.12 \mathrm{mmol}, 82 \%$ isolated yield). ${ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 8.50(\mathrm{~d}, J=6.0 \mathrm{~Hz}, 2 \mathrm{H}), 7.56-7.49(\mathrm{~m}, 2 \mathrm{H}), 7.40-7.31$ $(\mathrm{m}, 3 \mathrm{H}), 7.02(\mathrm{~d}, J=6.2 \mathrm{~Hz}, 2 \mathrm{H}), 5.82-5.72(\mathrm{~m}, 1 \mathrm{H}), 5.67(\mathrm{dt}, J=2.7,1.3 \mathrm{~Hz}, 1 \mathrm{H}), 5.52(\mathrm{~d}, J=$ $2.7 \mathrm{~Hz}, 1 \mathrm{H}), 5.16(\mathrm{dq}, J=17.2,1.6 \mathrm{~Hz}, 1 \mathrm{H}), 5.11(\mathrm{dq}, J=10.4,1.3 \mathrm{~Hz}, 1 \mathrm{H}), 4.14(\mathrm{dd}, J=7.9,5.3$ $\mathrm{Hz}, 1 \mathrm{H}), 3.75$ (ddt, $J=12.7,5.2,1.5 \mathrm{~Hz}, 1 \mathrm{H}), 3.55(\mathrm{ddt}, J=12.7,6.0,1.4 \mathrm{~Hz}, 1 \mathrm{H}), 2.61(\mathrm{dd}, J=$ $14.2,7.8 \mathrm{~Hz}, 1 \mathrm{H}), 2.35(\mathrm{dd}, J=14.3,5.3 \mathrm{~Hz}, 1 \mathrm{H}), 0.39(\mathrm{~s}, 3 \mathrm{H}), 0.38(\mathrm{~s}, 3 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( 126 MHz , ССС13) $\delta 151.3,149.7,145.5,137.9,134.3,133.9,129.7,129.1,127.8,121.7,116.9,79.4,69.8$, 44.3, -3.05, -3.08. HRMS (ESI) $m / z 324.1803\left[(\mathrm{M}+\mathrm{H})^{+}\right.$; calcd for $\left.\mathrm{C}_{20} \mathrm{H}_{26} \mathrm{NOSi}, 324.1784\right]$.

## Preparation of dimethyl(phenyl)(2-phenyl-3,6-dihydro-2H-pyran-4-yl)silane (3.1a)



Following General Procedure F, diene 3.12a ( $100 \mathrm{mg}, 0.31 \mathrm{mmol}$, 1 equiv) in benzene ( 6.2 mL ), and second-generation Grubbs catalyst ( $10.5 \mathrm{mg}, 0.0124 \mathrm{mmol}, 0.04$ equiv) at $80^{\circ} \mathrm{C}$ for 70 minutes, followed by silica gel chromatography (1:1 $\mathrm{CH}_{2} \mathrm{Cl}_{2} /$ hexanes $)$ afforded dihydropyran 3.1a as a colorless oil ( $89.9 \mathrm{mg}, 0.30 \mathrm{mmol}, 99 \%$ isolated yield). ${ }^{1} \mathrm{H} \mathrm{NMR}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.50$ $(\mathrm{m}, 2 \mathrm{H}), 7.36-7.29(\mathrm{~m}, 7 \mathrm{H}), 7.25(\mathrm{~m}, 1 \mathrm{H}), 6.10(\mathrm{~m}, 1 \mathrm{H}), 4.47(\mathrm{dd}, J=3.5,10.0 \mathrm{~Hz}, 1 \mathrm{H}), 4.40$ $(\mathrm{m}, 2 \mathrm{H}), 2.35-2.28(\mathrm{~m}, 1 \mathrm{H}), 2.26-2.21(\mathrm{~m}, 1 \mathrm{H}), 0.36(\mathrm{~s}, 6 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $\left.126 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ 142.7, 137.4, 136.1, 135.1, 134.0 (2 C), 129.1, 128.3 (2 C), 127.8 (2 C), 127.4, 125.8 (2 C), 75.6, 67.7, 34.5, -3.86, -3.98. IR (neat) $3067,2955,2901,2818,1427,1248,1115,833,819 \mathrm{~cm}^{-1}$. HRMS (EI) $m / z 294.1434\left[(\mathrm{M})^{+}\right.$; calcd for $\mathrm{C}_{19} \mathrm{H}_{22} \mathrm{OSi}$, 294.1440].

## Preparation of ethyldimethyl(2-phenyl-3,6-dihydro-2H-pyran-4-yl)silane (3.1b)



Following General Procedure F, diene 3.12b ( $69.6 \mathrm{mg}, 0.254 \mathrm{mmol}$, 1 equiv) in benzene ( 5.1 mL ), and second-generation Grubbs catalyst ( $8.6 \mathrm{mg}, 0.01 \mathrm{mmol}, 0.04$ equiv) at $80^{\circ} \mathrm{C}$ for 1 $h$, followed by silica gel chromatography ( $1: 1 \mathrm{CH}_{2} \mathrm{Cl}_{2} /$ hexanes $)$ afforded dihydropyran $\mathbf{3 . 1 b}$ as a
colorless oil ( $56.4 \mathrm{mg}, 0.231 \mathrm{mmol}, 91 \%$ isolated yield). ${ }^{1} \mathrm{H} \mathrm{NMR}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.35(\mathrm{~m}$, $4 \mathrm{H}), 7.26(\mathrm{~m}, 1 \mathrm{H}), 6.04(\mathrm{~m}, 1 \mathrm{H}), 4.48(\mathrm{dd}, J=3.5,10.0 \mathrm{~Hz}, 1 \mathrm{H}), 4.39(\mathrm{~m}, 2 \mathrm{H}), 2.31(\mathrm{~m}, 1 \mathrm{H})$, $2.24(\mathrm{~m}, 1 \mathrm{H}), 0.93(\mathrm{t}, J=8.0 \mathrm{~Hz}, 3 \mathrm{H}), 0.56(\mathrm{q}, J=8.0 \mathrm{~Hz}, 2 \mathrm{H}), 0.05(\mathrm{~s}, 6 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR (126 $\mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 142.9,135.9,134.6,128.4$ (2 C), 127.4, 125.8 (2 C), 75.7, 67.7, 34.8, 7.4, 6.1, 4.8. IR (film) 3032, 2953, 2816, 1246, 1124, 1026, $819 \mathrm{~cm}^{-1}$. HRMS (EI) $\mathrm{m} / \mathrm{z} 246.1426\left[(\mathrm{M})^{+}\right.$; calcd for $\mathrm{C}_{15} \mathrm{H}_{22} \mathrm{OSi}$, 246.1440].

Preparation of triethyl(2-phenyl-3,6-dihydro-2H-pyran-4-yl)silane (3.1c)


Following General Procedure F, diene 3.12c ( $110 \mathrm{mg}, 0.364 \mathrm{mmol}$, 1 equiv) in benzene ( 7.3 mL ), and second-generation Grubbs catalyst ( $12.3 \mathrm{mg}, 0.015 \mathrm{mmol}, 0.04$ equiv) at $80^{\circ} \mathrm{C}$ for 1 h , followed by silica gel chromatography ( $30 \% \mathrm{CH}_{2} \mathrm{Cl}_{2}$ in hexanes) afforded dihydropyran 3.1c as a colorless oil ( $101 \mathrm{mg}, 0.360 \mathrm{mmol}, 99 \%$ isolated yield). ${ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.35$ $(\mathrm{m}, 4 \mathrm{H}), 7.26(\mathrm{~m}, 1 \mathrm{H}), 6.03(\mathrm{~m}, 1 \mathrm{H}), 4.47(\mathrm{dd}, J=3.0,10.0 \mathrm{~Hz}, 1 \mathrm{H}), 4.40(\mathrm{~m}, 2 \mathrm{H}), 2.31(\mathrm{~m}, 1$ H), $2.22(\mathrm{~m}, 1 \mathrm{H}), 0.94(\mathrm{t}, J=8.0 \mathrm{~Hz}, 9 \mathrm{H}), 0.59(\mathrm{q}, J=8.0 \mathrm{~Hz}, 6 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 142.9,135.5,133.9,128.4$ (2 C), 127.4, 125.9 (2 C), 75.7, 67.8, 35.3, 7.4, 2.3. IR (film) 3030, 2953, 2814, 1454, 1126, 1026, $698 \mathrm{~cm}^{-1}$. HRMS (EI) $\mathrm{m} / \mathrm{z} 274.1751\left[(\mathrm{M})^{+}\right.$; calcd for $\mathrm{C}_{17} \mathrm{H}_{26} \mathrm{OSi}$, 274.1753].

## Preparation of dimethyl(phenyl)(2-(p-tolyl)-3,6-dihydro-2H-pyran-4-yl)siane (3.1d)



Following General Procedure F, diene 3.12d ( $95 \mathrm{mg}, 0.282 \mathrm{mmol}$, 1 equiv) in benzene ( 5.6 mL ), and second-generation Grubbs catalyst ( $9.6 \mathrm{mg}, 0.011 \mathrm{mmol}, 0.04$ equiv) at $80^{\circ} \mathrm{C}$ for 1 h , followed by silica gel chromatography ( $1: 1 \mathrm{CH}_{2} \mathrm{Cl}_{2} /$ hexanes $)$ afforded dihydropyran 3.1d as a colorless oil ( $89 \mathrm{mg}, \quad 0.279 \mathrm{mmol}, 99 \%$ isolated yield). ${ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.50(\mathrm{~m}, 2$ H), $7.34(\mathrm{~m}, 3 \mathrm{H}), 7.21(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 2 \mathrm{H}), 7.12(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 2 \mathrm{H}), 6.09(\mathrm{~m}, 1 \mathrm{H}), 4.44(\mathrm{dd}, J=$ $3.5,10.0 \mathrm{~Hz}, 1 \mathrm{H}), 4.39(\mathrm{~m}, 2 \mathrm{H}), 2.35-2.27(\mathrm{~m}, 1 \mathrm{H}), 2.31(\mathrm{~s}, 3 \mathrm{H}), 2.22(\mathrm{~m}, 1 \mathrm{H}), 0.35(\mathrm{~s}, 6 \mathrm{H})$. ${ }^{13}{ }^{13}$ NMR (126 MHz, $\mathrm{CDCl}_{3}$ ) $\delta 139.8,137.5,137.0,136.1,135.2,134.0$ (2 C), 129.1, 129.0 (2 C), 127.8 (2 C), 125.8 (2 C), 75.5, 67.7, 34.5, 21.1, -3.85, -3.98. IR (neat) 3013, 2920, 2814, 1427, $1248,1115,817 \mathrm{~cm}^{-1}$. HRMS (EI) $m / z 308.1596\left[\left(\mathrm{M}^{+}\right)\right.$; calcd for $\left.\mathrm{C}_{20} \mathrm{H}_{24} \mathrm{OSi}, 308.1596\right]$.

## Preparation of benzyldimethyl(2-(p-tolyl)-3,6-dihydro-2H-pyran-4-yl)silane (3.1e)



Following General Procedure F, diene 3.12e ( $113 \mathrm{mg}, 0.322 \mathrm{mmol}$, 1 equiv) in benzene ( 4 mL ), and second-generation Grubbs catalyst ( $10.9 \mathrm{mg}, 0.013 \mathrm{mmol}, 0.04$ equiv) at $80^{\circ} \mathrm{C}$ for 70 minutes, followed by silica gel chromatography ( $30 \% \mathrm{CH}_{2} \mathrm{Cl}_{2}$ in hexanes) afforded dihydropyran
3.1e as a colorless oil ( $82 \mathrm{mg}, 0.258 \mathrm{mmol}, 80 \%$ isolated yield). ${ }^{1} \mathrm{H}$ NMR $\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ $7.23-7.18(\mathrm{~m}, 4 \mathrm{H}), 7.15(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 2 \mathrm{H}), 7.06(\mathrm{t}, J=7.5 \mathrm{~Hz}, 1 \mathrm{H}), 6.98(\mathrm{~d}, J=7.5 \mathrm{~Hz}, 2 \mathrm{H})$, $6.01(\mathrm{~m}, 1 \mathrm{H}), 4.29(\mathrm{dd}, J=3.5,10.5 \mathrm{~Hz}, 1 \mathrm{H}), 4.36(\mathrm{~m}, 2 \mathrm{H}), 2.33(\mathrm{~s}, 3 \mathrm{H}), 2.29-2.21(\mathrm{~m}, 1 \mathrm{H})$, $2.14(\mathrm{~s}, 2 \mathrm{H}), 2.09(\mathrm{~m}, 1 \mathrm{H}), 0.05(\mathrm{~s}, 3 \mathrm{H}), 0.04(\mathrm{~s}, 3 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $\left.126 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 139.8$, 137.0, 135.5, 135.2, 129.0 (4 C), 128.2 (4 C), 125.7, 124.1, 75.5, 67.7, 34.9, 25.0, 21.1, -4.5, -4.6. IR (film) 3030, 2957, 1492, 1112, $833 \mathrm{~cm}^{-1}$. HRMS (EI) $m / z 322.1731\left[\left(\mathrm{M}^{+}\right)\right.$; calcd for $\mathrm{C}_{18} \mathrm{H}_{22} \mathrm{OSi}$, 322.1753].

## Preparation of ethyldimethyl(2-(o-tolyl)-3,6-dihydro-2H-pyran-4-yl)silane (3.1f)



Following General Procedure F, diene $\mathbf{3 . 1 2 f}$ ( $111 \mathrm{mg}, 0.385 \mathrm{mmol}$, 1 equiv) in benzene ( 8.5 mL ), and second-generation Grubbs catalyst ( $14.5 \mathrm{mg}, 0.017 \mathrm{mmol}, 0.04$ equiv) at $80^{\circ} \mathrm{C}$ for 1 h , followed by silica gel chromatography ( $45 \% \mathrm{CH}_{2} \mathrm{Cl}_{2}$ in hexanes) afforded dihydropyran 3.1f as a colorless oil ( $98.8 \mathrm{mg}, 0.381 \mathrm{mmol}, 99 \%$ isolated yield). ${ }^{1} \mathrm{H}$ NMR $\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.40$ $(\mathrm{d}, J=7.5 \mathrm{~Hz}, 1 \mathrm{H}), 7.17(\mathrm{t}, J=7.5 \mathrm{~Hz}, 1 \mathrm{H}), 7.10(\mathrm{~m}, 2 \mathrm{H}), 5.99(\mathrm{~m}, 1 \mathrm{H}), 4.60(\mathrm{dd}, J=3.5,10.0$ $\mathrm{Hz}, 1 \mathrm{H}), 4.34(\mathrm{~m}, 2 \mathrm{H}), 2.29(\mathrm{~s}, 3 \mathrm{H}), 2.24(\mathrm{~m}, 1 \mathrm{H}), 2.17(\mathrm{~m}, 1 \mathrm{H}), 0.88(\mathrm{t}, J=8.0 \mathrm{~Hz}, 3 \mathrm{H}), 0.51$ $(\mathrm{q}, J=8.0 \mathrm{~Hz}, 2 \mathrm{H}), 0.00(\mathrm{~s}, 6 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 140.9,136.2,134.55,134.51$, $130.2,127.2,126.3,125.4,72.7,67.9,33.2,19.1,7.4,6.1,-4.7,-4.8$. IR (film) 3370, 3009, 2953, 2814, 1460, 1246, 1124, 1032, $835 \mathrm{~cm}^{-1}$. HRMS (EI) $\mathrm{m} / \mathrm{z} 260.1582\left[\left(\mathrm{M}^{+}\right)\right.$; calcd for $\mathrm{C}_{16} \mathrm{H}_{24} \mathrm{OSi}$, 260.1596].

Preparation of ethyl(2-(3-methoxyphenyl)-3,6-dihydro-2H-pyran-4-yl)dimethylsilane (3.1g)


Following General Procedure F, diene 3.12g ( $100 \mathrm{mg}, 0.328 \mathrm{mmol}$, 1 equiv) in benzene ( 5 mL ), and second-generation Grubbs catalyst ( $11.1 \mathrm{mg}, 0.013 \mathrm{mmol}, 0.04$ equiv) at $80^{\circ} \mathrm{C}$ for 1 h , followed by silica gel chromatography ( $1: 1 \mathrm{CH}_{2} \mathrm{Cl}_{2} /$ hexanes $)$ afforded dihydropyran $\mathbf{3 . 1 g}$ as a colorless oil ( $78 \mathrm{mg}, 0.282 \mathrm{mmol}, 86 \%$ isolated yield). ${ }^{1} \mathrm{H} \mathrm{NMR}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.24(\mathrm{t}, J=$ $8.0 \mathrm{~Hz}, 1 \mathrm{H}), 6.93(\mathrm{~m}, 2 \mathrm{H}), 6.80(\mathrm{~m}, 1 \mathrm{H}), 6.02(\mathrm{~m}, 1 \mathrm{H}), 4.45(\mathrm{dd}, J=3.5,10.0 \mathrm{~Hz}, 1 \mathrm{H}), 4.38$ $(\mathrm{m}, 2 \mathrm{H}), 2.30(\mathrm{~m}, 1 \mathrm{H}), 2.23(\mathrm{~m}, 1 \mathrm{H}), 0.92(\mathrm{t}, J=8.0 \mathrm{~Hz}, 3 \mathrm{H}), 0.55(\mathrm{q}, J=8.0 \mathrm{~Hz}, 2 \mathrm{H}), 0.03(\mathrm{~s}$, $6 \mathrm{H}) .{ }^{13} \mathrm{C} \mathrm{NMR}\left(126 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 159.8,144.6,135.9,134.5,129.4,118.2,113.1,111.2,75.6$, 67.7, 55.2, 34.8, 7.4, 6.1, -4.8. IR (film) 3005, 2953, 1257, 1033, $775 \mathrm{~cm}^{-1}$. HRMS (EI) $\mathrm{m} / \mathrm{z}$ $276.1540\left[(\mathrm{M})^{+} ;\right.$calcd for $\left.\mathrm{C}_{16} \mathrm{H}_{24} \mathrm{O}_{2} \mathrm{Si}, 276.1546\right]$.

## Preparation of dimethyl(phenyl)(2-(ferrocenyl)-3,6-dihydro-2H-pyran-4-yl)silane (3.1h)



Following General Procedure F, diene 3.12h ( $180.3 \mathrm{mg}, 0.419 \mathrm{mmol}$, 1 equiv) in benzene $(8.4 \mathrm{~mL})$, and second-generation Grubbs catalyst ( $14.2 \mathrm{mg}, 0.0167 \mathrm{mmol}, 0.04$ equiv) at $80^{\circ} \mathrm{C}$ for 1 h , followed by silica gel chromatography $\left(7: 3 \mathrm{CH}_{2} \mathrm{Cl}_{2} /\right.$ hexanes $)$ afforded dihydropyran $\mathbf{3 . 1 h}$ as
an orange oil ( $175.3 \mathrm{mg}, 0.335 \mathrm{mmol}, 70 \%$ isolated yield). ${ }^{1} \mathrm{H}$ NMR ( $600 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.54(\mathrm{~m}$, $2 \mathrm{H}), 7.37(\mathrm{~m}, 3 \mathrm{H}), 6.07(\mathrm{~m}, 1 \mathrm{H}), 4.32(\mathrm{dd}, J=4.0,7.0 \mathrm{~Hz}, 1 \mathrm{H}), 4.29(\mathrm{~m}, 1 \mathrm{H}), 4.25(\mathrm{~m}, 1 \mathrm{H})$, $4.21(\mathrm{~m}, 1 \mathrm{H}), 4.11(\mathrm{~m}, 2 \mathrm{H}), 4.08(\mathrm{~m}, 1 \mathrm{H}), 4.06(\mathrm{~s}, 5 \mathrm{H}), 2.34(\mathrm{~m}, 2 \mathrm{H}), 0.40(\mathrm{~s}, 3 \mathrm{H}), 0.39(\mathrm{~s}, 3$ H). ${ }^{13} \mathrm{C}$ NMR (151 MHz, $\left.\mathrm{CDCl}_{3}\right) \delta 137.5,136.1,134.9,133.9$ (2 C), 129.1, 127.8 (2 C), 88.8, 71.7, 68.6 (5 C), 68.0, 67.6, 67.5, 66.8, 66.4, 32.4, -3.9, -4.0. IR (film) 3069, 2956, 1424, 1248, 1110, $815 \mathrm{~cm}^{-1}$. HRMS (EI) $m / z 402.1108$ [(M) ${ }^{+}$; calcd for $\left.\mathrm{C}_{23} \mathrm{H}_{26} \mathrm{OSiFe}, 402.1102\right]$.

Preparation of ethyldimethyl(2-(thiophen-2-yl)-3,6-dihydro-2H-pyran-4-yl)silane (3.1i)


Following General Procedure F, diene 3.12i ( $80 \mathrm{mg}, 0.285 \mathrm{mmol}$, 1 equiv) in benzene ( 3.6 mL ), and second-generation Grubbs catalyst ( $9.7 \mathrm{mg}, 0.0114 \mathrm{mmol}, 0.04$ equiv) at $80^{\circ} \mathrm{C}$ for 1 h , followed by silica gel chromatography ( $20-25 \% \mathrm{CH}_{2} \mathrm{Cl}_{2}$ in hexanes) afforded dihydropyran $\mathbf{3 . 1 i}$ as a colorless oil ( $43 \mathrm{mg}, 0.171 \mathrm{mmol}, 60 \%$ isolated yield). ${ }^{1} \mathrm{H}$ NMR $\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.24(\mathrm{~m}$, $1 \mathrm{H}), 6.99(\mathrm{~m}, 1 \mathrm{H}), 6.96(\mathrm{dd}, J=3.5,5.0 \mathrm{~Hz}, 1 \mathrm{H}), 6.00(\mathrm{~m}, 1 \mathrm{H}), 4.77(\mathrm{dd}, J=3.5,9.5 \mathrm{~Hz}, 1 \mathrm{H})$, $4.39(\mathrm{~m}, 1 \mathrm{H}), 4.33(\mathrm{~m}, 1 \mathrm{H}), 2.47(\mathrm{~m}, 1 \mathrm{H}), 2.37(\mathrm{~m}, 1 \mathrm{H}), 0.92(\mathrm{t}, J=8.0 \mathrm{~Hz}, 3 \mathrm{H}), 0.56(\mathrm{q}, J=$ 8.0 Hz, 2 H ), $0.05(\mathrm{~s}, 6 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR (126 MHz, $\left.\mathrm{CDCl}_{3}\right) \delta 145.9,135.3,134.4,126.5,124.6$, 123.7, 71.4, 67.3, 34.4, 7.4, 6.0, -4.8. IR (film) 2953, 1246, 1120, $833 \mathrm{~cm}^{-1}$. HRMS (EI) $\mathrm{m} / \mathrm{z}$ 252.0992 [(M) ${ }^{+}$; calcd for $\left.\mathrm{C}_{13} \mathrm{H}_{20} \mathrm{OSiS}, 252.1004\right]$.

## Preparation of dimethyl(phenyl)(2-propyl-3,6-dihydro-2H-pyran-4-yl)silane (3.1j)



Following General Procedure F, diene $\mathbf{3 . 1 2 j}$ ( $136 \mathrm{mg}, 0.471 \mathrm{mmol}$, 1 equiv) in benzene ( 9 mL ), and second-generation Grubbs catalyst ( $16 \mathrm{mg}, 0.019 \mathrm{mmol}, 0.04$ equiv) at $80^{\circ} \mathrm{C}$ for 1 h , followed by silica gel chromatography $\left(35 \% \mathrm{CH}_{2} \mathrm{Cl}_{2}\right.$ in hexanes) afforded dihydropyran $\mathbf{3 . 1 \mathbf { j }}$ as a colorless oil ( $92 \mathrm{mg}, 0.353 \mathrm{mmol}, 0.75 \%$ isolated yield). ${ }^{1} \mathrm{H} \mathrm{NMR}\left(600 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.49(\mathrm{~m}$, $2 \mathrm{H}), 7.34(\mathrm{~m}, 3 \mathrm{H}), 6.02(\mathrm{~m}, 1 \mathrm{H}), 4.21(\mathrm{~m}, 2 \mathrm{H}), 3.41(\mathrm{~m}, 1 \mathrm{H}), 1.96(\mathrm{~m}, 2 \mathrm{H}), 1.50(\mathrm{~m}, 1 \mathrm{H}), 1.46$ $-1.32(\mathrm{~m}, 3 \mathrm{H}), 0.89(\mathrm{t}, J=7.2 \mathrm{~Hz}, 3 \mathrm{H}), 0.33(\mathrm{~s}, 6 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $\left.151 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ 137.7, 136.2, 134.7, 134.0 (2 C), 129.0, 127.8 (2 C), 73.3, 67.1, 38.1, 32.4, 18.7, 14.1, -3.9, -4.0. IR (film) 3069, 2957, 2812, 1427, 1248, 1130, $833 \mathrm{~cm}^{-1}$. HRMS (EI) $\mathrm{m} / \mathrm{z} 260.1596\left[(\mathrm{M})^{+}\right.$; calcd for $\left.\mathrm{C}_{16} \mathrm{H}_{24} \mathrm{OSi}, 260.1596\right]$.

## Preparation of (2-cyclohexyl-3,6-dihydro-2H-pyran-4-yl)dimethyl(phenyl)silane (3.1k)



Following General Procedure F, diene 3.12k ( $87 \mathrm{mg}, 0.265 \mathrm{mmol}$, 1 equiv) in benzene ( 5.3 mL ), and second-generation Grubbs catalyst ( $9 \mathrm{mg}, 0.011 \mathrm{mmol}, 0.04$ equiv) at $80^{\circ} \mathrm{C}$ for 1 h , followed by silica gel chromatography $\left(35 \% \mathrm{CH}_{2} \mathrm{Cl}_{2}\right.$ in hexanes) afforded dihydropyran $\mathbf{3 . 1 k}$ as a colorless oil ( $82.3 \mathrm{mg}, 0.262 \mathrm{mmol}, 99 \%$ isolated yield). ${ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.48(\mathrm{~m}$,
$2 \mathrm{H}), 7.35(\mathrm{~m}, 3 \mathrm{H}), 6.01(\mathrm{~m}, 1 \mathrm{H}), 4.25-4.15(\mathrm{~m}, 2 \mathrm{H}), 3.14(\mathrm{ddd}, J=3.5,5.5,10.0 \mathrm{~Hz}, 1 \mathrm{H}), 2.03$ $(\mathrm{m}, 1 \mathrm{H}), 1.99-1.92(\mathrm{~m}, 2 \mathrm{H}), 1.70(\mathrm{~m}, 2 \mathrm{H}), 1.62(\mathrm{~m}, 2 \mathrm{H}), 1.33(\mathrm{~m}, 1 \mathrm{H}), 1.25-1.11(\mathrm{~m}, 3 \mathrm{H})$, $0.96(\mathrm{~m}, 2 \mathrm{H}), 0.32(\mathrm{~s}, 6 \mathrm{H}) .{ }^{13} \mathrm{C} \operatorname{NMR}\left(126 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 137.7,136.5,134.7,134.0(2 \mathrm{C})$, 129.0, 127.8 (2 C), $77.9,67.5,42.8,29.4,29.0,28.5,26.6,26.2,26.1,-3.8,-3.9$. IR (neat) 2924 , 2853, 1427, 1246, 1126, $835 \mathrm{~cm}^{-1}$. HRMS (EI) $m / z 300.1893\left[(\mathrm{M})^{+}\right.$; calcd for $\mathrm{C}_{19} \mathrm{H}_{28} \mathrm{OSi}$, 300.1909].

Preparation of (2-cyclopropyl-3,6-dihydro-2H-pyran-4-yl)dimethyl(phenyl)silane (3.11)


Following General Procedure F, diene $\mathbf{3 . 1 2 1}$ ( $82 \mathrm{mg}, 0.286 \mathrm{mmol}$, 1 equiv) in benzene ( 5.7 mL ), and second-generation Grubbs catalyst ( $9.7 \mathrm{mg}, 0.0114 \mathrm{mmol}, 0.04$ equiv) at $80^{\circ} \mathrm{C}$ for 1 h , followed by silica gel chromatography ( $20-25 \% \mathrm{CH}_{2} \mathrm{Cl}_{2}$ in hexanes) afforded dihydropyran $\mathbf{3 . 1 1}$ as a colorless oil ( $63 \mathrm{mg}, 0.243 \mathrm{mmol}, 85 \%$ isolated yield). ${ }^{1} \mathrm{H}$ NMR $\left(600 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.48(\mathrm{~m}$, $2 \mathrm{H}), 7.34(\mathrm{~m}, 3 \mathrm{H}), 5.99(\mathrm{~m}, 1 \mathrm{H}), 4.26(\mathrm{~m}, 1 \mathrm{H}), 4.18(\mathrm{~m}, 1 \mathrm{H}), 2.75(\mathrm{ddd}, J=3.0,7.8,9.6 \mathrm{~Hz}, 1$ H), 2.17-2.11(m, 1 H), 2.09-2.05 (m, 1 H), $0.87(\mathrm{~m}, 1 \mathrm{H}), 0.51(\mathrm{~m}, 1 \mathrm{H}), 0.45(\mathrm{~m}, 1 \mathrm{H}), 0.33(\mathrm{~s}$, 6 H ), 0.33 (overlapped) 0.17 (m, 1 H ). ${ }^{13} \mathrm{C}$ NMR ( $151 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 137.6,136.2,134.6,134.0$ (2 C), 129.0, 127.8 (2 C), 77.9, 67.2, 31.9. 15.7, 2.8, 1.8, -3.8, -3.9. IR (neat) 3070, 3007, 2957, 1427, 1248, 1122, $817 \mathrm{~cm}^{-1}$. HRMS (EI) $m / z 258.1429\left[(\mathrm{M})^{+}\right.$; calcd for $\mathrm{C}_{16} \mathrm{H}_{22} \mathrm{OSi}$, 258.1440].

## Preparation of (2-cyclopropyl-3,6-dihydro-2H-pyran-4-yl)triethylsilane (3.1m)



Following General Procedure F, diene 3.12m ( $106 \mathrm{mg}, 0.398 \mathrm{mmol}$, 1 equiv) in benzene ( 8 mL ), and second-generation Grubbs catalyst ( $13.6 \mathrm{mg}, 0.016 \mathrm{mmol}, 0.04$ equiv) at $80^{\circ} \mathrm{C}$ for 1 $h$, followed by silica gel chromatography (20-25\% $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ in hexanes) afforded dihydropyran 3.1m as a colorless oil ( $81.6 \mathrm{mg}, 0.342 \mathrm{mmol}, 86 \%$ isolated yield). ${ }^{1} \mathrm{H}$ NMR $\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ $5.90(\mathrm{~m}, 1 \mathrm{H}), 4.25(\mathrm{~m}, 1 \mathrm{H}), 4.17(\mathrm{~m}, 1 \mathrm{H}), 2.74(\mathrm{~m}, 1 \mathrm{H}), 2.13(\mathrm{~m}, 1 \mathrm{H}), 2.04(\mathrm{~m}, 1 \mathrm{H}), 0.91(\mathrm{t}, J$ $=8.0 \mathrm{~Hz}, 9 \mathrm{H}), 0.90(\mathrm{~m}$, heavily overlapped, 1 H$), 0.57(\mathrm{q}, J=8.0 \mathrm{~Hz}, 6 \mathrm{H}), 0.53-0.48(\mathrm{~m}, 2 \mathrm{H})$, $0.34(\mathrm{~m}, 1 \mathrm{H}), 0.20(\mathrm{~m}, 1 \mathrm{H}) .{ }^{13} \mathrm{C} \operatorname{NMR}\left(126 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 135.4,133.5,78.0,67.2,32.6,15.7$, 7.4, 2.9, 2.3, 1.8. IR (film) 3082, 3007, 2953, 1124, 1018, $731 \mathrm{~cm}^{-1}$. HRMS (EI) $m / z 238.1763$ [(M) ${ }^{+}$; calcd for $\left.\mathrm{C}_{14} \mathrm{H}_{26} \mathrm{OSi}, 238.1753\right]$.

Preparation of (2-(3-chlorophenyl)-3,6-dihydro-2H-pyran-4-yl)(ethyl)dimethylsilane (3.1n)


Following General Procedure F, diene 3.12n ( $123 \mathrm{mg}, 0.398 \mathrm{mmol}, 1$ equiv) in benzene ( 8 mL ), and second-generation Grubbs catalyst ( $13.5 \mathrm{mg}, 0.016 \mathrm{mmol}, 0.04$ equiv) at $80^{\circ} \mathrm{C}$ for 1 h , followed by silica gel chromatography $\left(30 \% \mathrm{CH}_{2} \mathrm{Cl}_{2}\right.$ in hexanes) afforded dihydropyran $\mathbf{3 . 1 n}$ as a colorless oil ( $93.8 \mathrm{mg}, 0.334 \mathrm{mmol}, 84 \%$ isolated yield). ${ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.36(\mathrm{t}, J$
$=1.5 \mathrm{~Hz}, 1 \mathrm{H}), 7.27-7.21(\mathrm{~m}, 3 \mathrm{H}), 6.01(\mathrm{~m}, 1 \mathrm{H}), 4.44(\mathrm{dd}, J=5.5,8.0 \mathrm{~Hz}, 1 \mathrm{H}), 4.36(\mathrm{q}, J=2.5$ $\mathrm{Hz}, 2 \mathrm{H}), 2.23(\mathrm{~m}, 2 \mathrm{H}), 0.91(\mathrm{t}, J=8.0 \mathrm{~Hz}, 3 \mathrm{H}), 0.54(\mathrm{q}, J=8.0 \mathrm{~Hz}, 2 \mathrm{H}), 0.03(\mathrm{~s}, 6 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR (126 MHz, $\mathrm{CDCl}_{3}$ ) $\delta 145.0,135.6,134.4,134.3,129.6,127.4,126.0,123.9,74.9,67.6,34.6,7.4$, 6.0, -4.8. IR (film) 3013, 2953, 2820, 1126, $835 \mathrm{~cm}^{-1}$. HRMS (EI) $\mathrm{m} / \mathrm{z} 280.1064\left[\left(\mathrm{M}^{+}\right)\right.$; calcd for $\left.\mathrm{C}_{15} \mathrm{H}_{21} \mathrm{OSiCl}, 280.1050\right]$.

## Preparation of 2-(4-(triethylsilyl)-3,6-dihydro-2H-pyran-2-yl)pyridine (3.10)



Following General Procedure F, diene $\mathbf{3 . 1 2 0}$ ( 334 mg , 1.1 mmol , 1 equiv) in benzene ( 40 mL ), and second-generation Grubbs catalyst ( $38 \mathrm{mg}, 0.044 \mathrm{mmol}, 0.04$ equiv) at $80^{\circ} \mathrm{C}$ for 1 h , followed by silica gel chromatography ( $30 \% \mathrm{CH}_{2} \mathrm{Cl}_{2}$ in hexanes) afforded dihydropyran $\mathbf{3 . 1 0}$ as a colorless oil ( $302 \mathrm{mg}, 1.09 \mathrm{mmol}, 99 \%$ isolated yield). ${ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 8.55$ (ddd, $J=5.7,1.8,0.8 \mathrm{~Hz}, 1 \mathrm{H}), 7.69(\mathrm{td}, J=7.7,1.8 \mathrm{~Hz}, 1 \mathrm{H}), 7.47(\mathrm{~d}, J=7.8 \mathrm{~Hz}, 1 \mathrm{H}), 7.17(\mathrm{ddd}, J=$ $7.6,4.8,1.2 \mathrm{~Hz}, 1 \mathrm{H}), 6.03(\mathrm{q}, J=2.7 \mathrm{~Hz}, 1 \mathrm{H}), 4.61(\mathrm{dd}, J=10.4,3.3 \mathrm{~Hz}, 1 \mathrm{H}), 4.43(\mathrm{dt}, J=4.6$, $2.3 \mathrm{~Hz}, 2 \mathrm{H}), 2.44(\mathrm{dq}, J=16.5,2.2 \mathrm{~Hz}, 1 \mathrm{H}), 2.29(\mathrm{ddq}, J=16.8,9.8,3.1 \mathrm{~Hz}, 1 \mathrm{H}), 0.92(\mathrm{t}, J=7.9$ $\mathrm{Hz}, 9 \mathrm{H}), 0.60(\mathrm{q}, J=7.9 \mathrm{~Hz}, 6 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 161.9,148.8,136.7,135.2$, 133.7, 122.2, 120.0, 76.3, 67.6, 33.6, 7.3, 2.2. HRMS (ESI) $m / z 274.1627\left[(\mathrm{M}-\mathrm{H})^{+}\right.$; calcd for $\mathrm{C}_{16} \mathrm{H}_{24} \mathrm{NOSi}$, 274.1627].

Synthesis of dimethyl(2-(naphthalen-2-yl)-3,6-dihydro-2H-pyran-4-yl)(phenyl)silane (3.1p)


Following General Procedure F, diene $\mathbf{3 . 1 2 p}$ ( 932 mg , 2.5 mmol , 1 equiv) in benzene ( 80 mL ), and second-generation Grubbs catalyst ( $85 \mathrm{mg}, 0.044 \mathrm{mmol}, 0.04$ equiv) at $80^{\circ} \mathrm{C}$ for 1 h , followed by silica gel chromatography ( $40 \% \mathrm{CH}_{2} \mathrm{Cl}_{2}$ in hexanes) afforded dihydropyran $\mathbf{3 . 1 p}$ as a colorless oil ( $492 \mathrm{mg}, 1.40 \mathrm{mmol}, 56 \%$ isolated yield). ${ }^{1} \mathrm{H} \mathrm{NMR}\left(500 \mathrm{MHz}, \mathrm{C}_{6} \mathrm{D}_{6}\right) \delta 7.76(\mathrm{~s}, 1 \mathrm{H})$, $7.63-7.55(\mathrm{~m}, 3 \mathrm{H}), 7.47(\mathrm{dq}, J=5.4,2.1 \mathrm{~Hz}, 2 \mathrm{H}), 7.44(\mathrm{dd}, J=8.5,1.7 \mathrm{~Hz}, 1 \mathrm{H}), 7.25-7.13(\mathrm{~m}$, $5 \mathrm{H}), 5.91(\mathrm{tt}, J=3.2,1.5 \mathrm{~Hz}, 1 \mathrm{H}), 4.53(\mathrm{dd}, J=10.0,3.3 \mathrm{~Hz}, 1 \mathrm{H}), 4.27(\mathrm{dtd}, J=17.2,3.1,1.0 \mathrm{~Hz}$, $1 \mathrm{H}), 4.19$ (dddd, $J=17.2,4.0,3.2,1.8 \mathrm{~Hz}, 1 \mathrm{H}), 2.43$ (dddt, $J=16.7,10.0,3.9,2.7 \mathrm{~Hz}, 1 \mathrm{H}), 2.30$ $(\mathrm{dtt}, J=17.1,3.3,1.2 \mathrm{~Hz}, 1 \mathrm{H}), 0.26(\mathrm{~s}, 3 \mathrm{H}), 0.25(\mathrm{~s}, 3 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR $\left(126 \mathrm{MHz}, \mathrm{C}_{6} \mathrm{D}_{6}\right) \delta 141.4$, $138.0,137.5,135.4,134.7,134.3,133.7,129.9,128.72,128.66,128.62,128.4,126.5,126.2,125.1$, $125.0,76.1,68.0,35.6,-3.4,-3.5$. IR (neat) $3057,2955,2818,1427,1248,1115,815 \mathrm{~cm}^{-1}$. HRMS (EI) $\mathrm{m} / \mathrm{z} 344.1580\left[(\mathrm{M})^{+}\right.$; calcd for $\left.\mathrm{C}_{23} \mathrm{H}_{24} \mathrm{OSi}, 344.1596\right]$.

Table 3.1: Conditions for ring closing metathesis


| Entry | S.M | Cat. (mol \%) | Add. | Solvent | Temp $\left({ }^{\circ} \mathrm{C}\right)$ | Time (h) | \% Conv |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathbf{1}$ | S2-q | A (0.04) | None | benzene | 80 | 1 | 0 |
| $\mathbf{2}$ | S2-q | A (0.04) | None | benzene | 80 | 12 | 0 |
| $\mathbf{3}$ | S2-q | A (0.04) | None | toluene | 110 | 12 | 0 |
| $\mathbf{4}$ | S2-q | A (0.04) | None | DCM | 25 | 24 | 0 |
| $\mathbf{5}$ | S2-q | B (0.04) | None | toluene | 60 | 12 | 0 |
| $\mathbf{6}$ | S2-r | A (0.04) | None | benzene | 80 | 1 | $5^{\text {a }}$ |
| $\mathbf{7}$ | S2-r | A (0.04) | TMSCl | $1,2-D C E$ | $90{ }^{\circ} \mathrm{C}$ | 4 | $20^{\mathrm{a}}$ |
|  |  |  |  |  |  |  |  |

${ }^{\text {a }}$ The product was not isolated but analyzed as a mixture with the starting material. S.M. $=$ starting material, Cat. $=$ catalyst, Add. $=$ additive. Cat. A = Grubbs II, cat. B = Hoveyda-Grubbs II. DCM $=$ dichloromethane. $\mathrm{TMSCl}=$ chlorotrimethylsilane. 1,2-DCE = 1,2-dichloroethane

### 3.10.3. Wittig rearrangements of 4-silyl-6-aryl dihydropyrans 3.1 - general procedure $\mathbf{G}$



Freshly prepared and purified dihydropyran $\mathbf{3 . 1}$ was dissolved in THF under nitrogen (concentration $=0.08 \mathrm{M}$, unless otherwise noticed) and the solution cooled down at $-78{ }^{\circ} \mathrm{C}$ (dry ice/acetone bath). $n$-Butyllithium (1.2 equiv, 1.6 M in hexanes) or sec-butyllithium (1.2 equiv, 1.4 M in cyclohexane) was added dropwise (1 drop/second) to give a colored solution. After the indicated time ( $\sim 3$ hours or $\sim 10$ minutes, respectively) the reaction was quenched by adding $\mathrm{NH}_{4} \mathrm{Cl}$ (sat), diluted with water and diethyl ether. The aqueous phase was extracted with diethyl ether three times. Combined organic extracts were washed with water, and brine. The solution was dried over $\mathrm{MgSO}_{4}$, filtered, concentrated in a rotavap at room temperature (or lower) and the residue purified by column chromatography ( $5-10 \% \mathrm{EtOAc}$ in hexanes) to afford the aldehyde and or alcohol as colorless oils.

### 3.10.4. Wittig rearrangements of 4-silyl-6-alkyl dihydropyrans 3.1 - general procedure $\mathbf{H}$



Freshly prepared and purified 3.1 was dissolved in THF under nitrogen (concentration = 0.08 M, unless otherwise noticed) and the solution cooled down at $-78^{\circ} \mathrm{C}$ (dry ice/acetone bath).
$n$-Butyllithium (1.2 equiv, 1.6 M in hexanes) or sec-butyllithium (1.2 equiv, 1.4 M in cyclohexane) was added dropwise ( 1 drop/second) with stirring to give a colored solution. After $\sim 25$ minutes the temperature was raised to $-10^{\circ} \mathrm{C}$ (unless otherwise indicated). After the indicated time (1.5-3 hours) the reaction was cooled down at $-78^{\circ} \mathrm{C}$ and quenched by adding $\mathrm{NH}_{4} \mathrm{Cl}$ (sat), diluted with water and diethyl ether. The aqueous phase was extracted with diethyl ether three times. Combined organic extracts were washed with water, and brine. The solution was dried over $\mathrm{MgSO}_{4}$, filtered, concentrated in a rotavap at room temperature (or lower) and the residue purified by column chromatography (4-5\% EtOAc in hexanes) to afford the aldehyde as a colorless oil.

## Preparation of 2-(1-(dimethyl(phenyl)silyl)-2-phenylcyclopropyl)acetaldehyde (3.2a)



Following General Procedure G, dihydropyran 3.1a ( $89 \mathrm{mg}, 0.302 \mathrm{mmol}, 1$ equiv) in THF ( 3.8 mL ) and $n$-butyllithium ( $0.23 \mathrm{~mL}, 0.36 \mathrm{mmol}, 1.2$ equiv) at $-78^{\circ} \mathrm{C}$ for 3 h , followed by workup and silica gel chromatography (5\% EtOAc in hexanes) afforded silylcyclopropane 3.2a as a colorless oil ( $70.7 \mathrm{mg}, 0.242 \mathrm{mmol}, 80 \%$ isolated yield, $d r=1.0: 0.3$ ). ${ }^{1} \mathrm{H} \mathrm{NMR}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$ mixture of diastereomers (1.0:0.3 ratio) $\delta 9.67(\mathrm{t}, J=2.0 \mathrm{~Hz}, 1 \mathrm{H}), 9.33(\mathrm{dd}, J=2.0,2.5 \mathrm{~Hz}, 0.3$ H), $7.55(\mathrm{~m}, 0,6 \mathrm{H}), 7.38(\mathrm{~m}, 0,9 \mathrm{H}), 7.33-7.17(\mathrm{~m}, 10.9 \mathrm{H}), 7.10(\mathrm{~m}, 0.6 \mathrm{H}), 2.66(\mathrm{dd}, J=2.0$, $17.0 \mathrm{~Hz}, 1 \mathrm{H}), 2.25(\mathrm{dd}, J=6.0,8.0 \mathrm{~Hz}, 0.3 \mathrm{H}$, partially overlapped), $2.22(\mathrm{dd}, J=6.0,8.0 \mathrm{~Hz}, 1$ H), $2.18(\mathrm{dd}, J=3.0,17.5 \mathrm{~Hz}, 0.3 \mathrm{H}), 2.01(\mathrm{dd}, J=2.0,17.0 \mathrm{~Hz}, 1 \mathrm{H}), 1.79(\mathrm{dd}, J=2.0,17.5 \mathrm{~Hz}$, $0.3 \mathrm{H}), 1.39(\mathrm{dd}, J=4.5,6.0 \mathrm{~Hz}, 1 \mathrm{H}), 1.16(\mathrm{dd}, J=5.5,8.0 \mathrm{~Hz}, 0.3 \mathrm{H}), 1.09(\mathrm{t}, J=5.0 \mathrm{~Hz}, 0.3 \mathrm{H})$, $0.93(\mathrm{dd}, J=4.5,8.5 \mathrm{~Hz}, 1 \mathrm{H}), 0.36(\mathrm{~s}, 0.9 \mathrm{H}), 0.35(\mathrm{~s}, 0.9 \mathrm{H}), 0.08(\mathrm{~s}, 3 \mathrm{H}),-0.23(\mathrm{~s}, 3 \mathrm{H}) .{ }^{13} \mathrm{C}$

NMR ( $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) major diastereomer $\delta$ 202.7, 139.1, 137.9, 134.0, 129.9, 129.0, 128.0, 127.7, 126.5, 53.2, 29.4, 15.7, -2.8, -3.4. Minor diastereomer (a substituted aromatic carbon could not be located) $\delta 203.2,137.8,134.1$ (2 C), 129.5, 129.3 (2 C), 128.2 (2 C), 127.9 (2 C), 126.4, 45.1, 24.9, 10.3, -4.35, -4.43. IR (neat) 3063, 2956, 1722, 1496. 1427, 1250, 1111, $816 \mathrm{~cm}^{-1}$. HRMS (EI) $m / z 294.1430\left[(\mathrm{M})^{+}\right.$; calcd for $\mathrm{C}_{19} \mathrm{H}_{22} \mathrm{OSi}$, 294.1440].

## Preparation of 2-(1-(ethyldimethylsilyl)-2-phenylcyclopropyl)acetaldehyde (3.2b)



Following General Procedure G, dihydropyran 3.1b ( $38 \mathrm{mg}, 0.154 \mathrm{mmol}$, 1 equiv) in THF ( 1.9 mL ) and sec-butyllithium ( $0.16 \mathrm{~mL}, 0.23 \mathrm{mmol}, 1.5$ equiv) at $-78^{\circ} \mathrm{C}$ for 40 minutes, followed by workup and silica gel chromatography ( $5 \% \mathrm{EtOAc}$ in hexanes) afforded silylcyclopropane $\mathbf{3 . 2 b}$ as a colorless oil $(70.7 \mathrm{mg}, 0.123 \mathrm{mmol}, 80 \%$ isolated yield, crude $d r=1.0: 0.26$, isolated $d r=$ 1.0:0.3). Mixture of diastereomers (1.0:0.3 ratio) ${ }^{1} \mathrm{H}$ NMR $\left.600 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 9.90(\mathrm{dd}, J=1.8$, 6.0 Hz, 1 H), $9.50(\mathrm{dd}, J=2.4,6.0 \mathrm{~Hz}, 0.3 \mathrm{H}), 7.29(\mathrm{~m}, 2 \mathrm{H}), 7.24(\mathrm{~m}, 2.6 \mathrm{H}), 7.17(\mathrm{~m}, 1.9 \mathrm{H})$, $2.74(\mathrm{dd}, J=6.0,16.8 \mathrm{~Hz}, 1 \mathrm{H}), 2.19(\mathrm{~m}, 1.6 \mathrm{H}), 2.01(\mathrm{dd}, J=1.8,16.8 \mathrm{~Hz}, 1 \mathrm{H}), 1.80(\mathrm{dd}, J=$ $2.4,17.4 \mathrm{~Hz}, 0.3 \mathrm{H}), 1.24(\mathrm{dd}, J=4.8,6.0 \mathrm{~Hz}, 1 \mathrm{H}), 1.13(\mathrm{dd}, J=5.4,7.8 \mathrm{~Hz}, 0.3 \mathrm{H}), 1.07(\mathrm{t}, J=$ $5.4 \mathrm{~Hz}, 0.3 \mathrm{H}), 0.97(\mathrm{t}, J=7.8 \mathrm{~Hz}, 0.9 \mathrm{H}), 0.86(\mathrm{dd}, J=4.8,8.4 \mathrm{~Hz}, 1.3 \mathrm{H}), 0.76(\mathrm{t}, J=7.8 \mathrm{~Hz}, 3$ H), $0.56(\mathrm{q}, J=7.8 \mathrm{~Hz}, 0.6 \mathrm{H}), 0.26(\mathrm{~m}, 2 \mathrm{H}), 0.00(\mathrm{~s}, 1.8 \mathrm{H}),-0.32(\mathrm{~s}, 3 \mathrm{H}),-0.42(\mathrm{~s}, 3 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR (126 MHz, $\mathrm{CDCl}_{3}$ ) Major diastereomer: $\delta$ 202.9, 139.4, 129.9 (2 C), 127.9 (2 C), 126.4, 53.4, 29.4, 15.3, 9.9, 7.3, 6.6, -3.8, -4.0. Minor diastereomer: $\delta$ 203.4, 138.1, 129.4 (2 C), 128.2 (2 C), 126.4, 45.4, 24.7, 13.7, 9.1, 7.4, 5.8, -5.0, -5.1. IR (film) 3059, 2955, 1724, 1454, 1250, 814
$\mathrm{cm}^{-1}$. HRMS (EI) $m / z 246.1431$ [(M) ${ }^{+}$; calcd for $\mathrm{C}_{15} \mathrm{H}_{22} \mathrm{OSi}$, 246.1440].

## Preparation of 2-(2-phenyl-1-(triethylsilyl)cyclopropyl)acetaldehyde (3.2c)



Following General Procedure G, dihydropyran 3.1c ( $89 \mathrm{mg}, 0.324 \mathrm{mmol}, 1$ equiv) in THF ( 4 mL ) and sec-butyllithium ( $0.27 \mathrm{~mL}, 0.389 \mathrm{mmol}, 1.2$ equiv) at $-78^{\circ} \mathrm{C}$ for 1 h , followed by workup and silica gel chromatography ( $5 \%$ EtOAc in hexanes) afforded silylcyclopropane 3.2c as a colorless oil ( $62 \mathrm{mg}, 0.227 \mathrm{mmol}, 70 \%$ isolated yield, crude $d r=11: 1$, isolated $d r>20: 1$ ). ${ }^{1} \mathrm{H}$ NMR (500 MHz, $\left.\mathrm{CDCl}_{3}\right) \delta 9.90$ (dd, $\left.J=2.5,3.0 \mathrm{~Hz}, 1 \mathrm{H}\right), 7.30(\mathrm{~m}, 2 \mathrm{H}), 7.24$ (m, 2 H ), 7.18 (m, $1 \mathrm{H}), 2.72$ (dd, A of ABX system, $J=2.5,17.0 \mathrm{~Hz}, 1 \mathrm{H}), 2.14(\mathrm{dd}, J=6.0,8.0 \mathrm{~Hz}, 1 \mathrm{H}), 2.04$ (dd, B of ABX system, $J=2.0,17.0 \mathrm{~Hz}, 1 \mathrm{H}), 1.32(\mathrm{dd}, J=4.5,5.5 \mathrm{~Hz}, 1 \mathrm{H}), 0.91(\mathrm{dd}, J=4.5,8.0 \mathrm{~Hz}$, $1 \mathrm{H}), 0.78(\mathrm{t}, J=8.0 \mathrm{~Hz}, 9 \mathrm{H}), 0.30(\mathrm{~m}, 3 \mathrm{H}), 0.15(\mathrm{~m}, 3 \mathrm{H}) .{ }^{13} \mathrm{C} \mathrm{NMR}\left(126 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 203.2$, 139.4, 129.7 (2 C), 127.9 (2 C), 126.4, 53.6, 28.7, 15.7, 9.4, 7.5, 3.2. IR (film) 3060, 2956, 1725, 1450, 1250, $814 \mathrm{~cm}^{-1}$. HRMS (EI) $m / z 274.1747\left[(\mathrm{M})^{+}\right.$; calcd for $\mathrm{C}_{17} \mathrm{H}_{26} \mathrm{OSi}$, 274.1753].

Preparation of 2-(1-(dimethyl(phenyl)silyl)-2-(p-tolyl)cyclopropyl)acetaldehyde (3.2d)


Following General Procedure G, dihydropyran 3.1d ( $108 \mathrm{mg}, 0.35 \mathrm{mmol}$, 1 equiv) in THF
( 4.5 mL ) and sec-butyllithium ( $0.3 \mathrm{~mL}, 0.42 \mathrm{mmol}, 1.2$ equiv) at $-78^{\circ} \mathrm{C}$ for 10 minutes, followed by workup and silica gel chromatography ( $5 \% \mathrm{EtOAc}$ in hexanes) afforded silylcyclopropane $\mathbf{3 . 2 d}$ as a colorless oil $(93.7 \mathrm{mg}, 0.305 \mathrm{mmol}, 87 \%$ isolated yield, crude $d r=1.0: 0.3$, isolated $d r=$ 1.0:0.4). Mixture of diastereomers (1:0.4 ratio) ${ }^{1} \mathrm{H} \operatorname{NMR}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 9.67(\mathrm{t}, J=2.0 \mathrm{~Hz}$, $1 \mathrm{H}), 9,34(\mathrm{t}, J=2.0 \mathrm{~Hz}, 0.4 \mathrm{H}), 7.55(\mathrm{~m}, 0.8 \mathrm{H}), 7.39-7.27(\mathrm{~m}, 6.2 \mathrm{H}), 7.14(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 2 \mathrm{H})$, $7.05(\mathrm{~m}, 2.8 \mathrm{H}), 6.99(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 0.8 \mathrm{H}), 2.64(\mathrm{dd}, J=2.0,17.0 \mathrm{~Hz}, 1 \mathrm{H}), 2.33(\mathrm{~s}, 2 \mathrm{H}), 2.30(\mathrm{~s}$, $0.8 \mathrm{H}), 2.22-2.16(\mathrm{~m}, 1.8 \mathrm{H}), 1.99(\mathrm{dd}, J=2.0,17.5 \mathrm{~Hz}, 1 \mathrm{H}), 1.81(\mathrm{dd}, J=2.0,17.5 \mathrm{~Hz}, 0.4 \mathrm{H})$, $1.38(\mathrm{t}, J=4.5 \mathrm{~Hz}, 1 \mathrm{H}), 1.14(\mathrm{dd}, J=5.0,8.0 \mathrm{~Hz}, 0.4 \mathrm{H}), 1.07(\mathrm{t}, J=5.5 \mathrm{~Hz}, 0.4 \mathrm{H}), 0.92(\mathrm{dd}, J$ $=4.5,8.0 \mathrm{~Hz}, 1 \mathrm{H}), 0.36(\mathrm{~s}, 1.2 \mathrm{H}), 0.35(\mathrm{~s}, 1.2 \mathrm{H}), 0.10(\mathrm{~s}, 3 \mathrm{H}),-0.21(\mathrm{~s}, 3 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR (126 $\mathrm{MHz}, \mathrm{CDCl}_{3}$ ) Major diastereomer: $\delta$ 202.6, 138.0, 135.97, 135.96, 134.1, 134.0 (2 C), 129.7 (2 C), 128.6 (2 C), 127.7 (2 C), 53.2, 29.0, 21.1, 15.7, 10.3, -2.7, -3.4. Minor diastereomer (one aromatic carbon could not be located): $\delta 203.2,136.8,135.94,134.6,129.4,129.2$ (2 C), 129.0 (2 C), 128.9 (2 C), 127.9 (2 C), 45.1, 24.5, 21.0, 13.9, 9.3, -4.2, -4.4. IR (film) 3033, 2954, 1719, 1490, 1250, $830 \mathrm{~cm}^{-1}$. HRMS (EI) $m / z 308.1591\left[\left(\mathrm{M}^{+}\right)\right.$; calcd for $\left.\mathrm{C}_{20} \mathrm{H}_{24} \mathrm{OSi}, 308.1596\right]$.

Preparation of 2-(1-(benzyldimethylsilyl)-2-(p-tolyl)cyclopropyl)acetaldehyde (3.2e)


Following General Procedure G, dihydropyran 3.1e ( $78 \mathrm{mg}, 0.242 \mathrm{mmol}, 1$ equiv) in THF ( 3 mL ) and sec-butyllithium ( $0.18 \mathrm{~mL}, 0.254 \mathrm{mmol}, 1.05$ equiv) at $-78^{\circ} \mathrm{C}$ for 15 minutes, followed by workup and silica gel chromatography ( $5 \% \mathrm{EtOAc}$ in hexanes) afforded silylcyclopropane $\mathbf{3 . 2 e}$ as a colorless oil $(71.7 \mathrm{mg}, 0.223 \mathrm{mmol}, 92 \%$ isolated yield, crude $d r=1.0: 0.5$, isolated $d r=$
1.0:0.24). Mixture of diastereomers (1.0:0.24 ratio) ${ }^{1} \mathrm{H} \operatorname{NMR}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 9.90(\mathrm{dd}, J=$ $2.0,2.5 \mathrm{~Hz}, 1 \mathrm{H}), 7.20(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 2 \mathrm{H}), 7.12(\mathrm{t}, J=7.5 \mathrm{~Hz}, 2 \mathrm{H}), 7.07(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 2 \mathrm{H})$, $7.00(\mathrm{t}, J=7.5 \mathrm{~Hz}, 1 \mathrm{H}), 6.78(\mathrm{~m}, 2 \mathrm{H}), 2.79(\mathrm{dd}, \mathrm{A}$ of ABX system, $J=3.0,17.5 \mathrm{~Hz}, 1 \mathrm{H}), 2.32$ $(\mathrm{s}, 3 \mathrm{H}), 2.18(\mathrm{~m}, 1 \mathrm{H}), 2.04(\mathrm{dd}, \mathrm{B}$ of ABX system, $J=2.5,17.5 \mathrm{~Hz}, 1 \mathrm{H}), 1.86(\mathrm{~d}, \mathrm{C}$ of CDX system, $J=13.5 \mathrm{~Hz}, 1 \mathrm{H}), 1.80(\mathrm{dd}, \mathrm{D}$ of CDX system, $J=13.5 \mathrm{~Hz}, 1 \mathrm{H}), 1.22(\mathrm{t}, J=5.5 \mathrm{~Hz}, 1$ H), $0.86(\mathrm{dd}, J=5.0,8.5 \mathrm{~Hz}, 1 \mathrm{H}),-0.32(\mathrm{~s}, 3 \mathrm{H}),-0.45(\mathrm{~s}, 3 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) major diastereomer $\delta$ 202.8, 139.7, 136.2, 136.1, 129.8 (2 C), 128.7 (2 C), 128.2 (2 C), 128.0 (2 C), 124.0, 53.4, 29.3, 24.8, 21.1, 15.5, 9.8, -3.3, -3.6. IR (film) 3024, 2957, 2720, 1722, 1493, 1250, $827 \mathrm{~cm}^{-1}$. HRMS (EI) $m / z 231.1197\left[\left(\mathrm{M}_{-} \mathrm{C}_{7} \mathrm{H}_{7}\right)^{+}\right.$; calcd for $\left.\mathrm{C}_{14} \mathrm{H}_{19} \mathrm{OSi}, 231.1205\right]$. The minor diastereomer was also partially purified as the major component. (1.0:0.2 ratio) ${ }^{1} \mathrm{H}$ NMR (500 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 9.50(\mathrm{dd}, J=2.0,3.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.21(\mathrm{t}, J=7.5 \mathrm{~Hz}, 2 \mathrm{H}), 7.07(\mathrm{~m}, 3 \mathrm{H}), 7.02(\mathrm{~m}, 2$ H), $6.98(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 2 \mathrm{H}), 2.32(\mathrm{~s}, 3 \mathrm{H}), 2.18(\mathrm{~m}, 4 \mathrm{H}), 1.87(\mathrm{dd}, J=2.0,17.5 \mathrm{~Hz}, 1 \mathrm{H}), 1.12$ $(\mathrm{dd}, J=5.5,8.0 \mathrm{~Hz}, 1 \mathrm{H}), 1.05(\mathrm{t}, J=5.5 \mathrm{~Hz}, 1 \mathrm{H}),-0.01(\mathrm{~s}, 3 \mathrm{H}),-0.02(\mathrm{~s}, 3 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR (126 $\mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 203.3,139.5,136.0,134.6,129.3$ (2 C), 128.9 (2 C), 128.3 (4 C), 124.2, 45.4, 24.6, 24.2, 21.0, 13.9, 8.9, -4.5, -4.7. IR (film) $3025,2956,2720,1722,1493,1251,827 \mathrm{~cm}^{-1}$. HRMS (EI) $m / z 231.1209\left[\left(\mathrm{M}-\mathrm{C}_{7} \mathrm{H}_{7}\right)^{+}\right.$; calcd for $\mathrm{C}_{14} \mathrm{H}_{19} \mathrm{OSi}$, 231.1205].

Preparation of 2-(1-(ethyldimethylsilyl)-2-(o-tolyl)cyclopropyl)acetaldehyde (3.2f)


Following General Procedure G, dihydropyran $\mathbf{1 f}(95 \mathrm{mg}, 0.365 \mathrm{mmol}, 1$ equiv) in THF (4.6 mL) and sec-butyllithium ( $0.3 \mathrm{~mL}, 0.438 \mathrm{mmol}, 1.2$ equiv) at $-78^{\circ} \mathrm{C}$ for 13 minutes, followed
by workup and silica gel chromatography ( $5 \%$ EtOAc in hexanes) afforded silylcyclopropane $\mathbf{2 f}$ as a colorless oil ( $86.5 \mathrm{mg}, 0.332 \mathrm{mmol}, 91 \%$ isolated yield, crude $d r=8.3: 1.0$, isolated $d r=20: 1$ ). ${ }^{1} \mathrm{H} \operatorname{NMR}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 9.88(\mathrm{t}, J=2.5 \mathrm{~Hz}, 1 \mathrm{H}), 7.12-7.05(\mathrm{~m}, 4 \mathrm{H}), 2.56(\mathrm{dd}, J=2.5,16.5$ $\mathrm{Hz}, 1 \mathrm{H}$ ), 2.37 (overlapped, dd, $J=3.0,16.5 \mathrm{~Hz}, 1 \mathrm{H}), 2.37(\mathrm{~s}, 3 \mathrm{H}), 1.95(\mathrm{dd}, J=6.0,8.0 \mathrm{~Hz}, 1$ H), $1.37(\mathrm{dd}, J=5.0,6.0 \mathrm{~Hz}, 1 \mathrm{H}), 0.97(\mathrm{dd}, J=5.0,8.5 \mathrm{~Hz}, 1 \mathrm{H}), 0.76(\mathrm{t}, J=8.0 \mathrm{~Hz}, 3 \mathrm{H}), 0.26$ (m, 2 H$),-0.30(\mathrm{~s}, 3 \mathrm{H}),-0.39(\mathrm{~s}, 3 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 202.8,138.8,137.5$, 129.7, $128.3,126.6,125.3,53.5,29.0,20.1,14.7,10.3,7.3,6.8,-3.7,-3.9$. IR (film) $3029,2953,1722$, 1124, $830 \mathrm{~cm}^{-1}$. HRMS (EI) $\mathrm{m} / \mathrm{z} 260.1605$ [( $\left.\mathrm{M}^{+}\right)$; calcd for $\left.\mathrm{C}_{16} \mathrm{H}_{24} \mathrm{OSi}, 260.1596\right]$.

Synthesis of 2-(1-(ethyldimethylsilyl)-2-(3-methoxyphenyl)cyclopropyl)acetaldehyde (3.2g)


Following General Procedure G, dihydropyran 3.1g ( $68.9 \mathrm{mg}, 0.249 \mathrm{mmol}$, 1 equiv) in THF ( 3.1 mL ) and sec-butyllithium ( $0.2 \mathrm{~mL}, 0.274 \mathrm{mmol}, 1.1$ equiv) at $-78^{\circ} \mathrm{C}$ for 30 minutes, followed by workup and silica gel chromatography (5-10\% EtOAc in hexanes) afforded silylcyclopropane $\mathbf{3 . 1 g}$ as a colorless oil $(41.7 \mathrm{mg}, 0.152 \mathrm{mmol}, 61 \%$ isolated yield, crude $d r=2: 1$, isolated $d r=1.1: 1.0$ ). Mixture of diastereomers (cis/trans $1.1: 1)^{1} \mathrm{H}$ NMR $\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ $9.89(\mathrm{t}, J=2.5 \mathrm{~Hz}, 1.1 \mathrm{H}), 9.51(\mathrm{dd}, J=2.0,3.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.18(\mathrm{t}, J=8.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.15(\mathrm{t}, J=8.0$ $\mathrm{Hz}, 1.1 \mathrm{H}), 6.87(\mathrm{~m}, 2.1 \mathrm{H}), 6.71(\mathrm{~m}, 4.2 \mathrm{H}), 3.78(\mathrm{~s}, 3.3 \mathrm{H}), 3.77(\mathrm{~s}, 3 \mathrm{H}), 2.74(\mathrm{dd}, J=3.0,17.5$ $\mathrm{Hz}, 1.1 \mathrm{H}), 2.23-2.13(\mathrm{~m}, 3.1 \mathrm{H}), 2.01(\mathrm{dd}, \mathrm{t}, J=2.0,17 \mathrm{~Hz}, 1.1 \mathrm{H}), 1.83(\mathrm{dd}, \mathrm{t}, J=2.5,17.5 \mathrm{~Hz}$, $1 \mathrm{H}), 1.22(\mathrm{dd}, J=4.5,5.5 \mathrm{~Hz}, 1.1 \mathrm{H}), 1.12(\mathrm{dd}, J=5.0,8.0 \mathrm{~Hz}, 1 \mathrm{H}), 1.05(\mathrm{t}, J=5.5 \mathrm{~Hz}, 1 \mathrm{H})$, $0.97(\mathrm{t}, J=8.0 \mathrm{~Hz}, 3 \mathrm{H}), 0.86(\mathrm{~m}, 1.1 \mathrm{H}), 0.77(\mathrm{t}, J=8.0 \mathrm{~Hz}, 3.3 \mathrm{H}), 0.55(\mathrm{q}, J=8.0 \mathrm{~Hz}, 2 \mathrm{H}), 0.28$
$(\mathrm{m}, 2.2 \mathrm{H}), 0.00(\mathrm{~s}, 6 \mathrm{H}),-0.29(\mathrm{~s}, 3.3 \mathrm{H}),-0.39(\mathrm{~s}, 3.3 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR (126 MHz, $\left.\mathrm{CDCl}_{3}\right)$ major diastereomer $\delta 202.9,159.3,141.1,128.9,121.8,115.5,112.0,29.5,15.5,9.9,7.3,6.7,-4.0,-5.0$. Minor diastereomer $\delta 203.5,159.5,139.8,129.2,122.3,115.5,111.4,24.8,13.9,9.1,7.4,5.8$, 3.7, -5.2. IR (film) 2955, 1722, 1601, 1255, 1045, $835 \mathrm{~cm}^{-1}$. HRMS (EI) $\mathrm{m} / \mathrm{z} 276.1550\left[(\mathrm{M})^{+}\right.$; calcd for $\left.\mathrm{C}_{16} \mathrm{H}_{24} \mathrm{O}_{2} \mathrm{Si}, 276.1546\right]$

Preparation of 2-(1-(dimethyl(phenyl)silyl)-2-ferrocenylcyclopropyl)acetaldehyde (3.2h)


Following General Procedure G, dihydropyran 3.1h ( $117 \mathrm{mg}, 0.291 \mathrm{mmol}, 1$ equiv) in THF ( 3.6 mL ) and sec-butyllithium ( $0.25 \mathrm{~mL}, 0.349 \mathrm{mmol}, 1.2$ equiv) at $-78^{\circ} \mathrm{C}$ for 10 minutes, followed by workup and silica gel chromatography ( $5 \%$ EtOAc in hexanes) afforded silylcyclopropane $\mathbf{3 . 2 h}$ as an orange oil ( $80.3 \mathrm{mg}, 0.201 \mathrm{mmol}, 69 \%$, isolated yield, crude $d r=9.4: 1$, isolated $d r=1.0: 0.1$ ). Mixture of diastereomers (1.0:0.1 ratio) ${ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) major diastereomer: $\delta 9.53$ $(\mathrm{t}, J=7.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.31(\mathrm{~m}, 2 \mathrm{H}), 7.28(\mathrm{~m}, 3 \mathrm{H}), 4.30(\mathrm{~m}, 1 \mathrm{H}), 4.09(\mathrm{~m}, 1 \mathrm{H}), 4.08(\mathrm{~s}, 5 \mathrm{H}), 3.92$ $(\mathrm{m}, 1 \mathrm{H}), 3.76(\mathrm{~m}, 1 \mathrm{H}), 2.42(\mathrm{dd}, \mathrm{A}$ of ABX system, $J=2.0,17.0 \mathrm{~Hz}, 1 \mathrm{H}), 1.97(\mathrm{dd}, \mathrm{B}$ of ABX system, $J=2.0,17.0 \mathrm{~Hz}, 1 \mathrm{H}), 1.93(\mathrm{dd}, J=6.5,9.0 \mathrm{~Hz}, 1 \mathrm{H}), 1.03(\mathrm{~d}, J=4.5,6.0 \mathrm{~Hz}, 1 \mathrm{H}), 0.87$ (dd, $J=5.0,8.5 \mathrm{~Hz}, 1 \mathrm{H}), 0.06(\mathrm{~s}, 3 \mathrm{H}),-0.04(\mathrm{~s}, 3 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta$ 202.8, 138.1, 134.2 (2 C), 129.0, 127.6 (2 C), 86.5, 70.1, 69.3, 68.7 ( 5 C), $68.2,66.0,53.3,24.9,16.8,11.2,-$ 2.5, -2.9. IR (film) $3091,2988,1722,1427,1250,1107,816 \mathrm{~cm}^{-1}$. HRMS (EI) $\mathrm{m} / \mathrm{z} 402.1093\left[\left(\mathrm{M}^{+}\right)\right.$; calcd for $\mathrm{C}_{23} \mathrm{H}_{26} \mathrm{OSiFe}$, 402.1102].

## Preparation of 2-(1-(ethyldimethylsilyl)-2-(thiophen-2-yl)cyclopropyl)acetaldehyde (3.2i)



Following General Procedure G, dihydropyran $\mathbf{3 . 1 i}$ ( $39.5 \mathrm{mg}, 0.157 \mathrm{mmol}, 1$ equiv) in THF ( 2 mL ) and sec-butyllithium ( $0.25 \mathrm{~mL}, 0.344 \mathrm{mmol}, 2.2$ equiv) at $-78^{\circ} \mathrm{C}$ for 1 h , followed by workup and silica gel chromatography ( $5 \%$ EtOAc in hexanes) afforded silylcyclopropane $\mathbf{3 . 2}$ as a colorless oil ( $28 \mathrm{mg}, 0.112 \mathrm{mmol}, 71 \%$ isolated yield, crude $d r=1.0: 0.5$, isolated $d r=1.0: 0.5$ ). Mixture of diastereomers (trans/cis 1.0:0.5). ${ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 9.84$ (dd, $J=2.0,3.0$ $\mathrm{Hz}, 0.5 \mathrm{H}), 9.57(\mathrm{dd}, J=2.0,3.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.10(\mathrm{~m}, 1.5 \mathrm{H}), 6.90(\mathrm{dd}, J=3.5,5.0 \mathrm{~Hz}, 1.0 \mathrm{H}), 0.87$ $(\mathrm{dd}, J=3.5,5.0 \mathrm{~Hz}, 0.5 \mathrm{H}), 6.81(\mathrm{dt}, J=1.5,3.5 \mathrm{~Hz}, 0.5 \mathrm{H}), 6.73(\mathrm{dt}, J=1.5,3.5 \mathrm{~Hz}, 1 \mathrm{H}), 2.65$ (dd, A of ABX system, $J=3.0,17.5 \mathrm{~Hz}, 0.5 \mathrm{H}), 2.30(\mathrm{dd}, \mathrm{C}$ of CDX system, $J=3.0,17.5 \mathrm{~Hz}, 1$ H), 2.24 (dd, $J=5.5,7.0 \mathrm{~Hz}, 1 \mathrm{H}$ ), 2.14 (dd, $J=6.0,7.5 \mathrm{~Hz}, 0.5 \mathrm{H}$ ), 2.02 (dd, B of ABX system, $J=2.0,17.0 \mathrm{~Hz}, 0.5 \mathrm{H}), 1.95(\mathrm{dd}, \mathrm{D}$ of CDX system, $J=2.0,17.5 \mathrm{~Hz}, 1 \mathrm{H}), 1.25(\mathrm{~m}, 1.5 \mathrm{H}), 1.01$ $(\mathrm{t}, J=5.0 \mathrm{~Hz}, 1 \mathrm{H}), 1.00(\mathrm{~m}, 0.5 \mathrm{H}), 0.97(\mathrm{t}, J=8.0 \mathrm{~Hz}, 3 \mathrm{H}), 0.81(\mathrm{t}, J=8.0 \mathrm{~Hz}, 1.5 \mathrm{H}), 0.54(\mathrm{q}$, $J=8.0 \mathrm{~Hz}, 2 \mathrm{H}), 0.40-0.28(\mathrm{~m}, 1 \mathrm{H}),-0.01(\mathrm{~s}, 3 \mathrm{H}), 0.02(\mathrm{~s}, 3 \mathrm{H}),-0.21(\mathrm{~s}, 1.5 \mathrm{H}),-0.32(\mathrm{~s}, 1.5 \mathrm{H})$. ${ }^{13} \mathrm{C}$ NMR ( $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) major diastereomer: $\delta 203.2,142.6,126.9,126.0,124.1,19.2,16.4$, 10.1, 7.4, 5.8, -5.0, -5.3. Minor diastereomer: $\delta$ 202.7, 144.1, 126.5, 126.2, 124.0, 23.3, 17.6, 11.2, 7.3, 6.4, -4.0, -4.3. IR (film) 2953, 1722, 1250, $833 \mathrm{~cm}^{-1}$. HRMS (EI) $\mathrm{m} / \mathrm{z} 252.1001\left[(\mathrm{M})^{+}\right.$; calcd for $\mathrm{C}_{13} \mathrm{H}_{20} \mathrm{OSiS}$, 252.1004].

## Preparation of 2-(1-(dimethyl(phenyl)silyl)-2-propylcyclopropyl)acetaldehyde (3.2j)



Following general procedure H , dihydropyran $\mathbf{3 . 1 j}$ ( $82.8 \mathrm{mg}, 0.318$, 1 equiv), in THF ( 4 mL ) and sec-butyllithium ( $0.27 \mathrm{~mL}, 0.381 \mathrm{mmol}, 1.2$ equiv) at $-78^{\circ} \mathrm{C}$ and warming to $-10^{\circ} \mathrm{C}$ for 2 h , followed by aqueous workup and silica gel chromatography ( $5 \% \mathrm{EtOAc}$ in hexanes) afforded silylcyclopropane $\mathbf{3 . 2 j}$ as a colorless oil $(68.6 \mathrm{mg}, 0.316 \mathrm{mmol}, 83 \%$ isolated yield, crude $d r=$ 4.3:1.0, isolated $d r=1.0: 0.2$ ). Mixture of diastereomers (cis/trans $=1.0: 0.2$ ratio). ${ }^{1} \mathrm{H}$ NMR ( 600 $\mathrm{MHz}, \mathrm{CDCl}_{3}$ ) major diastereomer (cis): $\delta 9.53(\mathrm{t}, J=2.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.50(\mathrm{~m}, 2 \mathrm{H}), 7.34(\mathrm{~m}, 3 \mathrm{H})$, 2.41 (dd, A of ABX system, $J=2.4,16.8 \mathrm{~Hz}, 1 \mathrm{H}), 1.80(\mathrm{dd}, \mathrm{B}$ of ABX system, $J=2.4,17.4 \mathrm{~Hz}$, $1 \mathrm{H}), 1.56(\mathrm{~m}, 1 \mathrm{H}), 1.38(\mathrm{~m}, 2 \mathrm{H}), 1.11(\mathrm{~m}, 1 \mathrm{H}), 0.88(\mathrm{t}, J=7.2 \mathrm{~Hz}, 3 \mathrm{H}), 0.81(\mathrm{~m}, 1 \mathrm{H}), 0,61$ (dd, C of CDX system, $J=4.2,8.4 \mathrm{~Hz}, 1 \mathrm{H}$ ), 0.56 (dd, D of CDX system, $J=4.2,5.4 \mathrm{~Hz}, 1 \mathrm{H}$ ), $0.36(\mathrm{~s}, 3 \mathrm{H}), 0.32(\mathrm{~s}, 3 \mathrm{H})$. Minor diastereomer (trans): $\delta 9.50(\mathrm{t}, J=2.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.48(\mathrm{~m}, 2 \mathrm{H})$, 2.38 (dd, A of ABX system, $J=3.0,17.4 \mathrm{~Hz}, 1 \mathrm{H}$ ), 2.23 (dd, B of ABX system, $J=2.4,17.4 \mathrm{~Hz}$, $1 \mathrm{H}), 0.24(\mathrm{~s}, 6 \mathrm{H})$, all other protons are overlapped with major diastereomer, presumably at 7.34 $(3 \mathrm{H}), 1.38(4 \mathrm{H}), 0.92-079(6 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $151 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) mixture of diastereomers (cis / trans $=1.0: 0.2$ ratio), Major (cis) diastereomer: $\delta 203.3,138.4,134.1$ (2 C), 129.1, 127.8 (2 C), 53.4, 33.5, 26.0, 23.3, 17.9, 13.9, 6.4, -1.4, -2.1. IR (neat) 3070, 2957, 2872, 1725, 1427, 1251, 1111, $816 \mathrm{~cm}^{-1}$. HRMS (EI) $m / z 260.1590\left[(\mathrm{M})^{+} ;\right.$calcd for $\left.\mathrm{C}_{16} \mathrm{H}_{24} \mathrm{OSi}, 260.1596\right]$.

## Preparation of 2-(2-cyclohexyl-1-(dimethyl(phenyl)silyl)cyclopropyl)acetaldehyde (3.2k)



Following general procedure $H$, dihydropyran $\mathbf{3 . 1 k}$ ( $67 \mathrm{mg}, 0.223$, 1 equiv), in THF ( 2.8 mL ) and $n$-butyllithium ( $0.28 \mathrm{~mL}, 0.446 \mathrm{mmol}, 2$ equiv) at $-78^{\circ} \mathrm{C}$ and warming to $-10^{\circ} \mathrm{C}$ for 2 h , followed by aqueous workup and silica gel chromatography (5\% EtOAc in hexanes) afforded silylcyclopropane 3.2k as a colorless oil $(50.4 \mathrm{mg}, 0.170 \mathrm{mmol}, 76 \%$ isolated yield, crude $d r=$ 1.6:1.0, isolated $d r=1.0: 0.6$ ). Mixture of diastereomers ( $\sim 1: 0.6$ ratio). ${ }^{1} \mathrm{H} \mathrm{NMR}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$ $\delta 9.54(\mathrm{dd}, J=2.0,3.0 \mathrm{~Hz}, 1 \mathrm{H}), 9.47(\mathrm{dd}, J=2.0,3.0 \mathrm{~Hz}, 0.6 \mathrm{H}), 7.49(\mathrm{~m}, 3.2 \mathrm{H}), 7.34(\mathrm{~m}, 4.8$ H), $2.54(\mathrm{dd}, J=3.0,17.0 \mathrm{~Hz}, 1 \mathrm{H}), 2.50(22, J=3.5,17.5 \mathrm{~Hz}, 0.6 \mathrm{H}), 2.06(\mathrm{dd}, J=2.0,17.0 \mathrm{~Hz}$, $0.6 \mathrm{H}), 1.84(\mathrm{~m}, 1 \mathrm{H}), 1.75-1.56(\mathrm{~m}, 8.80 \mathrm{H}), 1.17-0.93(\mathrm{~m}, 8.64 \mathrm{H}), 0.89-0.80(\mathrm{~m}, 2 \mathrm{H}), 0.70(\mathrm{~m}$, $1 \mathrm{H}), 0.63-0.52(\mathrm{~m}, 3.2 \mathrm{H}), 0.38(\mathrm{~s}, 3 \mathrm{H}), 0.31(\mathrm{~s}, 3 \mathrm{H}),-0.23(\mathrm{~d}, 3.6 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( 126 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta 203.9,203.7,138.2,137.1,134.1,134.0,129.3,129.1,127.8,127.7,53.2,44.5,39.1$, $37.6,34.3,33.43,33.41,33.38,33.3,26.6,26.4,26.33,26.31,26.30,26.0,25.9,16.7,14.1,6.8$, 5.3, -1.4, -2.2, -4.6, -4.7. HRMS (EI) $m / z 300.1909\left[(\mathrm{M})^{+}\right.$; calcd for $\left.\mathrm{C}_{19} \mathrm{H}_{28} \mathrm{OSi}, 300.1909\right]$.

Preparation of 2-(2-(dimethyl(phenyl)silyl)-[1,1'-bi(cyclopropan)]-2-yl)acetaldehyde (3.2l)


Following general procedure H , dihydropyran $\mathbf{3 . 1 1}(66.8 \mathrm{mg}, 0.257$, 1 equiv), in THF ( 3.2
mL ) and sec-butyllithium ( $0.22 \mathrm{~mL}, 0.309 \mathrm{mmol}, 1.2$ equiv) at $-78^{\circ} \mathrm{C}$ and warming to $-10^{\circ} \mathrm{C}$ for 2.5 h , followed by aqueous workup and silica gel chromatography ( $5 \% \mathrm{EtOAc}$ in hexanes) afforded silylcyclopropane $\mathbf{3 . 2 l}$ as a colorless oil ( $51.6 \mathrm{mg}, 0.188 \mathrm{mmol}, 73 \%$ isolated yield, crude $d r=1: 1$, isolated $d r=1: 1$ ). Mixture of diastereomers (1:1 ratio) ${ }^{1} \mathrm{H}$ NMR $\left(600 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ $9.58(\mathrm{t}, J=3.0 \mathrm{~Hz}, 1 \mathrm{H}), 9.49(\mathrm{t}, J=2.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.54(\mathrm{~m}, 2 \mathrm{H}), 7.47(\mathrm{~m}, 2 \mathrm{H}), 7.34(\mathrm{~m}, 6 \mathrm{H}), 2.49$ (dd, A of ABX system, $J=2.4,17.4 \mathrm{~Hz}, 1 \mathrm{H}), 2.40(\mathrm{dd}, \mathrm{B}$ of ABX system, $J=2.4,18.0 \mathrm{~Hz}, 1 \mathrm{H}$ ), 2.36 (dd, C of CDX system, $J=2.4,16.8 \mathrm{~Hz}, 1 \mathrm{H}$ ), 1.81 (dd, D of CDX system, $J=2.4,16.8 \mathrm{~Hz}$, $1 \mathrm{H}), 0.78-0.72(\mathrm{~m}, 2 \mathrm{H}), 0.71(\mathrm{t}, J=4.8 \mathrm{~Hz}, 1 \mathrm{H}), 0.61-0.57(\mathrm{~m}, 2 \mathrm{H}), 0.54-0.43(\mathrm{~m}, 6 \mathrm{H}), 0.40$ $(\mathrm{s}, 3 \mathrm{H}), 0.37(\mathrm{~m}, 3 \mathrm{H}), 0.31(\mathrm{t}, J=4.8 \mathrm{~Hz}, 1 \mathrm{H}), 0.26(\mathrm{~m}, 1 \mathrm{H}), 0.23(\mathrm{~s}, 3 \mathrm{H}), 0.22(\mathrm{~s}, 3 \mathrm{H}), 0.21$ ( m , overlapped with Me singlet at $0.22,1 \mathrm{H}$ ), $0.17-0.12(\mathrm{~m}, 2 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR $\left(151 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ 204.0, 203.1, 138.3, 137.1, 134.2 (2 C), 134.0 (2 C), 129.3, 129.1, 127.83 (2 C), 127.80 (2 C), $53.1,45.5,29.8,23.2,17.3,14.7,11.9,9.5,7.3,6.7,6.1,6.0,5.3,4.5,-1.9,-2.2,-4.52,-4.55$. IR (neat) $3071,3000,2959,2816,1722,1427,1250,1113,815 \mathrm{~cm}^{-1}$. HRMS (EI) $\mathrm{m} / \mathrm{z} 258.1440\left[(\mathrm{M})^{+}\right.$; calcd for $\mathrm{C}_{16} \mathrm{H}_{22} \mathrm{OSi}$, 258.1440].

## Preparation of 2-(2-(triethylsilyl)-[1,1'-bi(cyclopropan)]-2-yl)acetaldehyde (3.2m)



Following general procedure H , dihydropyran $\mathbf{3 . 1 m}(69.5 \mathrm{mg}, 0.292$, 1 equiv), in THF ( 3.6 mL ) and sec-butyllithium ( $0.25 \mathrm{~mL}, 0.35 \mathrm{mmol}, 1.2$ equiv) at $-78^{\circ} \mathrm{C}$ and warming to $-10^{\circ} \mathrm{C}$ for 3 h, followed by aqueous workup and silica gel chromatography (3.5\% EtOAc in hexanes) afforded silylcyclopropane $\mathbf{3 . 2 m}$ as a colorless oil $(52.4 \mathrm{mg}, 0.219 \mathrm{mmol}, 75 \%$ isolated yield, crude $d r=$
1.5:1.0, isolated $d r=1.0: 0.7$ ). Mixture of diastereomers (1:0.7 ratio, relative stereochemistry was not assigned) ${ }^{1} \mathrm{H}$ NMR ( $600 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) (1:0.7 ratio) $\delta 9.79(\mathrm{t}, J=2.5 \mathrm{~Hz}, 0.7 \mathrm{H}), 9.68(\mathrm{dd}, J=$ $2.5,3.5 \mathrm{~Hz}, 1 \mathrm{H}$ ), 2.49 (dd, A of ABX system, $J=2.5,17.0 \mathrm{~Hz}, 0.7 \mathrm{H}$ ), 2.42 (dd, C of CDX system, $J=3.0,16.5 \mathrm{~Hz}, 1 \mathrm{H}), 2.40(\mathrm{dd}, \mathrm{B}$ of ABX system, $J=2.5,17.0 \mathrm{~Hz}, 0.7 \mathrm{H}), 1.77(\mathrm{dd}, \mathrm{D}$ of CDX system, $J=2.5,17.0 \mathrm{~Hz}, 1 \mathrm{H}), 0.97(\mathrm{t}, J=8.0 \mathrm{~Hz}, 9 \mathrm{H}), 0.92(\mathrm{t}, J=8.0 \mathrm{~Hz}, 6.3 \mathrm{H}), 0.74(\mathrm{dd}, J=$ $4.5,8.0 \mathrm{~Hz}, 0.7 \mathrm{H}$ ), $0.69-0.60\left(\mathrm{~m}\right.$, heavily overlapped with $\mathrm{SiCH}_{2}$, not quantified), 0.60 (q, $J=8.0$ $\mathrm{Hz}, 6 \mathrm{H}), 0.52-0.47(\mathrm{~m}, \sim 6 \mathrm{H}), 0.46(\mathrm{q}, J=8.0 \mathrm{~Hz}, 4.2 \mathrm{H}), 0.37(\mathrm{~m}, 1 \mathrm{H}), 0.28-0.17(\mathrm{~m}, \sim 4 \mathrm{H})$. ${ }^{13} \mathrm{C}$ NMR ( $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) major diastereomer: $\delta$ 203.9, 53.5, 29.2, 17.4, 12.2, 7.6, 6.0, 5.3, 4.5, 3.7. Minor diastereomer: $\delta$ 204.5, 46.2, 22.8, 14.5, 9.6, 7.4, 6.5, 5.6, 4.3, 2.2. IR (film) 3078, 2999, 2953, 1724, 1458, 1018, $733 \mathrm{~cm}^{-1}$. HRMS (EI) $\mathrm{m} / \mathrm{z} 238.1756\left[(\mathrm{M})^{+}\right.$; calcd for $\mathrm{C}_{14} \mathrm{H}_{26} \mathrm{OSi}$, 238.1753].

Preparation of 2-(2-(3-chlorophenyl)-1-(ethyldimethylsilyl)cyclopropyl)acetaldehyde (2n), 5-(3-chlorophenyl)-3-(ethyldimethylsilyl)cyclopent-2-en-1-ol (3n), and 1-(3-chlorophenyl)-3-(ethyldimethylsilyl)cyclopent-3-en-1-ol (4n)


Following General Procedure G, dihydropyran 3.1n ( $90 \mathrm{mg}, 0.32 \mathrm{mmol}, 1$ equiv) in THF ( 4 mL ) and sec-butyllithium ( $0.24 \mathrm{~mL}, 0.336 \mathrm{mmol}, 1.05$ equiv) at $-78^{\circ} \mathrm{C}$ for 15 minutes, followed by workup and silica gel chromatography (6-15\% EtOAc in hexanes) afforded silylcyclopropane 3.2n $(16.4 \mathrm{mg}, 0.0544 \mathrm{mmol}, 17 \%$ isolated yield, crude $d r=5.4: 1$, isolated $d r=1.0: 0.12$ ),
secondary alcohol $\mathbf{3 . 3 n}(49.7 \mathrm{mg}, 0.173 \mathrm{mmol}, 54 \%$ isolated yield, $d r=20: 1)$, and tertiary alcohol $\mathbf{3 . 4 n}(6.3 \mathrm{mg}, 0.0192 \mathrm{mmol}, 6 \%$ isolated yield), all as colorless oils, together with unreacted $\mathbf{3 . 1} \mathbf{n}$ ( $21.5 \mathrm{mg}, 0.074 \mathrm{mmol}, 23 \%$ recovery).

Spectroscopic data for $\mathbf{3 . 2 n}$ : mixture of diastereomers (cis / trans $=1.0: 0.12$ ratio) ${ }^{1} \mathrm{H}$ NMR ( $600 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) major (cis) diastereomer: $\delta 9.87$ (dd, $J=1.8,3.0 \mathrm{~Hz}, 1 \mathrm{H}$ ), $7.28(\mathrm{~m}, 1 \mathrm{H}), 7.21$ (m, 1 H), 7.19-7.15 (m, 2 H), 2.78 (dd, $J=2.4,17.4 \mathrm{~Hz}, 1 \mathrm{H}), 2.12(\mathrm{dd}, J=6.6,8.4 \mathrm{~Hz}, 1 \mathrm{H}), 2.00$ $(\mathrm{dd}, J=1.8,17.4 \mathrm{~Hz}, 1 \mathrm{H}), 1.20(\mathrm{dd}, J=5.4,6.0 \mathrm{~Hz}, 1 \mathrm{H}), 0.87(\mathrm{dd}, J=5.4,8.4 \mathrm{~Hz}, 1 \mathrm{H}), 0.77(\mathrm{t}$, $J=7.8 \mathrm{~Hz}, 3 \mathrm{H}), 0.26(\mathrm{~m}, 2 \mathrm{H}),-0.31(\mathrm{~s}, 3 \mathrm{H}),-0.41(\mathrm{~s}, 3 \mathrm{H})$. Minor (trans) diastereomer: $\delta 9.51$ (dd, $J=1.8,3.0 \mathrm{~Hz}, 0.15 \mathrm{H}$ ), 7.20-7.14 (m, heavily overlapped with major diastereomer, 0.45 H ), $7.03(\mathrm{~m}, 0.15 \mathrm{H}), 2.17(\mathrm{~m}, 0.30 \mathrm{H}), 1.79(\mathrm{dd}, J=1.8,17.4 \mathrm{~Hz}, 0.15 \mathrm{H}), 1.14(\mathrm{dd}, J=5.4,7.8 \mathrm{~Hz}$, $0.15 \mathrm{H}), 1.03(\mathrm{t}, J=5.4 \mathrm{~Hz}, 0.15 \mathrm{H}), 0.96(\mathrm{t}, J=7.8 \mathrm{~Hz}, 0.45 \mathrm{H}), 0.55(\mathrm{q}, J=7.8 \mathrm{~Hz}, 0.30 \mathrm{H}), 0.01$ (s, 0.9 H ). ${ }^{13} \mathrm{C}$ NMR ( $151 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) major diastereomer $\delta 202.4,141.8,133.8,129.9,129.2$, $128.2,126.6,53.3,29.0,15.5,10.2,7.3,6.7,-3.7,-3.9$. IR (film) $3422,3061,2955,2876,1724$, 1250, $814 \mathrm{~cm}^{-1}$. HRMS (EI) $m / z 228.1042\left[(\mathrm{M})^{+}\right.$; calcd for $\left.\mathrm{C}_{15} \mathrm{H}_{21} \mathrm{OSiCl}, 280.1050\right]$.

Spectroscopic data for 3.3n: ${ }^{1} \mathrm{H}$ NMR $\left(600 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.22(\mathrm{t}, J=1.8 \mathrm{~Hz}, 1 \mathrm{H}), 7.20$ $(\mathrm{d}, J=7.2 \mathrm{~Hz}, 1 \mathrm{H}), 7.17(\mathrm{~m}, 1 \mathrm{H}), 7.11(\mathrm{~m}, 1 \mathrm{H}), 5.99(\mathrm{q}, J=1.8 \mathrm{~Hz}, 1 \mathrm{H}), 4.83(\mathrm{~m}, 1 \mathrm{H}), 3.12$ (m, 1 H ), 2.95 (ddt, $J=1.8 .8 .4,16.8 \mathrm{~Hz}, 1 \mathrm{H}), 2.41(\mathrm{ddt}, J=1.8,6.6,16.8 \mathrm{~Hz}, 1 \mathrm{H}), 1.80(\mathrm{~s}, 1 \mathrm{H})$, $0.95(\mathrm{t}, J=7.8 \mathrm{~Hz}, 3 \mathrm{H}), 0.60(\mathrm{q}, J=7.8 \mathrm{~Hz}, 2 \mathrm{H}), 0.09(\mathrm{~s}, 6 \mathrm{H}) .{ }^{13} \mathrm{C} \mathrm{NMR}\left(151 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ $147.5,146.3,141.8,134.3,129.8,127.3,126.4,125.4,86.4,54.9,43.0,7.4,6.6,-4.2,-4.3 . \operatorname{IR}$ (film) 3352, 2955, 1458, 1250, 1089, $837 \mathrm{~cm}^{-1}$. HRMS (EI) $\mathrm{m} / \mathrm{z} 263.1019\left[(\mathrm{M}-\mathrm{OH})^{+}\right.$; calcd for $\left.\mathrm{C}_{15} \mathrm{H}_{20} \mathrm{SiCl}, 263.1023\right]$.

Spectroscopic data for $\mathbf{3 . 4 n}:{ }^{1} \mathrm{H}$ NMR ( $600 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.49(\mathrm{t}, J=1.8 \mathrm{~Hz}, 1 \mathrm{H}), 7.34$ (ddd, $J=1.2,1.8,7.8 \mathrm{~Hz}, 1 \mathrm{H}), 7.24(\mathrm{~m}, 1 \mathrm{H}), 7.20(\mathrm{ddd}, J=1.2,2.4,7.8 \mathrm{~Hz}, 1 \mathrm{H}), 6.01(\mathrm{~m}, 1 \mathrm{H})$,
2.97 (dq, $J=1.8,18.0 \mathrm{~Hz}, 1 \mathrm{H}), 2.87(\mathrm{~m}, 1 \mathrm{H}), 2.80-2.75(\mathrm{~m}, 2 \mathrm{H}), 2.04(\mathrm{~s}, 1 \mathrm{H}), 0.94(\mathrm{t}, J=7.8$ $\mathrm{Hz}, 3 \mathrm{H}), 0.58(\mathrm{q}, J=7.8 \mathrm{~Hz}, 2 \mathrm{H}), 0.08(\mathrm{~s}, 6 \mathrm{H}) .{ }^{13} \mathrm{C} \operatorname{NMR}\left(151 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 149.0,142.3$, $137.8,134.1,129.4,126.8,125.4,123.1,82.9,53.9,52.4,7.4,6.7,-4.11,-4.13$. IR (film) 3397, 3031, 2955, 1253, $839 \mathrm{~cm}^{-1}$. HRMS (EI) $m / z 263.1009\left[(\mathrm{M}-\mathrm{OH})^{+}\right.$; calcd for $\mathrm{C}_{15} \mathrm{H}_{20} \mathrm{SiCl}$, 263.1023].

Preparation of trans-5-(pyridin-2-yl)-3-(triethylsilyl)cyclopent-2-en-1-ol (3.30), cis-5-(pyridin-2-yl)-3-(triethylsilyl)cyclopent-2-en-1-ol (3.3o'), and 6-(pyridin-2-yl)-4-(triethylsilyl)-5,6-dihydro-2H-pyran-2-one (3.5o)


Following General Procedure G, dihydropyran 3.1 o ( $248 \mathrm{mg}, 0.90 \mathrm{mmol}, 1$ equiv) in THF $(15 \mathrm{~mL})$ and 1.4 M sec-butyllithium ( $1.9 \mathrm{~mL}, 2.7 \mathrm{mmol}, 3.0$ equiv) at $-78^{\circ} \mathrm{C}$ for 2 hours, followed by workup and silica gel chromatography ( $35 \%$ EtOAc in hexanes) afforded silylcyclopentenols 3.3 o ( $56 \mathrm{mg}, 0.198 \mathrm{mmol}, 22 \%$ isolated yield, $d r=20: 1$ ), 3.3o' $(29 \mathrm{mg}, 0.099 \mathrm{mmol}, 11 \%$ isolated yield, $d r=20: 1$ ), and lactone $\mathbf{3 . 5 0}$ ( $40 \mathrm{mg}, 0.144 \mathrm{mmol}, 16 \%$ isolated), all as colorless oils.

Spectroscopic data for 3.30: ${ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 8.54$ (ddd, $J=4.9,1.9,0.9 \mathrm{~Hz}$, $1 \mathrm{H}), 7.62(\mathrm{td}, J=7.7,1.9 \mathrm{~Hz}, 1 \mathrm{H}), 7.21(\mathrm{~d}, J=7.8 \mathrm{~Hz}, 1 \mathrm{H}), 7.14(\mathrm{ddd}, J=7.6,4.9,1.2 \mathrm{~Hz}, 1 \mathrm{H})$, $6.07(\mathrm{dt}, J=2.7,1.6 \mathrm{~Hz}, 1 \mathrm{H}), 5.14(\mathrm{dq}, J=7.0,1.9 \mathrm{~Hz}, 1 \mathrm{H}), 3.36(\mathrm{td}, J=8.3,6.9 \mathrm{~Hz}, 1 \mathrm{H}), 2.94$ (ddt, $J=15.8,8.3,1.6 \mathrm{~Hz}, 1 \mathrm{H}), 2.59(\mathrm{ddt}, J=15.8,8.5,2.3 \mathrm{~Hz}, 1 \mathrm{H}), 0.96(\mathrm{t}, J=7.9 \mathrm{~Hz}, 9 \mathrm{H}), 0.64$ $(\mathrm{q}, J=8.0 \mathrm{~Hz}, 6 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR $\left(126 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 162.9,149.1,143.8,143.3,136.5,122.0,121.4$, 84.7, 57.0, 41.0, 7.4, 2.9. HRMS (ESI) $m / z 274.1630\left[(\mathrm{M}-\mathrm{H})^{+}\right.$; calcd for $\mathrm{C}_{16} \mathrm{H}_{24} \mathrm{NOSi}$ 274.1627]. Spectroscopic data for 3.30': ${ }^{1} \mathrm{H}$ NMR $\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 8.58(\mathrm{ddd}, J=4.9,1.8,0.9 \mathrm{~Hz}$,
$1 \mathrm{H}), 7.71(\mathrm{td}, J=7.7,1.8 \mathrm{~Hz}, 1 \mathrm{H}), 7.49-7.46(\mathrm{~m}, 1 \mathrm{H}), 7.20(\mathrm{ddd}, J=7.6,4.9,1.2 \mathrm{~Hz}, 1 \mathrm{H}), 6.09$ $(\mathrm{td}, J=2.8,1.2 \mathrm{~Hz}, 1 \mathrm{H}), 5.54(\mathrm{t}, J=2.1 \mathrm{~Hz}, 1 \mathrm{H}), 5.02(\mathrm{dd}, J=11.2,3.2 \mathrm{~Hz}, 1 \mathrm{H}), 3.13(\mathrm{~s}, 1 \mathrm{H})$, $2.44(\mathrm{ddd}, J=17.5,3.2,1.2 \mathrm{~Hz}, 1 \mathrm{H}), 2.25(\mathrm{dddd}, J=17.5,11.1,2.8,1.5 \mathrm{~Hz}, 1 \mathrm{H}), 0.94(\mathrm{t}, J=7.9$ $\mathrm{Hz}, 9 \mathrm{H}), 0.63(\mathrm{q}, J=8.1 \mathrm{~Hz}, 6 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 161.2,149.0,139.3,136.8$, 134.2, 122.4, 120.5, 89.6, 69.6, 33.2, 7.3, 2.2. HRMS (ESI) $m / z 276.1787\left[(\mathrm{M}+\mathrm{H})^{+}\right.$; calcd for $\mathrm{C}_{16} \mathrm{H}_{26} \mathrm{NOSi}$, 276.1784].

Spectroscopic data for $\mathbf{3 . 5 0}:{ }^{1} \mathrm{H}$ NMR $\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 8.56(\mathrm{dt}, J=4.9,1.3 \mathrm{~Hz}, 1 \mathrm{H})$, $7.75(\mathrm{td}, J=7.7,1.8 \mathrm{~Hz}, 1 \mathrm{H}), 7.59(\mathrm{~d}, J=7.9 \mathrm{~Hz}, 1 \mathrm{H}), 7.26-7.22(\mathrm{~m}, 1 \mathrm{H}), 6.27(\mathrm{dd}, J=2.4,1.1$ $\mathrm{Hz}, 1 \mathrm{H}), 5.49(\mathrm{dd}, J=10.6,4.1 \mathrm{~Hz}, 1 \mathrm{H}), 2.88(\mathrm{ddd}, J=18.1,4.1,1.1 \mathrm{~Hz}, 1 \mathrm{H}), 2.67(\mathrm{ddd}, J=18.1$, $10.7,2.4 \mathrm{~Hz}, 1 \mathrm{H}), 0.92(\mathrm{t}, J=7.9 \mathrm{~Hz}, 9 \mathrm{H}), 0.69(\mathrm{q}, J=7.6 \mathrm{~Hz}, 6 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR $\left(126 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$ $\delta 162.5,160.9,158.0,149.0,137.0,128.1,123.0,120.5,79.2,32.5,7.1,1.8$. HRMS (ESI) $\mathrm{m} / \mathrm{z}$ $290.1579\left[(\mathrm{M}+\mathrm{H})^{+}\right.$; calcd for $\left.\mathrm{C}_{16} \mathrm{H}_{24} \mathrm{NO}_{2} \mathrm{Si}, 290.1576\right]$.

Preparation of (E)-3-(dimethyl(phenyl)silyl)-5-(naphthalen-2-yl)pent-3-enal (3.6p), and ( $E$ )-3-(dimethyl(phenyl)silyl)-5-(naphthalen-2-yl)pent-4-enal (3.7p)


Following General Procedure G, dihydropyran 3.1p ( $150 \mathrm{mg}, 0.44 \mathrm{mmol}$, 1 equiv) in THF $(15 \mathrm{~mL})$ and 1.4 M sec-butyllithium ( $0.63 \mathrm{~mL}, 0.88 \mathrm{mmol}, 2.0$ equiv) at $-78{ }^{\circ} \mathrm{C}$ for 2 hours, followed by workup and silica gel chromatography (5\% EtOAc in hexanes) afforded aldehydes 3.6p ( $39.2 \mathrm{mg}, 0.114 \mathrm{mmol}, 26 \%$ isolated yield), $\mathbf{3 . 7} \mathbf{p}$ ( $20 \mathrm{mg}, 0.057 \mathrm{mmol}, 13 \%$ isolated yield) all as colorless oils.

Spectroscopic data for $\mathbf{3 . 6 p}:{ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 9.47(\mathrm{t}, J=2.2 \mathrm{~Hz}, 1 \mathrm{H}), 7.86$ $-7.77(\mathrm{~m}, 3 \mathrm{H}), 7.64-7.60(\mathrm{~m}, 1 \mathrm{H}), 7.57-7.52(\mathrm{~m}, 2 \mathrm{H}), 7.50-7.45(\mathrm{~m}, 2 \mathrm{H}), 7.43-7.35(\mathrm{~m}$, $3 \mathrm{H}), 7.32(\mathrm{dd}, J=8.4,1.8 \mathrm{~Hz}, 1 \mathrm{H}), 6.42(\mathrm{tt}, J=7.0,1.1 \mathrm{~Hz}, 1 \mathrm{H}), 3.65(\mathrm{~d}, J=7.0 \mathrm{~Hz}, 2 \mathrm{H}), 3.37$ (dd, $J=2.2,1.0 \mathrm{~Hz}, 2 \mathrm{H}), 0.42(\mathrm{~s}, 6 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 199.2,144.9,137.2,137.1$, $134.0,133.6,132.1,130.8,129.3,128.2,127.9,127.6,127.5,127.1,126.5,126.1,125.4,44.9$, 35.7, -3.3. HRMS (ESI) $m / z 344.1593$ [(M) ${ }^{+}$; calcd for $\left.\mathrm{C}_{23} \mathrm{H}_{24} \mathrm{OSi}, 344.1596\right]$.

Spectroscopic data for $\mathbf{3 . 7 p}:{ }^{1} \mathrm{H}$ NMR $\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 9.67(\mathrm{t}, J=2.1 \mathrm{~Hz}, 1 \mathrm{H}), 7.80$ - $7.73(\mathrm{~m}, 3 \mathrm{H}), 7.61(\mathrm{~s}, 1 \mathrm{H}), 7.56-7.51(\mathrm{~m}, 2 \mathrm{H}), 7.49(\mathrm{dd}, J=8.5,1.8 \mathrm{~Hz}, 1 \mathrm{H}), 7.47-7.37(\mathrm{~m}$, $5 \mathrm{H}), 6.39(\mathrm{~d}, J=15.9 \mathrm{~Hz}, 1 \mathrm{H}), 6.25(\mathrm{dd}, J=15.9,8.2 \mathrm{~Hz}, 1 \mathrm{H}), 2.65-2.43(\mathrm{~m}, 3 \mathrm{H}), 0.39(\mathrm{~s}, 6 \mathrm{H})$. ${ }^{13} \mathrm{C}$ NMR ( $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 202.9,136.1,135.0,134.0,133.6,132.6,130.3,129.6,128.9$, 128.1, 128.0, 127.8, 127.6, 126.2, 125.5, 125.2, 123.3, 42.7, 27.7, -4.5, -5.3. HRMS (ESI) $\mathrm{m} / \mathrm{z}$ $345.1676\left[(\mathrm{M}+\mathrm{H})^{+}\right.$; calcd for $\left.\mathrm{C}_{23} \mathrm{H}_{25} \mathrm{OSi}, 345.1675\right]$.

### 3.10.5. Rearrangement of 6-phenyl-5,6-dihydropyran

The 6-phenyl-5,6-dihydropyran (3.5) was synthesized following three steps from benzaldehyde (see the scheme below):


## Preparation of 1-phenylbut-3-en-1-ol (3.13)



Following a reported procedure ${ }^{61}$ with a slight modification, commercial zinc dust ( 3.92 g , $60 \mathrm{mmol}, 2.0$ equiv.) was weighed into a dry 250 mL round bottomed flask equipped with a magnetic stir bar. The flask was capped with a rubber septum and purged with nitrogen for approximately 5 minutes. Freshly distilled THF ( 80 mL ) was added into the flask followed by 3.9 mL of allyl bromide ( $5.45 \mathrm{~g}, 45 \mathrm{mmol}, 1.5$ equiv). The resulting mixture was stirred at room temperature for 30 minutes after which 2.9 mL of benzaldehyde ( $3.18 \mathrm{~g}, 30 \mathrm{mmol}, 1.0$ equiv.) was added dropwise. The resulting mixture was stirred at room temperature for 1 h then quenched by addition of 10 mL of saturated aqueous ammonium chloride solution. The mixture was diluted with 20 mL of diethyl ether and 10 mL of water, respectively. The layers were separated, and the aqueous layer was extracted with diethyl ether ( $20 \mathrm{~mL} \times 3$ ). The combined organic layers were washed with 10 mL saturated aqueous ammonium chloride, ( $10 \mathrm{~mL} \times 2$ ) water and 10 mL of saturated aqueous sodium chloride solution then dried over anhydrous magnesium sulfate. Filtration and concentration under reduced pressure afforded homoallylic alcohol $\mathbf{3 . 1 3}$ as a yellow oil ( $4.17 \mathrm{~g}, 28.2 \mathrm{mmol}, 94 \%$ crude yield) which was used in the next step without further purification. ${ }^{1} \mathrm{H}$ NMR $\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.36(\mathrm{~d}, J=4.4 \mathrm{~Hz}, 4 \mathrm{H}), 7.29(\mathrm{ddd}, J=8.8,4.9,3.9$ $\mathrm{Hz}, 1 \mathrm{H}), 5.88-5.76(\mathrm{~m}, 1 \mathrm{H}), 5.21-5.11(\mathrm{~m}, 2 \mathrm{H}), 4.73(\mathrm{dd}, J=7.6,5.3 \mathrm{~Hz}, 1 \mathrm{H}), 2.57-2.46(\mathrm{~m}$, 2H), $2.27(\mathrm{~s}, 1 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 143.8,134.4,128.3,127.5,125.8,118.3,73.3$, 43.7. IR (neat) $3342,3054,1638,1507,1270,1124,1042,817,744 \mathrm{~cm}^{-1}$. Spectroscopic data were
in agreement with those reported in literature. ${ }^{62}$

## Preparation of (1-(allyloxy)but-3-en-1-yl)benzene (3.14)



To a dry 250 mL round bottomed flask fitted with a magnetic stir bar was weighed 1.73 g of sodium hydride ( $60 \% \mathrm{w} / \mathrm{w}$ suspension in mineral oil, $75 \mathrm{mmol}, 3.0$ equiv). The flask was sealed with a rubber septum and purged with nitrogen. Freshly distilled THF ( 40 mL ) was then added to the flask via syringe. This was followed by addition of 4.3 mL of allyl bromide ( $6.05 \mathrm{~g}, 50 \mathrm{mmol}$, 2.0 equiv). The mixture was cooled to $0{ }^{\circ} \mathrm{C}$ using ice bath. To the cold suspension, 3.71 g of homoallylic 3.13 ( $25 \mathrm{mmol}, 1.0$ equiv) was added in a dropwise manner. After complete addition, the resulting mixtrure was allowed to warm up slowly to room temperature while stirring. The reaction was quenched after 5 hours by addition of 10 mL of saturated ammonium chloride solution. This was followed by addition of 20 mL of diethyl ether and 10 mL of water. The layers were separated, and the aqueous layer was extracted with diethyl ether ( $20 \mathrm{~mL} \times 3$ ). The combined organic layers were washed with 10 mL saturated ammonium chloride, ( $10 \mathrm{~mL} \times 2$ ) water and 10 mL of saturated sodium chloride solution then dried over anhydrous magnesium sulfate. Filtration and concentration under reduced pressure followed by a flash column chromatography purification $\left(40 \% \mathrm{CH}_{2} \mathrm{Cl}_{2}\right.$ in hexanes) afforded $4.12 \mathrm{~g}, 22 \mathrm{mmol}\left(88 \%\right.$ isolated yield) of 3.14. ${ }^{1} \mathrm{H}$ NMR ( 500 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.39-7.34(\mathrm{~m}, 2 \mathrm{H}), 7.34-7.27(\mathrm{~m}, 3 \mathrm{H}), 5.91(\mathrm{dddd}, J=17.3,10.4,6.0,5.1 \mathrm{~Hz}$, $1 \mathrm{H}), 5.80(\mathrm{ddt}, J=17.2,10.2,7.0 \mathrm{~Hz}, 1 \mathrm{H}), 5.26(\mathrm{dq}, J=17.3,1.7 \mathrm{~Hz}, 1 \mathrm{H}), 5.17(\mathrm{dq}, J=10.4,1.4$ $\mathrm{Hz}, 1 \mathrm{H}), 5.10-5.00(\mathrm{~m}, 2 \mathrm{H}), 4.35(\mathrm{dd}, J=7.5,5.9 \mathrm{~Hz}, 1 \mathrm{H}), 3.94(\mathrm{ddt}, J=12.8,5.1,1.6 \mathrm{~Hz}, 1 \mathrm{H})$, 3.79 (ddt, $J=12.8,6.1,1.4 \mathrm{~Hz}, 1 \mathrm{H}), 2.67-2.57(\mathrm{~m}, 1 \mathrm{H}), 2.44(\mathrm{dddt}, J=14.3,7.2,5.9,1.3 \mathrm{~Hz}$, 1H). ${ }^{13} \mathrm{C}$ NMR ( $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 141.9,134.9,134.8,128.3,127.5,126.7,116.8,116.7,81.1$,
69.4, 42.6. Spectroscopic data were in agreement with those reported in literature. ${ }^{63}$

## Preparation of 2-phenyl-3,6-dihydro-2H-pyran (3.8)



A dry 250 mL round bottomed flask fitted with a magnetic stir bar was sealed with a rubber septum and cooled under nitrogen. $136 \mathrm{mg}\left(0.16 \mathrm{mmol}, 0.008\right.$ equiv) of Grubbs catalyst $2^{\text {nd }}$ generation was weighed into a vial and dissolved in dry dichloromethane ( 5 mL ). The solution was transferred by means of syringe into the flask and additional 125 mL of dry dichloromethane added. This was followed by addition of 3.8 g ( $20 \mathrm{mmol}, 1.0$ equiv) of $\mathbf{3 . 1 4} \mathrm{in} 20 \mathrm{~mL}$ dry dichloromethane. The resulting mixture was stirred under nitrogen for 12 hours after which the dichloromethane was removed by rotorvap under reduced pressure. The crude material was purified by flash column chromatography ( $40 \% \mathrm{CH}_{2} \mathrm{Cl}_{2}$ in hexanes) to give $3.17 \mathrm{~g}, 19.8 \mathrm{mmol}$ ( $99 \%$ isolated yield) of dihydropyran 3.8. ${ }^{1} \mathrm{H}$ NMR $\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.44-7.35(\mathrm{~m}, 4 \mathrm{H}), 7.33$ $-7.27(\mathrm{~m}, 1 \mathrm{H}), 5.95(\mathrm{ddt}, J=9.9,5.4,2.2 \mathrm{~Hz}, 1 \mathrm{H}), 5.84(\mathrm{dtt}, J=10.9,3.1,1.5 \mathrm{~Hz}, 1 \mathrm{H}), 4.58(\mathrm{dd}$, $J=10.3,3.5 \mathrm{~Hz}, 1 \mathrm{H}), 4.45-4.34(\mathrm{~m}, 2 \mathrm{H}), 2.45-2.34(\mathrm{~m}, 1 \mathrm{H}), 2.33-2.22(\mathrm{~m}, 1 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR $\left(126 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 142.5,128.3,127.4,126.4,125.8,124.4,75.6,66.5,32.8$. HRMS (APCI) $\mathrm{m} / \mathrm{z}$ $160.0836\left[(\mathrm{M})^{+}\right.$; calcd for $\left.\mathrm{C}_{11} \mathrm{H}_{12} \mathrm{O}, 160.0888\right]$.

Preparation of 2-(2-phenylcyclopropyl)acetaldehyde (3.15) by Wittig rearrangement of 6-phenyl-5,6-dihydro-2H-pyran (3.8)


Following General Procedure G, dihydropyran 3.8 ( $321 \mathrm{mg}, 2.0 \mathrm{mmol}$, 1 equiv) in THF $(40 \mathrm{~mL})$ and sec -butyllithium (1.4 M in pentane) $\left(1.7 \mathrm{~mL}, 2.4 \mathrm{mmol}, 1.2\right.$ equiv) at $-78{ }^{\circ} \mathrm{C}$ for 1 hour, followed by workup and silica gel chromatography (5\% EtOAc in hexanes) afforded cyclopropane 3.15 ( $34.5 \mathrm{mg}, 0.22 \mathrm{mmol}, 11 \%$ isolated yield, $d r=4: 1$ ) and $161 \mathrm{mg}, 1.0 \mathrm{mmol}(50 \%$ of recovered starting material) 3.8. Spectroscopic data for compound 3.15: ${ }^{1} \mathrm{H}$ NMR ( 500 MHz , $\left.\mathrm{C}_{6} \mathrm{D}_{6}\right) \delta 9.28(\mathrm{t}, J=1.9 \mathrm{~Hz}, 1 \mathrm{H}), 7.11-7.05(\mathrm{~m}, 2 \mathrm{H}), 7.03-6.97(\mathrm{~m}, 1 \mathrm{H}), 6.91-6.83(\mathrm{~m}, 2 \mathrm{H})$, $1.75(\mathrm{ddd}, J=17.1,6.9,2.0 \mathrm{~Hz}, 1 \mathrm{H}), 1.67(\mathrm{ddd}, J=17.0,7.1,2.0 \mathrm{~Hz}, 1 \mathrm{H}), 1.27(\mathrm{dt}, J=9.1,4.9$ $\mathrm{Hz}, 1 \mathrm{H}), 0.93-0.85(\mathrm{~m}, 1 \mathrm{H}), 0.66(\mathrm{dt}, J=8.5,5.1 \mathrm{~Hz}, 1 \mathrm{H}), 0.34(\mathrm{dt}, J=8.6,5.2 \mathrm{~Hz}, 1 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR (126 MHz, $\left.\mathrm{C}_{6} \mathrm{D}_{6}\right) \delta 200.2,143.1,128.9,126.6,126.3,48.3,23.1,16.6,15.6$. HRMS (APCI) $m / z 161.0948\left[(\mathrm{M}+\mathrm{H})^{+}\right.$; calcd for $\left.\mathrm{C}_{11} \mathrm{H}_{13} \mathrm{O}, 161.0966\right]$.
3.10.6. Derivatization of 2-(2-phenyl-1-(triethylsilyl)cyclopropyl)acetaldehyde (3.2c)


Preparation of (E)-1-(2,4-dinitrophenyl)-2-(2-(2-phenyl-1-(triethylsilyl)cyclopropyl) ethylidene)hydrazine (3.16)

Compound 3.16 was prepared from aldehyde 3.2c $(d r>20: 1)$ utilizing the following procedure: To a 5 mL conical vial with a vane magnetic stir bar was weighed $50 \mathrm{mg}(0.25 \mathrm{mmol}$, 1.0 equiv) of 2,4-dinitrophenylhydrazine (DNPH). A solution of ethanol/water (5:2), 2 mL was added to the vial followed by $13.4 \mu \mathrm{~L}$ of concentrated sulfuric acid ( $0.25 \mathrm{mmol}, 1.0$ equiv) and the resulting mixture stirred for about 5 minutes to make a homogeneous solution. Compound 3.2c ( $68.7 \mathrm{mg}, 0.25 \mathrm{mmol}, 1.0$ equiv) in 1 mL ethanol was added dropwise to the vial and the mixture stirred for additional 10 minutes resulting in the formation of a deep orange precipitate. The mixture was filtered, and the residue rinsed with 10 mL solution of ethanol/water (5:2). The residue was dried for 12 hours under high vacuum to give $107 \mathrm{mg}, 0.235 \mathrm{mmol}(94 \%$ crude yield) of $\mathbf{3 . 1 6}$ as a single diastereomer. This product was recrystallized in ethanol/dichloromethane (4:1) and its crystal structure solved by X-ray crystallography and the results deposited to the Cambridge Crystallographic Data Centre and assigned CCDC 2031553. Spectroscopic and melting point data for compound 3.16: mp 115.5 to $117.0{ }^{\circ} \mathrm{C}$; ${ }^{1} \mathrm{H}$ NMR ( 500 MHz , DMSO- $d_{6}$ ) $\delta 11.52(\mathrm{~s}, 1 \mathrm{H}), 8.86$ $(\mathrm{d}, J=2.7 \mathrm{~Hz}, 1 \mathrm{H}), 8.37(\mathrm{dd}, J=9.7,2.7 \mathrm{~Hz}, 1 \mathrm{H}), 8.29(\mathrm{dd}, J=7.2,4.6 \mathrm{~Hz}, 1 \mathrm{H}), 7.91(\mathrm{~d}, J=9.7$ $\mathrm{Hz}, 1 \mathrm{H}), 7.34(\mathrm{~d}, J=7.1 \mathrm{~Hz}, 2 \mathrm{H}), 7.27(\mathrm{t}, J=7.5 \mathrm{~Hz}, 2 \mathrm{H}), 7.18(\mathrm{t}, J=7.3 \mathrm{~Hz}, 1 \mathrm{H}), 2.74(\mathrm{dd}, J=$ $14.8,7.3 \mathrm{~Hz}, 1 \mathrm{H}), 2.19(\mathrm{t}, J=7.1 \mathrm{~Hz}, 1 \mathrm{H}), 2.04(\mathrm{dd}, J=14.7,4.7 \mathrm{~Hz}, 1 \mathrm{H}), 1.30(\mathrm{dd}, J=5.9,4.5$ $\mathrm{Hz}, 1 \mathrm{H}), 0.97(\mathrm{dt}, J=8.2,4.8 \mathrm{~Hz}, 1 \mathrm{H}), 0.73(\mathrm{t}, J=7.9 \mathrm{~Hz}, 9 \mathrm{H}), 0.30(\mathrm{dq}, J=15.8,7.9 \mathrm{~Hz}, 3 \mathrm{H})$, 0.14 (dq, $J=15.7,7.9 \mathrm{~Hz}, 3 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $126 \mathrm{MHz}, \mathrm{DMSO}-d_{6}$ ) $\delta 154.6,144.8,139.6,136.6$, $129.9,129.6,128.8,127.8,126.3,123.2,116.3,42.2,27.5,15.3,12.2,7.5,2.8$. IR (neat) 3291 , $3107,2951,2872,1614,1588,1519,1496,1426,1324,1308,1281,1222,1138,1071 \mathrm{~cm}^{-1}$. HRMS (ESI) $m / z 455.2118\left[(\mathrm{M}+\mathrm{H})^{+}\right.$; calcd for $\left.\mathrm{C}_{23} \mathrm{H}_{31} \mathrm{~N}_{4} \mathrm{O}_{4} \mathrm{Si}, 455.2115\right]$.

## REFERENCES

(1) Felkin, H.; Tambute, A. 1,4-Alkyl shifts in the Wittig rearrangement of alkyl allyl ethers. Tetrahedron Lett. 1969, 821.
(2) Schöllkopf, U. Recent results in carbanion chemistry. Angew. Chem. Int. Ed. 1970, 9, 763.
(3) Briére, R.; Chérest, M.; Felkin, H.; Frajerman, C. A CIDNP study of the Wittig rearrangement of allylic ethers. Tetrahedron Lett. 1977, 18, 34893492.
(4) Rautenstrauch, V. Sigmatropic reactions in carbanions. The 5,6-dihydro-2H-pyran-2-ide cyclopropyl-enolate rearrangement. Helv. Chim. Acta 1972, 55, 594.
(5) Schlosser, M.; Strunk, S. The Wittig rearrangement as a practical method for aldehyde synthesis. Tetrahedron 1989, 45, 2649.
(6) Wang, F.; Wang, J.; Zhang, Y.; Yang, J. The [1,2]- and [1,4]-Wittig rearrangement. Tetrahedron 2020, 76, 130857.
(7) Felkin, H.; Frajerman, C. The [1,2] and [1,4] Wittig rearrangement of optically active allyl $\alpha$-phenylethyl ether. Tetrahedron Lett. 1977, 18, 3485.
(8) Tomooka, K.; Yamamoto, H.; Nakai, T. Stereoselective synthesis of highly functionalized C-alycosides based on acetal [1,2] and [1,4] Wittig rearrangements. Angew. Chem., Int. Ed. Engl. 2000, 39, 4500.
(9) Hodgson, D. M.; Tomooka, K.; Gras, E. Enantioselective synthesis by lithiation adjacent to oxygen and subsequent rearrangement. Top. Organomet. Chem. 2003, 5, 217.
(10) Onyeozili, E. N.; Maleczka, R. E., Jr. $\alpha$-substituted acylsilanes via a highly selective [1,4]Wittig rearrangement of $\alpha$-benzyloxyallylsilane. Chem. Commun. 2006, 2466.
(11) Onyeozili, E. N.; Mori-Quiroz, L. M.; Maleczka, R. E., Jr. [1,2]- and [1,4]-Wittig rearrangements of $\alpha$-alkoxysilanes: effect of substitutions at both the migrating benzylic carbon and the terminal $\mathrm{sp}^{2}$ carbon of the allyl moiety. Tetrahedron 2013, 69, 849.
(12) Mori-Quiroz, L. M.; Maleczka, R. E., Jr. Stereoconvergent [1,2]- and [1,4]-Wittig rearrangements of 2 -silyl-6-aryl-5,6-dihydropyrans: a tale of steric vs. electronic regiocontrol of divergents pathways. J. Org. Chem. 2015, 80, 1163.
(13) Paquette, L. A. Silyl-susbstituted cyclopropanes as versatile synthetic reagents. Chem. Rev. 1986, 86, 733-750.
(14) Lautens, M.; Delanghe, P. H. M. Diastereoselectivity in the hydroxyl-directed cyclopropanation of vinylorganometallic compounds. J. Org. Chem. 1992, 57, 798.
(15) Ukaji, Y.; Sada, K.; Inomata, K. Synthesis of silicon substituted cyclopropylmethyl alcohols in optically active form via asymmetric Simmons-Smith reaction of $\gamma$-silicon substituted allylic alcohols. Chem. Lett. 1993, 1227.
(16) Imai, N.; Sakamoto, K.; Takahashi, H.; Kobayashi, S. First catalytic and enantioselective synthesis of silyl and stannyl substituted cyclopropylmethanols. Tetrahedron Lett. 1994, 35, 7045.
(17) Lautens, M.; Delanghe, P. H. M. Diastereoselectivity in the cyclopropanation of 3,3bimetallic allylic alcohols. Preparation of diastereomeric cyclopropyl carbinols via a simple oxidation-reduction sequence. J. Org. Chem. 1995, 60, 2474.
(18) Berthon-Gelloz, G.; Marchant, M.; Straub, B. F.; Marko, I. E. Palladium-catalyzed cyclopropanation of alkenyl silanes by diazoalkanes: Evidence for a $\mathrm{Pd}^{0}$ mechanism. Chem. Eur. J. 2009, 15, 2923.
(19) Beaulieu, L.-P. B.; Delvos, L. B.; Charette, A. B. Dual role of silanol groups in cyclopropanation and Hiyama-Denmark cross-coupling reactions. Org. Lett. 2010, 12, 1348.
(20) Su, Y.; Li, Q.-F.; Zhao, Y.-M.; Gu, P. Preparation of optically active cis-cyclopropane carboxylates: Desilylation of the resulting silyl cyclopropanes. Org. Lett. 2016, 18, 4356.
(21) Lee, K.; Kim, S.-I.; Cha, J. K. Diastereoselective synthesis of trisubstituted cyclopropylstannanes. J. Org. Chem. 1998, 63, 9135.
(22) Mizojiri, R.; Urabe, H.; Sato, F. New synthesis of cyclopropanols via titanium(II)-mediated coupling of vinylsilanes and esters. Tetrahedron Lett. 1999, 40, 2557.
(23) Mizojiri, R.; Urabe, H.; Sato, F. Generation of a silylethylene-titanium alkoxide complex. A versatile reagent for silylethylation and silylethylidenation of unsaturated compounds. $J$. Org. Chem. 2000, 65, 6217.
(24) Aoyama, T.; Iwamoto, Y.; Nishigaki, S.; Shioiri, T. Reaction of trimethylsilyldiazomethane with olefins. Chem. Pharm. Bull. 1989, 37, 253.
(25) Goumri-Magnet, S.; Kato, T.; Gornitzka, H.; Bacereido, A.; Bertrand, G. Stereoselectivity and stereospecificity of cyclopropanation reactions with stable (phosphanyl)(silyl)carbenes. J. Am. Chem. Soc. 2000, 122, 4464.
(26) Hamaker, C. G.; Mirafzal, G. A.; Woo, L. K. Catalytic cyclopropanation with iron(II) complexes. Organometallics 2001, 20, 5171.
(27) Maas, G.; Alt, M.; Mayer, D.; Bergsträsser, U.; Sklenak, S.; Javier, P.; Apeloig, Y. Catalytic and photochemical cyclopropanation of alkenes with methyl
diazo(trialkylsilyl)acetates: Steric effects and thermodynamic stabilities of cyclopropanes. Organometallics 2001, 20, 4607.
(28) Takai, K.; Hirano, M.; Toshikawa, S. Preparation of cyclopropylsilanes from terminal alkenes with organochromium reagents. SynLett 2004, 1347.
(29) Sharma, V. B.; Jain, S. L.; Sain, B. Metallophthalocyanines catalyzed cyclopropanation of olefins with trimethylsilyldiazomethane: A facile and stereoselective synthesis of silylcyclopropanes. Cat. Comm. 2006, 7, 454.
(30) Concellón, J. M.; Rodríguez-Solla, H.; Méjica, C.; Blanco, E. G.; García-Granda, S.; Díaz, M. R. Stereospecific and stereoselective alkyl and silylcyclopropanation of $\alpha, \beta$ unsaturated amides. Org. Lett. 2008, 10, 349.
(31) Glass, A. C.; Morris, B. B.; Zakharov, L. N.; Liu, S.-Y. Synthesis of substituted naphthalenes via a catalytic ring-expansion rearrangement. Org. Lett. 2008, 10, 4855.
(32) Mata, S.; López, L. A.; Vicente, R. Synthesis of silylcyclopropanes through the catalytic generation of zinc silylcarbenoids from enynones. SynLett 2015, 26, 2685.
(33) Rubina, M.; Rubin, M.; Gevorgyan, V. Transition metal-catalyzed hydro-, sila-, and stannastannation of cyclopropenes: stereo- and regioselective approach towards multisubstituted cyclopropyl synthons. J. Am. Chem. Soc. 2002, 124, 11566.
(34) Trofimov, A.; Rubina, M.; Rubin, M.; Gevorgyan, V. Highly diastereo and regioselective transition metal-catalyzed additions of metal hydrides and bimetallic species to cyclopropenes: easy entry to multisubstituted cyclopropanes. J. Org. Chem. 2007, 72, 8910.
(35) Zhang, L.; Oestreich, M. Copper-catalyzed enantio- and diastereoselective addition of silicon nucleophiles to 3,3-disubstituted cyclopropenes. Chem. Eur. J. 2019, 25, 14304.
(36) Wang, H.; Zhang, G.; Zhang, Q.; Wang. Y.; Li, Y.; Xiong, T.; Zhang, Q. Copper-catalyzed non-directed hydrosilylation of cyclopropenes: highly diastereoselective synthesis of fully substituted cyclopropylsilanes. Chem. Commun. 2020, 56, 1819.
(37) Dian, L.; Marek, I. Cobalt-catalyzed diastereoselective and enantioselective hydrosilylation of achiral cyclopropenes. Org. Lett. 2020. 22, 4914.
(38) Lee, T.; Hartwig, J. F. Rhodium-catalyzed enantioselective silylation of cyclopropyl C-H bonds. Angew. Chem. Int. Ed. 2016, 55, 8723.
(39) Langer, P.; Freifeld, I. Chemo-, regio-, and diastereoselective synthesis of functionalized cyclopropanes by cyclization of dilithiated nitriles with epibromohydrin. Org. Lett. 2001, 3, 3903.
(40) Barluenga, J.; Fañanás, F. J.; Sanz, R.; Marcos, C. Diastereo- and enantioselective carbolithiation of allyl o-lithioaryl ethers. new chiral cyclopropane derivatives. Org. Lett 2002, 4, 2225.
(41) Maddess, M. L.; Mainetti, E.; Harrak, Y.; Brancour, C.; Devin, P.; Dhimane, A.-L.; Fensterbank, L.; Malacria, M. Preparation of fused polycyclic vinylcyclopropanes via radical cascade reactions. Chem. Commun. 2007, 936.
(42) Ito, H.; Kosaka. Y.; Nonoyama, K.; Sasaki, Y.; Sawamura, M. Synthesis of optically active boron-silicon bifunctional cyclopropane derivatives through enantioselective copper(I)catalyzed reaction of allylic carbonates with a diboron derivative. Angew. Chem. Int. Ed. 2008, 47, 7424.
(43) Shintani, R.; Fujie, R.; Takeda, M.; Nozaki, K. Silylative cyclopropanation of allyl phosphates with silylboronates. Angew. Chem. Int. Ed. 2014, 53, 6546.
(44) Trost, B. M.; Ball, Z. T. Markovnikov Alkyne Hydrosilylation Catalyzed by Ruthenium Complexes. J. Am. Chem. Soc. 2001, 123, 12726.
(45) Trost, B. M.; Ball, Z. T. Alkyne hydrosilylation catalyzed by a cationic ruthenium complex: efficient and general trans addition. J. Am. Chem. Soc. 2001, 123, 17644.
(46) Kawasaki, Y.; Ishikawa, Y.; Igawa, K.; Tomooka, K. Directing group-controlled hydrosilylation: regioselective functionalization of alkyne. J. Am. Chem. Soc. 2011, 133, 20712.
(47) Scholl, M.; Ding, S.; Lee, C. W.; Grubbs, R. H. synthesis and activity of a new generation of ruthenium-based olefin metathesis catalysts coordinated with 1,3-dimesityl-4,5-dihydroimidazol-2-ylidene ligands. Org. Lett. 1999, 1, 953.
(48) Vougiokalakis, G. C.; Grubbs, R. H. Ruthenium-based heterocyclic carbene-coordinated olefin metathesis catalysts. Chem. Rev. 2010, 100, 1746.
(49) Oxidation of $\mathbf{3 . 1 0}$ to $\mathbf{3 . 5 0}$ has been observed independently when $\mathbf{1 0}$ is stored at $-8{ }^{\circ} \mathrm{C}$ in the presence of air.
(50) Newcomb, M.; Glenn, A. G. A convenient method for kinetic studies of fast radical rearrangements. Rate constants and arrhenius function for the cyclopropylcarbinyl radical ring opening. J. Am. Chem. Soc. 1989, 111, 275.
(51) Wilt, J. W.; Lusztyk, J.; Peeran, M.; Ingold, K. U. Absolute rate constants for some intermolecular and intramolecular reactions of $\alpha$-, $\beta$-, and $\gamma$-silicon-substituted radicals. J. Am. Chem. Soc. 1988, 110, 281.
(52) Tello-Aburto, R.; Harned, A. M. Palladium-catalyzed reactions of cyclohexadienones: regioselective cyclizations triggered by alkyne acetoxylation. Org. Lett. 2011, 11, 3998.
(53) Lu, Y.; Woo, S. K.; Krische, M. J. Total synthesis of bryostatin 7 via $\mathrm{C}-\mathrm{C}$ bond-forming hydrogenation. J. Am. Chem. Soc. 2011, 133, 13876.
(54) Umaña, C. A.; Cabezas, J. A. Palladium-catalyzed one-pot conversion of aldehydes and ketones into 4-substituted homopropargyl alcohols and 5-en-3-yn-1-ols. J. Org. Chem. 2017, 82, 9505.
(55) Shu, C.; Liu, M.-Q.; Sun, Y.-Z.; Ye, L.-W. Efficient synthesis of $\gamma$-lactones via goldcatalyzed tandem cycloisomerization/oxidation. Org. Lett. 2012, 14, 4958.
(56) Freitas, J. J. R.; Couto, T. R.; Cavalcanti, I. H.; Regioselective propargylation of aldehydes using potassium allenyltrifluoroborate promoted by tonsil. Freitas, J. C. R.; Barbosa, Q. P. S.; Oliveira, R. A. Tetrahedron Lett. 2016, 57, 760.
(57) Sherwood, A. M.; Williamson, S. E.; Johnson, S. N.; Yilmaz, A.; Day, V. W.; Prisinzano, T. E. Scalable regioselective and stereoselective synthesis of functionalized $(E)$-4-iodobut-3-en-1-ols: gram-scale total synthesis of fungal decanolides and derivatives. J. Org. Chem. 2018, 83, 980.
(58) Donner, C. D.; Cuzzupe, A. N.; Falzon, C. L.; Gill, M. Investigations towards the synthesis of xylindein, a blue-green pigment from the fungus Chlorociboria aeruginosa. Tetrahedron 2012, 68, 2799.
(59) Ma, X.; Wang, J.-X.; Li, S.; Wang, K.-H.; Huang, D. One-pot, solvent-free regioselective addition reactions of propargyl bromide to carbonyl compounds mediated by $\mathrm{Zn}-\mathrm{Cu}$ couple. Tetrahedron 2009, 65, 8683.
(60) Barret, A. G. M.; Wan, P. W. H. A Convenient synthesis of $\gamma$-hydroxy- $\alpha$-methylene silanes. J. Org. Chem. 1996, 61, 8667.
(61) Ranu, C. B.; Majee, A.; Das, A. R. Facile and efficient synthesis of homoallylic alcohols using allyl bromide and commercial zinc dust. Tetrahedron Lett. 1995, 36, 4885
(62) Gualandi, A.; Rodeghiero, G.; Faraone, A.; Patuzzo, F.; Marchini, M.; Calogero, F.; Perciaccante, R.; Jansen, T. P.; Ceroni, P.; Cozzi, P. G. Allylation of aldehydes by dual photoredox and nickel catalysis. Chem. Commun. 2019, 55, 6838.
(63) Breen, C. P.; Parrish, C.; Shangguan, N.; Majumdar, S.; Murnen, H.; Jamison, T. F.; Bio, M. M. A scalable membrane pervaporation approach for continuous flow olefin metathesis. Org. Process Res. Dev. 2020, 24, 2298.

## APPENDIX

## Crystallographical information of compound 3.16

Crystal structure, Chirality not determined, but Regio-stereo chemistry is observed.

## Crystal data and experimental



Figure 3.2: Crystal structure of compound $\mathbf{3 . 1 6}$
Experimental. Single yellow needle crystals of $\mathbf{3 . 1 6}$ used as received. A suitable crystal with dimensions $0.21 \times 0.05 \times 0.03 \mathrm{~mm}^{3}$ was selected and mounted on a nylon loop with paratone oil on a XtaLAB Synergy, Dualflex, HyPix diffractometer. The crystal was kept at a steady $T=$ 100.00(10) K during data collection. The structure was solved with the ShelXT (Sheldrick, 2015) solution program using dual methods and by using Olex 2 (Dolomanov et al., 2009) as the graphical interface. The model was refined with ShelXL 2018/3 (Sheldrick, 2015) using full matrix least squares minimisation on $\boldsymbol{F}^{\mathbf{2}}$.

Crystal data. $\mathrm{C}_{23} \mathrm{H}_{30} \mathrm{~N}_{4} \mathrm{O}_{4} \mathrm{Si}, M_{r}=454.60$, monoclinic, $P 2_{1} / c$ (No. 14), $\mathrm{a}=18.5519(12) \AA, \mathrm{b}=$ $7.2993(5) \AA, \mathrm{c}=18.8084(12) \AA, b=114.662(8)^{\circ}, a=g=90^{\circ}, V=2314.6(3) \AA^{3}, T=100.00(10) \mathrm{K}$, $Z=4, Z^{\prime}=1, m\left(\mathrm{Cu} \mathrm{K}_{a}\right)=1.204$, 14074 reflections measured, 4597 unique $\left(\mathrm{R}_{\mathrm{int}}=0.0666\right)$ which were used in all calculations. The final $w R_{2}$ was 0.2855 (all data) and $R_{l}$ was $0.1072(\mathrm{I} \geq 2 s(\mathrm{I})$ ).

Table 3.2: Crystal data

| Compound | 3.16 |
| :---: | :---: |
| Formula | $\mathrm{C}_{23} \mathrm{H}_{30} \mathrm{~N}_{4} \mathrm{O}_{4} \mathrm{Si}$ |
| CCDC | 2031553 |
| $D_{\text {calc. }} / \mathrm{g} \mathrm{cm}^{-3}$ | 1.305 |
| $\mathrm{m} / \mathrm{mm}^{-1}$ | 1.204 |
| Formula Weight | 454.60 |
| Colour | yellow |
| Shape | needle |
| Size/mm ${ }^{3}$ | $0.21 \times 0.05 \times 0.03$ |
| T/K | 100.00(10) |
| Crystal System | monoclinic |
| Space Group | $P 21 / c$ |
| a/Å | 18.5519(12) |
| b/Å | 7.2993(5) |
| $c / \AA$ | 18.8084(12) |
| $a 1^{\circ}$ | 90 |
| $b 1^{\circ}$ | 114.662(8) |
| $g{ }^{\circ}$ | 90 |
| V/A ${ }^{3}$ | 2314.6(3) |
| Z | 4 |
| $Z^{\prime}$ | 1 |
| Wavelength/Å | 1.54184 |
| Radiation type | $\mathrm{Cu} \mathrm{K}_{a}$ |
| $\left.Q_{\text {min }}\right]^{\circ}$ | 4.725 |
| $Q_{\max }{ }^{\circ}$ | 77.629 |
| Measured Refl's. | 14074 |
| Indep't Refl's | 4597 |
| Refl's I $\geq 2 s(\mathrm{I})$ | 3836 |
| $R_{\text {int }}$ | 0.0666 |
| Parameters | 296 |
| Restraints | 0 |
| Largest Peak | 0.802 |
| Deepest Hole | -0.396 |
| GooF | 1.185 |
| $w R_{2}$ (all data) | 0.2855 |
| $w R_{2}$ | 0.2802 |
| $R_{l}$ (all data) | 0.1193 |
| $R_{1}$ | 0.1072 |

## Structure quality indicators

| Reflections: | d min (Cu) | 0.79 | (I) | 15.6 | Rint | 6.66\% | Empleter) | 99\% |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Refinement: | Shift | 0.000 | Max Pe | 0.8 | Min Pe | -0.4 | Goof | 1.185 |

Figure 3.3: Structure quality indicators
A yellow needle-shaped crystal with dimensions $0.21 \times 0.05 \times 0.03 \mathrm{~mm}^{3}$ was mounted on a nylon loop with paratone oil. Data were collected using a XtaLAB Synergy, Dualflex, HyPix diffractometer equipped with an Oxford Cryosystems low-temperature device, operating at $T=$ 100.00(10) K.

Data were measured using $w$ scans of $1.0^{\circ}$ per frame for $4.3 / 17.4 \mathrm{~s}$ using $\mathrm{Cu}_{a}$ radiation (microfocus sealed X-ray tube, $50 \mathrm{kV}, 1 \mathrm{~mA}$ ). The total number of runs and images was based on the strategy calculation from the program CrysAlisPro (Rigaku, V1.171.40.84a, 2020). The actually achieved resolution was $Q=77.629$.

Cell parameters were retrieved using the CrysAlisPro (Rigaku, V1.171.40.84a, 2020) software and refined using CrysAlisPro (Rigaku, V1.171.40.84a, 2020) on 4376 reflections, $31 \%$ of the observed reflections. Data reduction was performed using the CrysAlisPro (Rigaku, V1.171.40.84a, 2020) software which corrects for Lorentz polarization. The final completeness is 98.70 out to 77.629 in $Q$ CrysAlisPro 1.171.40.84a (Rigaku Oxford Diffraction, 2020) Numerical absorption correction based on gaussian integration over a multifaceted crystal model Empirical absorption correction using spherical harmonics, implemented in SCALE3 ABSPACK scaling algorithm.

The structure was solved in the space group $P 2_{1} / c$ (\# 14) by using dual methods using the ShelXT (Sheldrick, 2015) structure solution program. The structure was refined by Least Squares using version 2018/2 of XL (Sheldrick, 2008) incorporated in Olex2 (Dolomanov et al., 2009). All nonhydrogen atoms were refined anisotropically. Hydrogen atom positions were calculated geometrically and refined using the riding model, except for the hydrogen atom on the non-carbon atom(s) which were found by difference Fourier methods and refined isotropically when data permits.

CCDC 2031553 contains the supplementary crystallographic data for this paper. The data can be obtained free of charge from The Cambridge Crystallographic Data Centre via www.ccdc.cam.ac.uk/structures.

Model has Chirality at C1 (Centro SPGR) R Verify. Model has Chirality at $\mathbf{C 2}$ (Centro SPGR) $S$ Verify. In this centrosymmetric space group this means both, R,S and S,R confirmations are observed in the crystal studied.

There is a single molecule in the asymmetric unit, which is represented by the reported sum formula. In other words: Z is 4 and $\mathrm{Z}^{\prime}$ is 1 .


Figure 3.4: Crystal structure of $\mathbf{3 . 1 6}$ showing hydrogen bonding


Figure 3.5: Model has Chirality at C1 (Centro SPGR) R Verify. Model has Chirality at C2 (Centro SPGR) S Verify. In this centrosymmetric space group this means both, R,S and S,R confirmations are observed in the crystal studied


Figure 3.6: The following hydrogen bonding interactions with a maximum D-D distance of $2.9 \AA$ and a minimum angle of $120^{\circ}$ are present in 3.16: N2-O1: $2.634 \AA$


Figure 3.7: The $p-p$ interactions of 3.16. plane 2 to \#2@3_676 (1-X,2-Y,1-Z): The angle between these two planes is $0.000^{\circ}$, the centroid-centroid distance is $3.817 \AA$ and the shift distance is $1.962 \AA$


Figure 3.8: Packing diagram of $\mathbf{3 . 1 6}$
Data Plots: Diffraction data


Figure 3.9: Data plots: Diffraction data

Figure 3.9 (cont'd)



## Data Plots: Refinement and data




Figure 3.10: Data plots: Refinement and data
Table 3.3: Reflection statistics

| Total reflections (after <br> filtering) | 14898 |
| :--- | :--- |
| Completeness | 0.933 |
| hkl $_{\text {max }}$ collected | $(22,8,23)$ |
| $\mathrm{hkl}_{\text {max }}$ used | $(21,9,23)$ |
| Lim d $_{\text {max }}$ collected | 100.0 |
| $\mathrm{~d}_{\text {max }}$ used | 17.09 |
| Friedel pairs | 525 |
| Inconsistent | 29 |
| equivalents |  |
| $\mathrm{R}_{\text {sigma }}$ | 0.0642 |
| Omitted reflections | 0 |

Unique reflections ..... 4597
Mean I/s 10.59
$\mathrm{hk} 1_{\text {min }}$ collected
(-23, -9, -16)
$\mathrm{hkl}_{\text {min }}$ used $\quad(-23,0,0)$
Lim d ${ }_{\text {min }}$ collected 0.77
$\mathrm{d}_{\text {min }}$ used $\quad 0.79$
Friedel pairs merged 1
$\mathrm{R}_{\text {int }} \quad 0.0666$

Intensity transformed 0
Omitted by user 12
(OMIT hkl)
Multiplicity $\quad(4557,2079,970,341$, Maximum multiplicity 13 210, 109, 27, 2)
Removed systematic 812
absences

Filtered off
(Shel/OMIT)

Table 3.4: Fractional atomic coordinates $\left(\times 10^{4}\right)$ and equivalent isotropic displacement parameters $\left(\AA^{2} \times 10^{3}\right)$ for 3.16. $U_{e q}$ is defined as $1 / 3$ of the trace of the orthogonalised $U_{i j}$

| Atom | $\mathbf{x}$ | $\mathbf{y}$ | $\mathbf{z}$ | $\boldsymbol{U}_{\text {eq }}$ |
| :--- | :--- | :---: | :---: | :---: |
| Si1 | $8542.1(8)$ | $6859(2)$ | $8121.6(8)$ | $24.8(4)$ |
| O1 | $4210(2)$ | $7872(7)$ | $5945(2)$ | $36.5(10)$ |
| O2 | $3309(2)$ | $8675(7)$ | $4812(2)$ | $37.0(10)$ |
| O3 | $3938(3)$ | $9804(8)$ | $2654(3)$ | $48.3(13)$ |
| O4 | $5030(3)$ | $8764(8)$ | $2654(3)$ | $46.1(12)$ |
| N1 | $6295(3)$ | $5659(7)$ | $6397(3)$ | $31.3(11)$ |
| N2 | $5580(3)$ | $6558(8)$ | $6070(3)$ | $31.3(11)$ |
| N3 | $3985(3)$ | $8173(7)$ | $5231(3)$ | $31.6(11)$ |
| N4 | $4574(3)$ | $8999(8)$ | $2970(3)$ | $34.0(12)$ |
| C1 | $7861(3)$ | $5142(8)$ | $8281(3)$ | $27.2(12)$ |
| C2 | $7564(3)$ | $5259(9)$ | $8922(3)$ | $29.8(13)$ |
| C3 | $8222(3)$ | $3977(9)$ | $9013(3)$ | $31.9(13)$ |
| C4 | $7281(3)$ | $4185(9)$ | $7533(3)$ | $31.7(13)$ |
| C5 | $6530(3)$ | $5230(9)$ | $7122(3)$ | $30.8(13)$ |
| C6 | $5309(3)$ | $7133(8)$ | $5315(3)$ | $29.8(12)$ |
| C7 | $4551(3)$ | $7906(8)$ | $4893(3)$ | $27.8(12)$ |
| C8 | $4302(3)$ | $8517(9)$ | $4124(4)$ | $31.2(13)$ |
| C9 | $4816(3)$ | $8320(9)$ | $3766(3)$ | $32.2(13)$ |
| C10 | $5557(3)$ | $7522(9)$ | $4157(4)$ | $33.9(13)$ |
| C11 | $5800(3)$ | $6939(9)$ | $4907(3)$ | $32.4(13)$ |
| C12 | $7683(3)$ | $6914(9)$ | $9425(3)$ | $29.6(12)$ |
| C13 | $7321(3)$ | $8549(9)$ | $9101(3)$ | $29.3(13)$ |
| C14 | $7407(3)$ | $10100(9)$ | $9561(4)$ | $35.0(14)$ |
| C15 | $7883(4)$ | $10015(10)$ | $10365(4)$ | $36.3(14)$ |
| C16 | $8242(4)$ | $8385(10)$ | $10681(4)$ | $36.0(14)$ |
| C17 | $8146(3)$ | $6835(9)$ | $10229(3)$ | $30.7(13)$ |
| C18 | $7928(3)$ | $8512(9)$ | $7331(3)$ | $30.3(12)$ |
| C19 | $8372(4)$ | $10119(9)$ | $7168(4)$ | $35.0(14)$ |
| C20 | $9204(3)$ | $5423(9)$ | $7794(3)$ | $31.0(13)$ |
| C21 | $9739(4)$ | $4081(9)$ | $8418(4)$ | $34.7(14)$ |
| C22 | $9201(3)$ | $8121(9)$ | $9039(4)$ | $33.2(13)$ |
| C23 | $9973(3)$ | $8892(10)$ | $9031(4)$ | $36.7(14)$ |
|  |  |  |  |  |

Table 3.5: Anisotropic Displacement Parameters $\left(\times 10^{4}\right)$ for 3.16. The anisotropic displacement factor exponent takes the form: $-2 p^{2}\left[h^{2} a{ }^{* 2} \times U_{11}+\ldots+2 h k a * \times b^{*} \times U_{12}\right]$

| Atom | $\boldsymbol{U}_{\mathbf{1 1}}$ | $\boldsymbol{U}_{\mathbf{2 2}}$ | $\boldsymbol{U}_{\mathbf{3 3}}$ | $\boldsymbol{U}_{\mathbf{2 3}}$ | $\boldsymbol{U} \boldsymbol{U}_{\mathbf{1 3}}$ | $\boldsymbol{U}_{\mathbf{1 2}}$ |
| :--- | :--- | :--- | :--- | ---: | :--- | ---: |
| Si1 | $24.6(7)$ | $30.9(8)$ | $26.2(7)$ | $-0.4(6)$ | $17.9(6)$ | $-0.6(6)$ |
| O1 | $32(2)$ | $53(3)$ | $35(2)$ | $-1(2)$ | $23.3(18)$ | $0(2)$ |
| O2 | $26(2)$ | $51(3)$ | $42(2)$ | $4(2)$ | $21.0(18)$ | $4.1(19)$ |
| O3 | $34(2)$ | $75(4)$ | $40(2)$ | $11(2)$ | $20(2)$ | $10(2)$ |
| O4 | $44(2)$ | $69(3)$ | $39(2)$ | $0(2)$ | $31(2)$ | $2(2)$ |
| N1 | $26(2)$ | $35(3)$ | $35(3)$ | $-3(2)$ | $16(2)$ | $-1(2)$ |
| N2 | $25(2)$ | $40(3)$ | $33(3)$ | $0(2)$ | $16(2)$ | $5(2)$ |
| N3 | $29(2)$ | $36(3)$ | $40(3)$ | $-1(2)$ | $25(2)$ | $-1(2)$ |
| N4 | $29(2)$ | $45(3)$ | $33(3)$ | $-1(2)$ | $18(2)$ | $-3(2)$ |
| C1 | $25(3)$ | $33(3)$ | $31(3)$ | $1(2)$ | $20(2)$ | $-1(2)$ |
| C2 | $28(3)$ | $40(3)$ | $30(3)$ | $2(2)$ | $21(2)$ | $-2(2)$ |
| C3 | $34(3)$ | $36(3)$ | $32(3)$ | $4(3)$ | $20(2)$ | $0(3)$ |
| C4 | $29(3)$ | $36(3)$ | $36(3)$ | $-1(3)$ | $20(2)$ | $-4(3)$ |
| C5 | $29(3)$ | $37(3)$ | $35(3)$ | $-7(3)$ | $22(2)$ | $-6(2)$ |
| C6 | $29(3)$ | $33(3)$ | $35(3)$ | $-5(2)$ | $21(2)$ | $-6(2)$ |
| C7 | $24(3)$ | $30(3)$ | $36(3)$ | $-3(2)$ | $20(2)$ | $-2(2)$ |
| C8 | $25(3)$ | $33(3)$ | $40(3)$ | $-2(3)$ | $18(2)$ | $-2(2)$ |
| C9 | $31(3)$ | $37(3)$ | $36(3)$ | $-2(3)$ | $22(2)$ | $-8(3)$ |
| C10 | $29(3)$ | $43(4)$ | $40(3)$ | $-5(3)$ | $24(3)$ | $-4(3)$ |
| C11 | $25(3)$ | $41(4)$ | $38(3)$ | $-7(3)$ | $20(2)$ | $-4(3)$ |
| C12 | $24(3)$ | $41(3)$ | $33(3)$ | $4(3)$ | $22(2)$ | $0(2)$ |
| C13 | $25(3)$ | $45(4)$ | $30(3)$ | $6(3)$ | $23(2)$ | $3(2)$ |
| C14 | $32(3)$ | $40(3)$ | $47(3)$ | $8(3)$ | $30(3)$ | $5(3)$ |
| C15 | $35(3)$ | $45(4)$ | $38(3)$ | $-8(3)$ | $25(3)$ | $-5(3)$ |
| C16 | $36(3)$ | $48(4)$ | $30(3)$ | $-3(3)$ | $19(3)$ | $-1(3)$ |
| C17 | $29(3)$ | $43(4)$ | $29(3)$ | $4(3)$ | $21(2)$ | $1(3)$ |
| C18 | $29(3)$ | $36(3)$ | $34(3)$ | $2(2)$ | $21(2)$ | $3(2)$ |
| C19 | $40(3)$ | $39(4)$ | $36(3)$ | $6(3)$ | $25(3)$ | $2(3)$ |
| C20 | $26(3)$ | $40(3)$ | $33(3)$ | $2(3)$ | $18(2)$ | $-1(2)$ |
| C21 | $34(3)$ | $38(4)$ | $41(3)$ | $0(3)$ | $25(3)$ | $5(3)$ |
| C22 | $28(3)$ | $43(4)$ | $38(3)$ | $-6(3)$ | $23(2)$ | $-5(3)$ |
| C23 | $28(3)$ | $47(4)$ | $40(3)$ | $-5(3)$ | $19(3)$ | $-8(3)$ |
|  |  |  |  |  |  |  |

Table 3.6: Bond Lengths in $\AA$ for 3.16

| Atom | Atom | Length/Å |
| :--- | :--- | :--- |
| Si1 | C1 | $1.890(6)$ |
| Si1 | C18 | $1.885(6)$ |
| Si1 | C20 | $1.901(6)$ |
| Si1 | C22 | $1.887(6)$ |
| O1 | N3 | $1.249(6)$ |
| O2 | N3 | $1.227(6)$ |
| O3 | N4 | $1.227(7)$ |
| O4 | N4 | $1.233(6)$ |
| N1 | N2 | $1.374(7)$ |
| N1 | C5 | $1.285(8)$ |
| N2 | C6 | $1.361(8)$ |
| N3 | C7 | $1.450(6)$ |
| N4 | C9 | $1.458(8)$ |
| C1 | C2 | $1.524(7)$ |
| C1 | C3 | $1.516(8)$ |
| C1 | C4 | $1.537(8)$ |
| C2 | C3 | $1.491(8)$ |
| C2 | C12 | $1.492(9)$ |
| C4 | C5 | $1.490(8)$ |
| C6 | C7 | $1.412(8)$ |
| C6 | C11 | $1.422(7)$ |
| C7 | C8 | $1.394(8)$ |
| C8 | C9 | $1.385(8)$ |
| C9 | C10 | $1.388(9)$ |
| C10 | C11 | $1.358(9)$ |
| C12 | C13 | $1.380(9)$ |
| C12 | C17 | $1.396(8)$ |
| C13 | C14 | $1.394(9)$ |
| C14 | C15 | $1.398(9)$ |
| C15 | C16 | $1.372(10)$ |
| C16 | C17 | $1.381(9)$ |
| C18 | C19 | $1.536(8)$ |
| C20 | C21 | $1.533(8)$ |
| C22 | C23 | $1.545(8)$ |
|  |  |  |

## Citations

CrysAlisPro (ROD), Rigaku Oxford Diffraction, Poland (?).
CrysAlisPro Software System, Rigaku Oxford Diffraction, (2020).
O.V. Dolomanov and L.J. Bourhis and R.J. Gildea and J.A.K. Howard and H. Puschmann, Olex2: A complete structure solution, refinement and analysis program, J. Appl. Cryst., (2009), 42, 339-341.

Sheldrick, G.M., Crystal structure refinement with ShelXL, Acta Cryst., (2015), C71, 3-8.
Sheldrick, G.M., ShelXT-Integrated space-group and crystal-structure determination, Acta Cryst., (2015), A71, 3-8.

## Copies of NMR spectra


$\underbrace{\text { ion }}$

${ }^{1} \mathrm{H}$ NMR, $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$





3.101
${ }^{13} \mathrm{C}$ NMR, $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$

*)
*)

3.100
${ }^{1} \mathrm{H}$ NMR, $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$





3.11a
${ }^{1} \mathrm{H}$ NMR, $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$





##  <br> 


3.11c
${ }^{1} \mathrm{H}$ NMR, $600 \mathrm{MHz}, \mathrm{CDCl}_{3}$





Nơ


3.11 f
${ }^{1} \mathrm{H}$ NMR, $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$


3.11 f
${ }^{13} \mathrm{C}$ NMR, $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$






$3.11 i$
${ }^{1} \mathrm{H}$ NMR, $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$



3.11i
${ }^{13} \mathrm{C}$ NMR, $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$




```
200 190 180 170 160 150 140 130 120 110 100 90 clllllllllllllll
```




3.111
${ }^{1} \mathrm{H}$ NMR, $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$




## 


3.11m
${ }^{1} \mathrm{H}$ NMR, $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$




3.110
${ }^{1} \mathrm{H}$ NMR, $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$

${ }^{1} \mathrm{H}$ NMR, $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$




| 1 |  |  |  |  |  |  |  |  |  |  |  |  | 70 |  |  |  |  | 20 |  |  | 1s |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 200 | 190 | 180 | 170 | 160 | 150 | 140 | 130 | 120 | 110 | $100$ | 90 | 80 | 70 | 60 | 50 | 40 | 30 | 20 | 10 | 0 | -1C |



3.11q
${ }^{1} \mathrm{H}$ NMR, $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$






3.12a
${ }^{1} \mathrm{H}$ NMR, $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$



へヘヘヘヘベゥ


3．12b
${ }^{1} \mathrm{H}$ NMR， $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$






3.12c
${ }^{1} \mathrm{H} \mathrm{NMR}, 500 \mathrm{MHz}, \mathrm{CDCl}_{3}$



3.12c
${ }^{13} \mathrm{C}$ NMR, $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$


3.12d
${ }^{1} \mathrm{H}$ NMR, $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$








3.12g
${ }^{1} \mathrm{H}$ NMR, $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$


3.12g
${ }^{13} \mathrm{C}$ NMR, $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$




| 200 | 190 | 180 | 170 | 160 | 150 | 140 | 130 | 120 | 110 <br> 100 <br> f 1 <br> $(\mathrm{ppm})$ | 90 | 80 | 70 | 60 | 50 | 40 | 30 | 20 | 10 | 0 |
| :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- |



3.12i
${ }^{1} \mathrm{H}$ NMR, $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$





|  <br>  |
| :---: |
|  |  |
|  |  |
|  |  |


3.12k
${ }^{1} \mathrm{H}$ NMR, $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$






3.12m
${ }^{1} \mathrm{H}$ NMR, $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$





3.12n
${ }^{1} \mathrm{H}$ NMR, $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$



## 



3.120
${ }^{1} \mathrm{H}$ NMR, $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$


|  |  |  | $\frac{M}{⿱ ⿻ 丅 ⿵ 冂 ⿰ ⿱ 丶 丶 ⿱ 丶 丶 ⿱ 一 ⿱ ㇒ ⿵ 冂 ⿰ 丨 丨 一 心}$ |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: |
| N | ¢ |  |  | $\stackrel{\infty}{\infty}$ |  |
| ｜ | 1 ｜ | 1゙1く1 | くり， | । |  |


3.120
${ }^{1} \mathrm{H}$ NMR， $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$


|  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 200 | 190 | 180 | 170 | 160 | 150 | 140 | 130 | 120 | 110 |  |  | 80 | 70 | 60 | 50 | 40 | 30 | 20 | 10 | 0 | －1c |


${ }^{1} \mathrm{H}$ NMR， $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$







3．12q
${ }^{1} \mathrm{H}$ NMR， $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$




$$
\begin{array}{lllllllllllllllllllllllllllll}
\hline 230 & 220 & 210 & 200 & 190 & 180 & 170 & 160 & 150 & 140 & 130 & 120 & 110 & 100 & 90 & 80 & 70 & 60 & 50 & 40 & 30 & 20 & 10 & 0 & -10 \\
\mathrm{f1}(\mathrm{ppm})
\end{array}
$$





3.1b
${ }^{1} \mathrm{H}$ NMR, $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$









3.1d
${ }^{1} \mathrm{H}$ NMR, $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$







${ }^{1} \mathrm{H}$ NMR, $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$






3.1h
${ }^{1} \mathrm{H}$ NMR, $600 \mathrm{MHz}, \mathrm{CDCl}_{3}$




3.1
${ }^{1} \mathrm{H}$ NMR, $600 \mathrm{MHz}, \mathrm{CDCl}_{3}$











## 


3.1 m
${ }^{1} \mathrm{H}$ NMR, $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$



$210200190180170160150140130120 \begin{aligned} & 110100 \\ & \mathrm{f}(\mathrm{ppm})\end{aligned}$




3.10
${ }^{1} \mathrm{H}$ NMR, $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$


3.10
${ }^{13} \mathrm{C}$ NMR, $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$




3.1p
${ }^{1} \mathrm{H}$ NMR, $500 \mathrm{MHz}, \mathrm{C}_{6} \mathrm{D}_{6}$





3．2b $d r=1.0: 0.3$
${ }^{1} \mathrm{H}$ NMR， $600 \mathrm{MHz}, \mathrm{CDCl}_{3}$


| まし゚ |  |
| :---: | :---: |
| ～ | ¢ mimin |




3．2b $d r=1.0: 0.3$
${ }^{13} \mathrm{C}$ NMR， $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$

[^0]
3.2c $d r>20: 1$
${ }^{1} \mathrm{H}$ NMR, $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$

(



[^1]


| 210 | 190 | 170 | 150 | 130 | 110 <br> $f 1(\mathrm{ppm})$ | 90 | 70 | 50 | 30 | 10 | -1 |
| :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- |



| $\begin{aligned} & \text { N్ٍ } \\ & \underset{\sim}{\circ} \end{aligned}$ |  - | $\begin{aligned} & \text { Rofon } \\ & \end{aligned}$ | + |  |
| :---: | :---: | :---: | :---: | :---: |


$21020019018017016015014013012011010090 \quad 80$ f1 (ppm)


[^2]




3.2g $d r=1.1: 1.0$
${ }^{13} \mathrm{C}$ NMR, $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$


| 210 | 190 | 170 | 150 | 130 | 110 <br> $\mathrm{f}(\mathrm{ppm})$ | 90 | 70 | 50 | 30 | 10 | -1 |
| :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- |




3.2i $d r=1.0: 0.5$
${ }^{13} \mathrm{C}$ NMR, $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$


3.2j $d r=1.0: 0.2$
${ }^{1} \mathrm{H}$ NMR, $600 \mathrm{MHz}, \mathrm{CDCl}_{3}$

$\stackrel{\stackrel{\rightharpoonup}{0}}{\stackrel{1}{1}}$


i
en
en


3.2j $d r=1.0: 0.2$
${ }^{13} \mathrm{C}$ NMR, $151 \mathrm{MHz}, \mathrm{CDCl}_{3}$
$210200190180170160150140130120 \begin{gathered}110100 \\ \mathrm{f}(\mathrm{ppm})\end{gathered} \mathrm{go}$ ( 80


๗o


[^3] f1 (ppm)



[^4]
3.3n $d r>20: 1$
${ }^{1} \mathrm{H}$ NMR, $600 \mathrm{MHz}, \mathrm{CDCl}_{3}$

\[

$$
\begin{aligned}
& \text { 가 동 } \\
& \text { \& NiN }
\end{aligned}
$$
\]

No

3.3n $d r>20: 1$
${ }^{13} \mathrm{C}$ NMR, $151 \mathrm{MHz}, \mathrm{CDCl}_{3}$

[^5]

3.30 dr > 20: 1
${ }^{1} \mathrm{H}$ NMR, $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$


| ¢ | ~ $\mathrm{N}^{m}$ | ח | 人) |  |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| - | すio mi m | $\stackrel{\text { ¢ }}{\sim}$ | $\dot{\sim}$ | ¢ | N | ¢ٌ | ¢ |  |
| \| | $1>$ | 1 | $\checkmark$ | 1 | V | I | \| |  |


3.3o dr > 20 : 1 ${ }^{13} \mathrm{C}$ NMR, $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$



3.3o' dr > 20 : 1



$\stackrel{m}{\sim}$

3.3o' $d r>20$ : 1
${ }^{13} \mathrm{C}$ NMR, $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$



3.50
${ }^{1} \mathrm{H}$ NMR, $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$


3.50
${ }^{1} \mathrm{H}$ NMR, $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$




3.6p
${ }^{1} \mathrm{H}$ NMR, $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$


3.6p
${ }^{1} \mathrm{H}$ NMR, $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$



3.7p








3.8
${ }^{13} \mathrm{C}$ NMR, $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$




$\stackrel{\infty}{\infty}$

$3.15 d r=4$ : 1
${ }^{13} \mathrm{C}$ NMR, $126 \mathrm{MHz}, \mathrm{C}_{6} \mathrm{D}_{6}$





## CHAPTER 4.

## A [1,2]-WITTIG / m-CPBA TRIGGERED [1,2]-CARBON-TO-CARBON SILYL MIGRATION APPROACH TO $\alpha$-SILYL- $\beta$-HYDROXY CYCLOPENTANONES AND CYCLOHEXANONES

### 4.1. Introduction

The most explored [1,2]-silyl migrationinvolves the migration of the silicon group from carbon to oxygen (Brook rearrangement) ${ }^{1-3}$ or its reverse reaction (retro-Brook rearrangement). ${ }^{4-9}$ In contrast, the [1,2]-carbon-to-carbon silyl migration has remained largely unexplored. Reported examples of such [1,2]-carbon-to-carbon silyl migration involve the use of protic acids to trigger the migration. ${ }^{10-16}$ Other examples involve alkynyl silanes or silyl propagylic systems that are catalyzed by Lewis acids ${ }^{17,18}$ and/or transition metals. ${ }^{19-25}$ With recent work from our lab on Wittig rearrangements of silyldihydropyrans, ${ }^{26,27}$ we were looking to investigate the utility of the products formed. The [1,2]-silyl migration from carbon to carbon to generate $\alpha$-silyl aldehydes (Scheme $4.1 \mathrm{a})^{28}$ and ketones (Scheme 4.1b) ${ }^{29,30}$ has been previously reported and takes place via epoxide derivatives of acyclic $\alpha$-silyl allylic alcohols.


Scheme 4.1: [1,2]-carbon-to-carbon silyl migration triggered by epoxidation

In our quest to functionalize the olefin of the [1,2]-Wittig rearrangement products by epoxidation we saw an unexpected rearrangement involving a [1,2]-carbon-to-carbon silyl shift. We were interested to learn the generality of such migration in the context of the cyclic systems accessible by [1,2]-Wittig ring contraction of silyl cyclic ethers.

### 4.2. Synthesis of 2-silyl-6-aryl-5,6-dihydro-2H-pyrans

The 2-silyl-6-aryl-5,6-dihydro-2H-pyrans were synthesized following our earlier reported protocol. ${ }^{26}$ The synthesis began by allylation of benzaldehydes followed by conversion of the resulting homoallylic alcohols 4.1 to trichloroacetimidates 4.2. The trichloroacetimidates were then coupled with $\alpha$-hydroxy allyl silanes 4.3 in the presence of catalytic amount of a Lewis acid to form diastereomeric dienes 4.4. For most of the compounds reported herein, the syn diastereomer exhibited a lower $R_{f}$ value than its anti counterpart (dichloromethane/hexanes). The syn/anti dienes were then subjected to ring closing metathesis using Grubbs $2^{\text {nd }}$ generation catalyst leading to 2 -silyl-6-aryl-5,6-dihydro-2H-pyrans 4.5. The cis/trans diastereomers of pyrans 4.5 were separable by column chromatography. Generally, the trans diastereomer has lower $R_{f}$ value than its cis counterpart (dichloromethane/hexanes). The relative stereochemistry of the dihydropyrans was determined by ${ }^{1} \mathrm{H}$ NMR NOESY experiments.


Scheme 4.2: Synthesis of 2-silyl-6-aryl-5,6-dihydro-2H-pyrans

### 4.3. Wittig rearrangements of 2-silyl-6-aryl-5,6-dihydro-2H-pyrans

Since the trans diastereomers have been shown to be more reactive, ${ }^{26}$ they were subjected to Wittig rearrangement using $n$-butyllithium, whereas the cis diastereomers were reacted with secbutyllithium resulting in the stereoconvergent [1,2]- and [1,4]-Wittig rearrangement products (Table 4.1).

Table 4.1: Wittig rearrangement of 2-silyl-6-aryl-5,6-dihydro-2H-pyrans


| Entry | Substrate | Ar | $\mathrm{SiR}_{3}$ | R' | \% (4.6) | \% (4.7) |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 1 | $4.5 \mathrm{a}^{\mathrm{a}}$ | $\mathrm{C}_{6} \mathrm{H}_{5}$ | $\mathrm{SiMe}_{3}$ | H | $65^{\text {b }}$ | n.d |
| 2 | 4.5b | 4-Cl-C6 $\mathrm{H}_{4}$ | $\mathrm{SiMe}_{3}$ | H | 56 | 21 |
| 3 | 4.5c | $4-\mathrm{CF}_{3}-\mathrm{C}_{6} \mathrm{H}_{4}$ | $\mathrm{SiMe}_{3}$ | H | 70 | n.d. |
| 4 | 4.5d | 1-Naph | $\mathrm{SiMe}_{3}$ | H | 81 | n.d. |
| 5 | 4.5 e | 4-Ph-C6 $\mathrm{H}_{4}$ | $\mathrm{SiMe}_{3}$ | H | 75 | n.d. |
| 6 | 4.5 f | 2-Naph | $\mathrm{SiEt}_{3}$ | H | 73 | 17 |
| 7 | 4.5 g | $4-\mathrm{Cl}-\mathrm{C}_{6} \mathrm{H}_{6}$ | $\mathrm{SiMe}_{2} \mathrm{Ph}$ | H | 37 | n.d. |
| 8 | 4.5h | 4-OMe-C6 $\mathrm{H}_{6}$ | $\mathrm{SiMe}_{3}$ | Me | 85 | n.d. |
| 9 | 4.5i | $4-\mathrm{Cl}-\mathrm{C}_{6} \mathrm{H}_{6}$ | $\mathrm{SiMe}_{3}$ | Me | 63 | n.d. |

${ }^{\text {a }}$ Reaction performed on a $1: 1$ mixture of diastereomers as follows: 0.6 equiv $n-\mathrm{BuLi}$ added at -78 ${ }^{\circ} \mathrm{C}$ and stirred at $-78^{\circ} \mathrm{C}$ for 15 minutes, then 0.5 equiv sec-BuLi was added at $-78^{\circ} \mathrm{C}$ and stirred at room temperature for 3 hours after the cold bath was removed. ${ }^{\text {b }} \mathbf{4 . 6 a}$ and $4.6 a^{\prime}$ were formed. n.d. Not detected. It is also worth noting that we were looking at the substrates that would favor [1,2]- over [1,4]-Wittig rearrangements

### 4.4. The [1,2]-carbon-to-carbon silyl migration in cyclic system triggered by epoxidation

### 4.4.1. [1,2]-carbon-to-carbon silyl migration of silylcyclopentenols 4.6a and 4.6a'

Having the starting materials in hand, we began by subjecting cyclopentenol 4.6a to $m$-CPBA and $\mathrm{NaHCO}_{3}$ leading to the formation of cyclopentanone 4.8a as a single diastereomer in $83 \%$ yield. (Scheme 4.4, substrate 4.6a). Given that we were unable to detect any epoxide intermediate in the reaction and determine the stereoselectivity of this step, we questioned whether epimerization of the benzylic position might have taken place. To test this, we subjected an epimer of 4.6a (4.6a'), with aryl and hydroxy groups cis to each other, to our conditions that led to silyl migration with the resulting $\beta$-hydroxy and $\alpha$-aryl in cis relationship (Scheme 4.4, substrate 4.6a').


Scheme 4.3: [1,2]-carbon-to-carbon silyl migration of cyclopentenol 1a and 1a' ${ }^{a}$ Reaction conducted in absence of $\mathrm{NaHCO}_{3}$

Although the substrates employed in Scheme 4.3 were racemic, the relative stereochemistry of the diastereomeric products suggests that the epoxidation step is stereoselective and takes place syn to the tertiary hydroxyl group. In addition, the [1,2]-silyl migration leading to epoxide ring opening appears to occur in a syn fashion, in such a way that the relative stereochemistry of the silyl and aryl groups is conserved.

### 4.4.2. Substrate scope for [1,2]-carbon-to-carbon silyl migration in silylcyclopentenols

We next explored the substrate scope of this transformation (Scheme 4.4). We started by modifying the substituents of the aromatic appendage. Incorporating electron withdrawing groups at the para position of the phenyl ring resulted in higher yields (>98\%). Employing a sterically hindered aryl group (1-naphthyl) also led to higher yield (99\%) of cyclopentanone 4.8d as a single diastereomer.


Scheme 4.4: Substrate scope for [1,2]-carbon-to-carbon silyl migration in silylcyclopentenols ${ }^{\text {a }}$ Reaction conducted in absence of $\mathrm{NaHCO}_{3}$

Modifying the substituents on silicon to triethyl and dimethylphenyl groups respectively also
resulted in higher yields and greater diastereoselectivity (compounds 4.8e and 4.8f). We then explored the substrates with a methyl group at the olefin carbon proximal to silicon with varying aryl groups. All these resulted in high yields (compounds $\mathbf{4 . 8 g}, \mathbf{4 . 8 h}$ and $\mathbf{4 . 8 i}$ ). It is worth noting that compounds 4.8b through 4.8e did not require further purification by column chromatography or crystallization.

### 4.4.3. Substrate scope for [1,2]-carbon-to-carbon silyl migration in silylcyclohexenols

Next, we evaluated the behavior of analogous silyl cyclohexenols that were synthesized through [1,2]-Wittig rearrangement of 2-trimethylsilyl-2,5,6,7-tetrahydro-7-aryl-oxepins (see Chapter 2). The 6-aryl-2-trimethylsilylcyclohex-2-en-1-ols were then exposed to the standard conditions for silyl migration ( $m$ - CPBA and $\mathrm{NaHCO}_{3}$ ). This resulted in the formation of $\alpha$ -trimethylsilyl- $\beta$-hydroxy-6-arylcyclohexan-1-ones in high yields and high diastereoselectivities (Scheme 4.5).


Scheme 4.5: Substrate scope for [1,2]-carbon-to-carbon silyl migration in silylcyclohexenols ${ }^{\text {a }}$ Reaction performed on a mixture of diastereomers

### 4.5. Proposed reaction mechanism for the silyl migration

Mechanistically, the stereoselective epoxidation of the double bond by m-CPBA could be
controlled by intramolecular hydrogen bond interactions with the tertiary allylic alcohol. ${ }^{31-33}$ Furthermore, the epoxide opening may be triggered by the ring strain from a fused oxirane and the cyclopentane or cyclohexane skeleton. Since the same has also been seen in acyclic systems, ${ }^{28-30}$ the ring opening could also be as a result of the ability of the silicon atom to stabilize the development of positive charge at a beta carbon atom, also known as the $\beta$-silicon effect or silicon hyperconjugation. ${ }^{34-46}$ The silyl shift appears to be caused by a pinacol-type rearrangement, ${ }^{47}$ resulting in carbonyl formation. ${ }^{48-55}$

We therefore propose that after stereoselective epoxidation, the epoxide opens up leading to the formation of a carbocation $\beta$ to silyl group. This carbocation is then intramolecularly attacked by silicon through hyperconjugation resulting in silylium ion. Intramolecular proton abstraction from the alcohol by the resulting alkoxide followed by carbon oxygen double bond formation with concerted opening of the silyl heterocyclic ring furnishes the observed product.

We were unable to observe the epoxide intermediate in all the substrates herein. Thus, if this isomerization is a stepwise (epoxide formation and then ring opening to give a tertiary carbocation followed by silyl migration) or a concerted process (simultaneous epoxidation/silyl migration) is unknown at this point (Scheme 4.6).


Scheme 4.6: Proposed mechanistic pathways of the silyl shift
${ }^{\text {a }}$ Allylic alcohol directed epoxidation

### 4.6. Conclusion

In summary, a novel protocol to access $\alpha$-silyl $\beta$-hydroxy cyclopentanones and cyclohexanones by [1,2]-Wittig rearrangement followed by [1,2]-carbon-to-carbon silyl migration has been developed. This transformation occurs with high yields and excellent diastereoselectivities and is independent of the substitution at the aromatic group and more importantly, alkyl substitution at the olefin is not a requirement.

The reaction proceeded with poor yields in the absence of sodium bicarbonate $\left(\mathrm{NaHCO}_{3}\right)$. The purity of $m$-CPBA also played an important role in the conversion of the starting material into the desired products. Column chromatography could not be used to purify the products which had epimerizable proton at the carbon bearing the silyl group. In such cases, recrystallization was helpful. Lastly, from the mechanistic standpoint, whether this transformation occurs in a concerted or stepwise version is unknown at the moment.

### 4.7. Experimental section

### 4.7.1. General information

Unless otherwise noted, all reactions were run under a positive atmosphere of nitrogen in oven-
dried or flame-dried round-bottomed flasks or conical vials or disposable drum vials capped with rubber septa. Solvents were removed by rotary evaporation under reduced pressure at temperatures lower than $45^{\circ} \mathrm{C}$. Column chromatography was run on 230-400 mesh silica gel. Tetrahydrofuran and diethyl ether were distilled from sodium-benzophenone ketyl; dichloromethane, benzene, trimethylsilyl chloride were distilled from calcium hydride. Trimethylsilyltrifluoromethane sulfonate (TMSOTf) was redistilled and stored under nitrogen at $-10{ }^{\circ} \mathrm{C}$ before the reaction. Triethylsily chloride, dimethylphenylsilyl chloride, tert-butyllithium (1.7 M in pentane) and $\mathrm{BF}_{3} \cdot \mathrm{OEt}_{2}$ were used as received. $n$-Butyllithum ( 1.6 M or 2.5 M in hexanes) and sec-butyllithium (1.4 M in cyclohexane) were purchased from Aldrich and their concentration calculated by titration with diphenylacetic acid (average of three runs). ${ }^{1} \mathrm{H}$ NMR spectra was collected in 500 MHz and 600 MHz Varian instruments using $\mathrm{CDCl}_{3}$ as solvent, which was referenced at 7.26 ppm (residual chloroform proton) and ${ }^{13} \mathrm{C}$ NMR spectra was collected in $\mathrm{CDCl}_{3}$ at 126 MHz or 151 MHz and referenced at 77.0 ppm . High-resolution mass spectrometric analysis was run in TOF instruments.

### 4.7.2. Synthesis of 2-Silyl-5,6-Dihydropyrans

### 4.7.2.1. Preparation of aryl homoallylic alcohols 4.1 - peneral procedure A



Following a reported procedure ${ }^{56}$ with a slight modification, commercial zinc dust ( $5.23 \mathrm{~g}, 80$ mmol, 2.0 equiv.) was weighed into a dry 250 mL round-bottomed flask equipped with a magnetic stir bar. The flask was capped with a rubber septum and purged with nitrogen for approximately 5 minutes. Freshly distilled THF ( 100 mL ) was added into the flask followed by 5.2 mL of allyl
bromide ( $7.26 \mathrm{~g}, 60 \mathrm{mmol}, 1.5$ equiv). The resulting mixture was stirred at room temperature for 30 minutes after which the desired aryl aldehyde ( $40 \mathrm{mmol}, 1.0$ equiv.) was added dropwise. The resulting mixture was stirred at room temperature for 1 h then quenched by addition of 20 mL of saturated aqueous ammonium chloride solution. The mixture was diluted with 40 mL of diethyl ether and 20 mL of water, respectively. The layers were separated, and the aqueous layer was extracted with diethyl ether ( $40 \mathrm{~mL} x \mathrm{3}$ ). The combined organic layers were washed with 20 mL saturated aqueous ammonium chloride, ( $20 \mathrm{~mL} \times 2$ ) water and 20 mL of saturated aqueous sodium chloride solution then dried over anhydrous magnesium sulfate. Filtration and concentration under reduced pressure afforded aryl homoallylic alcohol 4.1 which was typically used in the next step without need for further purification.

### 4.7.2.2. Preparation of trichloroacetimidates 4.2 - general procedure $B$



Following our reported procedure with a slight modification, ${ }^{26}$ to a dry 250 mL roundbottomed flask fitted with a magnetic stir bar and sealed with a rubber septum was added 60 mL of dry dichloromethane under nitrogen. The desired homoallylic alcohol 4.1 ( $20 \mathrm{mmol}, 1.00$ equiv) in dichloromethane ( 20 mL ) was then transferred into the flask. This was followed by addition of 0.54 mL of 1,8-Diazabicyclo(5.4.0)undec-7-ene (DBU) ( $0.55 \mathrm{~g}, 3.6 \mathrm{mmol}, 0.18$ equiv.). After stirring for 5 minutes, the solution was cooled to $0^{\circ} \mathrm{C}$ on an ice bath. This was followed by dropwise addition of 2.8 mL of tricchloroacetonitrile ( $4.04 \mathrm{~g}, 28 \mathrm{mmol}, 1.40$ equiv.). After 12 hours, the resulting dark brown mixture was filtered through a plug of silica ( 5 cm thick) to remove
the dark residue. The filtrate was concentrated, and the crude mixture subjected to column chromatography (EtOAc/hexanes) to afford the desired trichloroacetimidate 4.2.

### 4.7.2.3. Alternative preparation of trichloroacetimidates 4.2 - general procedure $\mathbf{C}^{\mathbf{2 6}}$



Following our reported procedure, ${ }^{26} 240 \mathrm{mg}$ of sodium hydride $60 \% \mathrm{w} / \mathrm{w}$ dispersion in mineral oil ( $6 \mathrm{mmol}, 0.18$ equiv) was weighed into a dry 100 mL round-bottomed flask fitted with a magnetic stir bar and 20 mL of freshly distilled diethyl ether was added into the flask. The flask was sealed with a rubber septum and purged with nitrogen. The resulting grey suspension was cooled on an ice bath and the desired homoallylic alcohol 4.1 ( $30 \mathrm{mmol}, 1.00$ equiv) in dry diethyl ether ( 20 mL ) was then transferred into the flask slowly resulting in a fizzy reaction. The mixture was stirred at $0{ }^{\circ} \mathrm{C}$ for 10 minutes. This was followed by dropwise addition of 4.2 mL of trichloroacetonitrile ( $6.06 \mathrm{~g}, 42 \mathrm{mmol}, 1.40$ equiv.). The mixture turned dark brown after complete addition of the trichloroacetonitrile. The mixture was stirred at $0{ }^{\circ} \mathrm{C}$ for 20 minutes and then the ice bath was removed, and the mixture stirred at room temperature for 1 hour. The diethyl ether was then removed by rotorvap and methanol ( $0.25 \mathrm{~mL}, 6.0 \mathrm{mmol}, 0.18$ equiv.) in 15 mL pentane was added to the crude mixture. The mixture was further diluted with 40 mL pentane and filtered through a plug of celite ( 5 cm thick). The filtrate was concentrated, and the crude mixture subjected to column chromatography (EtOAc/hexanes) to afford the desired trichloroacetimidate 4.2.
4.7.2.4. Preparation of $\alpha$-hydroxy allyl silanes 4.3 - general procedure $D^{\mathbf{2 6}}$


A solution of the corresponding allylic alcohol in THF was cooled at $-78{ }^{\circ} \mathrm{C}$, and $n$ butyllithium ( 1.6 M or 2.5 M in hexanes) was added dropwise over 5 min . After 30 min the corresponding chlorosilane was added dropwise via syringe. After the resulting solution was stirred for a given amount of time (see individual compounds procedure below), sec-butyllithium or tert-butyllithium (see below for details) was added dropwise over $30-60 \mathrm{~min}$, and then the reaction was kept at the indicated temperature.

### 4.7.2.5. Preparation of diastereomeric dienes 4.4 - general procedure $\mathbf{E}^{\mathbf{2 6}}$



A dry 250 mL round-bottomed flask with a magnetic stir bar was sealed with a rubber septum and purged with nitrogen. A solution of the corresponding allylic alcohol 4.3 ( $10 \mathrm{mmol}, 1$ equiv) in 20 mL hexanes was transferred into the flask followed by a solution of the corresponding trichloroacetimidate 4.2 ( $15 \mathrm{mmol}, 1.5$ equiv.) in 20 mL hexanes. An additional 40 mL of hexanes was then added into the flask and the resulting mixture was cooled on ice bath to $0{ }^{\circ} \mathrm{C}$ while stirring. To the cold solution was added appropriate Lewis acid: $\mathrm{BF}_{3} \cdot \mathrm{OEt}_{2}(0.12 \mathrm{~mL}, 1 \mathrm{mmol}, 0.1$
equiv.) or TMSOTf ( $0.18 \mathrm{~mL}, 1 \mathrm{mmol}, 0.1$ equiv.). After complete addition, a thick precipitate was formed. The mixture was stirred at $0^{\circ} \mathrm{C}$ to room temperature for 6 hours and filtered through a plug of celite ( 5 cm thick) and the filtrate was transferred into a separating funnel. The filtrate was then washed with saturated solution of aqueous sodium bicarbonate ( $50 \mathrm{~mL} \times 3$ ), water ( 50 $\mathrm{mL} \times 2$ ) and brine ( 50 mL ) respectively. The organic layer was dried over anhydrous sodium sulfate and then filtered. The filtrate was concentrated under reduced pressure to afford diene 4.4 as a mixture of diastereomers. The resulting crude reaction mixture was purified by column chromatography (dichloromethane/hexanes). It is important to note that some diastereomers were separable by column chromatography and hence only one of them (syn) was taken to the next step ( RCM ). The diastereomers that could not be separated by column chromatography were taken to the next step as a mixture. The stereochemistry of these diastereomers were determined after the RCM reaction: syn diastereomers underwent RCM to form cis dihydropyrans or tetrahyrooxepins and vice versa. It is also worth noting that for most of the compounds reported herein, the syn diastereomer exhibited lower $R_{f}$ value than its anti counterpart (dichloromethane/hexanes).

### 4.7.2.6. Preparation of dihydropyrans 4.5 and via $R C M$ - general procedure $\mathbf{F}^{26}$



To a dry 250 mL round-bottomed flask with a magnetic stir bar was weighed 170 mg of Grubbs catalyst $2^{\text {nd }}$ generation ( $0.2 \mathrm{mmol}, 0.04$ equiv.) and the flask was sealed with a rubber septum and purged with nitrogen. This was followed by addition of 80 mL of dry dichloromethane and corresponding diene ( $5 \mathrm{mmol}, 1.0$ equiv) as a solution in 20 mL dry dichloromethane as a single
diastereomer (syn) or as a mixture of diastereomers (syn:anti $=1: 1$ ). The resulting mixture was stirred at room temperature for 12 hours. The mixture was concentrated under reduced pressure to afford dihydropyran 4.5. The resulting crude reaction mixture was purified by column chromatography (dichloromethane/hexanes). The cis and trans diastereomers were separable by column chromatography. The stereochemistry of these diastereomers were determined by 1D NOESY experiment. It is also worth noting that for most of the compounds reported here the trans diastereomer has lower $R_{f}$ value than its cis counterpart (dichloromethane/hexanes).

## Synthesis of 1-phenylbut-3-en-1-ol (4.1a)



Following general procedure A, commercial zinc dust ( $3.92 \mathrm{~g}, 60 \mathrm{mmol}, 2.0$ equiv), allyl bromide ( $3.9 \mathrm{~mL}, 45 \mathrm{mmol}, 1.5$ equiv), benzaldehyde ( $3.18 \mathrm{~g}, 30 \mathrm{mmol}, 1.0$ equiv.) and THF ( 80 mL ) homoallylic alcohol 4.1a was obtained as a yellow oil ( $4.17 \mathrm{~g}, 94 \%$ ) which was used in the next step without further purification. ${ }^{1} \mathrm{H} \operatorname{NMR}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta=7.36(\mathrm{~d}, J=4.4 \mathrm{~Hz}, 4 \mathrm{H})$, $7.29(\mathrm{ddd}, J=8.8,4.9,3.9 \mathrm{~Hz}, 1 \mathrm{H}), 5.88-5.76(\mathrm{~m}, 1 \mathrm{H}), 5.21-5.11(\mathrm{~m}, 2 \mathrm{H}), 4.73(\mathrm{dd}, J=7.6$, $5.3 \mathrm{~Hz}, 1 \mathrm{H}), 2.57-2.46(\mathrm{~m}, 2 \mathrm{H}), 2.27(\mathrm{~s}, 1 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR (126 MHz, $\left.\mathrm{CDCl}_{3}\right) \delta=143.8,134.4$, 128.3, 127.5, 125.8, 118.3, 73.3, 43.7. IR (FTIR, film, $\mathrm{cm}^{-1}$ ) $\tilde{v}=3342,3054,1638,1507,1270$, $1124,1042,817,744.4 .1 \mathrm{a}$ is a known compound and spectroscopic data are in agreement with those reported in literature. ${ }^{57}$

## Synthesis of 1-(4-chlorophenyl)but-3-en-1-ol (4.1b)



Following general procedure A , commercial zinc dust ( $5.88 \mathrm{~g}, 90 \mathrm{mmol}, 1.8$ equiv), allyl bromide ( $6.5 \mathrm{~mL}, 75 \mathrm{mmol}, 1.5$ equiv), 4-chlorobenzaldehyde ( $7.02 \mathrm{~g}, 50 \mathrm{mmol}, 1.0$ equiv.) and THF ( 100 mL ) homoallylic alcohol 4.1b was obtained as a yellow oil in THF ( $13 \mathrm{~g}, \sim 100 \%$ ) which was used in the next step without further purification. ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta=7.32(\mathrm{~d}, J$ $=8.6 \mathrm{~Hz}, 2 \mathrm{H}), 7.27(\mathrm{~d}, J=8.5 \mathrm{~Hz}, 2 \mathrm{H}), 5.81-5.72(\mathrm{~m}, 1 \mathrm{H}), 5.19-5.15(\mathrm{~m}, 1 \mathrm{H}), 5.14(\mathrm{q}, J=1.3$ $\mathrm{Hz}, 1 \mathrm{H}), 4.70(\mathrm{dd}, J=7.6,5.3 \mathrm{~Hz}, 1 \mathrm{H}), 2.60(\mathrm{~s}, 1 \mathrm{H}), 2.53-2.48(\mathrm{~m}, 1 \mathrm{H}), 2.48-2.42(\mathrm{~m}, 1 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta=142.0,133.8,133.1,128.5,127.2,118.8,72.6,43.7$. IR (FTIR, film, $\left.\mathrm{cm}^{-1}\right) \tilde{\mathrm{v}}=3348,3077,2905,1640,1492,1090,1012,916,824.4 .1 \mathrm{~b}$ is a known compound and spectroscopic data are in agreement with those reported in literature. ${ }^{57}$

## Synthesis of 1-(4-(trifluoromethyl)phenyl)but-3-en-1-ol (4.1c)



Following general procedure A, commercial zinc dust ( $13.08 \mathrm{~g}, 200 \mathrm{mmol}, 2.0$ equiv), allyl bromide ( $13 \mathrm{~mL}, 150 \mathrm{mmol}, 1.5$ equiv), 4-trifluoromethylbenzaldehyde ( $17.41 \mathrm{~g}, 100 \mathrm{mmol}, 1.0$
equiv.) and THF ( 200 mL ) homoallylic alcohol 4.1c was obtained as a yellow oil in THF (27.84 g , quantitative yield) which was used in the next step without further purification. ${ }^{1} \mathrm{H}-\mathrm{NMR}$ ( 500 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta=7.60(\mathrm{~d}, J=8.1 \mathrm{~Hz}, 2 \mathrm{H}), 7.46(\mathrm{~d}, J=8.5 \mathrm{~Hz}, 2 \mathrm{H}), 5.83-5.73(\mathrm{~m}, 1 \mathrm{H}), 5.19(\mathrm{~h}$, $J=1.8 \mathrm{~Hz}, 1 \mathrm{H}), 5.16(\mathrm{tq}, J=3.2,1.4 \mathrm{~Hz}, 1 \mathrm{H}), 4.79(\mathrm{dd}, J=8.1,4.7 \mathrm{~Hz}, 1 \mathrm{H}), 2.53(\mathrm{dddt}, J=14.0$, 6.2, 4.7, 1.3 Hz, 1H), $2.45(\mathrm{dtt}, J=14.1,7.9,1.1 \mathrm{~Hz}, 1 \mathrm{H}), 2.32(\mathrm{~s}, 1 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( 126 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta=147.7(\mathrm{~d}, J=1.4 \mathrm{~Hz}), 133.7,129.7(\mathrm{q}, J=32.4 \mathrm{~Hz}), 126.1,125.3(\mathrm{q}, J=3.8 \mathrm{~Hz}), 124.2$ $(\mathrm{q}, J=271.9 \mathrm{~Hz}), 119.2,72.5,43.9 .4 .1 \mathrm{c}$ is a known compound and spectroscopic data are in agreement with those reported in literature. ${ }^{57}$

## Synthesis of 1-(naphthalen-1-yl)but-3-en-1-ol (4.1d)



Following general procedure A , commercial zinc dust ( $11.77 \mathrm{~g}, 180 \mathrm{mmol}, 2.0$ equiv), allyl bromide ( $13 \mathrm{~mL}, 150 \mathrm{mmol}, 1.5$ equiv), 1-naphthaldehyde ( $15.62 \mathrm{~g}, 100 \mathrm{mmol}, 1.0$ equiv.) and THF ( 200 mL ) homoallylic alcohol 4.1d was obtained as a yellow oil in THF (39.11 g, quantitative yield) which was used in the next step without further purification. ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ $=8.11-8.04(\mathrm{~m}, 1 \mathrm{H}), 7.90(\mathrm{dd}, J=8.1,1.6 \mathrm{~Hz}, 1 \mathrm{H}), 7.80(\mathrm{dt}, J=8.1,1.1 \mathrm{~Hz}, 1 \mathrm{H}), 7.67(\mathrm{dt}, J=$ $7.2,1.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.56-7.47(\mathrm{~m}, 3 \mathrm{H}), 5.94(\mathrm{dddd}, J=16.9,10.2,7.6,6.5 \mathrm{~Hz}, 1 \mathrm{H}), 5.52(\mathrm{dd}, J=$ $8.4,4.1 \mathrm{~Hz}, 1 \mathrm{H}), 5.27-5.17(\mathrm{~m}, 2 \mathrm{H}), 2.77(\mathrm{dddt}, J=14.3,6.7,4.1,1.4 \mathrm{~Hz}, 1 \mathrm{H}), 2.62(\mathrm{dtt}, J=$ $14.3,8.6,1.1 \mathrm{~Hz}, 1 \mathrm{H}), 2.37(\mathrm{~s}, 1 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta=139.3,134.7,133.7,130.2$, $128.9,127.9,126.0,125.5,125.4,122.9,122.8,118.3,69.9,42.8 .4 .1 d$ is a known compound and
spectroscopic data are in agreement with those reported in literature. ${ }^{58}$

## Synthesis of 1-(naphthalen-2-yl)but-3-en-1-ol (4.1f)



Following general procedure A, commercial zinc dust ( $5.23 \mathrm{~g}, 80 \mathrm{mmol}, 2.0$ equiv), allyl bromide ( $5.2 \mathrm{~mL}, 60 \mathrm{mmol}, 1.5$ equiv), 2-naphthaldehyde ( $6.25 \mathrm{~g}, 40 \mathrm{mmol}, 1.0$ equiv.) and THF ( 100 mL ) homoallylic alcohol 4.1 f was obtained as a yellow oil in THF $(9.64 \mathrm{~g}$, quantitative yield) which was used in the next step without further purification. ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta=7.87$ $-7.82(\mathrm{~m}, 3 \mathrm{H}), 7.81(\mathrm{~s}, 1 \mathrm{H}), 7.54-7.45(\mathrm{~m}, 3 \mathrm{H}), 5.84(\mathrm{ddt}, J=17.1,10.1,7.1 \mathrm{~Hz}, 1 \mathrm{H}), 5.25-$ $5.13(\mathrm{~m}, 2 \mathrm{H}), 4.91(\mathrm{dd}, J=7.7,5.2 \mathrm{~Hz}, 1 \mathrm{H}), 2.68-2.55(\mathrm{~m}, 2 \mathrm{H}), 2.27(\mathrm{~s}, 1 \mathrm{H}) .{ }^{13} \mathrm{C}\left\{{ }^{1} \mathrm{H}\right\}-\mathrm{NMR}(126$ $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta=141.2,134.3,133.2,132.9,128.2,127.9,127.6,126.1,125.8,124.5,124.0$, 118.5, 73.3, 43.7. IR (FTIR, $\left.\mathrm{cm}^{-1}\right) \tilde{v}=3342,3054,1638,1507,1270,1124,1042,817,744$. S1-f is a known compound and spectroscopic data are in agreement with those reported in literature. ${ }^{57}$

## Synthesis of 1-(4-methoxyphenyl)but-3-en-1-ol (4.1h)



Following general procedure A , commercial zinc dust ( $5.88 \mathrm{~g}, 90 \mathrm{mmol}, 1.8$ equiv), allyl bromide ( $6.5 \mathrm{~mL}, 60 \mathrm{mmol}, 1.5$ equiv), 6.1 mL of $p$-anisaldehyde ( $6.81 \mathrm{~g}, 50 \mathrm{mmol}, 1.0$ equiv.) and THF ( 80 mL ) homoallylic alcohol 4.1h was obtained as a yellow oil in THF ( 13.44 g , quantitative yield) which was used in the next step without further purification. ${ }^{1} \mathrm{H}-\mathrm{NMR}$ (500 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta=7.27(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 2 \mathrm{H}), 6.87(\mathrm{~d}, J=8.8 \mathrm{~Hz}, 2 \mathrm{H}), 5.78(\mathrm{ddt}, J=17.3,10.2,7.1$ $\mathrm{Hz}, 1 \mathrm{H}), 5.17-5.09(\mathrm{~m}, 2 \mathrm{H}), 4.66(\mathrm{t}, J=6.6 \mathrm{~Hz}, 1 \mathrm{H}), 3.79(\mathrm{~s}, 3 \mathrm{H}), 2.53-2.44(\mathrm{~m}, 2 \mathrm{H}), 2.23(\mathrm{~s}$, 1H). ${ }^{13} \mathrm{C}$ NMR ( $\left.126 \mathrm{MHz}, \mathrm{CDCl}_{3}, \mathrm{ppm}\right) \delta=158.9,136.0,134.6,127.0,118.1,113.7,72.9,55.2$, 43.6. IR $\left(\mathrm{FTIR}^{2} \mathrm{~cm}^{-1}\right) \tilde{\mathrm{v}}=3395,2933,2835,1610,1510,1242,1173,1032,829.4 .1 \mathrm{~h}$ is a known compound and spectroscopic data are in agreement with those reported in literature. ${ }^{57}$

## Synthesis of 1-phenylbut-3-en-1-yl 2,2,2-trichloroacetimidate (4.2a)



Following general procedure B, 1-phenylbut-3-en-1-ol 4.1-a ( $2.23 \mathrm{~g}, 15 \mathrm{mmol}, 1.0$ equiv), DBU ( $0.4 \mathrm{~mL}, 2.7 \mathrm{mmol}, 0.18$ equiv), trichloroacetonitrile ( $2.1 \mathrm{~mL}, 21 \mathrm{mmol}, 1.4$ equiv.) and dichloromethane (40 mL), 1-phenylbut-3-en-1-yl 2,2,2-trichloroacetimidate 4.1a was obtained as a dark brown oil in dichloromethane ( $4.5 \mathrm{~g}, \sim 100 \%$ ) which was used in the next step without further purification. ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta=8.30(\mathrm{~s}, 1 \mathrm{H}), 7.45-7.40(\mathrm{~m}, 2 \mathrm{H}), 7.37$ (ddd, $J=7.7,6.7,1.3 \mathrm{~Hz}, 2 \mathrm{H}), 7.34-7.29(\mathrm{~m}, 1 \mathrm{H}), 5.90(\mathrm{dd}, J=7.9,5.4 \mathrm{~Hz}, 1 \mathrm{H}), 5.83(\mathrm{ddt}, J=17.2$, $10.2,6.9 \mathrm{~Hz}, 1 \mathrm{H}), 5.14(\mathrm{dq}, J=17.2,1.6 \mathrm{~Hz}, 1 \mathrm{H}), 5.10(\mathrm{ddt}, J=10.2,2.1,1.1 \mathrm{~Hz}, 1 \mathrm{H}), 2.81$ (dddt, $J=14.7,8.0,6.8,1.3 \mathrm{~Hz}, 1 \mathrm{H}), 2.66(\mathrm{dddd}, J=14.5,6.9,4.1,1.3 \mathrm{~Hz}, 1 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( 126 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta=161.5,139.6,133.0,128.4,128.0,126.2,118.2,91.6,80.1,41.0 .4 .1 \mathrm{a}$ is a known
compound and spectroscopic data are in agreement with those reported in literature. ${ }^{59}$

## Synthesis of 1-(4-chlorophenyl)but-3-en-1-yl 2,2,2-trichloroacetimidate (4.2b)



Following general procedure B, 1-(4-chlorophenyl)but-3-en-1-ol 4.1b (14.61 g, $80 \mathrm{mmol}, 1.0$ equiv), DBU ( $2.15 \mathrm{~mL}, 14.4 \mathrm{mmol}, 0.18$ equiv), trichloroacetonitrile ( $11.2 \mathrm{~mL}, 112 \mathrm{mmol}, 1.4$ equiv.) and dichloromethane (200 mL), 1-(4-chlorophenyl)but-3-en-1-yl 2,2,2trichloroacetimidate 4.2b was obtained as a dark brown oil in dichloromethane ( 29 g , quantitative yield) which was used in the next step without further purification. ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ $=8.31(\mathrm{~s}, 1 \mathrm{H}), 7.34(\mathrm{~s}, 4 \mathrm{H}), 5.86(\mathrm{dd}, J=7.8,5.6 \mathrm{~Hz}, 1 \mathrm{H}), 5.79(\mathrm{ddt}, J=17.1,10.2,6.9 \mathrm{~Hz}, 1 \mathrm{H})$, $5.17-5.06(\mathrm{~m}, 2 \mathrm{H}), 2.78$ (dddt, $J=14.6,8.0,6.8,1.3 \mathrm{~Hz}, 1 \mathrm{H}), 2.62$ (dddt, $J=14.2,7.0,5.6,1.2$ $\mathrm{Hz}, 1 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta=161.3,138.1,133.7,132.6,128.6,127.7,118.6,91.5$, 79.3, 40.8. 4.2b is a known compound and spectroscopic data are in agreement with those reported in literature. ${ }^{59}$

## Synthesis of 1-(4-(trifluoromethyl)phenyl)but-3-en-1-yl 2,2,2-trichloroacetimidate (4.2c)



Following general procedure C, 1-(4-trifluoromethylphenyl)but-3-en-1-ol 4.1c (10.81 g, 50 mmol, 1.0 equiv), $\mathrm{NaH} 60 \% \mathrm{w} / \mathrm{w}$ dispersion in mineral oil ( $400 \mathrm{mg}, 10 \mathrm{mmol}, 0.20$ equiv),
trichloroacetonitrile ( $5.51 \mathrm{~mL}, 55 \mathrm{mmol}, 1.1$ equiv.) and diethyl ether ( 15 mL ), $18.5 \mathrm{~g}, 51 \mathrm{mmol}$ (quantitative yield) of 1-(4-(trifluoromethyl)phenyl)but-3-en-1-yl 2,2,2-trichloroacetimidate 4.2c was obtained as a yellow oil after column chromatography, $R_{f}=0.3$ ( $5 \% \mathrm{EtOAc}$ in hexanes). ${ }^{1} \mathrm{H}$ NMR ( $\left.500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta=8.34(\mathrm{~s}, 1 \mathrm{H}), 7.63(\mathrm{~d}, J=8.1 \mathrm{~Hz}, 2 \mathrm{H}), 7.53(\mathrm{~d}, J=8.1 \mathrm{~Hz}, 2 \mathrm{H}), 5.93$ (dd, $J=7.7,5.5 \mathrm{~Hz}, 1 \mathrm{H}), 5.80(\mathrm{ddt}, J=17.2,10.3,7.0 \mathrm{~Hz}, 1 \mathrm{H}), 5.18-5.09(\mathrm{~m}, 2 \mathrm{H}), 2.79$ (dddt, $J$ $=14.6,8.0,6.8,1.3 \mathrm{~Hz}, 1 \mathrm{H}), 2.65(\mathrm{dddt}, J=14.1,6.9,5.5,1.3 \mathrm{~Hz}, 1 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( 126 MHz , $\left.\mathrm{CDCl}_{3}, \mathrm{ppm}\right) \delta=161.4,143.6(\mathrm{~d}, J=1.3 \mathrm{~Hz}), 132.4,130.2(\mathrm{q}, J=32.5 \mathrm{~Hz}), 126.5,125.5(\mathrm{q}, J=$ $3.8 \mathrm{~Hz}) 124.0(\mathrm{q}, J=272.2 \mathrm{~Hz}), 118.9,91.4,79.3,40.8 .4 .2 \mathrm{c}$ is a known compound and spectroscopic data are in agreement with those reported in literature. ${ }^{59}$

## Synthesis of 1-(naphthalen-1-yl)but-3-en-1-yl 2,2,2-trichloroacetimidate (4.2d)



Following general procedure C, 1-(naphthalen-1-yl)but-3-en-1-ol 4.1d (9.9 g, $50 \mathrm{mmol}, 1.0$ equiv), $\mathrm{NaH} 60 \% \mathrm{w} / \mathrm{w}$ dispersion in mineral oil ( $400 \mathrm{mg}, 10 \mathrm{mmol}, 0.20$ equiv), trichloroacetonitrile ( $5.51 \mathrm{~mL}, 55 \mathrm{mmol}, 1.1$ equiv.) and diethyl ether ( 15 mL ), $16.86 \mathrm{~g}, 49 \mathrm{mmol}$ ( $98 \%$ isolated yield) of 1-(naphthalen-1-yl)but-3-en-1-yl 2,2,2-trichloroacetimidate 4.2 d was obtained as a yellow oil after column chromatography, $R_{f}=0.4$ ( $30 \%$ DCM in hexanes). ${ }^{1} \mathrm{H}$ NMR $\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta=8.36(\mathrm{~s}, 1 \mathrm{H}), 8.20(\mathrm{~d}, J=8.5 \mathrm{~Hz}, 1 \mathrm{H}), 7.92(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.85(\mathrm{~d}, J$ $=8.2 \mathrm{~Hz}, 1 \mathrm{H}), 7.73(\mathrm{~d}, J=7.1 \mathrm{~Hz}, 1 \mathrm{H}), 7.60(\mathrm{ddd}, J=8.4,6.8,1.5 \mathrm{~Hz}, 1 \mathrm{H}), 7.57-7.48(\mathrm{~m}, 2 \mathrm{H})$, $6.73(\mathrm{dd}, J=8.2,4.8 \mathrm{~Hz}, 1 \mathrm{H}), 5.96(\mathrm{ddt}, J=17.1,10.2,6.9 \mathrm{~Hz}, 1 \mathrm{H}), 5.21(\mathrm{dq}, J=17.1,1.6 \mathrm{~Hz}$, $1 \mathrm{H}), 5.15$ (dt, $J=10.3,1.4 \mathrm{~Hz}, 1 \mathrm{H}), 2.97(\mathrm{dddt}, J=15.0,8.2,6.8,1.3 \mathrm{~Hz}, 1 \mathrm{H}), 2.89(\mathrm{dddt}, J=$
$14.6,7.3,4.8,1.3 \mathrm{~Hz}, 1 \mathrm{H}) .{ }^{13} \mathrm{C} \operatorname{NMR}\left(126 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta=161.5,135.5,133.7,133.4,130.1$, $128.9,128.5,126.3,125.6,125.3,123.5,123.0,118.1,91.7,77.5,40.5 . \mathrm{MS}(\mathrm{GC} / \mathrm{MS}): \mathrm{m} / \mathrm{z}(\%)=$ 181 (12.5) $\left[\mathrm{M}-\mathrm{Cl}_{3} \mathrm{CCONH}\right]^{+}: \mathrm{C}_{14} \mathrm{H}_{13}$ HRMS (ESI), $m / z[\mathrm{M}+\mathrm{H}]^{+}$calcd for $\mathrm{C}_{16} \mathrm{H}_{15} \mathrm{Cl}_{3} \mathrm{NO}:$ 342.0219 found: 342.0227 .

Synthesis of 1-(naphthalen-2-yl)but-3-en-1-yl 2,2,2-trichloroacetimidate (4.2f)


Following general procedure B, 1-(naphthalen-2-yl)but-3-en-1-ol 4.1f (5.95 g, $30 \mathrm{mmol}, 1.0$ equiv), $\operatorname{DBU}$ ( $0.81 \mathrm{~mL}, 5.4 \mathrm{mmol}, 0.18$ equiv), trichloroacetonitrile ( $4.21 \mathrm{~mL}, 42 \mathrm{mmol}, 1.4$ equiv.) and dichloromethane $(80 \mathrm{~mL}), 11.04 \mathrm{~g}, 32 \mathrm{mmol}$ (quantitative yield) of 1-(naphthalen-2-yl)but-3-en-1-yl 2,2,2-trichloroacetimidate $\mathbf{4 . 2 f}$ was obtained as a yellow oil after column chromatography, $R_{f}=0.3\left(4 \%\right.$ EtOAc in hexanes). ${ }^{1} \mathrm{H} \operatorname{NMR}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta=8.33(\mathrm{~s}, 1 \mathrm{H}), 7.90(\mathrm{~s}, 1 \mathrm{H}), 7.89$ - $7.83(\mathrm{~m}, 3 \mathrm{H}), 7.56(\mathrm{dd}, J=8.5,1.8 \mathrm{~Hz}, 1 \mathrm{H}), 7.54-7.46(\mathrm{~m}, 2 \mathrm{H}), 6.08(\mathrm{dd}, J=7.9,5.5 \mathrm{~Hz}, 1 \mathrm{H})$, $5.87(\mathrm{ddt}, J=17.1,10.2,6.9 \mathrm{~Hz}, 1 \mathrm{H}), 5.17(\mathrm{dq}, J=17.1,1.5 \mathrm{~Hz}, 1 \mathrm{H}), 5.12(\mathrm{ddt}, J=10.2,2.0,1.1$ $\mathrm{Hz}, 1 \mathrm{H}), 2.92$ (dddt, $J=14.7,8.0,6.8,1.3 \mathrm{~Hz}, 1 \mathrm{H}), 2.76(\mathrm{dddt}, J=14.2,6.9,5.6,1.3 \mathrm{~Hz}, 1 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR (126 MHz, $\left.\mathrm{CDCl}_{3}\right) \delta=161.5,136.9,133.1,133.02,133.00,128.3,128.1,127.7,126.2$, $126.1,125.5,124.0,118.3,91.7,80.2,40.9 .4 .2 \mathrm{f}$ is a known compound and spectroscopic data are in agreement with those reported in literature. ${ }^{26}$

Synthesis of 1-(4-methoxyphenyl)but-3-en-1-yl 2,2,2-trichloroacetimidate (4.2h)


Following general procedure C, 1-(4-methoxyphenyl)but-3-en-1-ol 4.1h ( $8.9 \mathrm{~g}, 50 \mathrm{mmol}, 1.0$ equiv), $\mathrm{NaH} 60 \% \mathrm{w} / \mathrm{w}$ dispersion in mineral oil ( $360 \mathrm{mg}, 9 \mathrm{mmol}, 0.18$ equiv), trichloroacetonitrile ( $5.51 \mathrm{~mL}, 55 \mathrm{mmol}, 1.1$ equiv.) and diethyl ether ( 15 mL ), $15.86 \mathrm{~g}, 49 \mathrm{mmol},(98 \%$ crude yield) of 1-(4-methoxyphenyl)but-3-en-1-yl 2,2,2-trichloroacetimidate 4.2h which was used in the next step without further purification. It is worth noting that compound 4.2h is unstable on silica and it breaks down to the starting alcohol. ${ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta=8.27(\mathrm{~s}, 1 \mathrm{H}), 7.35(\mathrm{~d}, J=8.5$ $\mathrm{Hz}, 2 \mathrm{H}), 6.90(\mathrm{~d}, J=8.8 \mathrm{~Hz}, 2 \mathrm{H}), 5.87-5.76(\mathrm{~m}, 2 \mathrm{H}), 5.13(\mathrm{dq}, J=17.2,1.6 \mathrm{~Hz}, 1 \mathrm{H}), 5.09(\mathrm{ddt}$, $J=10.3,2.1,1.1 \mathrm{~Hz}, 1 \mathrm{H}), 3.81(\mathrm{~s}, 3 \mathrm{H}), 2.80(\mathrm{dddt}, J=14.7,8.0,6.8,1.3 \mathrm{~Hz}, 1 \mathrm{H}), 2.62(\mathrm{dddt}, J=$ 14.2, 7.0, 5.7, 1.3 Hz, 1H). ${ }^{13} \mathrm{C}$ NMR (126 MHz, $\mathrm{CDCl}_{3}$ ) $\delta=161.5$, 159.3, 133.2, 131.6, 127.7, $118.1,113.7,91.7,79.9,55.2,40.9 .4 .2 \mathrm{~h}$ is a known compound and spectroscopic data are in agreement with those reported in literature. ${ }^{26}$

Synthesis of 1-(trimethylsilyl)prop-2-en-1-ol (4.3a)


Following general procedure D, a solution of allyl alcohol ( $3.49 \mathrm{~g}, 60 \mathrm{mmol}, 1$ equiv.) in THF ( 150 mL ) was cooled to $-78{ }^{\circ} \mathrm{C} . n-\mathrm{BuLi}(2.5 \mathrm{M}$ in hexanes, $26.4 \mathrm{~mL}, 66 \mathrm{mmol}$, 1.1 equiv.) was
added dropwise and the mixture stirred at $-78^{\circ} \mathrm{C}$ for 1 h . Freshly distilled chlorotrimethylsilane ( $7.6 \mathrm{~mL}, 60 \mathrm{mmol}, 1$ equiv.) was then added slowly from a syringe and the resulting mixture was stirred at $-78^{\circ} \mathrm{C}$ for 1.5 hours resulting in the formation of a white suspension. This was followed by addition of tert -BuLi (1.7 M in pentane, $43 \mathrm{~mL}, 72 \mathrm{mmol}, 1.2$ equiv.) dropwise via cannula and the reaction stirred for an additional 1.5 hours at $-78^{\circ} \mathrm{C}$. The reaction was quenched by the addition of aqueous $\mathrm{NH}_{4} \mathrm{Cl}$ and diluted with $\mathrm{Et}_{2} \mathrm{O}$, and the mixture was warmed to room temperature. After the layers were separated, the aqueous phase was washed with $\mathrm{Et}_{2} \mathrm{O}(3 \times 50 \mathrm{~mL})$. Then all the organic phases were combined, washed with $\mathrm{H}_{2} \mathrm{O}(50 \mathrm{~mL})$ and brine ( 50 mL ) respectively, and dried over anhydrous $\mathrm{MgSO}_{4}$. After filtration and concentration, the residue was purified by column chromatography, $R_{f}=0.5\left(30 \% \mathrm{Et}_{2} \mathrm{O}\right.$ in pentane) to afford $5.17 \mathrm{~g}, 47.4 \mathrm{mmol}$, ( $79 \%$ isolated yield) of compound 4.3a as a colorless liquid. ${ }^{1} \mathrm{H}$ NMR $\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta=6.03$ (ddd, $J=17.1,10.7,5.3 \mathrm{~Hz}, 1 \mathrm{H}), 5.07(\mathrm{ddd}, J=17.2,2.1,1.5 \mathrm{~Hz}, 1 \mathrm{H}), 4.99(\mathrm{dt}, J=10.8,1.8 \mathrm{~Hz}, 1 \mathrm{H})$, $4.02(\mathrm{dt}, J=5.3,2.1 \mathrm{~Hz}, 1 \mathrm{H}), 1.42(\mathrm{~s}, 1 \mathrm{H}), 0.04(\mathrm{~s}, 9 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR $\left(126 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta=139.9$, 109.4, 69.0, -4.3. IR (FTIR, film, $\mathrm{cm}^{-1}$ ) $\tilde{\mathrm{v}}=3405,2956,1632,1247,895.4 .3 \mathrm{a}$ is a known compound and spectroscopic data are in agreement with those reported in literature. ${ }^{60}$ Synthesis of 2-methyl-1-(trimethylsilyl)prop-2-en-1-ol (4.3b) ${ }^{61}$


Following general procedure D , a solution of 2-methylprop-2-en-1-ol (4.33 g, $60 \mathrm{mmol}, 1$ equiv.) in THF ( 110 mL ) was cooled to $-78^{\circ} \mathrm{C} . n-\mathrm{BuLi}(2.5 \mathrm{M}$ in hexanes, $26.4 \mathrm{~mL}, 66 \mathrm{mmol}, 1.1$ equiv.) was added dropwise and the mixture stirred at $-78{ }^{\circ} \mathrm{C}$ for 1 h . Freshly distilled
chlorotrimethylsilane ( $7.6 \mathrm{~mL}, 60 \mathrm{mmol}, 1$ equiv.) was then added slowly from a syringe and the resulting mixture was stirred at $-78{ }^{\circ} \mathrm{C}$ for 1.5 hours resulting in the formation of a white suspension. This was followed by addition of tert- $\mathrm{BuLi}(1.7 \mathrm{M}$ in pentane, $45.9 \mathrm{~mL}, 78 \mathrm{mmol}, 1.3$ equiv.) dropwise via cannula and the reaction warmed up slowly to $-35^{\circ} \mathrm{C}$ and stirred at this temperature for an additional 3.5 hours. The reaction was cooled down to $-78{ }^{\circ} \mathrm{C}$ and quenched by the addition of 4.5 mL acetic acid solution in 10 mL THF and the cold bath was removed. The reaction mixture was diluted with saturated aqueous $\mathrm{NaHCO}_{3}(50 \mathrm{~mL})$ solution and pentane (150 mL . After the layers were separated organic phase was washed with $\mathrm{H}_{2} \mathrm{O}(50 \mathrm{~mL} \times 3)$ and brine ( 50 mL ) respectively and dried over anhydrous $\mathrm{MgSO}_{4}$. After filtration and concentration, the residue was purified by column chromatography, $R_{f}=0.4\left(15 \% \mathrm{Et}_{2} \mathrm{O}\right.$ in hexanes) to afford 5.37 g , 37.2 mmol , ( $62 \%$ isolated yield) of compound $\mathbf{4 . 3 b}$ as a pale-yellow liquid. ${ }^{1} \mathrm{H}$ NMR ( 500 MHz , $\left.\mathrm{C}_{6} \mathrm{D}_{6}\right) \delta=4.85(\mathrm{tq}, J=1.7,0.8 \mathrm{~Hz}, 1 \mathrm{H}), 4.74(\mathrm{~h}, J=1.4 \mathrm{~Hz}, 1 \mathrm{H}), 3.63(\mathrm{~d}, J=1.4 \mathrm{~Hz}, 1 \mathrm{H}), 1.57$ $(\mathrm{dt}, J=1.5,0.6 \mathrm{~Hz}, 3 \mathrm{H}), 1.00(\mathrm{~s}, 1 \mathrm{H}), 0.08(\mathrm{~s}, 9 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR (126 MHz, $\left.\mathrm{C}_{6} \mathrm{D}_{6}\right) \delta=148.6,106.6$, $71.4,20.8,-3.3$. 4.3b is a known compound. ${ }^{61}$

## Synthesis of 1-(triethylsilyl)prop-2-en-1-ol (4.3c)



Following general procedure D , a solution of allyl alcohol ( $1.16 \mathrm{~g}, 20 \mathrm{mmol}, 1$ equiv.) in THF ( 35 mL ) was cooled to $-78^{\circ} \mathrm{C} . n-\mathrm{BuLi}(2.5 \mathrm{M}$ in hexanes, $10 \mathrm{~mL}, 24 \mathrm{mmol}, 1.2$ equiv.) was added dropwise and the mixture stirred at $-78^{\circ} \mathrm{C}$ for 1 h . Chlorotriethylsilane ( $3.7 \mathrm{~mL}, 22 \mathrm{mmol}, 1.1$ equiv.) was then added slowly from a syringe and the resulting mixture was stirred at $-78{ }^{\circ} \mathrm{C}$ to
room temperature for 18 hours resulting in the formation of a white suspension. The mixture was cooled back to $-78{ }^{\circ} \mathrm{C}$ followed by dropwise addition of sec-butyllithium (1.4 M in cyclohexane, $18.6 \mathrm{~mL}, 26 \mathrm{mmol}, 1.3$ equiv.) and the reaction stirred for an additional 2 hours at $-78{ }^{\circ} \mathrm{C}$ to -50 ${ }^{\circ} \mathrm{C}$. The reaction mixture was cooled back to $-78{ }^{\circ} \mathrm{C}$ and quenched by the addition of aqueous $\mathrm{NH}_{4} \mathrm{Cl}$ and diluted with $\mathrm{Et}_{2} \mathrm{O}$, and the mixture was warmed to room temperature. After the layers were separated, the aqueous phase was washed with $\mathrm{Et}_{2} \mathrm{O}(3 \times 20 \mathrm{~mL})$. Then all the organic phases were combined, washed with $\mathrm{H}_{2} \mathrm{O}(20 \mathrm{~mL})$ and brine ( 20 mL ) respectively, and dried over anhydrous $\mathrm{MgSO}_{4}$. After filtration and concentration, the residue was purified by column chromatography, $R_{f}=0.3$ ( $15 \% \mathrm{Et}_{2} \mathrm{O}$ in hexanes) to afford $2.97 \mathrm{~g}, 17.2 \mathrm{mmol},(86 \%$ isolated yield) of compound 4.3 c as a colorless liquid. ${ }^{1} \mathrm{H} \operatorname{NMR}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta=6.07(\mathrm{ddd}, J=17.1,10.7$, $5.2 \mathrm{~Hz}, 1 \mathrm{H}), 5.09(\mathrm{ddd}, J=17.2,2.2,1.6 \mathrm{~Hz}, 1 \mathrm{H}), 4.97(\mathrm{ddd}, J=10.7,2.1,1.6 \mathrm{~Hz}, 1 \mathrm{H}), 4.18(\mathrm{dt}$, $J=5.2,2.1 \mathrm{~Hz}, 1 \mathrm{H}), 1.30(\mathrm{~s}, 1 \mathrm{H}), 0.98(\mathrm{t}, J=8.0 \mathrm{~Hz}, 9 \mathrm{H}), 0.62(\mathrm{qd}, J=8.0,1.6 \mathrm{~Hz}, 6 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR $\left(126 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta=140.4,109.0,67.4,7.4,1.5 .4 .3 \mathrm{c}$ is a known compound and spectroscopic data are in agreement with those reported in literature. ${ }^{60}$

## Synthesis of 1-(dimethyl(phenyl)silyl)prop-2-en-1-ol (4.3d)



Following general procedure D , a solution of allyl alcohol ( $1.16 \mathrm{~g}, 20 \mathrm{mmol}, 1$ equiv.) in THF ( 35 mL ) was cooled to $-78^{\circ} \mathrm{C} . n-\mathrm{BuLi}(2.5 \mathrm{M}$ in hexanes, $10 \mathrm{~mL}, 24 \mathrm{mmol}, 1.2$ equiv.) was added dropwise and the mixture stirred at $-78{ }^{\circ} \mathrm{C}$ for 1 h . chlorodimethyl(phenyl)silane ( $3.7 \mathrm{~mL}, 22$ mmol, 1.1 equiv.) was then added slowly from a syringe and the resulting mixture was stirred at -
$78{ }^{\circ} \mathrm{C}$ to room temperature for 18 hours resulting in the formation of a white suspension. The mixture was cooled back to $-78{ }^{\circ} \mathrm{C}$ followed by dropwise addition of sec-butyllithium (1.4 M in cyclohexane, $18.6 \mathrm{~mL}, 26 \mathrm{mmol}, 1.3$ equiv.) and the reaction stirred for an additional 2 hours at $78{ }^{\circ} \mathrm{C}$ to $-50^{\circ} \mathrm{C}$. The reaction mixture was cooled back to $-78^{\circ} \mathrm{C}$ and quenched by the addition of aqueous $\mathrm{NH}_{4} \mathrm{Cl}$ and diluted with $\mathrm{Et}_{2} \mathrm{O}$, and the mixture was warmed to room temperature. After the layers were separated, the aqueous phase was washed with $\mathrm{Et}_{2} \mathrm{O}(3 \times 20 \mathrm{~mL})$. Then all the organic phases were combined, washed with $\mathrm{H}_{2} \mathrm{O}(20 \mathrm{~mL})$ and brine $(20 \mathrm{~mL})$ respectively, and dried over anhydrous $\mathrm{MgSO}_{4}$. After filtration and concentration the residue was purified by column chromatography, $R_{f}=0.4\left(15 \% \mathrm{Et}_{2} \mathrm{O}\right.$ in hexanes) to afford $2.73 \mathrm{~g}, 14.2 \mathrm{mmol},(71 \%$ isolated yield) of compound $\mathbf{4 . 3 d}$ as a colorless liquid. ${ }^{1} \mathrm{H} \mathrm{NMR}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta=7.64-$ $7.56(\mathrm{~m}, 2 \mathrm{H}), 7.45-7.36(\mathrm{~m}, 3 \mathrm{H}), 6.01(\mathrm{ddd}, J=17.1,10.7,5.2 \mathrm{~Hz}, 1 \mathrm{H}), 5.08(\mathrm{dt}, J=17.2,1.8$ $\mathrm{Hz}, 1 \mathrm{H}), 5.02(\mathrm{dt}, J=10.7,1.7 \mathrm{~Hz}, 1 \mathrm{H}), 4.23(\mathrm{dt}, J=5.1,2.0 \mathrm{~Hz}, 1 \mathrm{H}), 1.43(\mathrm{~s}, 1 \mathrm{H}), 0.37(\mathrm{~s}, 3 \mathrm{H})$, $0.35(\mathrm{~s}, 3 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR $\left(126 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta=139.3,136.0,134.2,129.5,127.8,110.0,68.4,-5.8$, -6.1. 4.3d is a known compound and spectroscopic data are in agreement with those reported in literature. ${ }^{62,63}$

Synthesis of syn/anti-trimethyl(2-methyl-1-((1-phenylbut-3-en-1-yl)oxy)allyl)silane (syn/anti-4.4a)


Compound 4.4a was prepared following general procedure E, a solution of 2-methyl-1-(trimethylsilyl)prop-2-en-1-ol 4.3b ( $2.47 \mathrm{~g}, 17.1 \mathrm{mmol}, 1$ equiv.) and 1-phenylbut-3-en-1-yl 2,2,2trichloroacetimidate $4.2 \mathrm{a}(10.0 \mathrm{~g}, 34.18 \mathrm{mmol}, 2.0$ equiv.), TMSOTf ( $0.46 \mathrm{~mL}, 2.6 \mathrm{mmol}, 0.15$ equiv.) and cyclohehane ( 80 mL ) for 12 hour followed by workup, concentration and column chromatography, $R_{f}$ for syn/anti-4.4d $=0.6(10 \% \mathrm{DCM}$ in hexanes) afforded $1.64 \mathrm{~g}, 6 \mathrm{mmol}(35 \%$ isolated yield) inseparable mixture of diastereomers of compound $\mathbf{4 . 4 d}$ (syn:anti $=1: 1$ ) as colorless liquid. ${ }^{1} \mathrm{H}$ NMR $\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta=7.35-7.26(\mathrm{~m}, 8 \mathrm{H}), 7.26-7.20(\mathrm{~m}, 2 \mathrm{H}), 5.81(\mathrm{ddt}, J=$ $17.2,10.2,7.1 \mathrm{~Hz}, 1 \mathrm{H}), 5.74-5.63(\mathrm{~m}, 1 \mathrm{H}), 5.03-4.95(\mathrm{~m}, 4 \mathrm{H}), 4.87-4.81(\mathrm{~m}, 1 \mathrm{H}), 4.69$ (dddd, $J=6.6,3.2,1.8,0.8 \mathrm{~Hz}, 3 \mathrm{H}), 4.34(\mathrm{dd}, J=6.4,5.4 \mathrm{~Hz}, 1 \mathrm{H}), 4.30(\mathrm{dd}, J=7.7,5.8 \mathrm{~Hz}, 1 \mathrm{H}), 3.79$ $(\mathrm{s}, 1 \mathrm{H}), 3.33(\mathrm{~s}, 1 \mathrm{H}), 2.59-2.52(\mathrm{~m}, 2 \mathrm{H}), 2.52-2.45(\mathrm{~m}, 1 \mathrm{H}), 2.40-2.33(\mathrm{~m}, 1 \mathrm{H}), 1.70-1.63$ $(\mathrm{m}, 3 \mathrm{H}), 1.53(\mathrm{dd}, J=1.4,0.8 \mathrm{~Hz}, 3 \mathrm{H}), 0.09(\mathrm{~s}, 9 \mathrm{H}), 0.00(\mathrm{~s}, 9 \mathrm{H}) .{ }^{13} \mathrm{C} \mathrm{NMR}\left(126 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ $=145.0,144.4,143.4,142.3,135.5,134.6,128.1,127.8,127.5,127.4,126.8,126.6,116.9,116.4$,
109.9, 109.5, 80.0, 79.1, 77.8, 75.4, 43.0, 40.5, 20.4, 20.3, -3.0, -3.2. Syn-4.4b and anti-4.4b are known compounds and spectroscopic data are in agreement with those reported in literature. ${ }^{26}$ Synthesis of syn/anti-(1-((1-(4-chlorophenyl)but-3-en-1-yl)oxy)allyl)trimethylsilane (syn/anti-4.4b)


Compound 4.4b was prepared following general procedure E with slight modification to minimize formation of the side product 4.10 b resulting from elimination. A solution of 1-(trimethylsilyl)prop-2-en-1-ol 4.3 a ( $1.96 \mathrm{~g}, 15 \mathrm{mmol}, 1$ equiv.) and 1-(4-chlorophenyl)but-3-en-1yl 2,2,2-trichloroacetimidate $\mathbf{4 . 2 b}(4.91 \mathrm{~g}, 15 \mathrm{mmol}, 1.0$ equiv) in dichloromethane ( 100 mL ) was cooled to $-78{ }^{\circ} \mathrm{C}$. TMSOTf ( $0.27 \mathrm{~mL}, 1.5 \mathrm{mmol}, 0.1$ equiv.) was added dropwise and the mixture stirred at $-78{ }^{\circ} \mathrm{C}$ for 6 hours. The rubber septum was removed, and 5 g of sodium bicarbonate was poured into the flask. The dry ice-acetone bath was removed, and the mixture was allowed to warm up to room temperature. The mixture was filtered and concentrated under reduced pressure to remove dichloromethane. Hexanes was then added to the resulting mixture resulting in the formation of white precipitate. Subsequent filtration and concentration furnished a residue which was purified by column chromatography, $R_{f}$ for $\mathbf{4 . 1 0 b}=0.8$ and $R_{f}$ for syn/anti-4.4b $=0.6(5 \%$

DCM in hexanes) to afford $2.15 \mathrm{~g}, 7.35 \mathrm{mmol}$, ( $49 \%$ isolated yield) inseparable mixture of diastereomers of compound 4.4b (syn:anti $=1: 1$ ) as colorless liquid and $1.26 \mathrm{~g}, 7.65 \mathrm{mmol}(51 \%$ isolated yield) elimination side product ( $E$ )-1-(buta-1,3-dien-1-yl)-4-chlorobenzene 4.10b ( $E: Z$ > 99:1).

Spectroscopic data for syn:anti-4.4b: ${ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta=7.30(\mathrm{~d}, J=8.4 \mathrm{~Hz}$, $2 \mathrm{H}), 7.27(\mathrm{~d}, J=8.6 \mathrm{~Hz}, 2 \mathrm{H}), 7.23(\mathrm{~d}, J=8.5 \mathrm{~Hz}, 2 \mathrm{H}), 7.18(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 2 \mathrm{H}), 5.82-5.60(\mathrm{~m}$, $4 \mathrm{H}), 5.06-4.95(\mathrm{~m}, 6 \mathrm{H}), 4.90(\mathrm{dt}, J=17.2,1.7 \mathrm{~Hz}, 1 \mathrm{H}), 4.85(\mathrm{dt}, J=10.6,1.6 \mathrm{~Hz}, 1 \mathrm{H}), 4.42(\mathrm{dd}$, $J=7.6,5.9 \mathrm{~Hz}, 1 \mathrm{H}), 4.35(\mathrm{t}, J=6.1 \mathrm{~Hz}, 1 \mathrm{H}), 3.80(\mathrm{dt}, J=7.2,1.5 \mathrm{~Hz}, 1 \mathrm{H}), 3.38(\mathrm{dt}, J=7.5,1.3$ $\mathrm{Hz}, 1 \mathrm{H}), 2.56-2.46(\mathrm{~m}, 2 \mathrm{H}), 2.40(\mathrm{dddt}, J=14.3,7.3,6.2,1.2 \mathrm{~Hz}, 1 \mathrm{H}), 2.36-2.28(\mathrm{~m}, 1 \mathrm{H}), 0.06$ (s, 9H), -0.01 (s, 9H). ${ }^{13} \mathrm{C}$ NMR ( $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta=142.1,140.9,137.6,137.3,134.9,134.3$, $133.0,132.5,128.7,128.3,128.0,127.9,117.2,116.8,113.1,112.1,80.3,78.5,75.9,73.1,42.9$, 41.3, -3.8, -4.0. Syn-4.4b and anti-4.4b are known compounds and spectroscopic data are in agreement with those reported in literature. ${ }^{26}$

Spectroscopic data for 4.10b: ${ }^{1} \mathrm{H}$ NMR $\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta=7.33(\mathrm{~d}, J=8.6 \mathrm{~Hz}, 2 \mathrm{H}), 7.29$ $(\mathrm{d}, J=8.6 \mathrm{~Hz}, 2 \mathrm{H}), 6.76(\mathrm{ddd}, J=15.2,10.2,0.9 \mathrm{~Hz}, 1 \mathrm{H}), 6.55-6.46(\mathrm{~m}, 2 \mathrm{H}), 5.36(\mathrm{dd}, J=16.4$, $1.6 \mathrm{~Hz}, 1 \mathrm{H}), 5.24-5.19(\mathrm{~m}, 1 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR (126 MHz, $\left.\mathrm{CDCl}_{3}\right) \delta=136.8,135.6,133.1,131.4$, $130.1,128.7,127.5,118.2 .4 .10 \mathrm{~b}$ is a known compound and spectroscopic data are in agreement with those reported in literature. ${ }^{64}$

Synthesis of syn/anti-(1-((1-(4-trifluromethylphenyl)but-3-en-1-yl)oxy)allyl) trimethylsilane (syn/anti-4.4c)


Compound 4.4c was prepared following general procedure E with slight modification to minimize formation of the side product as a result of elimination. A solution of 1-(trimethylsilyl)prop-2-en-1-ol 4.3a (2.61 g, $20 \mathrm{mmol}, \quad 1$ equiv.) and 1-(4-trifluoromethylphenyl)but-3-en-1-yl 2,2,2-trichloroacetimidate $\mathbf{4 . 2 c}(10.1 \mathrm{~g}, 28 \mathrm{mmol}, 1.4$ equiv.) in dichloromethane ( 100 mL ) was cooled to $-78^{\circ} \mathrm{C}$. TMSOTf ( $0.36 \mathrm{~mL}, 2.0 \mathrm{mmol}, 0.1$ equiv.) was added dropwise and the mixture stirred at $-78^{\circ} \mathrm{C}$ for 6 hours. The rubber septum was removed and 7 g of sodium bicarbonate was poured into the flask. The dry ice-acetone bath was removed and the mixture was allowed to warm up to room temperature. The mixture was filtered and concentrated under reduced pressure to remove dichloromethane. Hexanes was then added to the resulting mixture resulting in the formation of white precipitate. Subsequent filtration and concentration furnished a residue which was purified by column chromatography, $R_{f}$ for $\mathbf{4 . 1 0} \mathbf{c}=$ 0.7 and $R_{f}$ for syn/anti-4.4c $=0.5$ ( $100 \%$ hexanes) to afford $2.2 \mathrm{~g}, 6.6 \mathrm{mmol}(33 \%$ isolated yield $)$ inseparable mixture of diastereomers of compound $\mathbf{4 . 4} \mathbf{c}$ (syn:anti $=1: 1$ ) as colorless liquid and
$1.72 \mathrm{~g}, 8.7 \mathrm{mmol}$ ( $31 \%$ isolated yield based on the starting trichloroacetimidate) elimination side product $(E)$-1-(buta-1,3-dien-1-yl)-4-trifluoromethylbenzene 4.10c $(E: Z>99: 1)$.

Spectroscopic data for syn:anti-4.4c: ${ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta=7.59(\mathrm{~d}, J=8.1 \mathrm{~Hz}$, $2 \mathrm{H}), 7.56(\mathrm{~d}, J=8.1 \mathrm{~Hz}, 2 \mathrm{H}), 7.41(\mathrm{~d}, J=8.3 \mathrm{~Hz}, 2 \mathrm{H}), 7.37(\mathrm{~d}, J=8.1 \mathrm{~Hz}, 2 \mathrm{H}), 5.83-5.61(\mathrm{~m}$, $4 \mathrm{H}), 5.06(\mathrm{ddd}, J=10.6,2.0,1.2 \mathrm{~Hz}, 1 \mathrm{H}), 5.03-4.96(\mathrm{~m}, 5 \mathrm{H}), 4.91(\mathrm{dt}, J=17.2,1.8 \mathrm{~Hz}, 1 \mathrm{H})$, $4.87(\mathrm{ddd}, J=10.6,2.0,1.4 \mathrm{~Hz}, 1 \mathrm{H}), 4.52(\mathrm{dd}, J=7.6,5.8 \mathrm{~Hz}, 1 \mathrm{H}), 4.45(\mathrm{t}, J=6.0 \mathrm{~Hz}, 1 \mathrm{H}), 3.83$ $(\mathrm{dt}, J=7.3,1.5 \mathrm{~Hz}, 1 \mathrm{H}), 3.39(\mathrm{dt}, J=7.6,1.3 \mathrm{~Hz}, 1 \mathrm{H}), 2.57-2.48(\mathrm{~m}, 2 \mathrm{H}), 2.48-2.41(\mathrm{~m}, 1 \mathrm{H})$, $2.35(\mathrm{dddt}, J=14.1,7.1,5.8,1.2 \mathrm{~Hz}, 1 \mathrm{H}), 0.08(\mathrm{~s}, 9 \mathrm{H}), 0.01(\mathrm{~s}, 9 \mathrm{H}) .{ }^{13} \mathrm{C} \mathrm{NMR}\left(126 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ $=147.69(\mathrm{~d}, J=1.2 \mathrm{~Hz}), 146.65(\mathrm{~d}, J=1.2 \mathrm{~Hz}), 137.5,137.2,134.6,134.0,130.0-128.7(\mathrm{~m}, 2 \mathrm{C})$, $127.7-123.2(\mathrm{~m}, 2 \mathrm{C}), 127.5,126.7,125.12(\mathrm{q}, J=3.8 \mathrm{~Hz}), 124.85(\mathrm{q}, J=3.8 \mathrm{~Hz}), 117.5,117.0$, 113.3, 112.4, 80.2, 78.7, 76.1, 73.5, 42.9, 41.2, -3.8, -4.0. Syn-4.4c and anti-4.4c are known compounds and spectroscopic data are in agreement with those reported in literature. ${ }^{26}$

Spectroscopic data for 4.10c: ${ }^{1} \mathrm{H}$ NMR $\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta=7.57(\mathrm{~d}, J=8.2 \mathrm{~Hz}, 2 \mathrm{H}), 7.49$ $(\mathrm{d}, J=8.2 \mathrm{~Hz}, 2 \mathrm{H}), 6.87(\mathrm{dd}, J=15.7,10.5 \mathrm{~Hz}, 1 \mathrm{H}), 6.62-6.49(\mathrm{~m}, 2 \mathrm{H}), 5.42(\mathrm{~d}, J=16.9 \mathrm{~Hz}$, $1 \mathrm{H}), 5.28(\mathrm{~d}, J=10.0 \mathrm{~Hz}, 1 \mathrm{H}) .{ }^{13} \mathrm{C} \operatorname{NMR}\left(126 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta=140.58(\mathrm{q}, J=1.3 \mathrm{~Hz}), 136.7$, $132.0,131.2,129.25(\mathrm{q}, J=32.4 \mathrm{~Hz}), 126.5,125.55(\mathrm{q}, J=3.9 \mathrm{~Hz}), 124.21(\mathrm{q}, J=271.7 \mathrm{~Hz})$, 119.4. 4.10c is a known compound and spectroscopic data were in agreement with those reported in literature. ${ }^{65}$


Compound 4.4d was prepared following general procedure E, a solution of 1-(trimethylsilyl)prop-2-en-1-ol 4.3a ( $2.61 \mathrm{~g}, 20 \mathrm{mmol}, 1$ equiv.) and 1-(naphthalen-1-yl)but-3-en-1-yl 2,2,2-trichloroacetimidate $\mathbf{4 . 2 d}$ ( $9.6 \mathrm{~g}, 28 \mathrm{mmol}, 1.4$ equiv.), TMSOTf ( $0.36 \mathrm{~mL}, 2.0 \mathrm{mmol}$, 0.1 equiv.) and hexanes ( 50 mL ) for 12 hour followed by workup, concentration and column chromatography, $R_{f}$ for syn/anti-4.4d $=0.6(10 \% \mathrm{DCM}$ in hexanes $)$ afforded $2.11 \mathrm{~g}, 6.8 \mathrm{mmol}$ ( $34 \%$ isolated yield) inseparable mixture of diastereomers of compound $\mathbf{4 . 4 d}$ (syn:anti $=1: 1$ ) as colorless liquid. ${ }^{1} \mathrm{H}$ NMR $\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta=8.22(\mathrm{~d}, J=6.8 \mathrm{~Hz}, 1 \mathrm{H}), 8.12(\mathrm{~d}, J=8.1 \mathrm{~Hz}$, $1 \mathrm{H}), 7.92-7.86(\mathrm{~m}, 2 \mathrm{H}), 7.80(\mathrm{~d}, J=8.2 \mathrm{~Hz}, 1 \mathrm{H}), 7.77(\mathrm{~d}, J=8.2 \mathrm{~Hz}, 1 \mathrm{H}), 7.67-7.63(\mathrm{~m}, 1 \mathrm{H})$, $7.57-7.46(\mathrm{~m}, 7 \mathrm{H}), 5.94(\mathrm{ddt}, J=17.4,10.4,7.2 \mathrm{~Hz}, 1 \mathrm{H}), 5.89-5.80(\mathrm{~m}, 2 \mathrm{H}), 5.69(\mathrm{ddd}, J=$ $17.6,10.5,7.4 \mathrm{~Hz}, 1 \mathrm{H}), 5.31-5.23(\mathrm{~m}, 1 \mathrm{H}), 5.14(\mathrm{t}, J=6.0 \mathrm{~Hz}, 1 \mathrm{H}), 5.12-4.96(\mathrm{~m}, 7 \mathrm{H}), 4.93$
$(\mathrm{dt}, J=17.2,1.7 \mathrm{~Hz}, 1 \mathrm{H}), 4.80(\mathrm{dt}, J=10.5,1.6 \mathrm{~Hz}, 1 \mathrm{H}), 3.95(\mathrm{dt}, J=7.4,1.4 \mathrm{~Hz}, 1 \mathrm{H}), 3.52(\mathrm{dt}$, $J=7.8,1.3 \mathrm{~Hz}, 1 \mathrm{H}), 2.78-2.64(\mathrm{~m}, 3 \mathrm{H}), 2.60(\mathrm{dddt}, J=14.0,7.4,5.1,1.2 \mathrm{~Hz}, 1 \mathrm{H}), 0.13(\mathrm{~s}, 9 \mathrm{H})$, $0.04(\mathrm{~s}, 9 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta=139.5,137.8,137.7,135.7,135.2,133.8,133.7$, $131.5,130.4,128.8,128.7,127.7,127.3,125.6,125.5,125.4,125.29,125.28,125.1,124.3,123.7$, $116.7,116.3,113.6,112.2,78.8,76.6,73.4,42.4,41.3,-3.7,-3.9 . \operatorname{IR}\left(F T I R, \mathrm{~cm}^{-1}\right) \tilde{v}=3072,3050$, 2955, 2899, 1639, 1626, 1509, 1413, 1246, 909, 837, 775. HRMS (ESI), $m / z[\mathrm{M}+\mathrm{H}]^{+}$calcd for $\mathrm{C}_{20} \mathrm{H}_{27} \mathrm{OSi}$ : 311.1826; found: 311.1837.

Synthesis of syn/anti-(1-((1-([1,1'-biphenyl]-4-yl)but-3-en-1-yl)oxy)allyl)trimethylsilane (syn/anti-4.4e)


Compound 4.4e was prepared as per the following procedure: A mixture of syn/anti-4.4b ( $d r$ $=1: 1)(390 \mathrm{mg}, 1.3 \mathrm{mmol}, 1.0$ equiv), phenylboronic acid ( $240 \mathrm{mg}, 1.97 \mathrm{mmol}, 1.5$ equiv) and $\mathrm{K}_{3} \mathrm{PO}_{4} .2 \mathrm{H}_{2} \mathrm{O}$ ( 652 mg , 2.6 mmol , 2.0 equiv) in toluene ( 2 mL ) was degassed (3-freeze-pump-thaw actions) and then a solution of $\operatorname{Pd}(\mathrm{OAc})_{2}(3 \mathrm{mg}, 0.013 \mathrm{mmol}, 0.01$ equiv) and S-PHOS ( 10.8 mg , $0.026 \mathrm{mmol}, 0.02$ equiv) in THF ( 1 mL ) was added resulting In formation of red solution. The resulting mixture was heated in an oil bath at $100^{\circ} \mathrm{C}$ for 4.5 hours. The reaction mixture was concentrated and the residue subjected to column chromatography to give a total of $161 \mathrm{mg}, 0.478$
$\mathrm{mmol}\left(37 \%\right.$ isolated yield) of partially separable mixture of diastereomers. $R_{f}$ value for $\operatorname{syn}-\mathbf{4 . 4 e}=$ 0.36 and $R_{f}$ value for trans-4.4e $=0.40$.

Spectroscopic data for $\operatorname{syn}$-S4-e: ${ }^{1} \mathrm{H}$ NMR $\left(600 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta=7.62(\mathrm{dd}, J=8.3,1.1 \mathrm{~Hz}$, $2 \mathrm{H}), 7.56(\mathrm{~d}, J=8.2 \mathrm{~Hz}, 2 \mathrm{H}), 7.48-7.43(\mathrm{~m}, 2 \mathrm{H}), 7.39(\mathrm{~d}, J=8.1 \mathrm{~Hz}, 2 \mathrm{H}), 7.37-7.33(\mathrm{~m}, 1 \mathrm{H})$, $5.84-5.69(\mathrm{~m}, 2 \mathrm{H}), 5.09-5.01(\mathrm{~m}, 2 \mathrm{H}), 4.98(\mathrm{dt}, J=17.2,1.8 \mathrm{~Hz}, 1 \mathrm{H}), 4.90(\mathrm{dt}, J=10.6,1.7$ $\mathrm{Hz}, 1 \mathrm{H}), 4.47(\mathrm{t}, J=6.1 \mathrm{~Hz}, 1 \mathrm{H}), 3.87(\mathrm{dt}, J=7.1,1.5 \mathrm{~Hz}, 1 \mathrm{H}), 2.62-2.55(\mathrm{~m}, 1 \mathrm{H}), 2.55-2.47$ $(\mathrm{m}, 1 \mathrm{H}), 0.11(\mathrm{~s}, 9 \mathrm{H}) .{ }^{13} \mathrm{C}-\mathrm{NMR}\left(151 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta=142.7,141.1,139.7,137.9,134.8,128.7$, $127.0,127.0,127.0,126.6,117.0,112.0,80.6,75.7,41.4,-3.7$. Syn-4.4e is a known compound and spectroscopic data are in agreement with those reported in literature. ${ }^{26}$

Spectroscopic data for anti-4.4e: ${ }^{1} \mathrm{H}$ NMR $\left(600 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta=7.64(\mathrm{~d}, J=7.7 \mathrm{~Hz}, 2 \mathrm{H})$, $7.59(\mathrm{~d}, J=8.1 \mathrm{~Hz}, 2 \mathrm{H}), 7.46(\mathrm{t}, J=7.8 \mathrm{~Hz}, 2 \mathrm{H}), 7.39-7.32(\mathrm{~m}, 3 \mathrm{H}), 5.87(\mathrm{ddt}, J=17.2,10.2$, $7.0 \mathrm{~Hz}, 1 \mathrm{H}), 5.79$ (ddd, $J=17.5,10.6,7.4 \mathrm{~Hz}, 1 \mathrm{H}), 5.10-5.00(\mathrm{~m}, 4 \mathrm{H}), 4.52(\mathrm{dd}, J=7.8,5.7 \mathrm{~Hz}$, $1 \mathrm{H}), 3.52(\mathrm{~d}, J=7.4 \mathrm{~Hz}, 1 \mathrm{H}), 2.63-2.56(\mathrm{~m}, 1 \mathrm{H}), 2.45-2.38(\mathrm{~m}, 1 \mathrm{H}), 0.04(\mathrm{~s}, 9 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR $\left(151 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta=141.6,141.0,140.2,137.6,135.4,128.7,127.7,127.2,127.0,126.8,116.5$, 112.9, 78.9, 72.9, 43.07, -4.0. Trans-S4-e is a known compound and spectroscopic data are in agreement with those reported in literature. ${ }^{26}$

Synthesis of syn/anti-triethyl(1-((1-(naphthalen-2-yl)but-3-en-1-yl)oxy)allyl)silane (syn/anti-4.4f)


Compound 4.4 f was prepared following general procedure E with slight modification to minimize formation of the side product as a result of elimination. A solution of 1-(triethylsilyl)prop-2-en-1-ol 4.3 c ( $2.59 \mathrm{~g}, 15 \mathrm{mmol}, 1$ equiv.) and 1-(naphthalen-2-yl)but-3-en-1yl 2,2,2-trichloroacetimidate $4.2 \mathrm{f}(6.17 \mathrm{~g}, 18 \mathrm{mmol}, 1.2$ equiv.) in dichloromethane ( 120 mL ) was cooled to $-78{ }^{\circ} \mathrm{C}$. TMSOTf ( $0.28 \mathrm{~mL}, 1.5 \mathrm{mmol}, 0.1$ equiv.) was added dropwise and the mixture stirred at $-78^{\circ} \mathrm{C}$ for 6 hours. The rubber septum was removed and 7 g of sodium bicarbonate was poured into the flask. The dry ice-acetone bath was removed and the mixture was allowed to warm up to room temperature. The mixture was filtered and concentrated under reduced pressure to remove dichloromethane. Hexanes was then added to the resulting mixture resulting in the formation of white precipitate. Subsequent filtration and concentration furnished a residue which
was purified by column chromatography, $R_{f}$ for anti-4.4f $=0.6$ and $R_{f}$ for syn-4.4f $=0.4(10 \%$ DCM in hexanes) to afford a total of $2.48 \mathrm{~g}, 7.1 \mathrm{mmol}$ ( $47 \%$ isolated yield) of partially separable mixture of diastereomers of compound $\mathbf{4 . 4 f}$ (syn:anti=1:1) as colorless liquid

Spectroscopic data for syn-4.4f: ${ }^{1} \mathrm{H}$ NMR $\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta=7.86-7.79(\mathrm{~m}, 3 \mathrm{H}), 7.74(\mathrm{~s}$, $1 \mathrm{H}), 7.51-7.43(\mathrm{~m}, 3 \mathrm{H}), 5.80-5.68(\mathrm{~m}, 2 \mathrm{H}), 5.05-4.93(\mathrm{~m}, 3 \mathrm{H}), 4.82(\mathrm{dt}, J=10.5,1.6 \mathrm{~Hz}, 1 \mathrm{H})$, $4.55(\mathrm{t}, J=6.2 \mathrm{~Hz}, 1 \mathrm{H}), 4.08(\mathrm{dt}, J=7.4,1.5 \mathrm{~Hz}, 1 \mathrm{H}), 2.65(\mathrm{dddt}, J=14.4,7.3,6.0,1.3 \mathrm{~Hz}, 1 \mathrm{H})$, $2.58-2.48(\mathrm{~m}, 1 \mathrm{H}), 1.04(\mathrm{t}, J=7.9 \mathrm{~Hz}, 9 \mathrm{H}), 0.68(\mathrm{qd}, J=7.9,2.6 \mathrm{~Hz}, 6 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( 126 MHz, $\left.\mathrm{CDCl}_{3}\right) \delta=141.2,138.3,134.7,133.1,132.7,127.9,127.6,127.5,125.7,125.4,125.3,125.0$, $116.9,111.9,81.2,74.5,41.4,7.5,1.8$. Syn-4.4f is a known compound and spectroscopic data are in agreement with those reported in literature. ${ }^{26}$

Spectroscopic data for anti-4.4f: ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta=7.88-7.81(\mathrm{~m}, 3 \mathrm{H}), 7.67$ $(\mathrm{s}, 1 \mathrm{H}), 7.52-7.44(\mathrm{~m}, 3 \mathrm{H}), 5.91-5.78(\mathrm{~m}, 2 \mathrm{H}), 5.10-4.97(\mathrm{~m}, 4 \mathrm{H}), 4.63(\mathrm{dd}, J=7.8,5.7 \mathrm{~Hz}$, $1 \mathrm{H}), 3.62(\mathrm{dt}, J=7.7,1.4 \mathrm{~Hz}, 1 \mathrm{H}), 2.64(\mathrm{dddt}, J=14.1,8.0,6.8,1.3 \mathrm{~Hz}, 1 \mathrm{H}), 2.45(\mathrm{dddt}, J=14.2$, $7.1,5.7,1.2 \mathrm{~Hz}, 1 \mathrm{H}), 0.92(\mathrm{t}, J=7.9 \mathrm{~Hz}, 9 \mathrm{H}), 0.59(\mathrm{qd}, J=7.9,3.7 \mathrm{~Hz}, 6 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( 126 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta=139.7,137.9,135.4,133.10,133.05,128.0,127.8,127.7,126.6,125.9,125.6,125.1$, $116.5,112.9,79.1,71.5,42.9,7.4,1.6$. Anti-4.4c is a known compound and spectroscopic data are in agreement with those reported in literature. ${ }^{26}$

Synthesis of syn/anti-(1-((1-(4-chlorophenyl)but-3-en-1-yl)oxy)allyl) dimethyl(phenyl)silane (syn/anti-S4-g)


Compound 4.4c was prepared following general procedure E, a solution of 1-(dimethyl(phenyl)silyl)prop-2-en-1-ol $\quad 4.3 d \quad(2.40 \mathrm{~g}, \quad 13.5 \mathrm{mmol}, 1 \mathrm{equiv}$.$) and 1-(4-$ chlorophenyl)but-3-en-1-yl 2,2,2-trichloroacetimidate $\mathbf{4 . 2 b}(6.18 \mathrm{~g}, 18.9 \mathrm{mmol}, 1.4$ equiv.), $\mathrm{BF}_{3} \cdot \mathrm{OEt}_{2}(0.17 \mathrm{~mL}, 1.35 \mathrm{mmol}, 0.1$ equiv.) and hexanes $(80 \mathrm{~mL})$ for 12 hour followed by workup, concentration and column chromatography, $R_{f}$ for syn/anti-4.4g $=0.6(20 \% \mathrm{DCM}$ in hexanes $)$ afforded a total of $3.76 \mathrm{~g}, 10.5 \mathrm{mmol}$ ( $78 \%$ isolated yield) inseparable mixture of diastereomers of compound 4.4g (syn:anti $=1: 1$ ) as colorless liquid.

Spectroscopic data for syn/anti-4.4g: ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}, \mathrm{ppm}\right) \delta=7.64-7.58(\mathrm{~m}$, $2 \mathrm{H}), 7.53(\mathrm{dt}, J=6.5,1.6 \mathrm{~Hz}, 2 \mathrm{H}), 7.43-7.33(\mathrm{~m}, 6 \mathrm{H}), 7.26(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 2 \mathrm{H}), 7.21-7.16(\mathrm{~m}$, $4 \mathrm{H}), 6.96(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 2 \mathrm{H}), 5.82-5.53(\mathrm{~m}, 5 \mathrm{H}), 5.05(\mathrm{dt}, J=10.6,1.4 \mathrm{~Hz}, 1 \mathrm{H}), 5.03-4.88(\mathrm{~m}$, $6 \mathrm{H}), 4.86(\mathrm{dt}, J=10.6,1.7 \mathrm{~Hz}, 1 \mathrm{H}), 4.43(\mathrm{dd}, J=7.6,5.8 \mathrm{~Hz}, 1 \mathrm{H}), 4.28(\mathrm{t}, J=6.1 \mathrm{~Hz}, 1 \mathrm{H}), 3.99$
$(\mathrm{dt}, J=7.0,1.5 \mathrm{~Hz}, 1 \mathrm{H}), 3.58(\mathrm{dt}, J=7.4,1.3 \mathrm{~Hz}, 1 \mathrm{H}), 2.54-2.41(\mathrm{~m}, 2 \mathrm{H}), 2.39-2.28(\mathrm{~m}, 2 \mathrm{H})$, $0.39(\mathrm{~s}, 3 \mathrm{H}), 0.35(\mathrm{~s}, 3 \mathrm{H}), 0.32(\mathrm{~s}, 3 \mathrm{H}), 0.28(\mathrm{~s}, 3 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR $\left(126 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta=141.8,140.5$, $137.2,136.8,136.7,136.6,134.8,134.4,134.3,134.2,132.8,132.5,129.23,129.15,128.6,128.21$, $128.00,127.95,127.6,127.5,117.2,116.8,113.7,112.6,80.4,78.5,75.4,72.7,42.9,41.3,-5.28$, $-5.31,-5.7,-6.3 .4 .4 \mathbf{g}$ is a known compound and spectroscopic data are in agreement with those reported in literature. ${ }^{26}$

## Synthesis of syn/anti-(1-((1-(4-methoxyphenyl)but-3-en-1-yl)oxy)-2-methylallyl)trimethyl silane (syn/anti-4.4h)



Compound 4.4 h was prepared following general procedure E, a solution of 2-methyl-1-(trimethylsilyl)prop-2-en-1-ol $\mathbf{4 . 3 b}$ ( $1.44 \mathrm{~g}, 10 \mathrm{mmol}, 1$ equiv.) and 1-(4-methoxyphenyl)but-3-en-1-yl 2,2,2-trichloroacetimidate $4.2 \mathrm{~h}\left(4.84 \mathrm{~g}, 15 \mathrm{mmol}, 1.5\right.$ equiv.), $\mathrm{BF}_{3} \cdot \mathrm{OEt}_{2}(0.12 \mathrm{~mL}, 1.0$ mmol, 0.1 equiv.) and hexanes ( 80 mL ) for 12 hour followed by workup, concentration and column chromatography, $R_{f}$ for anti-4.4h $=0.7$ and $R_{f}$ for $\operatorname{syn}-\mathbf{4 . 4 h}=0.6(15 \% \mathrm{DCM}$ in hexanes $)$ afforded
a total of $1.15 \mathrm{~g}, 3.8 \mathrm{mmol}(38 \%$ isolated yield) partially separable mixture of diastereomers of compound $\mathbf{4 . 4 h}$ as colorless liquid.

Spectroscopic data for syn-4.4h: ${ }^{1} \mathrm{H}$ NMR $\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta=7.22(\mathrm{~d}, J=8.5 \mathrm{~Hz}, 2 \mathrm{H})$, $6.84(\mathrm{~d}, J=8.7 \mathrm{~Hz}, 2 \mathrm{H}), 5.68(\mathrm{ddt}, J=17.3,10.3,7.0 \mathrm{~Hz}, 1 \mathrm{H}), 5.01-4.94(\mathrm{~m}, 2 \mathrm{H}), 4.68(\mathrm{ddt}, J=$ 7.9, 2.2, 1.1 Hz, 2H), $4.28(\mathrm{t}, J=6.0 \mathrm{~Hz}, 1 \mathrm{H}), 3.80(\mathrm{~s}, 3 \mathrm{H}), 3.76(\mathrm{~s}, 1 \mathrm{H}), 2.59-2.49(\mathrm{~m}, 1 \mathrm{H}), 2.49$ - $2.41(\mathrm{~m}, 1 \mathrm{H}), 1.53(\mathrm{t}, J=1.1 \mathrm{~Hz}, 3 \mathrm{H}), 0.08(\mathrm{~s}, 9 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR $\left(126 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta=158.4$, $145.1,135.5,134.8,127.7,116.8,113.1,109.3,79.7,77.6,55.1,40.5,20.3,-3.0$. Syn-4.4h is a known compound and spectroscopic data are in agreement with those reported in literature. ${ }^{26}$

Spectroscopic data for anti-4.4h: ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta=7.17(\mathrm{~d}, J=8.6 \mathrm{~Hz}, 2 \mathrm{H})$, $6.87(\mathrm{~d}, J=8.7 \mathrm{~Hz}, 2 \mathrm{H}), 5.80(\mathrm{ddt}, J=17.3,10.2,7.1 \mathrm{~Hz}, 1 \mathrm{H}), 5.03-4.94(\mathrm{~m}, 2 \mathrm{H}), 4.81(\mathrm{dq}, J=$ $2.8,1.5 \mathrm{~Hz}, 1 \mathrm{H}), 4.68(\mathrm{dt}, J=2.2,1.0 \mathrm{~Hz}, 1 \mathrm{H}), 4.25(\mathrm{dd}, J=7.7,6.0 \mathrm{~Hz}, 1 \mathrm{H}), 3.82(\mathrm{~s}, 3 \mathrm{H}), 3.31$ $(\mathrm{s}, 1 \mathrm{H}), 2.54(\mathrm{dddt}, J=13.8,7.9,6.8,1.2 \mathrm{~Hz}, 1 \mathrm{H}), 2.34(\mathrm{dddt}, J=13.7,7.3,6.1,1.2 \mathrm{~Hz}, 1 \mathrm{H}), 1.68$ $-1.61(\mathrm{~m}, 3 \mathrm{H}),-0.01(\mathrm{~s}, 8 \mathrm{H}) .{ }^{13} \mathrm{C} \operatorname{NMR}\left(126 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta=158.9,144.4,135.7,134.3,128.6$, $116.3,113.4,109.8,78.4,75.0,55.1,43.0,20.4,-3.2$. Anti-4.4h is a known compound and spectroscopic data are in agreement with those reported in literature. ${ }^{26}$

Synthesis of syn/anti-(1-((1-(4-chlorophenyl)but-3-en-1-yl)oxy)-2-methylallyl)trimethyl silane (syn/anti-S4-i)


Compound 4.4 i was prepared following general procedure E , a solution of 2-methyl-1-(trimethylsilyl)prop-2-en-1-ol 4.3b (1.44 g, $10 \mathrm{mmol}, 1$ equiv.) and 1-(4-chlorophenyl)but-3-en-1-yl 2,2,2-trichloroacetimidate $\mathbf{4 . 2 b}\left(4.58 \mathrm{~g}, 14 \mathrm{mmol}, 1.4\right.$ equiv.), $\mathrm{BF}_{3} \cdot \mathrm{OEt}_{2}(0.12 \mathrm{~mL}, 1.0 \mathrm{mmol}$, 0.1 equiv.) and hexanes ( 60 mL ) for 12 hour followed by workup, concentration and column chromatography, $R_{f}$ for anti/syn- $\mathbf{4 . 4 i}=0.4(100 \%$ hexanes) afforded a total of $1.80 \mathrm{~g}, 5.8 \mathrm{mmol}$ ( $58 \%$ isolated yield) inseparable mixture of diastereomers of compound $\mathbf{4 . 4 i}$ as colorless liquid. Spectroscopic data for syn/anti-4.4i: ${ }^{1} \mathrm{H}$ NMR $\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta=7.32(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 2 \mathrm{H})$, $7.28(\mathrm{~d}, J=8.6 \mathrm{~Hz}, 2 \mathrm{H}), 7.24(\mathrm{~d}, J=8.5 \mathrm{~Hz}, 2 \mathrm{H}), 7.20(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 2 \mathrm{H}), 5.79(\mathrm{ddtd}, J=14.3$, $10.6,7.1,2.4 \mathrm{~Hz}, 1 \mathrm{H}), 5.66(\mathrm{ddt}, J=16.5,10.7,7.1 \mathrm{~Hz}, 1 \mathrm{H}), 5.19-5.12(\mathrm{~m}, 1 \mathrm{H}), 5.03-4.95(\mathrm{~m}$, $4 \mathrm{H}), 4.84(\mathrm{dq}, J=2.8,1.5 \mathrm{~Hz}, 1 \mathrm{H}), 4.72(\mathrm{qd}, J=1.6,0.9 \mathrm{~Hz}, 1 \mathrm{H}), 4.68(\mathrm{ddq}, J=4.2,2.1,0.9 \mathrm{~Hz}$, $2 \mathrm{H}), 4.36-4.27(\mathrm{~m}, 2 \mathrm{H}), 3.79(\mathrm{~s}, 1 \mathrm{H}), 3.30(\mathrm{~s}, 1 \mathrm{H}), 2.57-2.50(\mathrm{~m}, 2 \mathrm{H}), 2.50-2.42(\mathrm{~m}, 1 \mathrm{H}), 2.35$
(dddt, $J=13.8,7.3,6.1,1.2 \mathrm{~Hz}, 1 \mathrm{H}), 1.66(\mathrm{~s}, 3 \mathrm{H}), 1.54(\mathrm{~s}, 3 \mathrm{H}), 0.10(\mathrm{~s}, 9 \mathrm{H}), 0.02(\mathrm{~s}, 9 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR (126 MHz, $\left.\mathrm{CDCl}_{3}\right) \delta=144.7,144.1,141.8,140.8,135.0,134.0,133.0,132.4,128.9,128.3$, 128.0, 127.9, 117.3, 116.8, 110.1, 109.7, 79.2, 78.3, 77.9, 75.6, 42.9, 40.3, 20.4, 20.3, -3.0, -3.2. MS (GC/MS): $m / \mathrm{z}(\%)=308(0.02)[\mathrm{M}]^{+}, 143(33), 141(100), 113(25), 77(65)$.

Synthesis of cis/trans-trimethyl(3-methyl-6-phenyl-5,6-dihydro-2H-pyran-2-yl)silane (cis/trans-4.5a)


Compound 4.5a was prepared following general procedure F: Grubbs catalyst $2^{\text {nd }}$ generation ( $37 \mathrm{mg}, 0.043 \mathrm{mmol}, 0.04$ equiv.) and syn/anti-trimethyl(2-methyl-1-((1-phenylbut-3-en-1yl)oxy)allyl)silane, syn:anti $=1: 2,4.4 \mathbf{a}(297 \mathrm{mg}, 1.08 \mathrm{mmol}, 1$ equiv.) and dichloromethane (11 mL ) for 5 hours followed by concentration and column chromatography, $R_{f}$ for cis-4.5a $=0.7$ and $R_{f}$ for trans-4.5b $=0.4$ ( $10 \%$ and $30 \% \mathrm{DCM}$ in hexanes) afforded a total of $248 \mathrm{mg}, 1.00 \mathrm{mmol}$ ( $93 \%$ isolated yield) fully separable mixture of diastereomers of compound 4.5a as colorless liquid.

Spectroscopic data for cis-4.5a: ${ }^{1} \mathrm{H}$ NMR ( $\left.500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta=7.40-7.31(\mathrm{~m}, 4 \mathrm{H}), 7.29-$ $7.24(\mathrm{~m}, 1 \mathrm{H}), 5.52-5.47(\mathrm{~m}, 1 \mathrm{H}), 4.56(\mathrm{dd}, J=8.0,5.2 \mathrm{~Hz}, 1 \mathrm{H}), 4.06-4.01(\mathrm{~m}, 1 \mathrm{H}), 2.40-2.27$ $(\mathrm{m}, 2 \mathrm{H}), 1.66(\mathrm{dt}, J=2.6,1.3 \mathrm{~Hz}, 3 \mathrm{H}), 0.16(\mathrm{~s}, 9 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR $\left(126 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta=144.0,135.3$, $128.1,126.8,125.6,117.0,74.9,74.3,34.2,20.0,-2.6$. Cis-4.5a is a known compound and spectroscopic data are in agreement with those reported in literature. ${ }^{26}$

Spectroscopic data for trans-4.5a: ${ }^{1} \mathrm{H}$ NMR $\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta=7.38-7.27(\mathrm{~m}, 5 \mathrm{H}), 5.52$ $-5.45(\mathrm{~m}, 1 \mathrm{H}), 4.55(\mathrm{dd}, J=8.0,5.3 \mathrm{~Hz}, 1 \mathrm{H}), 4.03(\mathrm{~d}, J=2.8 \mathrm{~Hz}, 1 \mathrm{H}), 2.36-2.30(\mathrm{~m}, 2 \mathrm{H}), 1.67$
$-1.64(\mathrm{~m}, 3 \mathrm{H}), 0.16(\mathrm{~s}, 9 \mathrm{H}) .{ }^{13} \mathrm{C} \operatorname{NMR}\left(126 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta=142.7,135.4,128.2,127.2,126.0$, 115.8, 75.0, 72.9, 32.1, 20.4, -1.4. HRMS (ESI), $m / z\left[\mathrm{M}-\mathrm{H}^{-}\right]^{+}$calcd for $\mathrm{C}_{15} \mathrm{H}_{21} \mathrm{OSi}$ : 245.1362; found: 245.1353 .

Synthesis of cis/trans-(6-(4-chlorophenyl)-5,6-dihydro-2H-pyran-2-yl)trimethylsilane (cis/trans-4.5b)


Compound $\mathbf{4 . 5 b}$ was prepared following general procedure F : Grubbs catalyst $2^{\text {nd }}$ generation $(170 \mathrm{mg}, \quad 0.2 \mathrm{mmol}, \quad 0.04$ equiv. $)$ and syn/anti-(1-((1-(4-chlorophenyl)but-3-en-1yl)oxy)allyl)trimethylsilane, syn:anti $=1: 1,4.4 \mathrm{~b}(1.47 \mathrm{~g}, 5 \mathrm{mmol}, 1$ equiv.) and dichloromethane $(100 \mathrm{~mL})$ for 12 hours followed by concentration and column chromatography, $R_{f}$ for cis- $\mathbf{4 . 5 b}=$ 0.7 and $R_{f}$ for trans-4.5b $=0.4(10 \%$ and $30 \%$ DCM in hexanes) afforded a total of $1.16 \mathrm{~g}, 4.35$ $\mathrm{mmol}(87 \%$ isolated yield) fully separable mixture of diastereomers of compound $\mathbf{4 . 5 b}$ as colorless liquid.

Spectroscopic data for cis-4.5b: ${ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta=7.36-7.27(\mathrm{~m}, 4 \mathrm{H}), 5.85-$ $5.77(\mathrm{~m}, 2 \mathrm{H}), 4.38(\mathrm{dd}, J=10.1,3.1 \mathrm{~Hz}, 1 \mathrm{H}), 4.18(\mathrm{dddd}, J=5.0,3.6,2.2,1.5 \mathrm{~Hz}, 1 \mathrm{H}), 2.26-$ $2.19(\mathrm{~m}, 1 \mathrm{H}), 2.17-2.08(\mathrm{~m}, 1 \mathrm{H}), 0.10(\mathrm{~s}, 9 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $\left.126 \mathrm{MHz}, \mathrm{CDCl}_{3},\right) \delta=142.5,132.6$, 128.3, 128.1, 127.0, 120.4, 74.7, 71.7, 34.0, -4.0. Cis-4.5b is a known compound and spectroscopic data are in agreement with those reported in literature. ${ }^{26}$

Spectroscopic data for trans-4.5b: ${ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta=7.33-7.29(\mathrm{~m}, 4 \mathrm{H}), 5.85$ $-5.76(\mathrm{~m}, 2 \mathrm{H}), 4.72(\mathrm{dd}, J=6.4,4.5 \mathrm{~Hz}, 1 \mathrm{H}), 4.01-3.97(\mathrm{~m}, 1 \mathrm{H}), 2.47-2.40(\mathrm{~m}, 1 \mathrm{H}), 2.39-$
$2.32(\mathrm{~m}, 1 \mathrm{H}), 0.10(\mathrm{~s}, 9 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $\left.126 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta=140.5,132.9,128.3,128.2,128.0$, 119.7, 71.6, 69.7, 29.9, -3.1. Trans-4.5b is a known compound and spectroscopic data are in agreement with those reported in literature. ${ }^{26}$

## Synthesis of cis/trans-(6-(4-trifluoromethylphenyl)-5,6-dihydro-2H-pyran-2-yl)trimethyl silane (cis/trans-4.5c)



Compound 4.5 c was prepared following general procedure F : Grubbs catalyst $2^{\text {nd }}$ generation $(204 \mathrm{mg}, \quad 0.24 \mathrm{mmol}, \quad 0.04$ equiv.) and syn/anti-(1-((1-(4-trifluorophenyl)but-3-en-1yl)oxy)allyl)trimethylsilane, syn:anti $=1: 1,4.4 \mathrm{c}(1.97 \mathrm{~g}, 6 \mathrm{mmol}, 1$ equiv.) and dichloromethane $(100 \mathrm{~mL})$ for 12 hours followed by concentration and column chromatography, $R_{f}$ for cis-4.5c $=$ 0.7 and $R_{f}$ for trans-4.5c $=0.4(10 \%$ and $30 \% \mathrm{DCM}$ in hexanes) afforded a total of $1.74 \mathrm{~g}, 1.89$ mmol ( $96 \%$ isolated yield) fully separable mixture of diastereomers of compound $\mathbf{4 . 5 \mathrm { c }}$ as colorless liquids. Spectroscopic data for cis-4.5c: ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta=7.60(\mathrm{~d}, J=8.1 \mathrm{~Hz}, 2 \mathrm{H})$, $7.48(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 2 \mathrm{H}), 5.89-5.78(\mathrm{~m}, 2 \mathrm{H}), 4.47(\mathrm{dd}, J=10.3,3.1 \mathrm{~Hz}, 1 \mathrm{H}), 4.20(\mathrm{dddd}, J=5.0$, $3.5,2.3,1.6 \mathrm{~Hz}, 1 \mathrm{H}), 2.31-2.24(\mathrm{~m}, 1 \mathrm{H}), 2.19-2.11(\mathrm{~m}, 1 \mathrm{H}), 0.12(\mathrm{~s}, 9 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( 126 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta=148.0(\mathrm{t}, J=1.4 \mathrm{~Hz}), 129.2(\mathrm{q}, J=32.1 \mathrm{~Hz}), 128.1,125.9,125.1(\mathrm{q}, J=3.8 \mathrm{~Hz})$, 120.7,74.8, 71.7, 34.0, -4.0. Cis-4.5c is a known compound and spectroscopic data are in agreement with those reported in literature. ${ }^{26}$

Spectroscopic data for trans-4.5c: ${ }^{1} \mathrm{H} \operatorname{NMR}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta=7.61(\mathrm{~d}, J=8.1 \mathrm{~Hz}, 2 \mathrm{H})$, $7.52(\mathrm{~d}, J=8.1 \mathrm{~Hz}, 2 \mathrm{H}), 5.87-5.78(\mathrm{~m}, 2 \mathrm{H}), 4.80(\mathrm{dd}, J=6.6,4.5 \mathrm{~Hz}, 1 \mathrm{H}), 4.04(\mathrm{dddd}, J=3.4$,
2.6, 1.8, $0.6 \mathrm{~Hz}, 1 \mathrm{H}), 2.52-2.45(\mathrm{~m}, 1 \mathrm{H}), 2.43-2.35(\mathrm{~m}, 1 \mathrm{H}), 0.12(\mathrm{~s}, 9 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( 126 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta=146.2(\mathrm{~d}, J=1.5 \mathrm{~Hz}), 129.4,(\mathrm{q}, J=32.2 \mathrm{~Hz}), 128.3,126.8,125.15(\mathrm{q}, J=3.8 \mathrm{~Hz})$, 119.6,71.7, 70.0, 30.1, -3.1. Trans-4.5c is a known compound and spectroscopic data are in agreement with those reported in literature. ${ }^{26}$

Synthesis of cis/trans-trimethyl(6-(naphthalen-1-yl)-5,6-dihydro-2H-pyran-2-yl)silane (cis/trans-4.5d)


Compound 4.5 d was prepared following general procedure F : Grubbs catalyst $2^{\text {nd }}$ generation (204 mg, $0.24 \mathrm{mmol}, 0.04$ equiv.) and syn/anti-trimethyl(1-((1-(naphthalen-1-yl)but-3-en-1yl)oxy)allyl)silane, syn:anti $=1: 1,4.4 \mathrm{~d}\left(1.9 \mathrm{~g}, 6 \mathrm{mmol}, 1\right.$ equiv.) and benzene ( 100 mL ) at $85^{\circ} \mathrm{C}$ for 1 hour followed by concentration and column chromatography, $R_{f}$ for cis-4.5d $=0.7$ and $R_{f}$ for trans-4.5d $=0.4$ (20\% DCM in hexanes) afforded a total of $1.66 \mathrm{~g}, 5.88 \mathrm{mmol}$ ( $98 \%$ isolated yield $)$ fully separable mixture of diastereomers of compound $\mathbf{4 . 5 d}$ as colorless liquid.

Spectroscopic data for cis-4.5d: ${ }^{1} \mathrm{H}$ NMR $\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta=78.11(\mathrm{~d}, J=8.9 \mathrm{~Hz}, 1 \mathrm{H})$, $7.88(\mathrm{~d}, J=9.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.79(\mathrm{~d}, J=8.2 \mathrm{~Hz}, 1 \mathrm{H}), 7.67(\mathrm{~d}, J=7.1 \mathrm{~Hz}, 1 \mathrm{H}), 7.50(\mathrm{tddd}, J=10.2$, $8.4,3.8,2.3 \mathrm{~Hz}, 3 \mathrm{H}), 5.97-5.85(\mathrm{~m}, 2 \mathrm{H}), 5.13(\mathrm{dd}, J=9.9,3.3 \mathrm{~Hz}, 1 \mathrm{H}), 4.33(\mathrm{tdd}, J=4.0,2.3$, $1.2 \mathrm{~Hz}, 1 \mathrm{H}), 2.52-2.36(\mathrm{~m}, 2 \mathrm{H}), 0.17(\mathrm{~s}, 9 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR (126 MHz, $\left.\mathrm{CDCl}_{3}\right) \delta=139.6,133.6$, $130.6,128.7,128.0,127.5,125.59,125.56,125.2,123.7,123.0,121.5,73.5,72.0,33.4,-3.9$. IR (FTIR, $\left.\mathrm{cm}^{-1}\right) \tilde{\mathrm{v}}=3049,3027,2954,2897,1510,1385,1336,1245,1061,837,773 . \mathrm{MS}(\mathrm{GC} / \mathrm{MS}):$ $m / \mathrm{z}(\%)=282(5)[\mathrm{M}]^{+}, 281(7.5)[\mathrm{M}-\mathrm{H}]^{+}, 192(32), 191(21), 141$ (17.5), 73 (100). HRMS
(ESI), $m / z\left[\mathrm{M}-\mathrm{H}^{-}\right]^{+}$calcd for $\mathrm{C}_{18} \mathrm{H}_{21} \mathrm{OSi}$ 281.1362; found: 281.1360.
Spectroscopic data for trans-4.5d: ${ }^{1} \mathrm{H}$ NMR $\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta=8.33(\mathrm{~d}, J=8.3 \mathrm{~Hz}, 1 \mathrm{H})$, $7.89(\mathrm{~d}, J=7.8 \mathrm{~Hz}, 1 \mathrm{H}), 7.81(\mathrm{~d}, J=8.2 \mathrm{~Hz}, 1 \mathrm{H}), 7.58(\mathrm{~d}, J=7.1 \mathrm{~Hz}, 1 \mathrm{H}), 7.56-7.48(\mathrm{~m}, 2 \mathrm{H})$, $7.48-7.43(\mathrm{~m}, 1 \mathrm{H}), 5.95(\mathrm{dq}, J=10.6,3.8 \mathrm{~Hz}, 1 \mathrm{H}), 5.84(\mathrm{dq}, J=10.3,2.0 \mathrm{~Hz}, 1 \mathrm{H}), 5.55(\mathrm{t}, J=$ $5.1 \mathrm{~Hz}, 1 \mathrm{H}), 3.78(\mathrm{p}, J=3.2 \mathrm{~Hz}, 1 \mathrm{H}), 2.72(\mathrm{ddtd}, J=17.6,5.3,3.5,1.9 \mathrm{~Hz}, 1 \mathrm{H}), 2.61-2.52(\mathrm{~m}$, 1H), $0.09(\mathrm{~s}, 9 \mathrm{H}) .{ }^{13} \mathrm{C} \mathrm{NMR}\left(126 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta=136.7,133.9,131.7,128.6,128.3,128.1,125.8$, $125.4,125.0,124.5,124.4,120.4,70.0,68.3,29.7,-3.3 . \operatorname{IR}\left(\right.$ FTIR, $\left.\mathrm{cm}^{-1}\right) \tilde{\mathrm{v}}=3050,3027,2953$, 2898, 1509, 1336, 1245, 1048, 838, 774. HRMS (ESI), $m / z[\mathrm{M}+\mathrm{H}]^{+}$calcd for $\mathrm{C}_{18} \mathrm{H}_{23} \mathrm{OSi}$ : 283.1518; found: 283.1518 .

## Synthesis of cis/trans-(6-([1,1'-biphenyl]-4-yl)-5,6-dihydro-2H-pyran-2-yl)trimethylsilane (cis/trans-4.5e)



Compound 4.5 e was prepared following general procedure F : Grubbs catalyst $2^{\text {nd }}$ generation $(16 \mathrm{mg}, 0.02 \mathrm{mmol}, 0.04$ equiv.) and syn/anti-(1-((1-([1,1'-biphenyl]-4-yl)but-3-en-1yl)oxy)allyl)trimethylsilane, syn:anti $=2: 1,4.4 \mathrm{e}\left(161 \mathrm{mg}, 0.48 \mathrm{mmol}\right.$, 1 equiv.) and $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ (5 mL ) at room temperature for 3 hours followed by concentration and column chromatography, $R_{f}$ for cis-4.5e $=0.7$ and $R_{f}$ for trans-4.5e $=0.4(25 \% \mathrm{DCM}$ in hexanes) afforded a total of 92 mg , 0.30 mmol ( $63 \%$ isolated yield) fully separable mixture of diastereomers of compound $\mathbf{4 . 5 e}$ as colorless liquid.

Spectroscopic data for cis-4.5e: ${ }^{1} \mathrm{H}$ NMR $\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta=7.65-7.58(\mathrm{~m}, 4 \mathrm{H}), 7.50-$ $7.43(\mathrm{~m}, 4 \mathrm{H}), 7.39-7.34(\mathrm{~m}, 1 \mathrm{H}), 5.94-5.80(\mathrm{~m}, 2 \mathrm{H}), 4.49(\mathrm{dd}, J=9.8,3.4 \mathrm{~Hz}, 1 \mathrm{H}), 4.24(\mathrm{tdd}$,
$J=3.9,2.2,1.2 \mathrm{~Hz}, 1 \mathrm{H}), 2.36-2.22(\mathrm{~m}, 2 \mathrm{H}), 0.16(\mathrm{~s}, 9 \mathrm{H}) .{ }^{13} \mathrm{C} \operatorname{NMR}\left(126 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta=$ 143.1, 141.2, 139.9, 128.7, 128.1, 127.1, 127.1, 126.9, 126.1, 121.1, 75.2, 71.7, 34.1, -4.0. Cis4.5e is a known compound and spectroscopic data are in agreement with those reported in literature. ${ }^{26}$

Spectroscopic data for trans-4.5e: ${ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta=7.62-7.57(\mathrm{~m}, 4 \mathrm{H}), 7.49$ $-7.46(\mathrm{~m}, 2 \mathrm{H}), 7.44$ (dddd, $J=7.4,6.1,1.3,0.6 \mathrm{~Hz}, 2 \mathrm{H}), 7.37-7.32(\mathrm{~m}, 1 \mathrm{H}), 5.89-5.79(\mathrm{~m}$, $2 \mathrm{H}), 4.81(\mathrm{t}, J=5.6 \mathrm{~Hz}, 1 \mathrm{H}), 4.08(\mathrm{q}, J=2.9 \mathrm{~Hz}, 1 \mathrm{H}), 2.54-2.42(\mathrm{~m}, 2 \mathrm{H}), 0.14(\mathrm{~s}, 9 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR $\left(126 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta=141.2,141.0,140.1,128.7,128.2,127.2,127.09,127.05,127.0,120.0$, 72.1, 70.0, 30.2, -2.9. Trans-4.5e is a known compound and spectroscopic data are in agreement with those reported in literature. ${ }^{26}$

Synthesis of cis/trans-triethyl(6-(naphthalen-2-yl)-5,6-dihydro-2H-pyran-2-yl)silane (cis/trans-4.5f)


Compound 4.5f was prepared following general procedure F: Grubbs catalyst $2^{\text {nd }}$ generation (102 mg, $0.12 \mathrm{mmol}, 0.04$ equiv.) and syn/anti-triethyl(1-((1-(naphthalen-2-yl)but-3-en-1yl)oxy)allyl)silane, syn:anti $=1: 1,4.4 \mathrm{f}\left(1.06 \mathrm{~g}, 3 \mathrm{mmol}\right.$, 1 equiv.) and benzene ( 100 mL ) at $85^{\circ} \mathrm{C}$ for 1 hour followed by concentration and column chromatography, $R_{f}$ for cis $\mathbf{- 4 . 5 f}=0.7$ and $R_{f}$ for trans-4.5f $=0.4(10 \%$ and $20 \% \mathrm{DCM}$ in hexanes $)$ afforded a total of $883 \mathrm{mg}, 2.52 \mathrm{mmol}(84 \%$ isolated yield) fully separable mixture of diastereomers of compound $\mathbf{4 . 5 f}$ as colorless liquid.

Spectroscopic data for cis-4.5f: ${ }^{1} \mathrm{H}$ NMR $\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta=7.88-7.82(\mathrm{~m}, 3 \mathrm{H}), 7.81(\mathrm{~s}$,
$1 \mathrm{H}), 7.51(\mathrm{dd}, J=8.5,1.7 \mathrm{~Hz}, 1 \mathrm{H}), 7.49-7.42(\mathrm{~m}, 2 \mathrm{H}), 5.94-5.87(\mathrm{~m}, 1 \mathrm{H}), 5.82(\mathrm{ddt}, J=10.4$, $5.3,2.6 \mathrm{~Hz}, 1 \mathrm{H}), 4.57(\mathrm{dd}, J=9.0,4.3 \mathrm{~Hz}, 1 \mathrm{H}), 4.41(\mathrm{tdd}, J=4.3,3.1,2.2 \mathrm{~Hz}, 1 \mathrm{H}), 2.40-2.25$ $(\mathrm{m}, 2 \mathrm{H}), 1.06(\mathrm{t}, J=8.0 \mathrm{~Hz}, 9 \mathrm{H}), 0.72(\mathrm{q}, J=7.8 \mathrm{~Hz}, 6 \mathrm{H}) .{ }^{13} \mathrm{C} \operatorname{NMR}\left(126 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta=141.5$, $133.2,132.7,128.7,128.0,127.8,127.6,125.8,125.5,124.3,124.1,120.7,75.9,70.3,34.0,7.6$, 1.9. Cis-4.5f is a known compound and spectroscopic data are in agreement with those reported in literature. ${ }^{26}$

Spectroscopic data for trans-4.5f: ${ }^{1} \mathrm{H}$ NMR $\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta=7.87-7.83(\mathrm{~m}, 3 \mathrm{H}), 7.82$ $(\mathrm{s}, 1 \mathrm{H}), 7.56(\mathrm{dd}, J=8.5,1.8 \mathrm{~Hz}, 1 \mathrm{H}), 7.51-7.44(\mathrm{~m}, 2 \mathrm{H}), 5.88-5.79(\mathrm{~m}, 2 \mathrm{H}), 4.94(\mathrm{t}, J=5.3$ $\mathrm{Hz}, 1 \mathrm{H}), 4.18(\mathrm{td}, J=3.6,1.8 \mathrm{~Hz}, 1 \mathrm{H}), 2.63-2.48(\mathrm{~m}, 2 \mathrm{H}), 1.01(\mathrm{t}, J=7.9 \mathrm{~Hz}, 9 \mathrm{H}), 0.68(\mathrm{qd}, J=$ 8.0, 1.2 Hz, 6H). ${ }^{13} \mathrm{C}$ NMR $\left(126 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta=139.4,133.2,132.7,129.0,128.1,127.8,127.6$, $125.8,125.6,125.3,125.1,119.4,72.4,67.6,29.9,7.5,2.5$. Trans-4.5f is a known compound and spectroscopic data are in agreement with those reported in literature. ${ }^{26}$

Synthesis of cis/trans- (6-(4-chlorophenyl)-5,6-dihydro-2H-pyran-2-yl)dimethyl(phenyl) silane (cis/trans-4.5g)


Compound 4.5 g was prepared following general procedure F : Grubbs catalyst $2^{\text {nd }}$ generation $(200 \mathrm{mg}, \quad 0.2 \mathrm{mmol}, \quad 0.026$ equiv. $)$ and syn/anti-(1-((1-(4-chlorophenyl)but-3-en-1yl)oxy)allyl)dimethyl(phenyl)silane, syn:anti=1:1, 4.4g (3.21 g, $9 \mathrm{mmol}, 1$ equiv.) and benzene $(200 \mathrm{~mL})$ at $85^{\circ} \mathrm{C}$ for 2 hours followed by concentration and column chromatography, $R_{f}$ for cis$\mathbf{4 . 5 g}=0.7$ and $R_{f}$ for trans-4.5g $=0.4(30 \%$ DCM in hexanes) afforded a total of $2.96 \mathrm{~g}, 8.28 \mathrm{mmol}$ ( $92 \%$ isolated yield) fully separable mixture of diastereomers of compound $\mathbf{4 . 5 g}$ as colorless
liquids.
Spectroscopic data for cis-4.5g: ${ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta=7.67-7.61(\mathrm{~m}, 2 \mathrm{H}), 7.44-$ $7.36(\mathrm{~m}, 3 \mathrm{H}), 7.36-7.32(\mathrm{~m}, 2 \mathrm{H}), 7.32-7.29(\mathrm{~m}, 2 \mathrm{H}), 5.81(\mathrm{t}, J=1.9 \mathrm{~Hz}, 2 \mathrm{H}), 4.45-4.40(\mathrm{~m}$, $2 \mathrm{H}), 2.28-2.21(\mathrm{~m}, 1 \mathrm{H}), 2.18-2.10(\mathrm{~m}, 1 \mathrm{H}), 0.43(\mathrm{~s}, 3 \mathrm{H}), 0.42(\mathrm{~s}, 3 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( 126 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta=142.4,136.5,134.2,132.6,129.3,128.3,127.74,127.71,127.0,121.3,74.9,71.3$, 33.9, $-5.2,-6.0 . \mathrm{MS}(\mathrm{GC} / \mathrm{MS}): m / \mathrm{z}(\%)=328(1.5)[\mathrm{M}]^{+}, 250(4), 135$ (100), 107 (7), 75 (18). HRMS (ESI), $m / z\left[\mathrm{M}-\mathrm{H}^{-}\right]^{+}$calcd for $\mathrm{C}_{19} \mathrm{H}_{20}$ ClOSi: 327.0972; found: 327.0963 .

Spectroscopic data for trans-4.5g: ${ }^{1} \mathrm{H}$ NMR $\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta=7.63-7.56(\mathrm{~m}, 2 \mathrm{H}), 7.45$ - $7.36(\mathrm{~m}, 3 \mathrm{H}), 7.32(\mathrm{~d}, J=8.5 \mathrm{~Hz}, 2 \mathrm{H}), 7.26(\mathrm{~d}, J=8.5 \mathrm{~Hz}, 2 \mathrm{H}), 5.87-5.78(\mathrm{~m}, 2 \mathrm{H}), 4.63(\mathrm{t}, J$ $=5.6 \mathrm{~Hz}, 1 \mathrm{H}), 4.30-4.24(\mathrm{~m}, 1 \mathrm{H}), 2.42-2.29(\mathrm{~m}, 2 \mathrm{H}), 0.44(\mathrm{~s}, 3 \mathrm{H}), 0.42(\mathrm{~s}, 3 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR (126 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta=140.4,136.7,134.1,132.8,129.3,128.3,128.0,127.82,127.80,120.2,71.5$, $69.6,30.0,-4.5,-4.7$. Trans -4.5 g is a known compound and spectroscopic data are in agreement with those reported in literature. ${ }^{26}$

## Synthesis of trans-(6-(4-methoxyphenyl)-3-methyl-5,6-dihydro-2H-pyran-2-yl)trimethyl silane (trans-4.5h)



Compound 4.5 h was prepared following general procedure F : Grubbs catalyst $2^{\text {nd }}$ generation ( $51 \mathrm{mg}, 0.06 \mathrm{mmol}, 0.04$ equiv.) and syn-(1-((1-(4-methoxyphenyl)but-3-en-1-yl)oxy)-2methylallyl)trimethylsilane, syn-4.4h (457 mg, 1.5 mmol , 1 equiv.) and benzene ( 40 mL ) at $85^{\circ} \mathrm{C}$ for 2 hours followed by concentration and column chromatography, $R_{f}$ for trans- $\mathbf{4 . 5 h}=0.5(30 \%$ DCM in hexanes) afforded a total of $377 \mathrm{mg}, 1.37 \mathrm{mmol}$ ( $91 \%$ isolated yield) of compound trans-
4.5h as a colorless liquid. ${ }^{1} \mathrm{H}$ NMR $\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta=7.29(\mathrm{~d}, J=8.7 \mathrm{~Hz}, 2 \mathrm{H}), 6.88(\mathrm{~d}, J=$ $8.7 \mathrm{~Hz}, 2 \mathrm{H}), 5.48(\mathrm{ddq}, J=5.0,3.4,1.5 \mathrm{~Hz}, 1 \mathrm{H}), 4.50(\mathrm{dd}, J=8.9,4.2 \mathrm{~Hz}, 1 \mathrm{H}), 4.00(\mathrm{~s}, 1 \mathrm{H}), 3.80$ $(\mathrm{s}, 3 \mathrm{H}), 2.38-2.24(\mathrm{~m}, 2 \mathrm{H}), 1.65(\mathrm{~s}, 3 \mathrm{H}), 0.16(\mathrm{~s}, 9 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $\left.126 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta=158.8$, $135.4,134.9,127.3,115.9,113.6,74.9,72.6,55.2,32.0,20.4,-1.4$. Trans-4.5h is a known compound and spectroscopic data are in agreement with those reported in literature. ${ }^{26}$

## Synthesis of cis/trans-(6-(4-chlorophenyl)-3-methyl-5,6-dihydro-2H-pyran-2-yl)trimethyl silane (cis/trans-4.5i)



Compound 4.5 i was prepared following general procedure F : Grubbs catalyst $2^{\text {nd }}$ generation ( $70.04 \mathrm{mg}, 0.083 \mathrm{mmol}, 0.03$ equiv.) and syn/anti-(1-((1-(4-chlorophenyl)but-3-en-1-yl)oxy)-2methylallyl)trimethylsilane, syn:anti $=1: 1,4.4 i(850 \mathrm{mg}, 2.75 \mathrm{mmol}, 1$ equiv. $)$ and dichloromethane $(40 \mathrm{~mL})$ at room temperature for 12 hours followed by concentration and column chromatography, $R_{f}$ for cis-4.5i $=0.7$ and $R_{f}$ for trans-4.5i $=0.4(20 \%$ DCM in hexanes) afforded a total of $763 \mathrm{mg}, 2.70 \mathrm{mmol}$ ( $98 \%$ isolated yield) fully separable mixture of diastereomers of compound 4.5i as colorless liquid.

Spectroscopic data for cis-4.5i: ${ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta=7.33-7.27(\mathrm{~m}, 4 \mathrm{H}), 5.85-$ $5.77(\mathrm{~m}, 2 \mathrm{H}), 4.38(\mathrm{dd}, J=10.1,3.1 \mathrm{~Hz}, 1 \mathrm{H}), 4.18(\mathrm{dtd}, J=5.0,2.7,1.5 \mathrm{~Hz}, 1 \mathrm{H}), 2.26-2.19(\mathrm{~m}$, 1 H ), 2.13 (dddt, $J=14.8,8.3,2.7,1.6 \mathrm{~Hz}, 1 \mathrm{H}), 0.10(\mathrm{~s}, 9 \mathrm{H}) .{ }^{13} \mathrm{C} \operatorname{NMR}\left(126 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta=$ $142.5,132.6,128.2,128.1,127.0,120.8,74.7,71.6,34.0,-4.0 . \operatorname{IR}\left(\right.$ FTIR, $\left.\mathrm{cm}^{-1}\right) \tilde{v}=3057,2953$, 1492, 1380, 1243, 1012, 813. MS (GC/MS): m/z $(\%)=280(0.02)[\mathrm{M}]^{+}, 208(6), 68(100)$.

Spectroscopic data for trans-4.5i: ${ }^{1} \mathrm{H}$ NMR ( $\left.500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta=7.35-7.29(\mathrm{~m}, 4 \mathrm{H}), 5.84$
$-5.76(\mathrm{~m}, 2 \mathrm{H}), 4.72(\mathrm{dd}, J=6.4,4.5 \mathrm{~Hz}, 1 \mathrm{H}), 4.01-3.96(\mathrm{~m}, 1 \mathrm{H}), 2.47-2.40(\mathrm{~m}, 1 \mathrm{H}), 2.39-$ $2.32(\mathrm{~m}, 1 \mathrm{H}), 0.10(\mathrm{~s}, 9 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR (126 MHz, $\left.\mathrm{CDCl}_{3}\right) \delta=140.5,132.9,128.3,128.2,128.1$, 119.7, 71.6, 69.8, 30.0, -3.0. IR (FTIR, $\left.\mathrm{cm}^{-1}\right) ~ \tilde{v}=3056,2952,1492,1380,1242,1012,812 . \mathrm{MS}$ $(\mathrm{GC} / \mathrm{MS}): m / \mathrm{z}(\%)=280(0.015)[\mathrm{M}]^{+}, 208(5.8), 68(100)$.

### 4.7.3. Wittig rearrangements of trans-2-silyl-5,6-dihydro-6-aryl-(2H)-pyrans trans-4.5: general procedure $\mathbf{G}$



Following our reported procedure, ${ }^{8}$ freshly prepared and purified trans-2-silyl-5,6-dihydro-6-aryl-(2H)-pyran 4.5 was dissolved in THF under nitrogen (concentration 0.08 M , unless otherwise noted) and the solution cooled at $-78^{\circ} \mathrm{C}$ (dry ice/acetone bath), $n$-butyllithium ( 1.2 equiv, 1.6 M or 2.5 M in hexanes) was added dropwise ( $1 \mathrm{drop} / \mathrm{s}$ ) to give a colored solution. The reaction was quenched after the indicated time ( $10-30 \mathrm{~min}$ ) by adding saturated $\mathrm{NH}_{4} \mathrm{Cl}_{(\text {aq })}$ and diluted with $\mathrm{H}_{2} \mathrm{O}$ and diethyl ether. The aqueous phase was extracted with diethyl ether three times. Combined organic extracts were washed with saturated $\mathrm{NH}_{4} \mathrm{Cl}_{(\mathrm{aq})}, \mathrm{H}_{2} \mathrm{O}$, and brine. The solution was dried over magnesium sulfate, filtered, quickly concentrated in a rotovap at temperatures lower than 45 ${ }^{\circ}$ C. Column chromatography with EtOAc in hexanes afforded cyclopentenol 4.6. Other products including the ones resulting from [1,4]-Wittig rearrangement (4.7) were also observed (see individual substrate).

## Synthesis of 2-methyl-5-phenyl-1-(trimethylsilyl)cyclopent-2-en-1-ol (4.6a), -OH and phenyl trans to one another and (4.6a'), -OH and phenyl cis



Following general procedure G with slight modification, 370 mg of cis-trimethyl(3-methyl-6-phenyl-5,6-dihydro-2H-pyran-2-yl)silane, cis-4.5a and 370 mg of trans-trimethyl(3-methyl-6-phenyl-5,6-dihydro-2H-pyran-2-yl)silane, trans-4.5a were weighed into a 20 mL vial respectively making a mixture of $1: 1$ cis:trans diastereomeric ratio ( $740 \mathrm{mg}, 3.0 \mathrm{mmol}, 1.0$ equiv). This was dissolved in 20 mL dry THF and the resulting solution was transferred into a 100 mL round bottom flask via syringe. Additional 20 mL of THF was transferred into the flask and the resulting mixture was cooled to $-78^{\circ} \mathrm{C}$ on an acetone/dry-ice bath. This was followed by dropwise addition of 0.72 mL of $n$-butyllithium, 2.5 M in hexanes ( $1.8 \mathrm{mmol}, 0.6$ equiv.). The resulting mixture was stirred at $-78{ }^{\circ} \mathrm{C}$ for 15 minutes (approximate time taken by the trans diastereomer to react), then 3.22 mL of sec-butyllithium, 1.4 M in cyclohexane ( $4.5 \mathrm{mmol}, 1.5$ equiv.) was added to the flask at $78{ }^{\circ} \mathrm{C}$. The solution turned dark purple after complete addition of sec-butyllithium. The mixture was stirred at $-78^{\circ} \mathrm{C}$ for 1.5 hours after which the cold bath was removed, and the mixture allowed to warm up to room temperature. The mixture was stirred at room temperature for additional 3 hours and then cooled down to $-78^{\circ} \mathrm{C}$ and quenched by 10 mL water. The cold bath was removed again, and the mixture allowed to warm up to room temperature and diluted with diethyl ether. The resulting mixture was transferred into a separating funnel and the layers were separated. The aqueous layer was extracted with diethyl ether ( $25 \mathrm{~mL} \times 3$ ). The combined organic layers were washed with brine and dried over anhydrous $\mathrm{MgSO}_{4}$. The mixture was filtered, and the filtrate was
concentrated on a rotavapor under reduced pressure to give crude reaction mixture. Further purification by column chromatography, $R_{f}$ for $\mathbf{4 . 6} \mathbf{a}^{\prime}=0.71$ and $R_{f}$ for $\mathbf{4 . 6 a}=0.44(10 \% \mathrm{EtOAc}$ in hexanes) furnished $287 \mathrm{mg}, 1.2 \mathrm{mmol}$ ( $40 \%$ isolated yield) of $\mathbf{4 . 6 a}$ as a colorless liquid and 185 $\mathrm{mg}, 0.75 \mathrm{mmol}$ ( $25 \%$ isolated yield) of $\mathbf{4 . 6 a}^{\prime}$ as a colorless liquid.

Spectroscopic data for 4.6a: ${ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=7.46-7.40(\mathrm{~m}, 2 \mathrm{H}), 7.34-7.28$ (m, 2H), $7.28-7.23(\mathrm{~m}, 1 \mathrm{H}), 5.67(\mathrm{dp}, J=3.2,1.6 \mathrm{~Hz}, 1 \mathrm{H}), 3.44(\mathrm{dd}, J=10.6,7.6 \mathrm{~Hz}, 1 \mathrm{H}), 2.72$ $-2.63(\mathrm{~m}, 1 \mathrm{H}), 2.46(\mathrm{dddt}, J=15.4,7.7,3.1,1.5 \mathrm{~Hz}, 1 \mathrm{H}), 1.80(\mathrm{dt}, J=2.9,1.5 \mathrm{~Hz}, 3 \mathrm{H}), 1.47(\mathrm{~s}$, 1H), -0.27 (s, 9H). ${ }^{13} \mathrm{C}$ NMR (126 MHz, $\mathrm{CDCl}_{3}$ ): $\delta=145.6,140.8,128.6,128.1,126.7,124.7$, 84.8, 61.7, 32.8, 14.6, $-2.2 .{ }^{29} \mathrm{Si} \operatorname{NMR}\left(99 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta=1.44$. 4.6a is a known compound and the spectroscopic data are in agreement with those reported in the literature. ${ }^{26}$

Spectroscopic data for 4.6a': ${ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=7.34-7.28(\mathrm{~m}, 2 \mathrm{H}), 7.26-$ $7.20(\mathrm{~m}, 3 \mathrm{H}), 5.57(\mathrm{dq}, J=3.2,1.7 \mathrm{~Hz}, 1 \mathrm{H}), 3.71(\mathrm{dt}, J=8.2,1.6 \mathrm{~Hz}, 1 \mathrm{H}), 2.73-2.64(\mathrm{~m}, 1 \mathrm{H})$, 2.54 (ddq, $J=16.8,3.4,1.7 \mathrm{~Hz}, 1 \mathrm{H}), 1.75(\mathrm{dt}, J=3.0,1.6 \mathrm{~Hz}, 3 \mathrm{H}), 0.82(\mathrm{~s}, 1 \mathrm{H}), 0.15(\mathrm{~s}, 9 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR (126 MHz, $\mathrm{CDCl}_{3}$ ): $\delta=144.2,141.5,128.6,128.5,126.8,123.6,81.6,51.0,38.6,14.3,-$ 3.2. ${ }^{29} \mathrm{Si}$ NMR ( $99 \mathrm{MHz}, \mathrm{CDCl}_{3}, \mathrm{ppm}$ ): $\delta=4.01$. HRMS (ESI): $m / z[\mathrm{M}-\mathrm{OH}]^{+}$calcd for $\mathrm{C}_{15} \mathrm{H}_{21} \mathrm{Si}$ : 229.1413; found: 229.1406.

Synthesis of 5-(4-chlorophenyl)-1-(trimethylsilyl)cyclopent-2-en-1-ol (4.6b) and 2-(2-(4-chlorophenyl)cyclopropyl)-1-(trimethylsilyl)ethan-1-one (4.7b)


Following general procedure G, compounds 4.6b and 4.7b were prepared from trans-(6-(4-chlorophenyl)-5,6-dihydro-2H-pyran-2-yl)trimethylsilane, trans-4.5b (534 mg, $2.0 \mathrm{mmol}, 1.0$
equiv.), $n$-butyllithium 2.5 M in hexanes ( $1.0 \mathrm{~mL}, 2.4 \mathrm{mmol}, 1.2$ equiv.) and THF ( 25 mL ) for 15 minutes. The reaction was quenched by adding water instead of saturated $\mathrm{NH}_{4} \mathrm{Cl}_{\text {(aq) }}$. Workup, concentration and column chromatography, $R_{f}$ for $\mathbf{4 . 7 b}=0.7$ and $R_{f}$ for $\mathbf{4 . 6 b}=0.3(10 \% \mathrm{EtOAc}$ in hexanes) furnished $297 \mathrm{mg}, 1.12 \mathrm{mmol}$ ( $56 \%$ isolated yield) of $\mathbf{4 . 6 b}$ as a yellow liquid and 111 $\mathrm{mg}, 0.42 \mathrm{mmol}$ of $\mathbf{4 . 7 b}$ ( $21 \%$ isolated yield) as a colorless liquid.

Spectroscopic data for $\mathbf{4 . 6 b}:{ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=7.35(\mathrm{~d}, J=8.3 \mathrm{~Hz}, 2 \mathrm{H}), 7.27$ (d, $J=8.5 \mathrm{~Hz}, 2 \mathrm{H}), 5.99(\mathrm{ddd}, J=5.8,2.9,2.0 \mathrm{~Hz}, 1 \mathrm{H}), 5.77(\mathrm{ddd}, J=5.8,2.3,1.4 \mathrm{~Hz}, 1 \mathrm{H}), 3.41$ $(\mathrm{dd}, J=10.2,8.2 \mathrm{~Hz}, 1 \mathrm{H}), 2.72-2.59(\mathrm{~m}, 2 \mathrm{H}), 1.52(\mathrm{~s}, 1 \mathrm{H}),-0.27(\mathrm{~s}, 9 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( 126 MHz , $\left.\mathrm{CDCl}_{3}\right): \delta=139.0,137.4,132.3,130.4,129.8,128.2,84.2,59.3,35.3,-3.4 .4 .6 \mathrm{~b}$ is a known compound and the spectroscopic data are in agreement with those reported in the literature. ${ }^{26}$

Spectroscopic data for $\mathbf{4 . 6 b}:{ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=7.20(\mathrm{~d}, J=8.5 \mathrm{~Hz}, 2 \mathrm{H}), 7.01$ $(\mathrm{d}, J=8.4 \mathrm{~Hz}, 2 \mathrm{H}), 2.69(\mathrm{dd}, J=6.7,1.1 \mathrm{~Hz}, 2 \mathrm{H}), 1.60(\mathrm{dt}, J=8.9,4.9 \mathrm{~Hz}, 1 \mathrm{H}), 1.27(\mathrm{dtdd}, J=$ 8.6, 6.7, 5.7, $4.5 \mathrm{~Hz}, 1 \mathrm{H}), 0.94(\mathrm{dt}, J=8.6,5.2 \mathrm{~Hz}, 1 \mathrm{H}), 0.76(\mathrm{dt}, J=8.6,5.4 \mathrm{~Hz}, 1 \mathrm{H}), 0.20(\mathrm{~s}, 9 \mathrm{H})$. ${ }^{13} \mathrm{C}$ NMR ( $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=247.1,141.4,131.0,128.3,127.4,53.0,22.2,16.7,15.5,-3.2$. HRMS (APCI): $m / z[\mathrm{M}+\mathrm{H}]^{+}$calcd for $\mathrm{C}_{14} \mathrm{H}_{20} \mathrm{ClOSi}$ 267.0972; found: 267.0967.

## Synthesis of 5-(4-trifluorophenyl)-1-(trimethylsilyl)cyclopent-2-en-1-ol (4.6c)



Following general procedure $G$, compound 4.6 c was prepared from trans-(6-(4-trifluorophenyl)-5,6-dihydro-2H-pyran-2-yl)trimethylsilane trans-4.5c ( $841 \mathrm{mg}, 2.8 \mathrm{mmol}, 1.0$ equiv.), $n$-butyllithium 2.5 M in hexanes ( $1.4 \mathrm{~mL}, 3.36 \mathrm{mmol}, 1.2$ equiv.) and THF ( 40 mL ) for 20 minutes. Workup, concentration and column chromatography, $R_{f}=0.5$ ( $20 \% \mathrm{EtOAc}$ in hexanes)
furnished $578 \mathrm{mg}(70 \%)$ of $\mathbf{4 . 6 c}$ as a yellow liquid. Compound $\mathbf{4 . 6 c}$ is very unstable and upon storage at $-15{ }^{\circ} \mathrm{C}$ slowly converts to 4 -(4-(trifluoromethyl)phenyl)-3-(trimethylsilyl)cyclopent-2-en-1-one 4.11c.

Spectroscopic data for 4.6 c : ${ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=7.57$ (d, $J=8.8 \mathrm{~Hz}, 2 \mathrm{H}$ ), 7.54 $(\mathrm{d}, J=8.8 \mathrm{~Hz}, 2 \mathrm{H}), 6.01(\mathrm{ddd}, J=5.8,2.9,1.9 \mathrm{~Hz}, 1 \mathrm{H}), 5.79(\mathrm{ddd}, J=5.8,2.3,1.3 \mathrm{~Hz}, 1 \mathrm{H}), 3.50$ (dd, $J=10.3,8.0 \mathrm{~Hz}, 1 \mathrm{H}), 2.75(\mathrm{ddt}, J=15.8,10.3,2.1 \mathrm{~Hz}, 1 \mathrm{H}), 2.66(\mathrm{dddd}, J=15.9,8.0,2.9,1.3$ $\mathrm{Hz}, 1 \mathrm{H}), 1.56(\mathrm{~s}, 1 \mathrm{H}),-0.29(\mathrm{~s}, 9 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=144.75(\mathrm{~d}, J=1.4 \mathrm{~Hz})$, $137.4,130.4,128.8,129.02(\mathrm{q}, J=32.4 \mathrm{~Hz}), 124.97(\mathrm{q}, J=3.8 \mathrm{~Hz}), 124.28(\mathrm{q}, J=271.8 \mathrm{~Hz}), 84.2$, $59.7,35.1,-3.5 .4 .6 \mathrm{c}$ is a known compound and the spectroscopic data are in agreement with those reported in the literature. ${ }^{26}$

Spectroscopic data for $\mathbf{4 . 1 1 c}{ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=7.57(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 2 \mathrm{H}), 7.23$ $(\mathrm{d}, J=8.0 \mathrm{~Hz}, 2 \mathrm{H}), 6.52(\mathrm{~d}, J=1.9 \mathrm{~Hz}, 1 \mathrm{H}), 4.26(\mathrm{dt}, J=7.0,2.1 \mathrm{~Hz}, 1 \mathrm{H}), 2.89(\mathrm{dd}, J=19.2,7.1$ $\mathrm{Hz}, 1 \mathrm{H}), 2.28(\mathrm{dd}, J=19.1,2.3 \mathrm{~Hz}, 1 \mathrm{H}),-0.05(\mathrm{~s}, 9 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR $\left(126 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta=209.9$, $185.0,146.5(\mathrm{~d}, J=1.6 \mathrm{~Hz}), 142.3,129.6(\mathrm{q}, J=32.6 \mathrm{~Hz}), 128.0,125.8(\mathrm{q}, J=3.8 \mathrm{~Hz}), 124.0(\mathrm{q}$, $J=272.1 \mathrm{~Hz}), 50.3,45.4,-1.8 .{ }^{19} \mathrm{~F}$ NMR $\left(470 \mathrm{MHz}, \mathrm{CDCl}_{3}, \mathrm{ppm}\right): \delta=-62.47$.

## Synthesis of 5-(naphthalen-1-yl)-1-(trimethylsilyl)cyclopent-2-en-1-ol (4.6d)



Following general procedure G, compound 4.6c was prepared from trans-(6-naphthalen-1-yl)-5,6-dihydro-2H-pyran-2-yl)trimethylsilane trans-4.5d (701 mg, $2.48 \mathrm{mmol}, 1.0$ equiv.), $n$ butyllithium 2.5 M in hexanes ( $1.24 \mathrm{~mL}, 2.98 \mathrm{mmol}$, 1.2 equiv.) and THF ( 40 mL ) for 20 minutes. Workup, concentration and column chromatography, $R_{f}=0.4$ ( $10 \%$ EtOAc in hexanes) furnished
$568 \mathrm{mg}, 2.01 \mathrm{mmol}\left(81 \%\right.$ isolated yield) of $\mathbf{4 . 6 d}$ as a yellow liquid. ${ }^{1} \mathrm{H} \mathrm{NMR}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$ : $\delta=8.62(\mathrm{~d}, J=8.5 \mathrm{~Hz}, 1 \mathrm{H}), 7.86(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.78(\mathrm{~d}, J=8.2 \mathrm{~Hz}, 1 \mathrm{H}), 7.63(\mathrm{~d}, J=7.2 \mathrm{~Hz}$, $1 \mathrm{H}), 7.51$ (dddd, $J=22.6,8.0,6.8,1.3 \mathrm{~Hz}, 2 \mathrm{H}), 7.45(\mathrm{t}, J=7.7 \mathrm{~Hz}, 1 \mathrm{H}), 6.10(\mathrm{ddd}, J=5.7,2.9$, $1.9 \mathrm{~Hz}, 1 \mathrm{H}), 5.81(\mathrm{ddd}, J=5.8,2.4,1.3 \mathrm{~Hz}, 1 \mathrm{H}), 4.47(\mathrm{dd}, J=10.0,8.1 \mathrm{~Hz}, 1 \mathrm{H}), 3.04(\mathrm{ddt}, J=$ $16.0,10.0,2.1 \mathrm{~Hz}, 1 \mathrm{H}), 2.81$ (dddd, $J=16.2,8.1,2.8,1.3 \mathrm{~Hz}, 1 \mathrm{H}), 1.72(\mathrm{~s}, 1 \mathrm{H}),-0.43(\mathrm{~s}, 9 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR (126 MHz, $\mathrm{CDCl}_{3}$ ): $\delta=137.4,136.2,134.0,133.8,130.8,128.6,127.4,125.8,125.5$, $124.98,124.97,124.1,86.0,53.8,37.4,-3.6$. IR (FTIR, $\left.\mathrm{cm}^{-1}\right): \tilde{\mathrm{v}}=3423,3046,2952,1508,1396$, 1244, 832, 777. MS (GC/MS): m/z (\%) = $282(20)[\mathrm{M}]^{+}, 191(25), 165$ (18), 73 (100). HRMS (ESI): $m / z\left[\mathrm{M}-\mathrm{H}^{-}\right]^{+}$calcd for $\mathrm{C}_{18} \mathrm{H}_{21} \mathrm{OSi}$ : 281.1362; found: 281.1353.

Synthesis of 5-([1,1'-biphenyl]-4-yl)-1-(trimethylsilyl)cyclopent-2-en-1-ol (4.6e)


Following general procedure G, compound 4.6e was prepared from trans-(6-([1,1'-biphenyl]-4-yl)-5,6-dihydro-2H-pyran-2-yl)trimethylsilane trans-4.5e ( $67 \mathrm{mg}, 0.22 \mathrm{mmol}, 1.0$ equiv.), secbutyllithium 1.4 M in cyclohexane ( $0.46 \mathrm{~mL}, 0.66 \mathrm{mmol}, 3.0$ equiv.) and THF ( 2.4 mL ) for 3 hours. Workup, concentration and column chromatography, $R_{f}=0.4$ ( $10 \%$ EtOAc in hexanes) furnished $51 \mathrm{mg}, 0.165 \mathrm{mmol}\left(75 \%\right.$ isolated yield) of 4.6e as a white solid. ${ }^{1} \mathrm{H}$ NMR ( 500 MHz , $\left.\mathrm{CDCl}_{3}\right): \delta=7.66-7.62(\mathrm{~m}, 2 \mathrm{H}), 7.58(\mathrm{~d}, J=8.3 \mathrm{~Hz}, 2 \mathrm{H}), 7.50(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 2 \mathrm{H}), 7.48-7.43$ $(\mathrm{m}, 2 \mathrm{H}), 7.38-7.33(\mathrm{~m}, 1 \mathrm{H}), 6.04(\mathrm{ddd}, J=5.8,2.9,1.9 \mathrm{~Hz}, 1 \mathrm{H}), 5.82(\mathrm{ddd}, J=5.8,2.3,1.3 \mathrm{~Hz}$, $1 \mathrm{H}), 3.52(\mathrm{dd}, J=10.3,7.9 \mathrm{~Hz}, 1 \mathrm{H}), 2.81(\mathrm{ddt}, J=16.0,10.3,2.1 \mathrm{~Hz}, 1 \mathrm{H}), 2.70(\mathrm{dddd}, J=16.0$, 8.0, 2.9, $1.3 \mathrm{~Hz}, 1 \mathrm{H}), 1.58(\mathrm{~s}, 1 \mathrm{H}),-0.23(\mathrm{~s}, 9 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR (126 MHz, $\left.\mathrm{CDCl}_{3}\right): \delta=140.9,139.6$,
$137.3,130.5,128.9,128.7,127.1,126.9,126.8,84.5,59.7,35.4,-3.4 .4 .6 e$ is a known compound and the spectroscopic data are in agreement with those reported in the literature. ${ }^{26}$

## Synthesis of 5-(naphthalen-2-yl)-1-(triethylsilyl)cyclopent-2-en-1-ol (4.6f) and

 2-(2-(naphthalen-2-yl)cyclopropyl)-1-(triethylsilyl)ethan-1-one (4.7f)

Following general procedure G, compounds $\mathbf{4 . 6 f}$ and 4.7 f were prepared from trans-triethyl(6-(naphthalen-2-yl)-5,6-dihydro-2H-pyran-2-yl)silane trans-4.5f ( $650 \mathrm{mg}, 2.0 \mathrm{mmol}, 1.0$ equiv.), $n$ butyllithium 2.5 M in hexanes ( $1.0 \mathrm{~mL}, 2.4 \mathrm{mmol}, 1.2$ equiv.) and THF ( 25 mL ) for 15 minutes. The reaction was quenched by adding water instead of saturated $\mathrm{NH}_{4} \mathrm{Cl}_{(\mathrm{aq})}$. Workup, concentration and column chromatography, $R_{f}$ for $\mathbf{4 . 7 f}=0.7$ and $R_{f}$ for $\mathbf{4 . 6 f}=0.3(10 \% \mathrm{EtOAc}$ in hexanes $)$ furnished $473 \mathrm{mg}, 1.46 \mathrm{mmol}$ ( $73 \%$ isolated yield) of $\mathbf{4 . 6 f}$ as a yellow liquid and $110 \mathrm{mg}, 0.34$ mmol of $\mathbf{4 . 7 f}$ ( $17 \%$ isolated yield) as a colorless liquid.

Spectroscopic data for 4.6f: ${ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=7.88-7.76(\mathrm{~m}, 4 \mathrm{H}), 7.64(\mathrm{dd}, J$ $=8.5,1.7 \mathrm{~Hz}, 1 \mathrm{H}), 7.51-7.41(\mathrm{~m}, 2 \mathrm{H}), 6.03(\mathrm{ddd}, J=5.8,2.9,1.9 \mathrm{~Hz}, 1 \mathrm{H}), 5.91(\mathrm{ddd}, J=5.8$, $2.3,1.3 \mathrm{~Hz}, 1 \mathrm{H}), 3.60(\mathrm{dd}, J=10.0,7.7 \mathrm{~Hz}, 1 \mathrm{H}), 2.90(\mathrm{ddt}, J=16.0,9.9,2.1 \mathrm{~Hz}, 1 \mathrm{H}), 2.76$ (dddd, $J=16.0,7.8,2.9,1.3 \mathrm{~Hz}, 1 \mathrm{H}), 1.55(\mathrm{~s}, 1 \mathrm{H}), 0.80(\mathrm{t}, J=8.0 \mathrm{~Hz}, 9 \mathrm{H}), 0.28(\mathrm{qd}, J=7.9,1.9 \mathrm{~Hz}, 6 \mathrm{H})$. ${ }^{13} \mathrm{C}$ NMR (126 MHz, $\mathrm{CDCl}_{3}$ ): $\delta=138.5,138.3,133.3,132.5,130.5,127.9,127.7,127.6,127.4$, $126.2,125.8,125.3,85.9,60.6,35.8,7.8,2.3 .4 .6 f$ is a known compound and the spectroscopic data are in agreement with those reported in the literature. ${ }^{26}$

Spectroscopic data for 4.7f: ${ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=7.82-7.70(\mathrm{~m}, 3 \mathrm{H}), 7.54(\mathrm{~s}$, $1 \mathrm{H}), 7.46-7.36(\mathrm{~m}, 2 \mathrm{H}), 7.23(\mathrm{dd}, J=8.6,1.8 \mathrm{~Hz}, 1 \mathrm{H}), 2.80(\mathrm{dd}, J=17.2,6.2 \mathrm{~Hz}, 1 \mathrm{H}), 2.65(\mathrm{dd}$,
$J=17.2,7.0 \mathrm{~Hz}, 1 \mathrm{H}), 1.81(\mathrm{dt}, J=9.2,4.9 \mathrm{~Hz}, 1 \mathrm{H}), 1.51-1.44(\mathrm{~m}, 1 \mathrm{H}), 1.12(\mathrm{dt}, J=8.6,5.1 \mathrm{~Hz}$, $1 \mathrm{H}), 0.99(\mathrm{t}, J=7.9 \mathrm{~Hz}, 8 \mathrm{H}), 0.83(\mathrm{dt}, J=8.6,5.3 \mathrm{~Hz}, 1 \mathrm{H}), 0.80-0.70(\mathrm{~m}, 7 \mathrm{H}), 0.47(\mathrm{qd}, J=8.3$, $7.9,3.7 \mathrm{~Hz}, 1 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR $\left(126 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta=247.1,140.4,133.5,131.9,127.8,127.5,127.2$, $125.9,124.91,124.86,123.9,54.8,22.8,16.5,15.7,7.3,2.1 .4 .7 \mathbf{f}$ is a known compound, and the spectroscopic data are in agreement with those reported in the literature. ${ }^{26}$

## Synthesis of 5-(4-chlorophenyl)-1-(dimethyl(phenyl)silyl)cyclopent-2-en-1-ol (4.6g)



Following general procedure $G$, compound 4.6 g was prepared from trans- (6-(4-chlorophenyl)-5,6-dihydro-2H-pyran-2-yl)dimethyl(phenyl)silane trans-4.5g ( $987 \mathrm{mg}, 3.0 \mathrm{mmol}$, 1.0 equiv.), $n$-butyllithium 2.5 M in hexanes ( $1.5 \mathrm{~mL}, 3.6 \mathrm{mmol}, 1.2$ equiv.) and THF ( 10 mL ) for 10 minutes. Workup, concentration and column chromatography, $R_{f}=0.4$ ( $10 \% \mathrm{EtOAc}$ in hexanes) furnished 362 mg , 1.11 mmol ( $37 \%$ isolated yield) of $\mathbf{4 . 6 g}$ as a yellow liquid. ${ }^{1} \mathrm{H}$ NMR ( 500 MHz , $\left.\mathrm{CDCl}_{3}, \mathrm{ppm}\right): \delta=7.37-7.32(\mathrm{~m}, 1 \mathrm{H}), 7.32-7.25(\mathrm{~m}, 4 \mathrm{H}), 7.21-7.14(\mathrm{~m}, 4 \mathrm{H}), 6.01(\mathrm{ddd}, J=$ $5.7,2.9,1.9 \mathrm{~Hz}, 1 \mathrm{H}), 5.76(\mathrm{ddd}, J=5.8,2.4,1.3 \mathrm{~Hz}, 1 \mathrm{H}), 3.42(\mathrm{dd}, J=10.3,7.8 \mathrm{~Hz}, 1 \mathrm{H}), 2.57$ (dddd, $J=15.8,7.8,2.9,1.3 \mathrm{~Hz}, 1 \mathrm{H}), 2.48(\mathrm{ddt}, J=15.9,10.4,2.2 \mathrm{~Hz}, 1 \mathrm{H}), 1.63(\mathrm{~s}, 1 \mathrm{H}), 0.07(\mathrm{~s}$, $3 \mathrm{H}), 0.01(\mathrm{~s}, 3 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR (126 MHz, $\left.\mathrm{CDCl}_{3}\right): \delta=138.5,137.4,136.5,134.3,132.3,130.9$, $129.9,129.0,127.9,127.3,84.2,59.0,35.3,-4.9,-5.2 .4 .6 \mathrm{~g}$ is a known compound and the spectroscopic data are in agreement with those reported in the literature. ${ }^{26}$

Synthesis of 5-(4-methoxyphenyl)-2-methyl-1-(trimethylsilyl)cyclopent-2-en-1-ol (4.6h)


Following general procedure G, compound 4.6 h was prepared from trans-(6-(4-methoxyphenyl)-3-methyl-5,6-dihydro-2H-pyran-2-yl)trimethylsilane trans-4.5h (276 mg, 1.0 mmol, 1.0 equiv.), $n$-butyllithium 2.5 M in hexanes ( $0.5 \mathrm{~mL}, 1.2 \mathrm{mmol}, 1.2$ equiv.) and THF ( 10 $\mathrm{mL})$ for 20 minutes. Workup, concentration and column chromatography, $R_{f}=0.4(10 \% \mathrm{EtOAc}$ in hexanes) furnished $236 \mathrm{mg}, 0.85 \mathrm{mmol}\left(85 \%\right.$ isolated yield) of $\mathbf{4 . 6 h}$ as a yellow liquid. ${ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=7.32(\mathrm{~d}, J=8.5 \mathrm{~Hz}, 2 \mathrm{H}), 6.85(\mathrm{~d}, J=8.7 \mathrm{~Hz}, 2 \mathrm{H}), 5.65(\mathrm{dp}, J=3.3,1.6$ $\mathrm{Hz}, 1 \mathrm{H}), 3.80(\mathrm{~s}, 3 \mathrm{H}), 3.36(\mathrm{dd}, J=10.6,7.6 \mathrm{~Hz}, 1 \mathrm{H}), 2.60(\mathrm{ddp}, J=15.6,10.4,2.5 \mathrm{~Hz}, 1 \mathrm{H}), 2.43$ (dddt, $J=12.3,6.1,3.1,1.5 \mathrm{~Hz}, 1 \mathrm{H}), 1.78(\mathrm{dt}, J=2.8,1.5 \mathrm{~Hz}, 3 \mathrm{H}), 1.41(\mathrm{~s}, 1 \mathrm{H}),-0.27(\mathrm{~s}, 9 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR (126 MHz, $\mathrm{CDCl}_{3}$ ): $\delta=158.5,145.6,132.9,129.5,124.7,113.4,84.8,61.0,55.2,33.1,14.6$, -2.2. 4.6h is a known compound and the spectroscopic data are in agreement with those reported in the literature. ${ }^{26}$

Synthesis of 5-(4-chlorophenyl)-2-methyl-1-(trimethylsilyl)cyclopent-2-en-1-ol (4.6i) and 4-(4-chlorophenyl)-2-methyl-3-(trimethylsilyl)cyclopent-2-en-1-ol (4.12i)


Following general procedure $G$, compounds 4.6i and 4.12i were prepared from trans-(6-(4-chlorophenyl)-3-methyl-5,6-dihydro-2H-pyran-2-yl)trimethylsilane, trans-4.5i (617 mg, 2.2
mmol, 1.0 equiv.), $n$-butyllithium 2.5 M in hexanes ( $1.06 \mathrm{~mL}, 2.64 \mathrm{mmol}, 1.2$ equiv.) and THF $(30 \mathrm{~mL})$ for 15 minutes. Workup, concentration and column chromatography, $R_{f}$ for $\mathbf{4 . 6 i}=0.6$ and $R_{f}$ for $\mathbf{4 . 1 2 i}=0.3(15 \% \mathrm{EtOAc}$ in hexanes) furnished $388 \mathrm{mg}, 1.386 \mathrm{mmol}(63 \%$ isolated yield $)$ of 4.6i as a yellow liquid and $37 \mathrm{mg}, 0.132 \mathrm{mmol}$ of $\mathbf{4 . 1 2 \mathbf { i }}$ ( $6 \%$ isolated yield) as a colorless liquid. Compound 4.12i might have been formed from 4.6i during the workup.

Spectroscopic data for 4.6i: ${ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=7.33(\mathrm{~d}, J=8.6 \mathrm{~Hz}, 2 \mathrm{H}), 7.25$ $(\mathrm{d}, J=8.5 \mathrm{~Hz}, 2 \mathrm{H}), 5.63(\mathrm{dp}, J=3.3,1.6 \mathrm{~Hz}, 1 \mathrm{H}), 3.36(\mathrm{dd}, J=10.6,7.6 \mathrm{~Hz}, 1 \mathrm{H}), 2.61-2.52(\mathrm{~m}$, $1 \mathrm{H}), 2.41$ (dddt, $J=15.3,7.6,3.1,1.5 \mathrm{~Hz}, 1 \mathrm{H}), 1.76(\mathrm{dt}, J=2.9,1.5 \mathrm{~Hz}, 3 \mathrm{H}), 1.45(\mathrm{~s}, 1 \mathrm{H}),-0.29$ (s, 9H). ${ }^{13} \mathrm{C}$ NMR (126 MHz, $\mathrm{CDCl}_{3}$ ): $\delta=145.7,139.4,132.3,129.9,128.1,124.6,84.7,60.9$, 32.8, 14.5, -2.1. IR (FTIR, $\mathrm{cm}^{-1}$ ): $\tilde{\mathrm{v}}=3416,3092,3069,2955,1490,1401,1249,1091,1012,822$. MS (GC/MS): $m / \mathrm{z}(\%)=280(15)[\mathrm{M}]^{+}, 263$ (15), 169 (35), 97 (15), 73 (100). HRMS (ESI): $\mathrm{m} / \mathrm{z}$ [ $\mathrm{M}-\mathrm{OH}]^{+}$calcd for $\mathrm{C}_{15} \mathrm{H}_{20} \mathrm{ClSi}$ : 263.1023; found: 263.1010.

Spectroscopic data for 4.12i: ${ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=7.21(\mathrm{~d}, J=8.5 \mathrm{~Hz}, 2 \mathrm{H}), 6.97$ $(\mathrm{d}, J=8.4 \mathrm{~Hz}, 2 \mathrm{H}), 4.82(\mathrm{tdt}, J=7.2,2.0,1.0 \mathrm{~Hz}, 1 \mathrm{H}), 4.00(\mathrm{ddt}, J=7.6,3.7,1.7 \mathrm{~Hz}, 1 \mathrm{H}), 2.09$ (ddd, $J=6.8,5.2,2.2 \mathrm{~Hz}, 2 \mathrm{H}), 1.97(\mathrm{dd}, J=1.7,1.0 \mathrm{~Hz}, 3 \mathrm{H}), 1.70(\mathrm{~s}, 1 \mathrm{H}),-0.06(\mathrm{~s}, 9 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR $\left(126 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta=153.1,145.1,140.7,131.5,128.7,128.4,82.0,53.6,45.1,14.3,-0.5 . \mathrm{IR}$ (FTIR, $\mathrm{cm}^{-1}$ ): $\tilde{\mathrm{v}}=3400,2957,1489,1249,1090,823$. HRMS (ESI): $\mathrm{m} / \mathrm{z}[\mathrm{M}-\mathrm{OH}]^{+}$calcd for $\mathrm{C}_{15} \mathrm{H}_{20} \mathrm{ClSi}$ : 263.1023; found: 263.1005.

### 4.7.4. The [1,2]-carbon-to-carbon silyl migration ( 1.0 mmol scale) - general procedure $H$



A dry 50 mL round-bottom flask fitted with a magnetic stir bar was charged with 193 mg ( 2.3 mmol, 2.3 equiv) of $\mathrm{NaHCO}_{3}$. The flask was sealed with a rubber septum and kept under positive atmosphere of nitrogen gas. An amount corresponding to 1.0 mmol of alcohol from procedure I (starting material) was dissolved in 5 mL of dry $\mathrm{CH}_{2} \mathrm{Cl}_{2}$, and the solution transferred into the flask via syringe. An additional 5 mL of dry $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ was added into the flask via syringe. Into a separate vial, 302.5 mg of $77 \% \mathrm{w} / \mathrm{w} \mathrm{m}$-CPBA ( $1.35 \mathrm{mmol}, 1.35$ equiv) was weighed and 5 mL dry $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ transferred into the vial to dissolve the $m$ - CPBA . The resulting solution was transferred into the flask via syringe. The resulting mixture (in the flask) was stirred for $1-3$ hours monitoring by TLC. Typically, formation of a white suspension indicated the end of the reaction. The reaction mixture was transferred into a separating funnel and diluted with $50 \mathrm{~mL} \mathrm{CH}_{2} \mathrm{Cl}_{2}$. The mixture was washed with 20 mL saturated $\mathrm{Na}_{2} \mathrm{SO}_{3(\mathrm{aq})}, 20 \mathrm{~mL}$ saturated $\mathrm{NaHCO}_{3(\mathrm{aq})}$ and 20 mL water respectively. The aqueous layers were combined and extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(30 \mathrm{~mL} \mathrm{X} 2)$. All the organic layers were combined and washed with 20 mL saturated $\mathrm{NaCl}_{(\mathrm{aq})}$ and dried over anhydrous $\mathrm{Na}_{2} \mathrm{SO}_{4}$. Filtration and concentration of the filtrate under reduced pressure afforded the desired product. In most cases, purification by column chromatography was not necessary.

## Synthesis of 3-hydroxy-2-methyl-5-phenyl-2-(trimethylsilyl)cyclopentan-1-one (4.8a)



Applying general procedure H to $\mathrm{NaHCO}_{3},(6.5 \mathrm{mg}, 0.09 \mathrm{mmol}, 1.2$ equiv.), 2-methyl-5-phenyl-1-(trimethylsilyl)cyclopent-2-en-1-ol 4.6a ( $16 \mathrm{mg}, 0.064 \mathrm{mmol}, 1.0$ equiv), $m$-CPBA ( $77 \%$ $\mathrm{w} / \mathrm{w}, 15.7 \mathrm{mg}, 0.08 \mathrm{mmol}, 1.1$ equiv.), and DCM ( 2 mL ) afforded $14 \mathrm{mg}, 0.053 \mathrm{mmol}$ ( $83 \%$ crude yield) of 4.8a as a white solid. No further purification was necessary judging by the ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$

NMR analysis of the crude reaction mixture. The crystal structure of compound 4.8a was solved by X-ray crystallography and the results deposited to the Cambridge Crystallographic Data Centre and assigned CCDC 2175610. ${ }^{1} \mathrm{H}$ NMR ( $600 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=7.35-7.31(\mathrm{~m}, 2 \mathrm{H}), 7.31-7.28$ (m, 2H), $7.25-7.22(\mathrm{~m}, 1 \mathrm{H}), 4.57(\mathrm{~d}, J=4.1 \mathrm{~Hz}, 1 \mathrm{H}), 3.78(\mathrm{dd}, J=12.6,8.5 \mathrm{~Hz}, 1 \mathrm{H}), 2.52$ (ddd, $J=13.6,8.6,1.3 \mathrm{~Hz}, 1 \mathrm{H}), 2.42(\mathrm{ddd}, J=13.6,12.5,4.2 \mathrm{~Hz}, 1 \mathrm{H}), 1.75(\mathrm{~s}, 1 \mathrm{H}), 1.30(\mathrm{~s}, 3 \mathrm{H}), 0.06$ (s, 9H). ${ }^{13} \mathrm{C}$ NMR (151 MHz, $\mathrm{CDCl}_{3}$ ): $\delta=218.1,137.0,128.4,127.8,126.7,72.5,50.7,50.1,36.7$, 13.8, -2.3.

Synthesis of epi-3-hydroxy-2-methyl-5-phenyl-2-(trimethylsilyl)cyclopentan-1-one (4.8a')


Applying general procedure H to $\mathrm{NaHCO}_{3},(96.6 \mathrm{mg}, 1.15 \mathrm{mmol}, 2.30$ equiv.), epi-2-methyl-5-phenyl-1-(trimethylsilyl)cyclopent-2-en-1-ol 4.6a' ( $123.2 \mathrm{mg}, 0.5 \mathrm{mmol}, 1.0$ equiv), $m$-CPBA ( $77 \% \mathrm{w} / \mathrm{w}, 151.3 \mathrm{mg}, 0.68 \mathrm{mmol}, 1.35$ equiv.) , and $\operatorname{DCM}(8 \mathrm{~mL})$ afforded after workup 130 mg , 0.495 mmol ( $99 \%$ crude yield) $\mathbf{4 . 8} \mathbf{a}^{\prime}$ as a white solid. The crystal structure of compound $\mathbf{4 . 8} \mathbf{a}^{\prime}$ was solved by X-ray crystallography and the results deposited to the Cambridge Crystallographic Data Centre and assigned CCDC 2158501. ${ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=7.35-7.28(\mathrm{~m}, 2 \mathrm{H}), 7.27$ $-7.20(\mathrm{~m}, 3 \mathrm{H}), 4.54(\mathrm{dd}, J=5.7,4.8 \mathrm{~Hz}, 1 \mathrm{H}), 3.39(\mathrm{dd}, J=9.9,8.2 \mathrm{~Hz}, 1 \mathrm{H}), 2.75(\mathrm{ddd}, J=13.9$, $9.9,5.8 \mathrm{~Hz}, 1 \mathrm{H}), 2.19(\mathrm{ddd}, J=13.9,8.2,4.8 \mathrm{~Hz}, 1 \mathrm{H}), 1.24(\mathrm{~s}, 3 \mathrm{H}), 0.14(\mathrm{~s}, 9 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR (126 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta=218.3,139.2,128.6,128.4,126.7,72.7,53.9,49.8,38.6,12.2,-3.6 .{ }^{29} \mathrm{Si}$ NMR (99 MHz, $\mathrm{CDCl}_{3}$ ): $\delta=6.79 . \mathrm{HRMS}(\mathrm{ESI}): m / z[\mathrm{M}+\mathrm{Na}]^{+}$calcd for $\mathrm{C}_{15} \mathrm{H}_{22} \mathrm{NaO}_{2} \mathrm{Si}: 285.1287$; found: 285.1282 .

## Synthesis of 5-(4-chlorophenyl)-3-hydroxy-2-(trimethylsilyl)cyclopentan-1-one (4.8b)



Applying general procedure H to $\mathrm{NaHCO}_{3},(116 \mathrm{mg}, 1.38 \mathrm{mmol}, 2.30$ equiv.), 5-(4-chlorophenyl)-1-(trimethylsilyl)cyclopent-2-en-1-ol 4.6b (160 mg, $0.6 \mathrm{mmol}, 1.0$ equiv), $m$-CPBA ( $77 \% \mathrm{w} / \mathrm{w}, 182 \mathrm{mg}, 0.81 \mathrm{mmol}, 1.35$ equiv.), and $\mathrm{DCM}(10 \mathrm{~mL})$ afforded $170 \mathrm{mg}, 0.6 \mathrm{mmol}(100 \%$ crude yield) of $\mathbf{4 . 8 b}$ as a white solid. No further purification was necessary judging by the ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR analysis of the crude reaction mixture. ${ }^{1} \mathrm{H}$ NMR $\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta=7.29(\mathrm{~d}, J=8.5$ $\mathrm{Hz}, 2 \mathrm{H}), 7.16(\mathrm{~d}, J=8.3 \mathrm{~Hz}, 2 \mathrm{H}), 4.68(\mathrm{~d}, J=3.9 \mathrm{~Hz}, 1 \mathrm{H}), 3.79(\mathrm{dd}, J=12.5,8.3 \mathrm{~Hz}, 1 \mathrm{H}), 2.49$ (dddd, $J=13.6,8.2,3.0,1.2 \mathrm{~Hz}, 1 \mathrm{H}), 2.30(\mathrm{dd}, J=3.0,1.6 \mathrm{~Hz}, 1 \mathrm{H}), 2.26(\mathrm{ddd}, J=13.6,12.7,4.1$ $\mathrm{Hz}, 1 \mathrm{H}), 2.03(\mathrm{~s}, 1 \mathrm{H}), 0.12(\mathrm{~s}, 9 \mathrm{H}) .{ }^{13} \mathrm{C} \operatorname{NMR}\left(126 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta=215.5,135.6,132.7,129.3$, 128.6, 70.2, 54.0, 50.4, 38.9, -1.6. MS (GC/MS): $m / \mathrm{z}(\%)=282(0.08)[\mathrm{M}]^{+}, 192(45), 157(20)$, 129 (100).

Synthesis 5-(4-trifluoromethylphenyl)-3-hydroxy-2-(trimethylsilyl)cyclopentan-1-one (4.8c)


Applying general procedure H to $\mathrm{NaHCO}_{3}$, ( $193 \mathrm{mg}, 2.3 \mathrm{mmol}, 2.30$ equiv.), 5-(4-trifluoromethylphenyl)-1-(trimethylsilyl)cyclopent-2-en-1-ol 4.6c ( $301 \mathrm{mg}, 1.0 \mathrm{mmol}, 1.0$ equiv), $m$-CPBA ( $77 \% \mathrm{w} / \mathrm{w}, 303 \mathrm{mg}, 1.35 \mathrm{mmol}, 1.35$ equiv.), and DCM ( 10 mL ) afforded $309 \mathrm{mg}, 0.98$ mmol ( $98 \%$ crude yield) of $\mathbf{4 . 8 c}$ as a white solid. No further purification was necessary judging by the ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR analysis of the crude reaction mixture. ${ }^{1} \mathrm{H}$ NMR $\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta=7.58$
$(\mathrm{d}, J=8.1 \mathrm{~Hz}, 2 \mathrm{H}), 7.34(\mathrm{~d}, J=8.1 \mathrm{~Hz}, 2 \mathrm{H}), 4.70(\mathrm{~d}, J=3.8 \mathrm{~Hz}, 1 \mathrm{H}), 3.89(\mathrm{dd}, J=12.8,8.2 \mathrm{~Hz}$, $1 \mathrm{H}), 2.53$ (dddd, $J=13.5,8.2,3.0,1.1 \mathrm{~Hz}, 1 \mathrm{H}), 2.36-2.28(\mathrm{~m}, 2 \mathrm{H}), 2.15(\mathrm{~s}, 1 \mathrm{H}), 0.13(\mathrm{~s}, 9 \mathrm{H})$. ${ }^{13} \mathrm{C}$ NMR ( $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=215.1,141.2(\mathrm{~d}, J=1.5 \mathrm{~Hz}), 129.1(\mathrm{q}, J=32.4 \mathrm{~Hz}), 128.3,125.4$ $(\mathrm{q}, J=3.8 \mathrm{~Hz}), 70.2,54.1,50.9,38.7,-1.6 . \mathrm{MS}(\mathrm{GC} / \mathrm{MS}): m / \mathrm{z}(\%)=316(0.12)[\mathrm{M}]^{+}, 240(35)$, 212 (100), 115 (35)

Synthesis of 3-hydroxy-5-(naphthalen-1-yl)-2-(trimethylsilyl)cyclopentan-1-one (4.8d)


Applying general procedure H to $\mathrm{NaHCO}_{3},(193 \mathrm{mg}, 2.3 \mathrm{mmol}, 2.30$ equiv.), 5 -(naphthalen-1-yl)-1-(trimethylsilyl)cyclopent-2-en-1-ol 4.6d ( $283 \mathrm{mg}, 1.0 \mathrm{mmol}, 1.0$ equiv), $m$-CPBA ( $77 \%$ $\mathrm{w} / \mathrm{w}, 300 \mathrm{mg}, 1.34 \mathrm{mmol}, 1.34$ equiv.), and DCM ( 10 mL ) afforded $285 \mathrm{mg}, 0.96 \mathrm{mmol}(96 \%$ crude yield) of $\mathbf{4 . 8 d}$ as a white solid. No further purification was necessary judging by the ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR analysis of the crude reaction mixture. ${ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=8.04(\mathrm{~d}, J=8.5$ $\mathrm{Hz}, 1 \mathrm{H}), 7.85(\mathrm{dd}, J=7.4,2.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.76(\mathrm{~d}, J=8.2 \mathrm{~Hz}, 1 \mathrm{H}), 7.54-7.46(\mathrm{~m}, 2 \mathrm{H}), 7.45-7.40$ $(\mathrm{m}, 1 \mathrm{H}), 7.21(\mathrm{~d}, J=7.1 \mathrm{~Hz}, 1 \mathrm{H}), 4.71-4.62(\mathrm{~m}, 2 \mathrm{H}), 2.63(\mathrm{dddd}, J=13.6,8.2,3.0,1.2 \mathrm{~Hz}, 1 \mathrm{H})$, $2.40(\mathrm{dd}, J=3.1,1.6 \mathrm{~Hz}, 1 \mathrm{H}), 2.30(\mathrm{td}, J=13.2,4.1 \mathrm{~Hz}, 1 \mathrm{H}), 2.22(\mathrm{~s}, 1 \mathrm{H}), 0.20(\mathrm{~s}, 8 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=216.5,134.4,134.0,132.7,128.7,127.5,126.0,125.7,125.3,123.9,123.8$, $70.5,54.6,47.8,40.2,-1.5 . \mathrm{MS}(\mathrm{GC} / \mathrm{MS}): m / \mathrm{z}(\%)=298(0.03)[\mathrm{M}]^{+}, 208(100), 179(80), 165$ (70), 152 (45), 128 (20), 89 (30). HRMS (ESI): $m / z[\mathrm{M}-\mathrm{OH}]^{+}$calcd for $\mathrm{C}_{18} \mathrm{H}_{21} \mathrm{OSi}$ 281.1356; found: 281.1360 .

## Synthesis of 5-([1,1'-biphenyl]-4-yl)-3-hydroxy-2-(trimethylsilyl)cyclopentan-1-one (4.8e)



Applying general procedure H to $\mathrm{NaHCO}_{3},(8.7 \mathrm{mg}, 0.1 \mathrm{mmol}, 1.2$ equiv.), 5-([1,1'-biphenyl]-4-yl)-1-(trimethylsilyl)cyclopent-2-en-1-ol (4.6e) ( $26.5 \mathrm{mg}, 0.086 \mathrm{mmol}, 1.0$ equiv), $m$-CPBA ( $77 \% \mathrm{w} / \mathrm{w}, 21.2 \mathrm{mg}, 0.095 \mathrm{mmol}, 1.1$ equiv.), and DCM ( 10 mL ) afforded $25.1 \mathrm{mg}, 0.077 \mathrm{mmol}$ ( $90 \%$ crude yield) of $\mathbf{4 . 8 d}$ as a white solid. No further purification was necessary judging by the ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR analysis of the crude reaction mixture. ${ }^{1} \mathrm{H}$ NMR ( $600 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=7.58-$ $7.55(\mathrm{~m}, 4 \mathrm{H}), 7.45-7.41(\mathrm{~m}, 2 \mathrm{H}), 7.36-7.32(\mathrm{~m}, 1 \mathrm{H}), 7.30(\mathrm{~d}, J=8.1 \mathrm{~Hz}, 2 \mathrm{H}), 4.73(\mathrm{~d}, J=3.9$ $\mathrm{Hz}, 1 \mathrm{H}), 3.89(\mathrm{dd}, J=12.6,8.2 \mathrm{~Hz}, 1 \mathrm{H}), 2.58-2.52(\mathrm{~m}, 1 \mathrm{H}), 2.38(\mathrm{td}, J=13.1,4.1 \mathrm{~Hz}, 1 \mathrm{H}), 2.35$ - $2.32(\mathrm{~m}, 1 \mathrm{H}), 1.74(\mathrm{~s}, 1 \mathrm{H}), 0.17(\mathrm{~s}, 9 \mathrm{H}) .{ }^{13} \mathrm{C} \operatorname{NMR}\left(151 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta=215.7,140.9,139.8$, $136.4,128.7,128.4,127.3,127.2,127.0,70.5,54.0,51.0,39.3,-1.5$.

Synthesis of 3-hydroxy-5-(naphthalen-2-yl)-2-(triethylsilyl)cyclopentan-1-one (4.8f)


Applying general procedure H to $\mathrm{NaHCO}_{3},(174 \mathrm{mg}, 2.07 \mathrm{mmol}, 2.30$ equiv.), 5 -(naphthalen-2-yl)-1-(triethylsilyl)cyclopent-2-en-1-ol 4.6 f ( $292 \mathrm{mg}, 0.9 \mathrm{mmol}, 1.0$ equiv), $m$-CPBA ( $77 \% \mathrm{w} / \mathrm{w}$, $274 \mathrm{mg}, 1.22 \mathrm{mmol}, 1.35$ equiv.), and $\mathrm{DCM}(10 \mathrm{~mL})$ afforded 308 mg (quantitative yield) of $\mathbf{4 . 8 f}$ as a white solid. No further purification was necessary judging by the ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR analysis of the crude reaction mixture. ${ }^{1} \mathrm{H} \operatorname{NMR}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta=7.87-7.77(\mathrm{~m}, 3 \mathrm{H}), 7.67(\mathrm{~s}, 1 \mathrm{H})$,
$7.51-7.43(\mathrm{~m}, 2 \mathrm{H}), 7.41(\mathrm{~d}, J=8.5 \mathrm{~Hz}, 1 \mathrm{H}), 4.73(\mathrm{~d}, J=4.0 \mathrm{~Hz}, 1 \mathrm{H}), 4.00(\mathrm{dd}, J=12.7,8.2 \mathrm{~Hz}$, $1 \mathrm{H}), 2.60-2.51(\mathrm{~m}, 1 \mathrm{H}), 2.49-2.41(\mathrm{~m}, 2 \mathrm{H}), 2.14-1.98(\mathrm{~m}, 1 \mathrm{H}), 0.97(\mathrm{t}, J=7.9 \mathrm{~Hz}, 9 \mathrm{H}), 0.68$ $(\mathrm{q}, J=7.7 \mathrm{~Hz}, 6 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=216.2,134.8,133.3,132.4,128.0,127.7$, $127.5,126.5,126.3,125.9,125.6,70.3,51.2,50.7,38.9,7.3,3.3$. IR (FTIR, $\mathrm{cm}^{-1}$ ): $\tilde{\mathrm{v}}=3463,2952$, 2933, 2899, 2873, 1697, 1415, 1242, 1115, 1011, 802, 740. HRMS (ESI): $m / z[\mathrm{M}+\mathrm{H}]^{+}$calcd for $\mathrm{C}_{21} \mathrm{H}_{29} \mathrm{O}_{2} \mathrm{Si}$ : 341.1937; found: 341.1934.

## Synthesis of 5-(4-chlorophenyl)-2-(dimethyl(phenyl)silyl)-3-hydroxycyclopentan-1-one (4.8g)



Applying general procedure H to $\mathrm{NaHCO}_{3},(174 \mathrm{mg}, 2.07 \mathrm{mmol}, 2.30$ equiv.), 5-(4-chlorophenyl)-1-(dimethyl(phenyl)silyl)cyclopent-2-en-1-ol 4.6 g ( $296 \mathrm{mg}, 0.9 \mathrm{mmol}, 1.0$ equiv), $m$-CPBA ( $77 \% \mathrm{w} / \mathrm{w}, 274 \mathrm{mg}, 1.22 \mathrm{mmol}, 1.35$ equiv.), and DCM ( 10 mL ) afforded $307 \mathrm{mg}, 0.99$ mmol ( $99 \%$ crude yield) of $\mathbf{4 . 8 g}$ as a white solid. Attempted further purification by column chromatography, $R_{f}=0.6$ ( $50 \%$ EtOAc in hexanes) led to some of the product to undergo epimerization at the carbon bearing the silyl group followed by subsequent Peterson olefination.

Spectroscopic data for $\mathbf{4 . 8 g}:{ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=7.46-7.40(\mathrm{~m}, 3 \mathrm{H}), 7.38-7.32$ $(\mathrm{m}, 2 \mathrm{H}), 7.17(\mathrm{~d}, J=8.5 \mathrm{~Hz}, 2 \mathrm{H}), 6.73(\mathrm{~d}, J=8.5 \mathrm{~Hz}, 2 \mathrm{H}), 4.68(\mathrm{~d}, J=3.6 \mathrm{~Hz}, 1 \mathrm{H}), 3.70(\mathrm{dd}, J=$ $12.7,8.1 \mathrm{~Hz}, 1 \mathrm{H}), 2.47(\mathrm{dd}, J=3.0,1.5 \mathrm{~Hz}, 1 \mathrm{H}), 2.29(\mathrm{dddd}, J=13.5,8.2,3.0,1.3 \mathrm{~Hz}, 1 \mathrm{H}), 1.95$ $(\mathrm{td}, J=13.1,4.1 \mathrm{~Hz}, 1 \mathrm{H}), 1.55(\mathrm{~s}, 1 \mathrm{H}), 0.49(\mathrm{~s}, 3 \mathrm{H}), 0.45(\mathrm{~s}, 3 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta$ $=215.1,135.4,135.3,134.0,132.5,130.0,129.3,128.4,128.3,70.3,53.6,50.7,38.3,-3.2,-3.6$. IR (FTIR, $\mathrm{cm}^{-1}$ ): $\tilde{\mathrm{v}}=3426,2957,1708,1491,1250,1044,789,699 . \operatorname{MS}(\mathrm{GC} / \mathrm{MS}): m / \mathrm{z}(\%)=344$ (0.02) $[\mathrm{M}]^{+}, 271$ (100), 193 (90), 89 (35).

Spectroscopic data for $\mathbf{4 . 1 2 g}$ (Peterson olefination product): ${ }^{1} \mathrm{H}$ NMR $\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta=$ $7.87(\mathrm{dt}, J=5.6,2.7 \mathrm{~Hz}, 1 \mathrm{H}), 7.30(\mathrm{~d}, J=8.5 \mathrm{~Hz}, 2 \mathrm{H}), 7.08(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 2 \mathrm{H}), 6.30(\mathrm{dt}, J=5.7$, $2.1 \mathrm{~Hz}, 1 \mathrm{H}), 3.53(\mathrm{dd}, J=7.0,2.6 \mathrm{~Hz}, 1 \mathrm{H}), 3.26(\mathrm{ddt}, J=19.6,7.1,2.5 \mathrm{~Hz}, 1 \mathrm{H}), 2.78(\mathrm{dq}, J=19.6$, $2.5 \mathrm{~Hz}, 1 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR (126 MHz, $\left.\mathrm{CDCl}_{3}\right): \delta=209.1,164.1,137.6,133.4,133.0,132.8,132.5$, 50.1, 38.5. MS (GC/MS): $m / \mathrm{z}(\%)=192(0.1)[\mathrm{M}]^{+}, 178(100), 115(65), 75(75)$.

Synthesis 3-hydroxy-5-(4-methoxyphenyl)-2-methyl-2-(trimethylsilyl)cyclopentan-1-one (4.8h)


Applying general procedure H to $\mathrm{NaHCO}_{3},(135 \mathrm{mg}, 1.61 \mathrm{mmol}, 2.30$ equiv.), 5-(4-methoxyphenyl)-2-methyl-1-(trimethylsilyl)cyclopent-2-en-1-ol $\mathbf{4 . 6 h}$ ( $194 \mathrm{mg}, 0.7 \mathrm{mmol}, 1.0$ equiv), $m$-CPBA ( $77 \% \mathrm{w} / \mathrm{w}, 212 \mathrm{mg}, 0.91 \mathrm{mmol}, 1.35$ equiv.), and $\mathrm{DCM}(10 \mathrm{~mL})$ afforded after column chromatography, $R_{f}=0.5(40 \%$ EtOAc in hexanes) $194 \mathrm{mg}, 0.67 \mathrm{mmol}(95 \%$ isolated yield) of $\mathbf{4 . 8 h}$ as a white solid. ${ }^{1} \mathrm{H}$ NMR ( $\left.500 \mathrm{MHz}, \mathrm{CDCl}_{3}, \mathrm{ppm}\right): \delta=7.21(\mathrm{~d}, J=8.8 \mathrm{~Hz}, 2 \mathrm{H})$, $6.87(\mathrm{~d}, J=8.8 \mathrm{~Hz}, 2 \mathrm{H}), 4.55(\mathrm{~d}, J=3.9 \mathrm{~Hz}, 1 \mathrm{H}), 3.78(\mathrm{~s}, 3 \mathrm{H}), 3.71(\mathrm{dd}, J=12.6,8.5 \mathrm{~Hz}, 1 \mathrm{H})$, $2.48(\mathrm{ddd}, J=13.7,8.6,1.3 \mathrm{~Hz}, 1 \mathrm{H}), 2.36(\mathrm{ddd}, J=13.5,12.6,4.2 \mathrm{~Hz}, 1 \mathrm{H}), 1.93(\mathrm{~s}, 1 \mathrm{H}), 1.29(\mathrm{~s}$, $3 \mathrm{H}), 0.05(\mathrm{~s}, 9 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=218.6,158.3,129.1,128.7,113.8,72.4,55.2$, 50.6, 49.4, 36.8, 13.8, -2.3. HRMS (ESI): $\mathrm{m} / \mathrm{z}[\mathrm{M}+\mathrm{H}]^{+}$calcd for $\mathrm{C}_{16} \mathrm{H}_{25} \mathrm{O}_{3} \mathrm{Si}: 293.1573$; found: 293.1567.

Synthesis of 5-(4-chlorophenyl)-3-hydroxy-2-methyl-2-(trimethylsilyl)cyclopentan-1-one (4.8i)


Applying general procedure H to $\mathrm{NaHCO}_{3}$, $(212 \mathrm{mg}, 2.53 \mathrm{mmol}, 2.30$ equiv.), 5-(4-chlorophenyl)-2-methyl-1-(trimethylsilyl)cyclopent-2-en-1-ol $4.6 \mathbf{( 3 0 9} \mathrm{mg}, 1.1 \mathrm{mmol}, 1.0$ equiv), $m-\operatorname{CPBA}(77 \% \mathrm{w} / \mathrm{w}, 333 \mathrm{mg}, 1.49 \mathrm{mmol}, 1.35$ equiv. $)$, and $\mathrm{DCM}(10 \mathrm{~mL})$ afforded $321 \mathrm{mg}, 1.08$ $\mathrm{mmol}\left(98 \%\right.$ crude yield) of $\mathbf{4 . 8 i}$ as a white solid. ${ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=7.30(\mathrm{~d}, J=8.6$ $\mathrm{Hz}, 2 \mathrm{H}), 7.25(\mathrm{~d}, J=8.6 \mathrm{~Hz}, 2 \mathrm{H}), 4.56(\mathrm{~d}, J=3.9 \mathrm{~Hz}, 1 \mathrm{H}), 3.74(\mathrm{dd}, J=12.6,8.5 \mathrm{~Hz}, 1 \mathrm{H}), 2.51$ (ddd, $J=13.6,8.5,1.2 \mathrm{~Hz}, 1 \mathrm{H}), 2.36(\mathrm{ddd}, J=13.5,12.6,4.2 \mathrm{~Hz}, 1 \mathrm{H}), 1.78(\mathrm{~s}, 1 \mathrm{H}), 1.29(\mathrm{~s}, 3 \mathrm{H})$, $0.04(\mathrm{~s}, 9 \mathrm{H}) .{ }^{13} \mathrm{C}\left\{{ }^{1} \mathrm{H}\right\}$ NMR ( $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=217.6,135.4,132.5,129.0,128.5,72.3,50.9$, 49.3, 36.4, 13.7, -2.3. IR (FTIR, $\mathrm{cm}^{-1}$ ): $\tilde{\mathrm{v}}=3435,2975,2953,2895,1696,1487,1444,1405,1251$, 1012, 836, 820. HRMS (ESI): $m / z[\mathrm{M}+\mathrm{H}]^{+}$calcd for $\mathrm{C}_{15} \mathrm{H}_{22} \mathrm{ClO}_{2}$ Si: 297.1078; found: 297.1077. Synthesis of 6-(4-chlorophenyl)-3-hydroxy-2-(trimethylsilyl)cyclohexan-1-one (4.9a)


Applying general procedure H to $\mathrm{NaHCO}_{3}$, (193 mg, $2.3 \mathrm{mmol}, 2.30$ equiv.), 4'-chloro-2-(trimethylsilyl)-1,2,5,6-tetrahydro-[1,1'-biphenyl]-2-ol 2.5b (281 mg, $1.0 \mathrm{mmol}, 1.0$ equiv), $m$ CPBA ( $77 \% \mathrm{w} / \mathrm{w}, 303 \mathrm{mg}, 1.35 \mathrm{mmol}, 1.35$ equiv.), and DCM ( 10 mL ) afforded after workup and recrystallization ( $50 \%$ EtOAc in hexanes) $321 \mathrm{mg}, 0.93 \mathrm{mmol}$ ( $93 \%$ crude yield) of 4.9 a as white
crystals. The crystal structure of compound 4.9a was solved by X-ray crystallography and the results deposited to the Cambridge Crystallographic Data Centre and assigned CCDC 1890864. ${ }^{1} \mathrm{H} \operatorname{NMR}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta=7.30(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 2 \mathrm{H}), 7.06(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 2 \mathrm{H}), 4.53(\mathrm{dq}, J=$ $3.3,1.6 \mathrm{~Hz}, 1 \mathrm{H}), 3.32(\mathrm{dd}, J=13.2,5.2 \mathrm{~Hz}, 1 \mathrm{H}), 2.62(\mathrm{t}, J=2.0 \mathrm{~Hz}, 1 \mathrm{H}), 2.53-2.42(\mathrm{~m}, 1 \mathrm{H})$, $2.11-1.97(\mathrm{~m}, 3 \mathrm{H}), 1.95(\mathrm{~s}, 1 \mathrm{H}), 0.18(\mathrm{~s}, 9 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=208.8,137.8$, 132.6, 130.1, 128.5, 70.3, 57.1, 56.6, 30.8, 28.0, -1.4.

## Synthesis of 3-hydroxy-6-(4-methoxyphenyl)-2-(trimethylsilyl)cyclohexan-1-one (4.9b)



Applying general procedure H to $\mathrm{NaHCO}_{3}$, ( $58 \mathrm{mg}, 0.7 \mathrm{mmol}, 2.30$ equiv.), 6-(methoxyphenyl-2-yl)-1-(trimethylsilyl)cyclohex-2-en-1-ol (2.5c), $d r=2: 1,(83 \mathrm{mg}, 0.3 \mathrm{mmol}, 1.0$ equiv), $m$-CPBA ( $77 \% \mathrm{w} / \mathrm{w}, 92 \mathrm{mg}, 0.4 \mathrm{mmol}, 1.35$ equiv.), and DCM ( 5 mL ) afforded after workup 89 mg (quantitative crude yield) of $\mathbf{4 . 9 b}$ as a mixture of diastereomers ( $d r=2: 1$ ). ${ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=7.15(\mathrm{~d}, J=8.7 \mathrm{~Hz}, 2 \mathrm{H}), 7.05(\mathrm{~d}, J=8.7 \mathrm{~Hz}, 2 \mathrm{H}), 6.90-6.83(\mathrm{~m}, 4 \mathrm{H})$, $4.53(\mathrm{~h}, J=1.5 \mathrm{~Hz}, 1 \mathrm{H}), 4.27(\mathrm{dt}, J=5.8,4.1 \mathrm{~Hz}, 1 \mathrm{H}), 3.79(\mathrm{~s}, 3 \mathrm{H}), 3.79(\mathrm{~s}, 3 \mathrm{H}), 3.67(\mathrm{dd}, J=8.0$, $6.3 \mathrm{~Hz}, 1 \mathrm{H}), 3.30(\mathrm{dd}, J=13.2,5.2 \mathrm{~Hz}, 1 \mathrm{H}), 2.62(\mathrm{t}, J=1.9 \mathrm{~Hz}, 1 \mathrm{H}), 2.52-2.43(\mathrm{~m}, 1 \mathrm{H}), 2.38$ $(\mathrm{dt}, J=4.6,1.4 \mathrm{~Hz}, 1 \mathrm{H}), 2.37-2.30(\mathrm{~m}, 1 \mathrm{H}), 2.20-2.12(\mathrm{~m}, 1 \mathrm{H}), 2.12-1.85(\mathrm{~m}, 7 \mathrm{H}), 0.18(\mathrm{~s}$, $9 \mathrm{H}),-0.03(\mathrm{~s}, 9 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=210.6,209.5,158.5,158.4,131.4,129.9$, $129.7,129.1,113.84,113.77,70.5,70.2,57.1,56.4,55.22,55.20,54.4,52.9,31.0,30.6,28.2,24.3$, -1.37, -1.4. HRMS (ESI): $m / z[\mathrm{M}+\mathrm{H}]^{+}$calcd for $\mathrm{C}_{16} \mathrm{H}_{25} \mathrm{O}_{3} \mathrm{Si}$ : 293.1573; found: 293.1571.

Synthesis of 3-hydroxy-6-(4-methoxyphenyl-3-d)-2-(trimethylsilyl)cyclohexan-1-one (4.9b- $d_{1}$ )


Applying general procedure H to $\mathrm{NaHCO}_{3},(135 \mathrm{mg}, 1.6 \mathrm{mmol}, 2.30$ equiv.), 4'-methoxy-2-(trimethylsilyl)-1,2,5,6-tetrahydro-[1,1'-biphenyl]-3'- $d$-2-ol $\left(\mathbf{2 . 5 c}-\boldsymbol{d}_{\boldsymbol{1}}\right), d r=12: 1$, $(194 \mathrm{mg}, 0.7$ mmol, 1.0 equiv), $m$-CPBA ( $77 \% \mathrm{w} / \mathrm{w}, 212 \mathrm{mg}, 0.95 \mathrm{mmol}, 1.35$ equiv.), and DCM ( 10 mL ) afforded after workup $199 \mathrm{mg}, 0.679 \mathrm{mmol}$ ( $97 \%$ crude yield) of $\mathbf{4 . 9 \mathrm { b }}-\boldsymbol{d}_{\mathbf{l}}, d r=12: 1$. A portion of the crude material was recrystallized in a mixture of hexanes and EtOAc (1:1) to give a single diastereomer of $\mathbf{4 . 9 b}-\boldsymbol{d}_{\boldsymbol{1}} .{ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=7.05(\mathrm{dq}, J=4.4,2.3 \mathrm{~Hz}, 2 \mathrm{H}), 6.87(\mathrm{~d}$, $J=9.1 \mathrm{~Hz}, 1 \mathrm{H}), 4.52(\mathrm{dq}, J=3.3,1.5 \mathrm{~Hz}, 1 \mathrm{H}), 3.79(\mathrm{~s}, 3 \mathrm{H}), 3.30(\mathrm{dd}, J=13.2,5.3 \mathrm{~Hz}, 1 \mathrm{H}), 2.62$ $(\mathrm{t}, J=2.0 \mathrm{~Hz}, 1 \mathrm{H}), 2.55-2.41(\mathrm{~m}, 1 \mathrm{H}), 2.23-1.93(\mathrm{~m}, 4 \mathrm{H}), 0.18(\mathrm{~s}, 9 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( 126 MHz, $\left.\mathrm{CDCl}_{3}\right): \delta=209.6,158.3,131.4,129.7,129.6,113.8,70.4,57.1,56.4,55.2,30.9,28.2,-1.4 .{ }^{29} \mathrm{Si}$ NMR (99 MHz, $\mathrm{CDCl}_{3}$ ): $\delta=$ 2.24. $\mathrm{HRMS}(\mathrm{ESI}): m / z[\mathrm{M}+\mathrm{H}]^{+}$calcd for $\mathrm{C}_{16} \mathrm{H}_{24} \mathrm{DO}_{3} \mathrm{Si}: 294.1630$; found: 294.1628.

Synthesis of 3-hydroxy-6-(naphthalen-2-yl)-2-(trimethylsilyl)cyclohexan-1-one (4.9c)


Applying general procedure H to $\mathrm{NaHCO}_{3}$, ( $155 \mathrm{mg}, 1.84 \mathrm{mmol}, 2.30$ equiv.), 6-(naphthalen-2-yl)-1-(trimethylsilyl)cyclohex-2-en-1-ol (2.5d), $d r=4: 1,(237 \mathrm{mg}, 0.8 \mathrm{mmol}, 1.0$ equiv), $m$ -

CPBA ( $77 \% \mathrm{w} / \mathrm{w}, 242 \mathrm{mg}, 1.08 \mathrm{mmol}, 1.35$ equiv.), and DCM ( 10 mL ) afforded after workup 242 $\mathrm{mg}, 0.784 \mathrm{mmol}(98 \%$ crude yield) of 4.9 c as a mixture of diastereomers $(d r=4: 1)$. A portion of the crude material was recrystallized in a mixture of ethyl acetate and hexanes (1:1). The resulting crystals were for the major diastereomer which crushed out as brown crystals of a single diastereomer. ${ }^{1} \mathrm{H}$ NMR $\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta=7.86-7.75(\mathrm{~m}, 3 \mathrm{H}), 7.59(\mathrm{~s}, 1 \mathrm{H}), 7.49-7.41(\mathrm{~m}$, $2 \mathrm{H}), 7.28(\mathrm{dd}, J=8.5,1.7 \mathrm{~Hz}, 1 \mathrm{H}), 4.57(\mathrm{t}, J=2.2 \mathrm{~Hz}, 1 \mathrm{H}), 3.53(\mathrm{dd}, J=13.1,5.5 \mathrm{~Hz}, 1 \mathrm{H}), 2.67$ $(\mathrm{d}, J=1.8 \mathrm{~Hz}, 1 \mathrm{H}), 2.62(\mathrm{td}, J=12.9,4.6 \mathrm{~Hz}, 1 \mathrm{H}), 2.18-2.02(\mathrm{~m}, 3 \mathrm{H}), 1.88(\mathrm{~s}, 1 \mathrm{H}), 0.22(\mathrm{~s}, 9 \mathrm{H})$. ${ }^{13}{ }^{13}$ NMR (126 MHz, $\left.\mathrm{CDCl}_{3}\right): \delta=209.1,137.1,133.5,132.6,127.8,127.7,127.6,127.23,127.19$, $125.8,125.5,70.5,57.4,57.2,30.9,28.1,-1.3$. IR (FTIR, $\left.\mathrm{cm}^{-1}\right): \tilde{v}=3434,3055,3021,2938,2923$, 2882, 2863, 1665, 1336, 1287, 1251, 1147, 1074, 956, 836, 740. HRMS (ESI): $m / z[\mathrm{M}+\mathrm{H}]^{+}$calcd for $\mathrm{C}_{19} \mathrm{H}_{25} \mathrm{O}_{2} \mathrm{Si}$ : 313.1624; found: 313.1623.

## REFERENCES

(1) Brook, A. G. Isomerism of some $\alpha$-hydroxysilanes to silyl ethers. J. Am. Chem. Soc. 1958, 80, 1886.
(2) Brook, A. G. Molecular rearrangements of organosilicon compounds. Acc. Chem. Res. 1974, 7, 77.
(3) Page, P. C. B.; Klair, S. S.; Rosenthal, S. Synthesis and chemistry of acyl silanes. Chem. Soc. Rev. 1990, 19, 147.
(4) West, R.; Gornowicz, G. A. New anionic rearrangements XIII. Reactions of tertbutyllithium with organosilanes. J. Organomet. Chem. 1971, 28, 25.
(5) Thadani, A. N.; Huang, Y.; Rawal, V. H. Expedient, high-yielding synthesis of silylsubstituted salen ligands. Org. Lett. 2007, 9, 3873.
(6) He, Y.; Hu, H.; Xie, X.; She, X. Regioselective lithiation/retro-Brook rearrangement via direct deprotonation. Tetrahedron 2013, 69, 559.
(7) He, Y.; Ma, B.; Yang, J.; Xie, X.; She, X. Regioselective lateral or vinyl C-H lithiation/1,5-retro-Brook rearrangement via quinolyl or pyridyl ring directed deprotonation. Tetrahedron 2013, 69, 5545.
(8) Kapeller, D. C.; Brecker, L.; Hammerschmidt, F. Configurational stability of oxymethyllithiums as intermediates in intramolecular rearrangements. Chem. Eur. J. 2007, 13, 9582.
(9) Bariak, V.; Malastova, A.; Almassy, A.; Sebesta, R. Retro-Brook rearrangement of ferrocene-derived silyl ethers. Chem. Eur. J. 2015, 21, 13445.
(10) Hudrlik, P. F.; Hudrlik, A. M.; Kulkarni, A. K. Elimination reactions of $\alpha, \beta-$ dihydroxysilanes: Stereospecific synthesis of silyl enol ethers from vinylsilanes. J. Am. Chem. Soc. 1985, 107, 4260.
(11) Cunico, R. F. The silapinacol rearrangement: Conversion of $\alpha, \beta$-dihydroxysilanes into $\alpha$ silyl carbonyl compounds. Tetrahedron Lett. 1986, 27, 4269.
(12) Pornet, J. Migration 1,2 d'un groupe triméthylsilyle au niveau d'un cation vinylique: synthèse de diènes conjugués silylés à partir de bis(triméthylsilyl)-1,4 alcynes-2. J. Organomet. Chem. 1988, 340, 273.
(13) Miura, K.; Okajima, S.; Hondo, T.; Nakagawa, T.; Takahashi, T.; Hosomi, a. acidcatalyzed cyclization of vinylsilanes bearing a hydroxy group: A new method for stereoselective synthesis of disubstituted tetrahydrofurans. J. Am. Chem. Soc. 2000, 122, 11348.
(14) Miura, K.; Hondo, T.; Nakagawa, T.; Takahashi, T.; Hosomi, A. Acid-catalyzed cyclization of vinylsilanes bearing an amino group. Stereoselective synthesis of pyrrolidines. Org. Lett. 2000, 2, 385.
(15) Miura, K.; Hondo, T.; Okajima, S.; Nakagawa, T.; Takahashi, T.; Hosomi, A. 1, 2-silylmigrative cyclization of vinylsilanes bearing a hydroxy group: stereoselective synthesis of multisubstituted tetrahydropyrans and tetrahydrofurans1. J. Org. Chem. 2002, 67, 6082.
(16) Miura, K.; Takahashi, T.; Hondo, T.; Hosomi, A. 1,2-Silyl-migrative cyclization of vinylsilanes bearing an amino group. Chirality, 2003, 15, 41.
(17) Mergardt, M.; Weber, K.; Adiwidjaja, G.; Schaumann, E. Tandem silyl migrations leading to $\alpha, \beta$-bis-silylated enals and enones. Angew. Chem. Int. Ed. 1991, 30, 1687.
(18) Sakaguchi, K.; Higashino, M.; Ohfune, Y. Acid-catalyzed rearrangement of $\alpha$ hydroxytrialkylsilanes. Tetrahedron 2003, 59, 6647.
(19) Werner, H.; Baum, M.; Schneider, D.; Windmüller, B. Reactions of alkynylsilanes with [ $\left.\mathrm{RhCl}\left(\mathrm{PiPr}_{3}\right)_{2}\right]$ : The synthesis of four-coordinate (alkyne)-, alkynyl-, and vinylidenerhodium (I) and five-and six-coordinate alkynylhydridorhodium (III) complexes from $\mathrm{RC} \equiv \mathrm{CSiMe}_{3}$, Me3SiC $\equiv \mathrm{CSiMe}_{3}$, and $\mathrm{HC} \equiv \mathrm{CSiMe}_{3}$. Organometallics 1994, 13, 1089.
(20) Katayama, H.; Ozawa, F. Convenient routes to vinylideneruthenium dichlorides with basic and bulky tertiary phosphine ligands ( $\mathrm{PiPr}_{3}$ and $\mathrm{PCy}_{3}$ ). Organometallics 1998, 17, 5190.
(21) Huang, D.; Streib, W. E.; Bollinger, J. C.; Caulton, K. G.; Winter, R. F.; Scheiring, T. 14electron four-coordinate Ru (II) carbyl complexes and their five-coordinate precursors: Synthesis, double agostic interactions, and reactivity. J. Am. Chem. Soc. 1999, 121, 8087.
(22) Katayama, H.; Wada, C.; Taniguchi, K.; Ozawa, F. Effect of Substituents on the formation of vinylideneruthenium(II) complexes. X-ray structures of $\mathrm{RuCl}_{2}\{\overline{=}=\mathrm{C}(\mathrm{Z}) \mathrm{Ph}\}(\mathrm{dcpmp})$ $\left(\mathrm{Z}=\mathrm{H}, \mathrm{SiMe}_{3} ;\right.$ dcpmp $\left.=\mathrm{C}_{5} \mathrm{H}_{3} \mathrm{~N}\left(\mathrm{CH}_{2} \mathrm{PCy}_{2}\right)_{2}\right)$. Organometallics, 2002, 21, 3285.
(23) Seregin, I. V.; Gevorgyan, V. Gold-catalyzed 1,2-migration of silicon, tin, and germanium en route to C-2 substituted fused pyrrole-containing heterocycles. J. Am. Chem. Soc. 2006, 128, 12050.
(24) Rooke, D. A.; Ferreira, E. M. Stereoselective syntheses of trisubstituted olefins via platinum catalysis: $\alpha$-Silylenones with geometrical complementarity. J. Am. Chem. Soc. 2010, 132, 11926.
(25) Allegretti, P. A.; Ferreira, E. M. Generation of $\alpha, \beta$-unsaturated platinum carbenes from homopropargylic alcohols: Rearrangements to polysubstituted furans. Org. Lett. 2011, 13, 5924.

Mori-Quiroz, L. M.; Maleczka, R, E., Jr. Stereoconvergent [1,2]- and [1,4]-wittig rearrangements of 2-silyl-6-aryl-5,6-dihydropyrans: A tale of steric vs electronic regiocontrol of divergent pathways. J. Org. Chem. 2015, 80, 1163.
(27) Mori-Quiroz, L. M.; Maloba, E. W.; Maleczka, R. E., Jr. Silylcyclopropanes by selective [1,4]-Wittig rearrangement of 4-silyl-5,6-dihydropyrans. Org. Lett. 2021, 23, 5724.
(28) Scheller, M. E.; Schweizer, W. B.; Frei, B. Syntheses of $\alpha, \beta$-epoxy silyl ketones. Helv. Chim. Acta 1989, 72, 264.
(29) Fässler, J.; Enev, V.; Bienz, S. A novel stereoselective reaction cascade leading from $\alpha$ silylated allylic alcohols to aldol-type products. Helv. Chim. Acta 1999, 82, 561.
(30) Honda, M.; Nakamura, T.; Sumigawa, T.; Kunimoto, K.-K.; Segi, M. Stereoselective synthesis of $1,2,3$-triol derivatives from $\alpha, \beta$-unsaturated acylsilanes. Heteroat. Chem. 2014, 25, 565.
(31) Davies, S. G.; Fletcher, A. M.; Thomson, J. E. Hydrogen bond directed epoxidation: Diastereoselective olefinic oxidation of allylic alcohols and amines. Org. Biomol. Chem. 2014, 12, 4544.
(32) Brennan, M. B.; Davies, S. G.; Fletcher, A. M.; Lee, J. A.; Roberts, P. M.; Russell, A. J.; Thomson, J. E. Epoxidation of trans-4-aminocyclohex-2-en-1-ol derivatives: Competition of hydroxy-directed and ammonium-directed pathways. Aust. J. Chem. 2015, 68, 610.
(33) Brambilla, M.; Brennan, M. B.; Csatayová, K.; Davies, S. G.; Fletcher, A. M.; Kennett, A. M. R.; Lee, J. A.; Roberts, P. M.; Russell, A. J.; Thomson, J. E. Probing competitive and co-operative hydroxyl and ammonium hydrogen-bonding directed epoxidations. J. Org. Chem. 2017, 82, 10297.
(34) Ushakav, S. N.; Itenberg, A. M. The synthesis of triethylvinylsilicane. Zh. Obshch. Khim. 1937, 7, 2495.
(35) Sommer, L. H.; Dorfman, E.; Goldberg, G. M.; Whitmore, F. C. The reactivity with alkali of chlorine - carbon bonds alpha, beta and gamma to silicon. J. Am. Chem. Soc. 1946, 68, 488.
(36) Sommer, L. H.; Bailey, D. L.; Whitmore, F. C. Further studies of $\beta$-eliminations involving silicon. J. Am. Chem. Soc. 1948, 70, 2869.
(37) Sommer, L. H.; Baughman, G. A. Siliconium ions and carbonium ions as reaction intermediates. J. Am. Chem. Soc. 1961, 83, 3346.
(38) Traylor, T. G.; Hanstein, W.; Berwin, H. J.; Clinton, N. A.; Brown, R. S. Vertical stabilization of cations by neighboring $\sigma$ bonds. General considerations. J. Am. Chem. Soc. 1971, 93, 5715.
(39) Davis, D. D.; Jacocks, H. M., III. Deoxymetalation reactions. The mechanisms of deoxysilylation of mono-trimethylsilyl-and bis-trimethylsilyl-substituted alcohols and a comparison to the mechanism of deoxystannylation and deoxyplumbylation. J. Organomet. Chem. 1981, 206, 33.
(40) Colvin, E. W. Silicon in Organic Synthesis; Butterworths: London, 1981.
(41) Lambert, J. B.; Wang, G. T.; Finzel, R. B.; Teramura, D. H. Stabilization of positive charge by $\beta$-silicon. J. Am. Chem. Soc. 1987, 109, 7838.
(42) Lambert, J. B.; Chelius, E. C. The $\beta$ effect of silicon in the synperiplanar geometry. J. Am. Chem. Soc. 1990, 112, 8120.
(43) Lambert, J. B. Tetrahedron report number 273: The interaction of silicon with positively charged carbon. Tetrahedron 1990, 46, 2677.
(44) White, J. M. Reactivity and ground state effects of silicon in organic chemistry. Aust. J. Chem. 1995, 48, 1227.
(45) Lambert, J. B.; Liu, X. The $\beta$-effect of silicon in the orthogonal geometry. J. Organomet. Chem. 1996, 521, 203.
(46) Lambert, J. B.; Zhao, Y.; Robert W. Emblidge, R. W.; Salvador, L. A.; Liu, X.; So, J-H.; Chelius, E. C. The $\beta$ effect of silicon and related manifestations of $\sigma$ conjugation. Acc. Chem. Res. 1999, 32, 183.
(47) Jeon, S-J.; Walsh, P. J. Asymmetric addition of alkylzinc reagents to cyclic $\alpha, \beta$-unsaturated ketones and a tandem enantioselective addition/diastereoselective epoxidation with dioxygen. J. Am. Chem. Soc. 2003, 125, 9544.
(48) Blanksby, S. J.; Ellison, G. B. Bond dissociation energies of organic molecules. Acc. Chem. Res. 2003, 36, 255.
(49) Feng, Y.; Huang, H.; Liu, L.; Guo, Q.-X. Homolytic bond dissociation energies associated with acyl radicals and electron demands of acyl groups. Phys. Chem. Chem. Phys. 2003, 5, 685.
(50) Zavitsas, A. A. The relation between bond lengths and dissociation energies of CarbonCarbon bonds. J. Phys. Chem. A 2003, 107, 897.
(51) McGivern, S. W.; Derecskei-Kovacs, A.; North, S. W.; Francisco, J. S. Computationally efficient methodology to calculate $\mathrm{C}-\mathrm{H}$ and $\mathrm{C}-\mathrm{X}(\mathrm{X}=\mathrm{F}, \mathrm{Cl}$, and Br$)$ bond dissociation energies in haloalkanes. J. Phys Chem. A 2000, 104, 436.
(52) Jursic, B. S. Reliability of hybrid density theory-semiempirical approach for evaluation of bond dissociation energies. J. Chem. Soc. Perkin Trans. 1999, 2, 369.
(53) Verevkin, S. P.; Krashnykh, E. L.; Wright, J. S. Thermodynamic properties of benzyl halides: enthalpies of formation, strain enthalpies, and carbon-halogen bond dissociation enthalpies. Phys. Chem. Chem. Phys. 2003, 5, 2605.
(54) Tannenbaum, S. The Si-C bond energy in alkylsilanes. J. Am. Chem. Soc. 1954, 76, 1027.
(55) Grunenberg, J. Intrinsic bond strengths of multiple $\mathrm{C}-\mathrm{C}, \mathrm{Si}-\mathrm{Si}$, and $\mathrm{C}-\mathrm{Si}$ bonds. Angew. Chem. Int. Ed. 2001, 40, 4027.
(56) Ranu, C. B.; Majee, A.; Das, A. R. Facile and efficient synthesis of homoallylic alcohols using allyl bromide and commercial zinc dust. Tetrahedron Lett. 1995, 36, 4885.
(57) Gualandi, A.; Rodeghiero, G.; Faraone, A.; Patuzzo, F.; Marchini, M.; Calogero, F.; Perciaccante, R.; Jansen, T. P.; Ceroni, P.; Cozzi, P. G. Chem. Commun. 2019, 55, 6838.
(58) Gomes, P.; Gosmini, C.; Perichon, J. Allylation of carbonyl compounds by allylic acetates using a cobalt halide as catalyst. Synthesis, 2003, 12, 1909.
(59) Zhu, R.; Yu, K.; Gu, Z. N-bromoacetamide-mediated domino cyclization and elimination of homoallylic trichloroacetimidates: A novel approach toward the synthesis of 1-bromo-2-amino-3-butene derivatives. Org. Biomol. Chem. 2014, 12, 6653.
(60) Reddy, G. P.; Reddy, J. S.; Das, S.; Roisnel, T.; Yadav, J. S.; Chandrasekhar, S.; Grée, R. Synthesis of acylsilanes via nickel-catalyzed reactions of $\alpha$-hydroxyallylsilanes. Org. Lett. 2013, $15,1524$.
(61) Danheiser, R. L.; Fink, D. M.; Okano, K.; Tsai, Y. M.; Szczepanski, S. W. A practical and efficient synthesis of $\alpha, \beta$-unsaturated acylsilanes. J. Org. Chem. 1985, 50, 5393.
(62) An, I.; Onyeozili, E. N.; Maleczka, R. E., Jr. Enzymatic kinetic resolution of $\alpha$ hydroxysilanes. Tetrahedron Asymmetry 2010, 21, 527.
(63) Leonard, N. M.; Woerpel, K. A. Formation of medium-sized nitrogen heterocycles from $\gamma-$ silyloxy- $\gamma$-lactams. J. Org. Chem. 2009, 74, 6915.
(64) Schneider, D. F.; Venter, A. C. Initial interaction of triphenylphosphonium-2-propenylide with prenal prior to wittig olefination. Synth. Commun. 1999, 29, 3067.

Ji, D.-W.; He, G.-C.; Zhang, W.-S.; Zhao, C.-Y.; Hu, Y.-C.; Chen, Q.-A. Nickel-catalyzed allyl-allyl coupling reactions between 1,3-dienes and allylboronates. Chem. Commun. 2020, 56, 7431.

## APPENDIX

## Crystallographic document of 4.8a'

Crystal structure of crystal provided hydrogen atom on the Oxygen atom was found and refined isotropically.

## Crystal data and experimental




Figure 4.1: Crystal structure of compound 4.8a'
Experimental. Single colourless block-shaped crystals of 4.8a' used as received. A suitable crystal with dimensions $0.29 \times 0.19 \times 0.04 \mathrm{~mm}^{3}$ was selected and mounted on a nylon loop with paratone oil on a XtaLAB Synergy, Dualflex, HyPix diffractometer. The crystal was kept at a steady $T=$ 99.99(10) K during data collection. The structure was solved with the ShelXT (Sheldrick, 2015) solution program using dual methods and by using Olex2 1.5 (Dolomanov et al., 2009) as the graphical interface. The model was refined with ShelXL 2018/3 (Sheldrick, 2015) using full matrix least squares minimisation on $\boldsymbol{F}^{\mathbf{2}}$.

Crystal data. $\mathrm{C}_{15} \mathrm{H}_{22} \mathrm{O}_{2} \mathrm{Si}, M_{r}=262.41$, monoclinic, $P 2_{1} / n$ (No. 14), $\mathrm{a}=6.40749$ (6) $\AA, \mathrm{b}=$
 $99.99(10) \mathrm{K}, Z=4, Z^{\prime}=1, m\left(\mathrm{Cu} \mathrm{K}_{a}\right)=1.355,14998$ reflections measured, 2900 unique $\left(\mathrm{R}_{\mathrm{int}}=\right.$ 0.0328 ) which were used in all calculations. The final $w R_{2}$ was 0.0844 (all data) and $R_{l}$ was 0.0324 $(\mathrm{I} \geq 2 s(\mathrm{I})$ ).

Table 4.2: Crystal data

| Compound | 4.8a' |
| :---: | :---: |
| Formula | $\mathrm{C}_{15} \mathrm{H}_{22} \mathrm{O}_{2} \mathrm{Si}$ |
| CCDC | 2158501 |
| $D_{\text {calc. }} / \mathrm{g} \mathrm{cm}^{-3}$ | 1.195 |
| $\mathrm{m} / \mathrm{mm}^{-1}$ | 1.355 |
| Formula Weight | 262.41 |
| Color | colourless |
| Shape | block-shaped |
| Size/mm ${ }^{3}$ | $0.29 \times 0.19 \times 0.04$ |
| T/K | 99.99(10) |
| Crystal System | monoclinic |
| Space Group | $P 2_{1} / n$ |
| a/Å | 6.40749(6) |
| b/Å | 23.6316(2) |
| c/Å | 9.64907(10) |
| $a 1^{\circ}$ | 90 |
| $b{ }^{\circ}$ | 93.6810(8) |
| $g{ }^{\circ}$ | 90 |
| V/A ${ }^{3}$ | 1458.04(2) |
| Z | 4 |
| Z' | 1 |
| Wavelength/Å | 1.54184 |
| Radiation type | $\mathrm{Cu} \mathrm{K}_{a}$ |
| $Q_{\text {min }}{ }^{\circ}$ | 3.741 |
| $Q_{\max }{ }^{\circ}$ | 77.169 |
| Measured Refl's. | 14998 |
| Indep't Refl's | 2900 |
| Refl's I $\geq 2 s$ (I) | 2729 |
| $R_{\text {int }}$ | 0.0328 |
| Parameters | 171 |
| Restraints | 0 |
| Largest Peak | 0.359 |
| Deepest Hole | -0.263 |
| GooF | 1.054 |
| $w R_{2}$ (all data) | 0.0844 |
| $w R_{2}$ | 0.0828 |
| $R_{l}$ (all data) | 0.0342 |
| $R_{1}$ | 0.0324 |

## Structure quality indicators

| Reflections: | $\begin{gathered} d \min \\ 2 \Theta=154.3^{\circ} \end{gathered}$ | 0.79 | $1 / \sigma()$ | 44.6 | Rint | 3.28\% | Full $135.4^{\circ}$ $94 \%$ to $154.3^{\circ}$ | 100 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Refinement: | Shift | 0.001 | Max Peak | 0.4 | Min Peak | -0.3 | GooF | 1.054 |

Figure 4.2: Structure quality indicators
A colourless block-shaped-shaped crystal with dimensions $0.29 \times 0.19 \times 0.04 \mathrm{~mm}^{3}$ was mounted on a nylon loop with paratone oil. Data were collected using a XtaLAB Synergy, Dualflex, HyPix diffractometer equipped with an Oxford Cryosystems low-temperature device, operating at $T=$ 99.99(10) K.

MSU Data were measured using $w$ scans using $\mathrm{Cu} \mathrm{K}_{a}$ radiation (micro-focus sealed X-ray tube, $50 \mathrm{kV}, 1 \mathrm{~mA}$ ). The total number of runs and images was based on the strategy calculation from the program CrysAlisPro 1.171.41.123a (Rigaku OD, 2022). The achieved resolution was $Q=$ 77.169 .

Cell parameters were retrieved using the CrysAlisPro 1.171.41.123a (Rigaku OD, 2022) software and refined using CrysAlisPro 1.171.41.123a (Rigaku OD, 2022) on 9207 reflections, $61 \%$ of the observed reflections. Data reduction was performed using the CrysAlisPro 1.171.41.123a (Rigaku OD, 2022) software which corrects for Lorentz polarization. The final completeness is 100.00 out to 77.169 in $Q$ CrysAlisPro 1.171.41.123a (Rigaku Oxford Diffraction, 2022) Numerical absorption correction based on gaussian integration over a multifaceted crystal model Empirical absorption correction using spherical harmonics, implemented in SCALE3 ABSPACK scaling algorithm.

The structure was solved in the space group $P 2_{1} / n$ (\# 14) by using dual methods using the ShelXT (Sheldrick, 2015) structure solution program. The structure was refined by Least Squares ShelXL incorporated in Olex2 software program. All non-hydrogen atoms were refined anisotropically. Hydrogen atom positions were calculated geometrically and refined using the riding model, except for the hydrogen atom on the non-carbon atom(s) which were found by difference Fourier methods and refined isotropically when data permits.

CCDC 2158501 contains the supplementary crystallographic data for this paper. The data can be obtained free of charge from The Cambridge Crystallographic Data Centre via www.ccdc.cam.ac.uk/structures.

There is a single molecule in the asymmetric unit, which is represented by the reported sum formula. In other words: Z is 4 and $\mathrm{Z}^{\prime}$ is 1 .


Figure 4.3: Drawing of molecule 4.8a' with labeled heteroatoms


Figure 4.4: Drawing of molecule 4.8a' with labeling scheme. although crystal is in the
Centro-symmetric space group. Model has Chirality at C1 S, Model has Chirality at C2 (Centro SPGR) R; Model has Chirality at C4 (Centro SPGR) R Verify


Figure 4.5: The following hydrogen bonding interactions with a maximum D-D distance of 3.1 $\AA$ and a minimum angle of $110^{\circ}$ are present in 4.8a': O1-O2_1: $2.766 \AA$


Figure 4.6: Packing diagram of 4.8a'

## Data Plots: Diffraction Data



Figure 4.7: Data plots: Diffraction data

Figure 4.7 (cont'd)


## Data plots: Refinement and data



Figure 4.8: Data plots: Refinement and data
Table 4.3: Reflection statistics
Total reflections (after 15216 Unique reflections 2900
filtering)

| Completeness | 0.942 |
| :--- | :--- |
| hkl $1_{\text {max }}$ collected | $(3,28,12)$ |
| hkl $l_{\text {max }}$ used | $(7,28,12)$ |
| Lim d $\mathrm{d}_{\text {max }}$ collected | 100.0 |
| $\mathrm{~d}_{\text {max }}$ used | 11.82 |
| Friedel pairs | 1065 |
| Inconsistent | 1 |

Mean I/s 28.51
$\mathrm{hkl}_{\text {min }}$ collected $\quad(-7,-28,-11)$
hkl $1_{\text {min }}$ used $\quad(-7,0,0)$
Lim d ${ }_{\text {min }}$ collected 0.77
$\mathrm{d}_{\text {min }}$ used 0.79
Friedel pairs merged 1
$\mathrm{R}_{\text {int }} \quad 0.0328$

Removed systematic
absences

Intensity transformed 0
Omitted by user 0
(OMIT hkl)
Multiplicity $\quad(2746,1774,1016, \quad$ Maximum multiplicity 17 548, 292, 196, 88, 39, $12,1)$
218
0.0224

0

Filtered off
0
(Shel/OMIT)

## Selected crystal pictures



Figure 4.9: Selected crystal pictures

Table 4.4: Fractional atomic coordinates $\left(\times 10^{4}\right)$ and equivalent isotropic displacement parameters $\left(\AA^{2} \times 10^{3}\right)$ for 4.8a' $\mathrm{a}_{\text {eq }}$ is defined as $1 / 3$ of the trace of the orthogonalised $U_{i j}$

| Atom | $\mathbf{x}$ | $\mathbf{y}$ | $\mathbf{z}$ | $\boldsymbol{U}_{\boldsymbol{e q}}$ |
| :--- | :--- | :--- | :---: | :--- |
| Si1 | $5412.1(5)$ | $5875.8(2)$ | $2579.5(3)$ | $16.40(11)$ |
| O1 | $1373.5(13)$ | $7131.8(4)$ | $3587.1(10)$ | $19.4(2)$ |
| O2 | $5249.2(15)$ | $7133.1(4)$ | $641.8(9)$ | $25.0(2)$ |
| C1 | $3755.8(17)$ | $6557.1(5)$ | $2397.3(12)$ | $14.4(2)$ |
| C2 | $3185.8(17)$ | $6783.0(5)$ | $3819.5(12)$ | $15.0(2)$ |
| C3 | $5083.5(18)$ | $7140.1(5)$ | $4360.9(12)$ | $16.3(2)$ |
| C4 | $6153.3(18)$ | $7355.3(5)$ | $3070.5(12)$ | $16.2(2)$ |
| C5 | $5059.6(18)$ | $7028.5(5)$ | $1861.9(12)$ | $16.2(2)$ |
| C6 | $1877.2(18)$ | $6417.1(5)$ | $1378.6(13)$ | $18.2(3)$ |
| C7 | $6148.2(18)$ | $7987.9(5)$ | $2829.4(12)$ | $16.5(2)$ |
| C8 | $4333.8(19)$ | $8310.1(5)$ | $2918.8(14)$ | $21.1(3)$ |
| C9 | $4372(2)$ | $8891.3(6)$ | $2701.3(14)$ | $24.4(3)$ |
| C10 | $6212(2)$ | $9157.6(6)$ | $2383.6(14)$ | $25.2(3)$ |
| C11 | $8012(2)$ | $8841.1(6)$ | $2295.0(13)$ | $23.7(3)$ |
| C12 | $7983.4(19)$ | $8260.2(5)$ | $2515.9(12)$ | $19.3(3)$ |
| C13 | $7676(2)$ | $5928.8(6)$ | $3893.8(14)$ | $22.7(3)$ |
| C14 | $6373(2)$ | $5707.3(6)$ | $841.9(14)$ | $26.3(3)$ |
| C15 | $3645(2)$ | $5300.8(6)$ | $3150.7(16)$ | $28.2(3)$ |

Table 4.5: Anisotropic displacement parameters $\left(\times 10^{4}\right)$ for 4.8a'. The anisotropic displacement factor exponent takes the form: $-2 p^{2}\left[\mathrm{~h}^{2} \mathrm{a}^{* 2} \times U_{11}+\ldots+2 \mathrm{hka}^{*} \times \mathrm{b}^{*} \times U_{12}\right]$

| Atom | $\boldsymbol{U}_{11}$ | $\boldsymbol{U}_{\mathbf{2 2}}$ | $\boldsymbol{U}_{33}$ | $\boldsymbol{U}_{23}$ | $\boldsymbol{U}_{\mathbf{1 3}}$ | $\boldsymbol{U}_{\mathbf{1 2}}$ |
| :--- | :--- | :--- | :--- | ---: | ---: | ---: |
| Si1 | $16.73(18)$ | $13.66(18)$ | $18.89(19)$ | $-0.61(12)$ | $1.90(12)$ | $0.52(11)$ |
| O1 | $16.9(4)$ | $21.6(5)$ | $19.9(5)$ | $-4.7(4)$ | $2.3(3)$ | $3.4(3)$ |
| O2 | $37.6(5)$ | $21.8(5)$ | $16.3(5)$ | $0.0(4)$ | $6.7(4)$ | $-8.8(4)$ |
| C1 | $15.4(5)$ | $14.1(6)$ | $13.8(5)$ | $0.3(4)$ | $0.8(4)$ | $-0.2(4)$ |
| C2 | $15.9(5)$ | $14.0(6)$ | $15.4(6)$ | $-0.1(4)$ | $1.7(4)$ | $0.7(4)$ |
| C3 | $19.2(6)$ | $14.7(6)$ | $15.0(6)$ | $-0.1(4)$ | $-0.3(4)$ | $-0.8(4)$ |
| C4 | $15.0(5)$ | $15.9(6)$ | $17.5(6)$ | $-0.5(4)$ | $0.9(4)$ | $-0.5(4)$ |
| C5 | $17.6(5)$ | $13.9(6)$ | $17.2(6)$ | $-0.1(4)$ | $3.0(4)$ | $1.2(4)$ |
| C6 | $18.6(6)$ | $18.8(6)$ | $16.9(6)$ | $-0.8(5)$ | $-0.9(4)$ | $-1.2(4)$ |
| C7 | $20.1(6)$ | $15.9(6)$ | $13.2(5)$ | $-1.1(4)$ | $-1.0(4)$ | $-2.3(4)$ |
| C8 | $19.7(6)$ | $19.1(6)$ | $24.2(6)$ | $0.8(5)$ | $-0.7(5)$ | $-1.8(5)$ |
| C9 | $27.4(7)$ | $19.7(7)$ | $25.6(7)$ | $0.8(5)$ | $-1.6(5)$ | $3.7(5)$ |
| C10 | $39.4(8)$ | $15.0(6)$ | $20.7(6)$ | $0.9(5)$ | $-1.4(5)$ | $-4.2(5)$ |
| C11 | $29.5(7)$ | $21.7(7)$ | $20.2(6)$ | $-0.4(5)$ | $3.3(5)$ | $-9.6(5)$ |
| C12 | $21.2(6)$ | $21.2(6)$ | $15.7(6)$ | $-1.6(5)$ | $2.2(4)$ | $-3.0(5)$ |
| C13 | $21.2(6)$ | $21.9(7)$ | $24.8(7)$ | $0.7(5)$ | $-0.7(5)$ | $3.9(5)$ |
| C14 | $26.8(7)$ | $26.7(7)$ | $25.4(7)$ | $-6.6(5)$ | $2.7(5)$ | $4.9(5)$ |
| C15 | $26.7(7)$ | $17.0(6)$ | $41.1(8)$ | $4.5(6)$ | $2.8(6)$ | $-1.1(5)$ |

Table 4.6: Bond lengths in $\AA$ for $4.8 \mathbf{a}^{\prime}$

| Atom | Atom | Length/A |
| :--- | :--- | :--- |
| Si1 | C1 | $1.9302(12)$ |
| Si1 | C13 | $1.8689(13)$ |
| Si1 | C14 | $1.8659(14)$ |
| Si1 | C15 | $1.8741(14)$ |
| O1 | C2 | $1.4299(14)$ |
| O2 | C5 | $1.2167(15)$ |
| C1 | C2 | $1.5381(16)$ |
| C1 | C5 | $1.5037(16)$ |
| C1 | C6 | $1.5404(16)$ |
| C2 | C3 | $1.5433(16)$ |
| C3 | C4 | $1.5455(16)$ |
| C4 | C5 | $1.5305(16)$ |
| C4 | C7 | $1.5130(16)$ |
| C7 | C8 | $1.3970(17)$ |
| C7 | C12 | $1.3906(17)$ |
| C8 | C9 | $1.3898(19)$ |
| C9 | C10 | $1.388(2)$ |
| C10 | C11 | $1.382(2)$ |
| C11 | C12 | $1.3896(19)$ |

Table 4.7: Bond angles in ${ }^{\circ}$ for 4.8a'

| Atom | Atom | Atom | Angle $^{\circ}$ |
| :--- | :--- | :--- | :--- |
| C13 | Si1 | C1 | $113.84(5)$ |
| C13 | Si1 | C15 | $107.84(6)$ |
| C14 | Si1 | C1 | $107.91(6)$ |
| C14 | Si1 | C13 | $109.72(6)$ |
| C14 | Si1 | C15 | $110.49(7)$ |
| C15 | Si1 | C1 | $107.01(6)$ |
| C2 | C1 | Si1 | $111.65(8)$ |
| C2 | C1 | C6 | $115.04(9)$ |
| C5 | C1 | Si1 | $109.57(8)$ |
| C5 | C1 | C2 | $102.61(9)$ |
| C5 | C1 | C6 | $111.63(10)$ |
| C6 | C1 | Si1 | $106.34(8)$ |
| O1 | C2 | C1 | $107.29(9)$ |
| O1 | C2 | C3 | $110.62(9)$ |
| C1 | C2 | C3 | $105.35(9)$ |
| C2 | C3 | C4 | $106.72(9)$ |
| C5 | C4 | C3 | $104.05(9)$ |
| C7 | C4 | C3 | $116.90(10)$ |
| C7 | C4 | C5 | $112.64(10)$ |
| O2 | C5 | C1 | $125.11(11)$ |
| O2 | C5 | C4 | $124.44(11)$ |
| C1 | C5 | C4 | $110.44(10)$ |
| C8 | C7 | C4 | $121.51(11)$ |
| C12 | C7 | C4 | $119.81(11)$ |
| C12 | C7 | C8 | $118.68(12)$ |
| C9 | C8 | C7 | $120.43(12)$ |
| C10 | C9 | C8 | $120.33(13)$ |
| C11 | C10 | C9 | $119.50(12)$ |
| C10 | C11 | C12 | $120.38(12)$ |
| C11 | C12 | C7 | $120.68(12)$ |

Table 4.8: Torsion angles in ${ }^{\circ}$ for $4.8 \mathbf{a}^{\prime}$

| Atom | Atom | Atom | Atom | ${\text { Angle }{ }^{\circ}}^{\text {At }}$ |
| :--- | :--- | :--- | :--- | ---: |
| Si1 | C1 | C2 | O1 | $-157.81(7)$ |
| Si1 | C1 | C2 | C3 | $84.30(9)$ |
| Si1 | C1 | C5 | O2 | $88.25(13)$ |
| Si1 | C1 | C5 | C4 | $-90.80(10)$ |
| O1 | C2 | C3 | C4 | $-88.74(11)$ |
| C1 | C2 | C3 | C4 | $26.88(12)$ |
| C2 | C1 | C5 | O2 | $-153.01(12)$ |
| C2 | C1 | C5 | C4 | $27.94(11)$ |
| C2 | C3 | C4 | C5 | $-9.65(12)$ |
| C2 | C3 | C4 | C7 | $115.23(11)$ |
| C3 | C4 | C5 | O2 | $169.31(11)$ |
| C3 | C4 | C5 | C1 | $-11.63(12)$ |
| C3 | C4 | C7 | C8 | $-46.77(16)$ |
| C3 | C4 | C7 | C12 | $132.98(11)$ |
| C4 | C7 | C8 | C9 | $179.60(12)$ |
| C4 | C7 | C12 | C11 | $-179.81(11)$ |
| C5 | C1 | C2 | O1 | $84.93(10)$ |
| C5 | C1 | C2 | C3 | $-32.97(11)$ |
| C5 | C4 | C7 | C8 | $73.66(14)$ |
| C5 | C4 | C7 | C12 | $-106.58(12)$ |
| C6 | C1 | C2 | O1 | $-36.50(13)$ |
| C6 | C1 | C2 | C3 | $-154.39(10)$ |
| C6 | C1 | C5 | O2 | $-29.28(16)$ |
| C6 | C1 | C5 | C4 | $151.67(10)$ |
| C7 | C4 | C5 | O2 | $41.74(16)$ |
| C7 | C4 | C5 | C1 | $-139.20(10)$ |
| C7 | C8 | C9 | C10 | $0.4(2)$ |
| C8 | C7 | C12 | C11 | $-0.04(18)$ |
| C8 | C9 | C10 | C11 | $-0.5(2)$ |
| C9 | C10 | C11 | C12 | $0.3(2)$ |
| C10 | C11 | C12 | C7 | $-0.02(19)$ |
| C12 | C7 | C8 | C9 | $-0.16(18)$ |
|  |  |  |  |  |

Table 4.9: Hydrogen fractional atomic coordinates $\left(\times 10^{4}\right)$ and equivalent isotropic displacement parameters $\left(\AA^{2} \times 10^{3}\right)$ for $4.8 \mathbf{a}^{\prime} . U_{e q}$ is defined as $1 / 3$ of the trace of the orthogonalised $U_{i j}$

| Atom | $\mathbf{x}$ | $\mathbf{y}$ | $\mathbf{z}$ | $\boldsymbol{U}_{\boldsymbol{e q}}$ |
| :--- | :--- | :--- | :--- | :--- |
| H1 | $1160(30)$ | $7293(8)$ | $4300(20)$ | $37(5)$ |
| H2 | 2918.1 | 6465.78 | 4470.56 | 18 |
| H3A | 6066.73 | 6906.39 | 4951.14 | 20 |
| H3B | 4620.53 | 7462.89 | 4918.67 | 20 |
| H4 | 7644.54 | 7228.26 | 3159.86 | 19 |
| H6A | 2378.83 | 6319.69 | 470.41 | 27 |
| H6B | 950.92 | 6746.65 | 1282.43 | 27 |
| H6C | 1106.54 | 6095.74 | 1734.89 | 27 |
| H8 | 3065.66 | 8130.8 | 3129.68 | 25 |
| H9 | 3132.55 | 9107.3 | 2770.38 | 29 |
| H10 | 6233.34 | 9554.51 | 2228.3 | 30 |
| H11 | 9276.76 | 9021.76 | 2081.83 | 28 |
| H12 | 9230.1 | 8046.79 | 2452.04 | 23 |
| H13A | 8530.96 | 5585.71 | 3860.43 | 34 |
| H13B | 7158.42 | 5969.13 | 4822.53 | 34 |
| H13C | 8526.92 | 6259.24 | 3687.77 | 34 |
| H14A | 7187.58 | 6027.02 | 520.06 | 39 |
| H14B | 5177.97 | 5637.08 | 178.99 | 39 |
| H14C | 7259.43 | 5369.18 | 914.01 | 39 |
| H15A | 4479.27 | 4966.18 | 3419.7 | 42 |
| H15B | 2630.72 | 5202.86 | 2384.79 | 42 |
| H15C | 2901.18 | 5432.54 | 3946.23 | 42 |

Table 4.10: Hydrogen bond information for 4.8a'

| $\mathbf{D}$ | $\mathbf{H}$ | $\mathbf{A}$ | $\mathbf{d}(\mathbf{D}-\mathbf{H}) / \AA$ | $\mathbf{d}(\mathbf{H}-\mathbf{A}) / \AA$ | $\mathbf{d}(\mathbf{D}-\mathbf{A}) / \AA$ | D-H-A/deg |
| :--- | :--- | :--- | ---: | ---: | ---: | :---: |
| O 1 | H 1 | $\mathrm{O} 2^{1}$ | $0.81(2)$ | $1.99(2)$ | $2.7659(13)$ | $161.8(19)$ |

${ }^{1}-1 / 2+x, 3 / 2-y, 1 / 2+z$

## Citations

CrysAlisPro (Rigaku, V1.171.41.123a, 2022)
CrysAlisPro (ROD), Rigaku Oxford Diffraction, Poland (?).
O.V. Dolomanov and L.J. Bourhis and R.J. Gildea and J.A.K. Howard and H. Puschmann, Olex2: A complete structure solution, refinement and analysis program, J. Appl. Cryst., (2009), 42, 339-341.

Sheldrick, G.M., Crystal structure refinement with ShelXL, Acta Cryst., (2015), C71, 3-8.
Sheldrick, G.M., ShelXT-Integrated space-group and crystal-structure determination, Acta Cryst., (2015), A71, 3-8.

## Crystallographic document of 4.9a

Structure from thin plate, shows some disorder in the $\mathrm{SiMe}_{3}$ portion, but the stereochemistry of the diastereomer is determined, both enantiomers are present in the crystal.



Figure 4.10: Crystal structure of compound 4.9a
Experimental. Single colourless plate-shaped crystals of 4.9a were used as received. A suitable crystal $0.34 \times 0.20 \times 0.02 \mathrm{~mm}^{3}$ was selected and mounted on a nylon loop with paratone oil on a Bruker APEX-II CCD diffractometer. The crystal was kept at a steady $T=173$ (2) K during data collection. The structure was solved with the XT (Sheldrick, 2015) structure solution program using the Intrinsic Phasing solution method and by using Olex2 (Dolomanov et al., 2009) as the graphical interface. The model was refined with version 2018/3 of ShelXL (Sheldrick, 2015) using Least Squares minimization.

Crystal data. $\mathrm{C}_{15} \mathrm{H}_{21} \mathrm{ClO}_{2} \mathrm{Si}, M_{r}=296.86$, monoclinic, $P 2_{1} / n$ (No. 14), $\mathrm{a}=6.4892(6) \AA$, $\mathrm{b}=$ 22.023(2) $\AA, \mathrm{c}=11.0546(11) \AA, \beta=94.8430(10)^{\circ}, \alpha=\gamma=90^{\circ}, V=1574.2(3) \AA^{3}, T=173(2) \mathrm{K}$, $Z=4, Z^{\prime}=1, \mu\left(\mathrm{MoK}_{\alpha}\right)=0.315,12974$ reflections measured, 2988 unique $\left(R_{\text {int }}=0.0378\right)$ which were used in all calculations. The final $w R_{2}$ was 0.1393 (all data) and $R_{l}$ was 0.0509 (I > 2(I)).

Table 4.11: Crystal data

| Compound | 4.9a |
| :---: | :---: |
| CCDC | 1890864 |
| Formula | $\mathrm{C}_{15} \mathrm{H}_{21} \mathrm{ClO}_{2} \mathrm{Si}$ |
| $D_{\text {calc. }} / \mathrm{g} \mathrm{cm}^{-3}$ | 1.253 |
| $\mu / \mathrm{mm}^{-1}$ | 0.315 |
| Formula Weight | 296.86 |
| Color | colourless |
| Shape | plate |
| Size/mm ${ }^{3}$ | $0.34 \times 0.20 \times 0.02$ |
| T/K | 173(2) |
| Crystal System | monoclinic |
| Space Group | $P 2{ }_{1} / n$ |
| $a / \AA$ | 6.4892(6) |
| b/A | 22.023(2) |
| c/Å | 11.0546(11) |
| $\alpha{ }^{\circ}$ | 90 |
| $\beta 1^{\circ}$ | 94.8430(10) |
| $\gamma 1{ }^{\circ}$ | 90 |
| V/A ${ }^{3}$ | 1574.2(3) |
| Z | 4 |
| Z' | 1 |
| Wavelength/Å | 0.710730 |
| Radiation type | $\mathrm{MoK}_{\square}$ |
| $\Theta_{\text {min }}{ }^{\circ}$ | 1.849 |
| $\Theta_{\max }{ }^{\circ}$ | 25.703 |
| Measured Refl. | 12974 |
| Independent Refl | . 2988 |
| Reflections with | I2231 |
| > 2(I) |  |
| $R_{\text {int }}$ | 0.0378 |
| Parameters | 201 |
| Restraints | 0 |
| Largest Peak | 0.503 |
| Deepest Hole | -0.384 |
| GooF | 1.047 |
| $w R_{2}$ (all data) | 0.1393 |
| $w R_{2}$ | 0.1267 |
| $R_{1}$ (all data) | 0.0691 |
| $R_{1}$ | 0.0509 |

## Structure iuality indicators

| Reflections: | d min (Mo) 0.82 | /1/ | 31.4 |  | 3.78\% | complete | 100\% |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Refinement: | Shift 0.000 | Max Peak | 0.5 | Min | -0.4 | Goo | 1.047 |

Figure 4.11: Structure quality indicators
A colourless plate-shaped crystal with dimensions $0.34 \times 0.20 \times 0.02 \mathrm{~mm}^{3}$ was mounted on a nylon loop with paratone oil. Data were collected using a Bruker APEX-II CCD diffractometer equipped with an Oxford Cryosystems low-temperature device, operating at $T=173(2) \mathrm{K}$.

Data were measured using $\phi$ and $\omega$ scans of $-0.50^{\circ}$ per frame for 299.33 s using $\mathrm{MoK}_{\square}$ radiation (sealed tube, $50 \mathrm{kV}, 40 \mathrm{~mA}$ ). The total number of runs and images was based on the strategy calculation from the program COSMO (BRUKER, V1.61, 2009). The actually achieved resolution was $\Theta=25.703$.

Cell parameters were retrieved using the SAINT (Bruker, V8.34A, after 2013) software and refined using SAINT (Bruker, V8.34A, after 2013) on 4999 reflections, $39 \%$ of the observed reflections. Data reduction was performed using the SAINT (Bruker, V8.34A, after 2013) software which corrects for Lorentz polarization. The final completeness is 100.00 out to 25.703 in $\square$ SADABS-2014/5 (Bruker,2014/5) was used for absorption correction. $w R_{2}$ (int) was 0.0586 before and 0.0488 after correction. The Ratio of minimum to maximum transmission is 0.8603 . The $\square / 2$ correction factor is 0.00150 .

The structure was solved in the space group $P 2_{1} / n$ (\# 14) by Intrinsic Phasing using the XT (Sheldrick, 2015) structure solution program. The structure was refined by Least Squares using version 2018/3 of XL (Sheldrick, 2015) incorporated in Olex 2 (Dolomanov et al., 2009). All nonhydrogen atoms were refined anisotropically. Hydrogen atom positions were calculated geometrically and refined using the riding model, except for the Hydrogen atom on the nitrogen atom which was found by difference Fourier methods and refined isotropically.
$\beta$ CCDC 1890864 contains the supplementary crystallographic data for this paper. The data can be obtained free of charge from The Cambridge Crystallographic Data Centre via www.ccdc.cam.ac.uk/structures.

There is a single molecule in the asymmetric unit, which is represented by the reported sum formula. In other words: Z is 4 and Z ' is 1 .


$$
0_{i}^{000} 0
$$

Figure 4.12: Drawing of 4.9a with representative atoms


Figure 4.13: Model has Chirality at C1 (Centro SPGR) R Verify; Model has Chirality at C3 (Centro SPGR) S Verify; Model has Chirality at C6 (Centro SPGR) S Verify


Figure 4.14: Drawing of 4.9a


Figure 4.15: The following hydrogen bonding interactions with a maximum D-D distance of $2.9 \AA$ and a minimum angle of $120^{\circ}$ are present in 4.9a: O2-O1_1:2.765 $\AA$


Figure 4.16: Packing diagram of 4.9a

## Data Plots: Diffraction Data



Figure 4.17: Data plots: Diffraction data

## Data Plots: Refinement and Data




Figure 4.18: Data plots: Refinement and data
Table 4.12: Reflection statistics

Total reflections (after 13207
filtering)
Completeness $\quad 1.0$
$\mathrm{hkl}_{\text {max }}$ collected $\quad(7,26,13)$
$\mathrm{hk} 1_{\text {max }}$ used $\quad(7,26,13)$
Lim $\mathrm{d}_{\text {max }}$ collected $\quad 100.0$
$\mathrm{d}_{\text {max }}$ used $\quad 11.02$
Friedel pairs 5258
Inconsistent 0
equivalents

| $\mathrm{R}_{\text {sigma }}$ | 0.0318 |
| :--- | :--- |
| Omitted reflections | 0 |

$\begin{array}{ll}\text { Multiplicity } & (9289,1854,70) \\ \begin{array}{l}\text { Removed systematic } \\ \text { absences }\end{array} & 233\end{array}$

Unique reflections 2988
Mean I/ $\sigma \quad 15.77$
$\mathrm{hkl}_{\text {min }}$ collected $\quad(-7,-26,-13)$
hk $l_{\text {min }}$ used $\quad(-7,0,0)$
Lim d ${ }_{\text {min }}$ collected 0.36
$\mathrm{d}_{\text {min }}$ used 0.82
Friedel pairs merged 1
$\mathrm{R}_{\text {int }} \quad 0.0378$

Intensity transformed 0
Omitted by user 0
(OMIT hkl)
Maximum multiplicity 10
Filtered off 0
(Shel/OMIT)

Images of the crystal on the diffractometer


Figure 4.19: Images of the crystal on the diffractometer

Table 4.13: Fractional atomic coordinates $\left(\times 10^{4}\right)$ and equivalent isotropic displacement parameters $\left(\AA^{2} \times 10^{3}\right)$ for 4.9a. $U_{e q}$ is defined as $1 / 3$ of the trace of the orthogonalised $U_{i j}$

| Atom | $\mathbf{x}$ | $\mathbf{y}$ | $\mathbf{z}$ | $\boldsymbol{U}_{\boldsymbol{e q}}$ |
| :--- | :---: | :--- | :--- | :--- |
| Cl1 | $2851(2)$ | $4768.2(4)$ | $6691.2(8)$ | $100.7(4)$ |
| Si1 | $6323.1(11)$ | $8825.0(3)$ | $5073.8(7)$ | $45.7(2)$ |
| O1 | $7102(3)$ | $7491.2(8)$ | $6575.6(13)$ | $43.3(4)$ |
| O2 | $10432(2)$ | $7651.3(8)$ | $3777.8(14)$ | $44.1(4)$ |
| C1 | $7880(3)$ | $8094.2(10)$ | $4911.5(19)$ | $33.5(5)$ |
| C2 | $6865(3)$ | $7568.5(10)$ | $5476.5(19)$ | $33.6(5)$ |
| C3 | $5430(3)$ | $7168.3(10)$ | $4666(2)$ | $35.7(5)$ |
| C4 | $6340(4)$ | $7041.4(11)$ | $3447(2)$ | $39.9(6)$ |
| C5 | $6904(4)$ | $7630.2(11)$ | $2838.0(19)$ | $40.2(6)$ |
| C6 | $8530(3)$ | $7981.9(11)$ | $3626.4(19)$ | $35.3(5)$ |
| C7 | $4784(4)$ | $6583.3(11)$ | $5254(2)$ | $40.3(6)$ |
| C8 | $2744(4)$ | $6391.5(12)$ | $5093(2)$ | $49.2(6)$ |
| C9 | $2146(5)$ | $5831.8(14)$ | $5525(3)$ | $60.2(8)$ |
| C10 | $3595(6)$ | $5471.4(13)$ | $6132(3)$ | $62.5(8)$ |
| C11 | $5619(5)$ | $5644.3(13)$ | $6322(3)$ | $64.0(8)$ |
| C12 | $6215(5)$ | $6204.1(12)$ | $5873(2)$ | $51.8(7)$ |
| C13 | $4312(6)$ | $8926.7(19)$ | $3748(4)$ | $70.7(12)$ |
| C13B | $6760(20)$ | $9400(6)$ | $3999(15)$ | $70.7(12)$ |
| C14 | $8236(7)$ | $9442.7(18)$ | $4983(4)$ | $73.6(13)$ |
| C14B | $7290(30)$ | $9181(8)$ | $6693(13)$ | $78(5)$ |
| C15 | $5170(11)$ | $8818(2)$ | $6506(5)$ | $111(3)$ |
| C15B | $3560(20)$ | $8679(6)$ | $5240(20)$ | $95(7)$ |

Table 4.14: Anisotropic displacement parameters $\left(\times 10^{4}\right)$ 4.9a. The anisotropic displacement factor exponent takes the form: $-2 \pi^{2}\left[\mathrm{~h}^{2} \mathrm{a}^{* 2} \times U_{11}+\ldots+2 \mathrm{hka}^{*} \times \mathrm{b}^{*} \times U_{12}\right] \$$

| Atom | $\boldsymbol{U}_{\mathbf{I I}}$ | $\boldsymbol{U}_{\mathbf{2 2}}$ | $\boldsymbol{U}_{\mathbf{3 3}}$ | $\boldsymbol{U}_{23}$ | $\boldsymbol{\boldsymbol { U } _ { \mathbf { 1 3 } }}$ | $\boldsymbol{U}_{\mathbf{1 2}}$ |
| :--- | :---: | :---: | :---: | :---: | :---: | :---: |
| C11 | $186.6(12)$ | $55.6(5)$ | $61.4(5)$ | $9.1(4)$ | $18.8(6)$ | $-49.9(6)$ |
| Si1 | $55.6(5)$ | $36.3(4)$ | $46.7(4)$ | $3.8(3)$ | $13.4(3)$ | $6.0(3)$ |
| O1 | $50.8(10)$ | $53.2(11)$ | $27.1(8)$ | $7.2(7)$ | $10.3(7)$ | $4.3(8)$ |
| O2 | $39.0(9)$ | $63.8(11)$ | $30.8(8)$ | $-2.8(8)$ | $10.3(7)$ | $7.3(8)$ |
| C1 | $34.6(11)$ | $38.4(12)$ | $28.2(11)$ | $0.5(9)$ | $7.0(9)$ | $1.9(10)$ |
| C2 | $32.7(12)$ | $38.2(12)$ | $31.1(11)$ | $3.6(9)$ | $10.0(9)$ | $8.3(9)$ |
| C3 | $34.8(12)$ | $37.7(13)$ | $35.3(12)$ | $5.0(9)$ | $7.3(9)$ | $2.0(10)$ |
| C4 | $48.4(14)$ | $42.0(14)$ | $29.6(12)$ | $0.5(10)$ | $5.0(10)$ | $-5.6(11)$ |
| C5 | $48.0(14)$ | $46.6(14)$ | $26.0(11)$ | $4.4(10)$ | $3.2(10)$ | $-3.5(11)$ |
| C6 | $40.3(12)$ | $39.8(13)$ | $26.8(11)$ | $4.0(9)$ | $7.8(9)$ | $0.1(10)$ |
| C7 | $49.5(14)$ | $38.4(13)$ | $35.1(12)$ | $1.4(10)$ | $15.8(10)$ | $-0.4(11)$ |
| C8 | $56.7(16)$ | $46.4(15)$ | $45.9(14)$ | $-2.7(12)$ | $13.5(12)$ | $-6.0(12)$ |
| C9 | $72.7(19)$ | $55.9(18)$ | $54.7(17)$ | $-6.2(14)$ | $20.6(15)$ | $-21.2(16)$ |
| C10 | $104(3)$ | $40.7(16)$ | $45.3(16)$ | $-0.6(12)$ | $24.0(16)$ | $-22.8(16)$ |
| C11 | $95(2)$ | $45.4(17)$ | $51.6(17)$ | $13.0(13)$ | $9.1(16)$ | $4.1(16)$ |
| C12 | $61.8(17)$ | $42.7(15)$ | $52.0(15)$ | $10.1(12)$ | $10.2(13)$ | $2.2(12)$ |
| C13 | $67(2)$ | $55(2)$ | $88(3)$ | $2(2)$ | $-2(2)$ | $15.3(19)$ |
| C13B | $67(2)$ | $55(2)$ | $88(3)$ | $2(2)$ | $-2(2)$ | $15.3(19)$ |
| C14 | $89(3)$ | $42(2)$ | $87(3)$ | $-7(2)$ | $-2(2)$ | $0(2)$ |
| C14B | $90(11)$ | $83(11)$ | $59(9)$ | $-35(8)$ | $-10(8)$ | $30(9)$ |
| C15 | $187(7)$ | $77(3)$ | $83(4)$ | $21(3)$ | $84(4)$ | $70(4)$ |
| C15B | $48(8)$ | $33(7)$ | $210(20)$ | $-17(10)$ | $32(11)$ | $2(6)$ |

Table 4.15: Bond lengths in $\AA$ for 4.9a

| Atom | Atom | Length/Å |
| :--- | :--- | :--- |
| C11 | C10 | $1.750(3)$ |
| Si1 | C1 | $1.917(2)$ |
| Si1 | C13 | $1.892(4)$ |
| Si1 | C13B | $1.776(15)$ |
| Si1 | C14 | $1.850(4)$ |
| Si1 | C14B | $2.006(13)$ |
| Si1 | C15 | $1.807(4)$ |
| Si1 | C15B | $1.847(13)$ |
| O1 | C2 | $1.224(2)$ |
| O2 | C6 | $1.431(3)$ |
| C1 | C2 | $1.494(3)$ |
| C1 | C6 | $1.536(3)$ |
| C2 | C3 | $1.518(3)$ |
| C3 | C4 | $1.541(3)$ |
| C3 | C7 | $1.518(3)$ |
| C4 | C5 | $1.520(3)$ |
| C5 | C6 | $1.522(3)$ |
| C7 | C8 | $1.387(4)$ |
| C7 | C12 | $1.386(4)$ |
| C8 | C9 | $1.389(4)$ |
| C9 | C10 | $1.363(5)$ |
| C10 | C11 | $1.366(4)$ |
| C11 | C12 | $1.396(4)$ |

Table 4.16: Bond angles in ${ }^{\circ}$ for 4.9a

| Atom | Atom | Atom | Angle ${ }^{\circ}$ |
| :--- | :--- | :--- | :--- |
| C1 | Si1 | C14B | $106.5(4)$ |
| C13 | Si1 | C1 | $111.19(15)$ |
| C13B | Si1 | C1 | $114.8(5)$ |
| C13B | Si1 | C14B | $105.2(8)$ |
| C13B | Si1 | C15B | $113.7(8)$ |
| C14 | Si1 | C1 | $104.54(16)$ |
| C14 | Si1 | C13 | $107.0(2)$ |
| C15 | Si1 | C1 | $109.55(16)$ |
| C15 | Si1 | C13 | $111.7(3)$ |
| C15 | Si1 | C14 | $112.6(3)$ |
| C15B | Si1 | C1 | $112.8(4)$ |
| C15B | Si1 | C14B | $102.4(9)$ |
| C2 | C1 | Si1 | $110.93(14)$ |
| C2 | C1 | C6 | $115.28(18)$ |
| C6 | C1 | Si1 | $114.20(15)$ |
| O1 | C2 | C1 | $120.0(2)$ |
| O1 | C2 | C3 | $121.6(2)$ |
| C1 | C2 | C3 | $118.23(18)$ |
| C2 | C3 | C4 | $111.10(18)$ |
| C7 | C3 | C2 | $114.71(19)$ |
| C7 | C3 | C4 | $111.12(18)$ |
| C5 | C4 | C3 | $110.92(19)$ |
| C4 | C5 | C6 | $111.19(18)$ |
| O2 | C6 | C1 | $105.99(17)$ |
| O2 | C6 | C5 | $110.94(19)$ |
| C5 | C6 | C1 | $112.36(18)$ |
| C8 | C7 | C3 | $119.9(2)$ |
| C12 | C7 | C3 | $121.7(2)$ |
| C12 | C7 | C8 | $118.2(2)$ |
| C7 | C8 | C9 | $121.2(3)$ |
| C10 | C9 | C8 | $119.0(3)$ |
| C9 | C10 | C11 | $119.3(3)$ |
| C10 | C10 | C11 | $121.8(3)$ |
| C7 | C11 | C11 | $118.9(3)$ |
| C12 | C11 | $118.9(3)$ |  |
| C120.9(3) |  |  |  |
| C1 |  |  |  |

Table 4.17: Hydrogen fractional atomic Coordinates $\left(\times 10^{4}\right)$ and equivalent isotropic displacement parameters $\left(\AA^{2} \times 10^{3}\right)$ for 4.9a. $U_{e q}$ is defined as $1 / 3$ of the trace of the orthogonalised $U_{i j}$

| Atom | x | y | z | $U_{e q}$ |
| :---: | :---: | :---: | :---: | :---: |
| H2 | 10915.64 | 7610.88 | 3101.1 | 66 |
| H1 | 9203 | 8159.27 | 5426.81 | 40 |
| H3 | 4136.26 | 7407.92 | 4470.65 | 43 |
| H4A | 7589.06 | 6784.64 | 3589.99 | 48 |
| H4B | 5314.99 | 6816.39 | 2904.89 | 48 |
| H5A | 5649.47 | 7883.61 | 2683.25 | 48 |
| H5B | 7441.42 | 7537.68 | 2046.5 | 48 |
| H6 | 8779.35 | 8381.02 | 3232.94 | 42 |
| H8 | 1737.16 | 6647.54 | 4680.66 | 59 |
| H9 | 748.55 | 5702.08 | 5398.98 | 72 |
| H11 | 6603.92 | 5387.82 | 6751.76 | 77 |
| H12 | 7620.93 | 6326.8 | 5992.76 | 62 |
| H13A | 4992.41 | 8934.14 | 2989.88 | 106 |
| H13B | 3573.51 | 9309.95 | 3839.56 | 106 |
| H13C | 3327.57 | 8588.59 | 3728.08 | 106 |
| H13D | 8255.56 | 9458.59 | 3967.81 | 106 |
| H13E | 6132.1 | 9780.36 | 4239.55 | 106 |
| H13F | 6147.94 | 9277.1 | 3196.09 | 106 |
| H14A | 9210.09 | 9435.85 | 5710.1 | 110 |
| H14B | 7518.71 | 9834.57 | 4929.13 | 110 |
| H14C | 8992.71 | 9384.82 | 4260.84 | 110 |
| H14D | 6986.93 | 8898.16 | 7338.5 | 117 |
| H14E | 6576.98 | 9566.59 | 6803.47 | 117 |
| H14F | 8786.9 | 9252.73 | 6728.92 | 117 |
| H15A | 4045.06 | 8519.58 | 6470.27 | 167 |
| H15B | 4620.13 | 9221.41 | 6665.2 | 167 |
| H15C | 6220.55 | 8707.45 | 7158.11 | 167 |
| H15D | 2887.08 | 8536.91 | 4460.84 | 142 |
| H15E | 2892.3 | 9054.56 | 5477.09 | 142 |
| H15F | 3426.12 | 8367.89 | 5859.76 | 142 |

Table 4.18: Hydrogen bond information for 4.9a

| $\mathbf{D}$ | $\mathbf{H}$ | $\mathbf{A}$ | $\mathbf{d}(\mathbf{D}-\mathbf{H}) / \AA$ | $\mathbf{d}(\mathbf{H}-\mathbf{A}) / \AA$ | $\mathbf{d}(\mathbf{D}-\mathbf{A}) / \AA$ | D-H-A/deg |
| :--- | :--- | :--- | :---: | :---: | :---: | :---: |
| O 2 | H 2 | $\mathrm{O}^{1}$ | 0.84 | 1.93 | $2.765(2)$ | 178.2 |
| $\overline{{ }^{1} 1 / 2+\mathrm{x}, 3 / 2-\mathrm{y},-1 / 2+\mathrm{z}}$ |  |  |  |  |  |  |

Table 4.19: Atomic occupancies for all atoms that are not fully occupied in 4.9a

| Atom | Occupancy |
| :--- | ---: |
| C13 | $0.776(4)$ |
| H13A | $0.776(4)$ |
| H13B | $0.776(4)$ |
| H13C | $0.776(4)$ |
| C13B | $0.224(4)$ |
| H13D | $0.224(4)$ |
| H13E | $0.224(4)$ |
| H13F | $0.224(4)$ |
| C14 | $0.776(4)$ |
| H14A | $0.776(4)$ |
| H14B | $0.776(4)$ |
| H14C | $0.776(4)$ |
| C14B | $0.224(4)$ |
| H14D | $0.224(4)$ |
| H14E | $0.224(4)$ |
| H14F | $0.224(4)$ |
| C15 | $0.776(4)$ |
| H15A | $0.776(4)$ |
| H15B | $0.776(4)$ |
| H15C | $0.776(4)$ |
| C15B | $0.224(4)$ |
| H15D | $0.224(4)$ |
| H15E | $0.224(4)$ |
| H15F | $0.224(4)$ |

## Citations

COSMO-V1.61 - Software for the CCD Detector Systems for Determining Data Collection Parameters, Bruker axs, Madison, WI (2000).
O.V. Dolomanov and L.J. Bourhis and R.J. Gildea and J.A.K. Howard and H. Puschmann, Olex2: A complete structure solution, refinement and analysis program, J. Appl. Cryst., (2009), 42, 339-341.

Sheldrick, G.M., Crystal structure refinement with ShelXL, Acta Cryst., (2015), C27, 3-8.
Software for the Integration of CCD Detector System Bruker Analytical X-ray Systems, Bruker axs, Madison, WI (after 2013).

## Copies of NMR spectra




${ }^{1} \mathrm{H}$ NMR, $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$



























4.3a
${ }^{1} \mathrm{H}$ NMR, $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$


$\stackrel{\stackrel{\circ}{\%}}{\underset{i}{i}}$

SiMe
4.3a
${ }^{13} \mathrm{C}\left\{{ }^{1} \mathrm{H}\right\}$ NMR, $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$








SiMe
4.3d
${ }^{1} \mathrm{H}$ NMR, $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$



4.3d
${ }^{13} \mathrm{C}\left\{{ }^{1} \mathrm{H}\right\}$ NMR, $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$



syn/anti-4.4a dr = 1:1
${ }^{1} \mathrm{H}$ NMR, $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$




syn/anti-4.4b $d r=1: 1$
${ }^{13} \mathrm{C}\left\{{ }^{1} \mathrm{H}\right\}$ NMR, $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$

| 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 210 | 200 | 190 | 180 | 170 | 160 | 150 | 140 | 130 | 120 | 110 | $\begin{gathered} 100 \\ \mathrm{f} 1(\mathrm{ppm}) \end{gathered}$ | 90 | 80 | 70 | 60 | 50 | 40 | 30 | 20 | 10 | 0 | -10 |




syn/anti-4.4c $d r=1: 1$
${ }^{1} \mathrm{H}$ NMR, $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$





$$
\text { II } \mid 111
$$


syn/anti-4.4d dr = 1:1
${ }^{1} \mathrm{H}$ NMR, $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$


どん

syn／anti－4．4d dr＝1：1
${ }^{13} \mathrm{C}\left\{{ }^{1} \mathrm{H}\right\}$ NMR， $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$

 $\xrightarrow{\omega}$

$$
\left\|\|_{1} \quad / \quad l_{1 /} \quad 1 \quad 1\right.
$$


${ }^{1} \mathrm{H}$ NMR, $600 \mathrm{MHz}, \mathrm{CDCl}_{3}$







anti-4.4f
${ }^{1} \mathrm{H}$ NMR, $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$



| 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 210 | 200 | 190 | 180 | 170 | 160 | 150 | 140 | 130 | 120 | 110 | 100 | 90 | 80 | 70 | 60 | 50 | 40 | 30 | 20 | 10 | 0 | -10 |





```
%
```


111111111

anti-4.4h
${ }^{1} \mathrm{H}$ NMR, $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$




|  | 管 管管 |  |  |  |
| :---: | :---: | :---: | :---: | :---: |
| すよす jo did |  |  | ¢مٌ | مّ¢ |
|  | $\xrightarrow{+1}$ | ｜｜ | V | V |


syn／anti－4．4i $d r=1: 1$
${ }^{13} \mathrm{C}\left\{{ }^{1} \mathrm{H}\right\}$ NMR， $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$

cis-4.5a
${ }^{1} \mathrm{H}$ NMR, $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$











$\stackrel{\circ}{0}$

trans-4.5c
${ }^{13} \mathrm{C}\left\{{ }^{1} \mathrm{H}\right\}$ NMR, $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$



cis-4.5d
${ }^{1} \mathrm{H}$ NMR, $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$






cis-4.5e
${ }^{1} \mathrm{H}$ NMR, $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$



trans-4.5e
${ }^{1} \mathrm{H}$ NMR, $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$











${ }^{1} \mathrm{H}$ NMR, $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$








${ }^{1} \mathrm{H}$ NMR, $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$



4.6a
${ }^{29}$ Si NMR, $99 \mathrm{MHz}, \mathrm{CDCl}_{3}$


4.6a'
${ }^{1} \mathrm{H}$ NMR, $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$









4.6c
${ }^{1} \mathrm{H}$ NMR, $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$



${ }^{1} \mathrm{H}$ NMR, $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$




|  |  |
| :---: | :---: |
|  |  |


4.11c
${ }^{13} \mathrm{C}\left\{{ }^{1} \mathrm{H}\right\}$ NMR, $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$


${ }^{19} \mathrm{~F}$ NMR, $470 \mathrm{MHz}, \mathrm{CDCl}_{3}$

|  |  |  |  |  |  |  |  |  |  | , |  |  |  | I |  |  |  |  |  |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 30 | 20 | 10 | 0 | -10 | -20 | -30 | -40 | -50 | -60 | -70 |  | $\mathrm{ppm})^{-90}$ | -100 | -110 | -120 | -130 | -140 | -150 | -160 | -170 | -180 | -190 | -21 |


${ }^{1} \mathrm{H}$ NMR, $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$




4.6e
${ }^{1} \mathrm{H}$ NMR, $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$






 ( $\underbrace{\text { r }}$

4.6 g





4.6i
${ }^{1} \mathrm{H}$ NMR, $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$







4.8a
${ }^{13} \mathrm{C}\left\{{ }^{1} \mathrm{H}\right\}$ NMR, $151 \mathrm{MHz}, \mathrm{CDCl}_{3}$



| $\stackrel{\stackrel{\rightharpoonup}{W}}{\sim}$ | 碼 |  |  |  | \% | $\stackrel{\square}{\sim}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |


4.8a'
${ }^{13} \mathrm{C}\left\{{ }^{1} \mathrm{H}\right\}$ NMR, $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$


4.8a'
${ }^{29} \mathrm{Si} \mathrm{NMR}, 99 \mathrm{MHz}, \mathrm{CDCl}_{3}$


${ }^{13} \mathrm{C}\left\{{ }^{1} \mathrm{H}\right\}$ NMR, $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$



4.8c
${ }^{1} \mathrm{H}$ NMR, $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$



4.8c
${ }^{13} \mathrm{C}\left\{{ }^{1} \mathrm{H}\right\}$ NMR, $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$

4.8d
${ }^{1} \mathrm{H}$ NMR, $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$




4.8e
${ }^{1} \mathrm{H}$ NMR, $600 \mathrm{MHz}, \mathrm{CDCl}_{3}$






[^6]


| 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 230 | 220 | 210 | 200 | 190 | 180 | 170 | 160 | 150 | 140 | 130 | 120 | 110 | 100 | 90 | 80 | 70 | 60 | 50 | 40 | 30 | 20 | 10 | 0 | -10 |







[^7]






|  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 230 | 220 | 210 | 200 | 190 | 180 | 170 | 160 | 150 | 140 | 130 | 120 | $\stackrel{110}{\mathrm{f}_{1}(\mathrm{~nm})}$ | 100 | 90 | 80 | 70 | 60 | 50 | 40 | 30 | 20 | 10 | 0 | -10 |



4.9b-d $d_{1}$
${ }^{1} \mathrm{H}$ NMR, $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$



4.9b-d $1_{1}$
${ }^{29}$ Si NMR, $99 \mathrm{MHz}, \mathrm{CDCl}_{3}$

## $\prod_{10}$




$$
j_{1} l_{1}
$$


4.9c
${ }^{1} \mathrm{H}$ NMR, $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$



## CHAPTER 5.

## SERENDIPITOUS [1,2]-CARBON-TO-CARBON SILYL MIGRATION IN $\alpha$-HYRDOXY ALLYL SILANES: ACCESS TO $\alpha$-SILYL ALKANALS

### 5.1. Introduction

As we have seen in chapter 4, carbon to carbon silyl migration is rare compared to carbon to oxygen (Brook rearrangement) ${ }^{1-4}$ or oxygen to carbon (retro-Brook rearrangement) ${ }^{5-10}$ silyl shifts. Aldehydes and ketones with a silyl group at the $\alpha$-position are good precursors for vinyl silyl ethers, ${ }^{11}$ aldol reactions, ${ }^{12} \alpha$-amino ketones, ${ }^{13}$ and Peterson olefinations ${ }^{14,15}$ (Scheme 5.1)


Scheme 5.1: Selected synthetic applications of $\alpha$-silyl ketones and aldehydes
Despite their synthetic usefulness, few methods of accessing $\alpha$-silyl ketones and aldehydes have been reported. Example of these methods are lithiation followed by silyl vinylation, ${ }^{11}$
synthesis from 3-silyl-2,3-epoxy alcohols, ${ }^{16}$ and catalytic asymmetric Roskamp reaction ${ }^{17}$ (Scheme 5.2).


Scheme 5.2: Selected syntheses of $\alpha$-silyl ketones and aldehydes

### 5.2. Serendipitous [1,2]-carbon-to-carbon silyl migration

During preparation of one of the intermediates for Wittig rearrangement in Chapter 4 (Scheme
5.3), a peak at around 9.5 ppm in ${ }^{1} \mathrm{H}$ NMR corresponding to an aldehyde was observed (Figure
5.1).


Scheme 5.3: Synthesis of 2-methyl-1-(trimethylsilyl)prop-2-en-1-ol 4.3b by retro-Brook rearrangement


Figure 5.1: ${ }^{1} \mathrm{H}$ NMR of crude reaction mixture
At first, the peak was thought to be from a byproduct or impurity in the crude reaction mixture. After purification by column chromatography, the peak remained as a $\sim 1: 1$ mixture with the compound of interest (Figure 5.2).


Figure 5.2: ${ }^{1} \mathrm{H}$ NMR of purified material in $\mathrm{CDCl}_{3}$
At a glance, it was unclear whether this unknown compound had the same $R_{f}$ value as the compound of interest or if it was being formed from that compound in the NMR tube. To further investigate this, the sample in the NMR tube was left overnight ( $\sim 12$ hours) and resubmitted for NMR analysis the following day. Surprisingly, the aldehyde peak had increased in intensity, overshadowing the compound of interest (Figure 5.3).

Figure 5.3: ${ }^{1} \mathrm{H}$ NMR of purified material in $\mathrm{CDCl}_{3}$ after 12 hours
Having determined that the unknown compound was being formed from the compound of interest, we embarked on deducing its structure. It turned out to be 2-methyl-2(trimethylsilyl)propanal, which is formed as a result of an irreversible [1,2]-carbon-to-carbon silyl migration of 2-methyl-1-(trimethylsilyl)prop-2-en-1-ol (Scheme 5.4).


Scheme 5.4: The [1,2]-carbon-to-carbon silyl migration of 2-methyl-1-(trimethylsilyl)prop-2-en-1-ol

Although this rearrangement looks trivial, we were surprised to find only one report of such migration in the literature (Scheme 5.5). ${ }^{18}$ In this paper, the authors stated, "This unusual rearrangement reaction was NOT appreciably accelerated by the presence of acids $(\mathrm{HCl}$ gas, TsOH ), and was completely inhibited by weak bases (pyridine, $\mathrm{Et}_{3} \mathrm{~N}$ ) and even THF. We are not
aware of a precedent for this rearrangement process."


Scheme 5.5: Silyl migration of compound $\mathbf{5 . 2}$ in DCM
With this information, we decided to do further experiments to determine why this migration was occurring. Knowing that over many months of storage at room temperature, deuterated chloroform can become acidic, ${ }^{19}$ we acquired the ${ }^{1} \mathrm{H}$ NMR in deuterated chloroform that was stored over $\mathrm{K}_{2} \mathrm{CO}_{3}$. Compound 5.1a did not form immediately, but the transformation of compound 4.3b to 5.1a slowly occurred. Thus, the chloroform stored over $\mathrm{K}_{2} \mathrm{CO}_{3}$ only slows the reaction but it does not stop the process. Notably even with new bottle of chloroform, the aldehyde was formed overnight. However, compound 4.3b was stable in deuterated benzene and the aldehyde was not formed even after weeks (Figure 5.4).


Figure 5.4: ${ }^{1} \mathrm{H}$ NMR of purified material in $\mathrm{C}_{6} \mathrm{D}_{6}$

### 5.3. Substrate scope for [1,2]-carbon-to-carbon silyl migration

Next, we looked at the substrate scope of this migration. We began by modifying the substituents on silicon: Moving from trimethyl to triethyl, tripropyl and diphenyl methyl groups all effected the migration in varying amount of time (Scheme 5.6, substrates 5.1b-5.1d). Having an isopropyl group in place of methyl at the olefin carbon proximal to silicon also worked well (Scheme 5.6, substrates 5.1e and 5.1f). Finally, [1,2]-silyl migration also occurred on a trisubstituted olefin 5.1g.


Scheme 5.6: Substrate scope for [1,2]-carbon to carbon migration in NMR tube ${ }^{\text {a Conversion ( }} \mathrm{y} \%$ ) was determined by ${ }^{1} \mathrm{H}$ NMR integration

We then tested the importance of having an alkyl substituent on the olefin carbon $\beta$ to silicon. The silyl migration was not observed in compounds $\mathbf{4 . 3}$ a and $\mathbf{5 . 3}$ even after days of sitting in NMR tube with $\mathrm{CDCl}_{3}$ as a solvent (Scheme 5.7).


Scheme 5.7: Substrate scope limitation
When a mixture of $\mathbf{5 . 3}$ and $\mathbf{5 . 1 f}$ was dissolved in $\mathrm{CDCl}_{3}$ from a bottle that contained $\mathrm{K}_{2} \mathrm{CO}_{3}$, both alcohols could be observed at time zero. After 18 hours, alcohol 5.1f had converted to aldehyde 5.2 f while alcohol 5.3 remained unchanged. On the fifth day, alcohol 5.1 f had completely transformed to aldehyde 5.2f while silyl migration was not detected on alcohol 5.3. (Figure 5.5)


Figure 5.5: ${ }^{1} \mathrm{H}$ NMR of compounds $\mathbf{5 . 1 f}$ and $\mathbf{5 . 3}$ over time

### 5.4. Proposed reaction mechanism of the [1,2]-carbon-to-carbon silyl migration

With the above observations of importance of having the alkyl group at the olefin carbon $\beta$ to the silyl group, we propose that this reaction is acid catalyzed and proceeds by the olefin abstracting proton followed by subsequent silyl migration which is facilitated by carbonyl formation (Scheme 5.8).


Scheme 5.7: Proposed reaction mechanism of the silyl shift

### 5.5. Attempted mechanistic investigation of the [1,2]-carbon-to-carbon silyl migration

To investigate whether the proposed mechanism is concerted or stepwise. Compound $\mathbf{5 . 1 f}$ was studied since the silyl migration would create a new stereogenic center. An enantiomerically enriched 5.1f will lead to $\mathbf{5 . 2 f}$ with inversion / retention of stereochemistry (concerted) or erosion of stereochemistry (stepwise) with carbocation intermediate (Scheme 5.8).


Scheme 5.8: Reaction design to investigate mechanistic pathways
With the above idea, we began with synthesizing the enantiomerically enriched $\mathbf{5 . 1 f}$ by first oxidizing the racemic $\mathbf{5 . 1 f}$ followed by asymmetric reduction of the resulting acyl silane $\mathbf{5 . 4 f}$ using Corey-Bakshi-Shibata (CBS) catalyst ${ }^{36}$ to generate 5.1f* (Scheme 5.9).


Scheme 5.9: Synthesis of enantioenriched 5.1f
With compound 5.1f* at hand, we subjected it to conditions leading to [1,2]-carbon-to-carbon silyl shift. Unfortunately, attempts to derivatize the resulting aldehyde with the purpose of determining absolute stereochemistry were unsuccessful. Aldehyde 5.2 f was reacted with $2,4-$ dinitrophenyl hydrazine with intention of forming 2,4-dinitrophenyl hydrazone (5.5), but instead, desilylated hydrazone 5.6 was formed (Scheme 5.10). The proposed mechanism is shown in Scheme 5.11.


Scheme 5.10: Attempted derivatization of $\mathbf{5 . 2 f}$


Scheme 5.11: Proposed reaction mechanism for the formation of the observed product 5.6
Whether the mechanism of the [1,2]-carbon to carbon silyl migration is stepwise (involving carbocation formation) or concerted is not known at this time. Formation of carbocation intermediate is more likely since it would be doubly stabilized: a tertiary carbocation ${ }^{20-22}$ and $\beta$ to silicon. ${ }^{23-35}$ Further investigations will be performed by others.

### 5.6. Unexpected $S_{\mathrm{N}}$ 2-like reaction between the O -silylated 2-phenylprop-2-en-1-ol and butyl lithiums

Knowing the importance of having a substituent other than proton on the olefin carbon $\beta$ to silicon, we embarked on synthesizing substrates containing aromatic groups at this position. In our
quest to expand the substrate scope of the silyl migration, we saw an unexpected transformation involving $\mathrm{S}_{\mathrm{N}} 2$-like reaction between the O -silylated 2-phenylprop-2-en-1-ol and butyl lithiums (Scheme 5.12). To the best of our knowledge, this reaction had no literature precedence. Use of tert-butyllithium and sec-butyllithium resulted in the formation of products 5.8a and $\mathbf{5 . 8} \mathbf{b}$ in high yields. However, when we employed $n$-butyllithium, product $\mathbf{5 . 8} \mathbf{c}$ was formed as a mixture with desilylated alcohol $\mathbf{5 . 9 h}$ which could have resulted from unreacted starting material during workup. Lastly, there exists a challenge to access molecules with all C -sp ${ }^{3}$ quaternary center by $\mathrm{C}-\mathrm{C}$ coupling reactions. ${ }^{37-50}$ This discovery will be helpful in accessing such products.


Scheme 5.12: Unexpected $\mathrm{S}_{\mathrm{N}} 2$-like reactions
${ }^{\text {a }}$ The product was formed as a mixture with 2-phenyl ally alcohol

### 5.7. Conclusion

In summary, we have serendipitously discovered an irreversible [1,2]-carbon-to-carbon silyl migration leading to $\alpha$-silylated alkanals. This migration proceeds even when the substituents on silicon are modified but requires an alkyl group at the olefin carbon $\beta$ to silicon. Preliminary studies have also shown that the migration is acid catalyzed.

Further experimental studies to determine the mechanistic pathway of this migration are
underway. Lastly, we have seen unexpected $\mathrm{S}_{\mathrm{N}} 2$-like reaction between the O -silylated 2-phenylprop-2-en-1-ol and butyl lithiums in attempted retro-Brook rearrangement to expand the substrate scope. This reaction not only takes place on an in situ generated O-Silylated alcohol but also occurs when the O -silyl alcohol is prepared separately.

### 5.8. Experimental section

### 5.8.1. General information

Unless otherwise noticed all reactions were run under a positive atmosphere of nitrogen in oven- dried (at least 4 hours) or flame-dried round bottom flasks or disposable drum vials capped with rubber septa. Solvents were removed by rotary evaporation at temperatures lower than $45^{\circ} \mathrm{C}$. Column chromatography was run on 230-400 mesh silica gel. Tetrahydrofuran and diethyl ether were distilled from sodium-benzophenone ketyl; dichloromethane and trimethylsilyl chloride were distilled from calcium hydride. Triethylsilyl chloride, tripropylsilyl chloride, and diphenylmethylsilyl chloride were used as received. $n$-Butyllithum ( 2.5 M in hexanes) and secbutyllithium (1.4 M in cyclohexane) were purchased from Aldrich and their concentration calculated by titration with diphenylacetic acid (average of three runs). ${ }^{1} \mathrm{H}$ NMR spectra was collected in 500 MHz Varian instruments using $\mathrm{CDCl}_{3}$ as solvent, which was referenced at 7.26 ppm (residual chloroform proton) and ${ }^{13} \mathrm{C}$ NMR spectra was collected in $\mathrm{CDCl}_{3}$ at 126 MHz or 151 MHz and referenced at 77.0 ppm . Another deuterated solvent used for NMR analysis was benzene (referenced at 7.16 for ${ }^{1} \mathrm{HNMR}$ and 128.39 for ${ }^{13} \mathrm{CNMR}$ ). High resolution mass spectrometric (HRMS) analysis was run in TOF instruments.

### 5.8.2. Synthesis of ally alcohols 5.9 from propargyl alcohols: General procedure $A$



Following a reported procedure, ${ }^{51}$ for a 50 mmol scale reaction, to a dry 500 mL 3-neck round bottomed flask fitted with a magnetic stir bar was weighed 4.01 g of magnesium powder (165 mmol, 3.3 equiv), 3 crystals of iodine and 125 mL of freshly distilled dry THF. The two side necks of the flask were sealed with rubber septa and a reflux condenser was attached to the middle neck and the whole system purged with nitrogen after placing an oil bath was placed underneath the flask. On a separate 250 mL round bottomed flask a solution corresponding to $150 \mathrm{mmol}(3.0$ equiv) of alkyl/aryl halide in 100 mL dry THF was prepared. The 250 mL flask was sealed with a rubber septum and purged with nitrogen. The 250 mL flask was then connected to the 500 mL flask via canula. The alkyl/aryl halide solution was transferred in a dropwise manner for $\sim 30$ minutes to the 500 mL flask via canula while stirring and monitoring both the temperature of the oil bath and the reaction mixture in the 500 mL flask. After complete addition, the temperature of the oil bath had risen to $35-40^{\circ} \mathrm{C}$. The canula was removed and the oil bath was heated to $75^{\circ} \mathrm{C}$ to allow the Grignard reagent to form over 1.5 hours. The heat was then turned off and the reaction mixture allowed to cool down to room temperature slowly without removing the oil bath. This was followed by addition of 1.43 g of copper (I) iodide ( $7.5 \mathrm{mmol}, 0.15$ equiv) which was done by removing one of the side neck rubber septum, quickly adding the CuI and replacing the septum fast enough to minimize contact with air. The resulting mixture was stirred at room temperature for 30 minutes after which 2.91 mL ( 50 mmol , 1 equiv) of propargyl alcohol as a solution in 25
mL dry THF was added dropwise via syringe. After complete addition, the mixture was heated to $75^{\circ} \mathrm{C}$ and stirred at this temperature for 24 hours. This was followed by turning off the heat and allowing the mixture to cool to room temperature. The oil bath was removed and replaced with an ice bath to cool down the mixture further to $0^{\circ} \mathrm{C}$. The mixture was quenched by slow addition of 80 mL of water. The reaction mixture was transferred to a 1000 mL separatory funnel and diluted with 100 mL ethyl ether. The layers were separated, and the aqueous layer was extracted with ether ( 80 mL X 3 ). Combined organic layers were washed with 80 mL water and 80 mL brine respectively then dried over anhydrous magnesium sulfate. The mixture was filtered, and the filtrate was concentrated on a rotorvap under reduced pressure to afford allyl alcohol 5.9 which was purified by column chromatography (Hexanes/EtOAc).

### 5.8.3. Preparation of $\alpha$-hydroxy allyl silanes 4.3 - general procedure $B:^{\mathbf{5 2}}$



A solution of the corresponding allylic alcohol in THF was cooled at $-78^{\circ} \mathrm{C}$, and $n$ butyllithium (1.6 M or 2.5 M in hexanes) was added dropwise over 5 min . After 30 min the corresponding chlorosilane was added dropwise via syringe. After the resulting solution was stirred for a given amount of time (see individual compounds procedure below), sec-butyllithium or tert-butyllithium (see below for details) was added dropwise over $30-60 \mathrm{~min}$, and then the reaction was kept at the indicated temperature.

Synthesis of 3-methyl-2-methylenebutan-1-ol (5.9e) and ( $E$ )-4-methylpent-2-en-1-ol (5.10)


Following the general procedure A with slight modification compounds $\mathbf{5 . 9}$ e and $\mathbf{5 . 1 0}$ were synthesized from 8.00 g ( $330 \mathrm{mmol}, 3.3$ equiv) of magnesium, 28 mL ( $300 \mathrm{mmol}, 3.0$ equiv) of isopropyl bromide, $2.86 \mathrm{~g} \mathrm{CuI}(15 \mathrm{mmol}, 0.15$ equiv) and $5.85 \mathrm{~mL}(100 \mathrm{mmol}, 1.0$ equiv) of propargyl alcohol: After reflux (formation of the Grignard reagent) the Grignard reagent was allowed to cool down to around 40 degrees Celsius. On a separate flask, copper (I) iodide and propargyl alcohol were stirred at $0^{\circ} \mathrm{C}$ in THF ( 100 mL ). The Grignard reagent was transferred to this flask via cannula and the reaction proceeded at room temperature for 24 hours and then quenched at minus 10 degrees Celsius, worked up and solvent evaporated. Purification by column chromatography ( $30 \% \mathrm{Et}_{2} \mathrm{O}$ in hexanes) afforded total of $7428 \mathrm{mg}, 74 \mathrm{mmol}$ ( $74 \%$ isolated yield) of 5.9e and 5.10, as a 5:1 mixture: a $74 \%$ yield. 803 mg of another side product $\mathbf{5 . 1 1}$ was also formed (see structure below).

Spectroscopic data for 5.9e: ${ }^{1} \mathrm{H}$ NMR $\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 4.97(\mathrm{q}, J=1.5 \mathrm{~Hz}, 1 \mathrm{H}), 4.86(\mathrm{p}$, $J=1.2 \mathrm{~Hz}, 1 \mathrm{H}), 4.08(\mathrm{t}, J=1.3 \mathrm{~Hz}, 2 \mathrm{H}), 2.29(\mathrm{hept}, J=6.4 \mathrm{~Hz}, 1 \mathrm{H}), 2.18(\mathrm{~s}, 1 \mathrm{H}), 1.04(\mathrm{~d}, J=6.9$ $\mathrm{Hz}, 6 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 155.1,106.8,64.7,30.9$, 21.7. Spectroscopic data were similar to those reported in the literature. ${ }^{53}$ The minor product $\mathbf{5 . 1 0}$ spectroscopic data also matched the literature report. ${ }^{54}$


Spectroscopic data for $(E)$-2-isopropyl-4-methylenepent-2-ene-1,5-diol 5.11: ${ }^{1} \mathrm{H}$ NMR (500 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 5.33(\mathrm{dt}, J=10.2,1.1 \mathrm{~Hz}, 1 \mathrm{H}), 5.26(\mathrm{q}, J=1.5 \mathrm{~Hz}, 1 \mathrm{H}), 4.91(\mathrm{dd}, J=1.9,1.0 \mathrm{~Hz}$, $1 \mathrm{H}), 4.10-4.06(\mathrm{~m}, 2 \mathrm{H}), 4.02(\mathrm{~d}, J=1.1 \mathrm{~Hz}, 2 \mathrm{H}), 3.43(\mathrm{~s}, 2 \mathrm{H}), 2.48(\mathrm{dp}, J=10.1,6.6 \mathrm{~Hz}, 1 \mathrm{H})$, $0.92(\mathrm{~d}, J=6.7 \mathrm{~Hz}, 6 \mathrm{H}) .{ }^{13} \mathrm{C} \operatorname{NMR}\left(126 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 146.2,138.4,136.2,114.7,67.4,65.6$, 27.9, 23.1.

## Synthesis of (E)-2-methylbut-2-en-1-ol (5.9g)



Following procedure A: This time around having the propargyl alcohol and CuI in a 500 mL RB flask at $0^{\circ} \mathrm{C}, \mathrm{MeMgCl}$ as a solution in THF (3M) was added. Accidentally, the flask cracked while quenching and spilled the reaction mixture in the hood. A small amount of the reaction mixture was recovered and worked up. Purification by column chromatography afforded 731 mg as a mixture of $\mathbf{5 . 9 g}$ and the starting propargyl alcohol $(\mathbf{5 . 9 g} 66 \% \mathrm{~W} / \mathrm{W}$ with but-2-yn-1-ol). Spectroscopic data for $5.9 \mathrm{~g}:{ }^{1} \mathrm{H} \operatorname{NMR}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 5.46(\mathrm{q}, J=6.8 \mathrm{~Hz}, 1 \mathrm{H}), 3.96(\mathrm{~s}, 2 \mathrm{H})$, $2.14(\mathrm{~s}, 2 \mathrm{H}), 1.63(\mathrm{~s}, 3 \mathrm{H}), 1.59(\mathrm{~d}, J=6.7 \mathrm{~Hz}, 3 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 135.3,120.5$, $68.8,13.2,13.0$. Spectroscopic data were similar to those reported in the literature. ${ }^{55}$

## Synthesis of 2-phenylprop-2-en-1-ol (5.9h)



Applying procedure A to $4 \mathrm{~g}(165 \mathrm{mmol}, 3.3$ equiv $)$ of magnesium, $15.8 \mathrm{~mL}(150 \mathrm{mmol}, 3.0$ equiv) of bromobenzene, $1.43 \mathrm{~g} \mathrm{CuI}(7.5 \mathrm{mmol}, 0.15$ equiv) and $2.88 \mathrm{~mL}(50 \mathrm{mmol}, 1.0$ equiv) of propargyl alcohol, alcohol 5.9 h was prepared in quantitative yield: ${ }^{1} \mathrm{H} \mathrm{NMR}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ $7.48-7.43(\mathrm{~m}, 2 \mathrm{H}), 7.40-7.34(\mathrm{~m}, 2 \mathrm{H}), 7.34-7.29(\mathrm{~m}, 1 \mathrm{H}), 5.48(\mathrm{q}, J=1.0 \mathrm{~Hz}, 1 \mathrm{H}), 5.36(\mathrm{q}, J$ $=1.4 \mathrm{~Hz}, 1 \mathrm{H}), 4.53(\mathrm{dd}, J=1.5,0.9 \mathrm{~Hz}, 2 \mathrm{H}), 2.22(\mathrm{~s}, 1 \mathrm{H}) .{ }^{13} \mathrm{C} \mathrm{NMR}\left(126 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 147.1$, $138.4,128.4,127.8,126.0,112.5,64.7$. Spectroscopic data were similar to those reported in the literature. ${ }^{56}$

## Attempted synthesis of 2-benzylprop-2-en-1-ol (5.9i)



Applying general procedure A to $3.65 \mathrm{~g}(150 \mathrm{mmol}, 3.0$ equiv $)$ of magnesium, $16.04 \mathrm{~mL}(135$ mmol, 2.7 equiv) of benzyl bromide, $1.43 \mathrm{~g} \mathrm{CuI}(7.5 \mathrm{mmol}, 0.15$ equiv) and 2.88 mL ( 50 mmol , 1.0 equiv) of propargyl alcohol, compound $\mathbf{5 . 9}$ i could not be obtained. Instead, 1,2-diphenylethane 5.12i was formed in quantitative yield. Spectroscopic data for $\mathbf{5 . 1 2 i}:{ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.42-7.34(\mathrm{~m}, 4 \mathrm{H}), 7.33-7.25(\mathrm{~m}, 6 \mathrm{H}), 3.03(\mathrm{~s}, 4 \mathrm{H}) .{ }^{13} \mathrm{C} \mathrm{NMR}\left(126 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ 141.7, 128.4, 128.3, 125.9, 37.9. Spectroscopic data were similar to those reported in the literature. ${ }^{57}$

## Synthesis of 2-methyl-1-(triethylsilyl)prop-2-en-1-ol (5.1b)



Following general procedure B, a solution of 1.26 mL 2-methyl allyl alcohol ( $1081.7 \mathrm{mg}, 15$ mmol, 1 equiv.) in THF ( 25 mL ) was cooled to $-78^{\circ} \mathrm{C} . n-\mathrm{BuLi}(2.4 \mathrm{M}$ in hexanes, $7.5 \mathrm{~mL}, 18$ mmol, 1.2 equiv.) was added dropwise and the mixture stirred at $-78{ }^{\circ} \mathrm{C}$ for 1 h . Chlorotriethylsilane ( $2.52 \mathrm{~mL}, 15 \mathrm{mmol}, 1.0$ equiv.) was then added slowly from a syringe and the resulting mixture was stirred at $-78^{\circ} \mathrm{C}$ to room temperature for 18 hours resulting in the formation of a white suspension. The mixture was cooled back to $-78{ }^{\circ} \mathrm{C}$ followed by dropwise addition of sec-butyllithium (1.4 M in cyclohexane, $12.9 \mathrm{~mL}, 18 \mathrm{mmol}, 1.2$ equiv.) and the reaction stirred for an additional 2 hours at $-78^{\circ} \mathrm{C}$ to $-50^{\circ} \mathrm{C}$. The reaction mixture was cooled back to $-78{ }^{\circ} \mathrm{C}$ and quenched by the addition of aqueous $\mathrm{NH}_{4} \mathrm{Cl}$ and diluted with $\mathrm{Et}_{2} \mathrm{O}$, and the mixture was warmed to room temperature. After the layers were separated, the aqueous phase was washed with $\mathrm{Et}_{2} \mathrm{O}$ (3 $\times 20 \mathrm{~mL})$. Then all the organic phases were combined, washed with $\mathrm{H}_{2} \mathrm{O}(20 \mathrm{~mL})$ and brine (20 mL ) respectively, and dried over anhydrous $\mathrm{MgSO}_{4}$. After filtration and concentration, the residue was purified by column chromatography, $R_{f}=0.3\left(15 \% \mathrm{Et}_{2} \mathrm{O}\right.$ in hexanes) to afford $1.4 \mathrm{~g}, 7.5 \mathrm{mmol}$ ( $50 \%$ isolated yield) of compound $\mathbf{5 . 1 b}$ as a colorless liquid. ${ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{C}_{6} \mathrm{D}_{6}$ ) $\delta 4.88$ $(\mathrm{tq}, J=1.7,0.8 \mathrm{~Hz}, 1 \mathrm{H}), 4.74(\mathrm{~h}, J=1.3 \mathrm{~Hz}, 1 \mathrm{H}), 3.80(\mathrm{~s}, 1 \mathrm{H}), 1.61(\mathrm{dt}, J=1.3,0.7 \mathrm{~Hz}, 3 \mathrm{H}), 1.23$ $(\mathrm{s}, 1 \mathrm{H}), 1.02(\mathrm{t}, J=7.9 \mathrm{~Hz}, 9 \mathrm{H}), 0.65(\mathrm{qd}, J=14.9,7.6 \mathrm{~Hz}, 6 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $\left.126 \mathrm{MHz}, \mathrm{C}_{6} \mathrm{D}_{6}\right) \delta$ 149.4, 107.1, 70.1, 21.0, 8.1, 2.8.

## Synthesis of 2-methyl-1-(tripropylsilyl)prop-2-en-1-ol (5.1c)



Following general procedure B , a solution of 0.84 mL 2-methyl allyl alcohol $(721.1 \mathrm{mg}, 10$ mmol, 1 equiv.) in THF ( 25 mL ) was cooled to $-78^{\circ} \mathrm{C} . n-\operatorname{BuLi}(2.4 \mathrm{M}$ in hexanes, $5 \mathrm{~mL}, 12 \mathrm{mmol}$, 1.2 equiv.) was added dropwise and the mixture stirred at $-78^{\circ} \mathrm{C}$ for 1 h . Chlorotripropylsilane ( $2.2 \mathrm{~mL}, 10 \mathrm{mmol}, 1.0$ equiv.) was then added slowly from a syringe and the resulting mixture was stirred at $-78^{\circ} \mathrm{C}$ to room temperature for 18 hours resulting in the formation of a white suspension. The mixture was cooled back to $-78^{\circ} \mathrm{C}$ followed by dropwise addition of sec-butyllithium (1.4 M in cyclohexane, $8.6 \mathrm{~mL}, 12 \mathrm{mmol}, 1.2$ equiv.) and the reaction stirred for an additional 2 hours at $-78{ }^{\circ} \mathrm{C}$ to $-50^{\circ} \mathrm{C}$. The reaction mixture was cooled back to $-78{ }^{\circ} \mathrm{C}$ and quenched by the addition of aqueous $\mathrm{NH}_{4} \mathrm{Cl}$ and diluted with $\mathrm{Et}_{2} \mathrm{O}$, and the mixture was warmed to room temperature. After the layers were separated, the aqueous phase was washed with $\mathrm{Et}_{2} \mathrm{O}(3 \times 20 \mathrm{~mL})$. Then all the organic phases were combined, washed with $\mathrm{H}_{2} \mathrm{O}(20 \mathrm{~mL})$ and brine $(20 \mathrm{~mL})$ respectively, and dried over anhydrous $\mathrm{MgSO}_{4}$. After filtration and concentration, the residue was purified by column chromatography ( $10 \% \mathrm{Et}_{2} \mathrm{O}$ in hexanes) to afford $491 \mathrm{mg}, 2.2 \mathrm{mmol}(22 \%$ isolated yield) of compound 5.1c as a colorless liquid. ${ }^{1} \mathrm{H}$ NMR $\left(500 \mathrm{MHz}, \mathrm{C}_{6} \mathrm{D}_{6}\right) \delta 4.89(\mathrm{tt}, J=1.7,0.7 \mathrm{~Hz}, 1 \mathrm{H})$, $4.75(\mathrm{p}, J=1.4 \mathrm{~Hz}, 1 \mathrm{H}), 3.80(\mathrm{~s}, 1 \mathrm{H}), 1.63(\mathrm{dd}, J=1.6,0.7 \mathrm{~Hz}, 3 \mathrm{H}), 1.42(\mathrm{dddd}, J=14.4,12.8$, $11.1,7.3 \mathrm{~Hz}, 6 \mathrm{H}), 1.01(\mathrm{t}, J=7.3 \mathrm{~Hz}, 9 \mathrm{H}), 0.97(\mathrm{~s}, 1 \mathrm{H}), 0.74-0.59(\mathrm{~m}, 6 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( 126 MHz , $\left.\mathrm{C}_{6} \mathrm{D}_{6}\right) \delta 149.4,107.2,70.5,21.1,19.3,18.3,14.7$.

## Synthesis of 2-methyl-1-(methyldiphenylsilyl)prop-2-en-1-ol (5.1d)



Following general procedure B, a solution of 1.26 mL 2-methyl allyl alcohol ( $1081.7 \mathrm{mg}, 15$ mmol, 1 equiv.) in THF ( 25 mL ) was cooled to $-78^{\circ} \mathrm{C} . n-\mathrm{BuLi}(2.4 \mathrm{M}$ in hexanes, $7.5 \mathrm{~mL}, 18$ mmol, 1.2 equiv.) was added dropwise and the mixture stirred at $-78{ }^{\circ} \mathrm{C}$ for 1 h . Chloromethyldiphenylsilane ( $3.15 \mathrm{~mL}, 15 \mathrm{mmol}, 1.0$ equiv.) was then added slowly from a syringe and the resulting mixture was stirred at $-78^{\circ} \mathrm{C}$ to room temperature for 18 hours resulting in the formation of a white suspension. The mixture was cooled back to $-78^{\circ} \mathrm{C}$ followed by dropwise addition of sec-butyllithium (1.4 M in cyclohexane, $12.9 \mathrm{~mL}, 18 \mathrm{mmol}, 1.2$ equiv.) and the reaction stirred for an additional 2 hours at $-78^{\circ} \mathrm{C}$ to $-50^{\circ} \mathrm{C}$. The reaction mixture was cooled back to -78 ${ }^{\circ} \mathrm{C}$ and quenched by the addition of aqueous $\mathrm{NH}_{4} \mathrm{Cl}$ and diluted with $\mathrm{Et}_{2} \mathrm{O}$, and the mixture was warmed to room temperature. After the layers were separated, the aqueous phase was washed with $\mathrm{Et}_{2} \mathrm{O}(3 \times 20 \mathrm{~mL})$. Then all the organic phases were combined, washed with $\mathrm{H}_{2} \mathrm{O}(20 \mathrm{~mL})$ and brine ( 20 mL ) respectively, and dried over anhydrous $\mathrm{MgSO}_{4}$. After filtration and concentration, the residue was purified by column chromatography ( $12 \% \mathrm{Et}_{2} \mathrm{O}$ in hexanes) to afford $391 \mathrm{mg}, 1.5$ $\mathrm{mmol}\left(10 \%\right.$ isolated yield) of compound $\mathbf{5 . 1 d}$ as a colorless liquid. ${ }^{1} \mathrm{H} \mathrm{NMR}\left(500 \mathrm{MHz}, \mathrm{C}_{6} \mathrm{D}_{6}\right) \delta$ $7.73-7.64(\mathrm{~m}, 2 \mathrm{H}), 7.61-7.54(\mathrm{~m}, 2 \mathrm{H}), 7.21-7.10(\mathrm{~m}, 6 \mathrm{H}), 4.86(\mathrm{dq}, J=1.7,0.9 \mathrm{~Hz}, 1 \mathrm{H}), 4.72$ (h, $J=1.4 \mathrm{~Hz}, 1 \mathrm{H}), 4.22(\mathrm{~s}, 1 \mathrm{H}), 1.34(\mathrm{~s}, 3 \mathrm{H}), 1.21(\mathrm{~s}, 1 \mathrm{H}), 0.55(\mathrm{~s}, 3 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( 126 MHz , $\left.\mathrm{C}_{6} \mathrm{D}_{6}\right) \delta 148.3,136.1,135.8,135.8,130.1,128.5,128.4,108.7,70.8,21.6,-6.1$.

## Synthesis of 3-methyl-2-methylene-1-(trimethylsilyl)butan-1-ol (5.1e) and (E)-4-methyl-1-(trimethylsilyl)pent-2-en-1-ol (5.13)



Following general procedure B , a solution of 3-methyl-2-methylenebutan-1-ol 5.9e and $(E)$ -4-methylpent-2-en-1-ol 5.10e ( $501 \mathrm{mg}, 5.0 \mathrm{mmol}$, 1 equiv.) in THF ( 10 mL ) was cooled to -78 ${ }^{\circ} \mathrm{C} . n-\mathrm{BuLi}(2.4 \mathrm{M}$ in hexanes, $2.5 \mathrm{~mL}, 12 \mathrm{mmol}, 1.2$ equiv.) was added dropwise and the mixture stirred at $-78{ }^{\circ} \mathrm{C}$ for 1 h . Chlorotrimethylsilane ( $0.65 \mathrm{~mL}, 5.0 \mathrm{mmol}, 1.0$ equiv.) was then added slowly from a syringe and the resulting mixture was stirred at $-78^{\circ} \mathrm{C}$ for 1.5 hours resulting in the formation of a white suspension. This was followed by dropwise addition of tert-butyllithium (1.6 M in pentane, $4.4 \mathrm{~mL}, 7.0 \mathrm{mmol}, 1.4$ equiv.) and the reaction stirred for an additional 1.5 hours at $-78{ }^{\circ} \mathrm{C}$ and quenched by the addition of aqueous $\mathrm{NH}_{4} \mathrm{Cl}$ and diluted with $\mathrm{Et}_{2} \mathrm{O}$, and the mixture was warmed to room temperature. After the layers were separated, the aqueous phase was washed with $\mathrm{Et}_{2} \mathrm{O}(3 \times 20 \mathrm{~mL})$. Then all the organic phases were combined, washed with $\mathrm{H}_{2} \mathrm{O}(20 \mathrm{~mL})$ and brine ( 20 mL ) respectively, and dried over anhydrous $\mathrm{MgSO}_{4}$. After filtration and concentration, the residue was purified by column chromatography ( $10 \% \mathrm{Et}_{2} \mathrm{O}$ in hexanes) to afford $51 \mathrm{mg}, 0.3 \mathrm{mmol}$ ( $6 \%$ isolated yield) of compound $\mathbf{5 . 1} \mathbf{e}$ as a colorless liquid. The very low isolated yield could be attributed to impure starting material. Alcohol $\mathbf{5 . 1 3}$ was observed in crude reaction mixture but could not be isolated. Spectroscopic data for 5.1e: ${ }^{1} \mathrm{H} N \mathrm{NRR}\left(500 \mathrm{MHz}, \mathrm{C}_{6} \mathrm{D}_{6}\right)$ $\delta 4.94(\mathrm{dd}, J=1.5,1.0 \mathrm{~Hz}, 1 \mathrm{H}), 4.82(\mathrm{q}, J=1.0 \mathrm{~Hz}, 1 \mathrm{H}), 3.74(\mathrm{~s}, 1 \mathrm{H}), 1.86(\mathrm{hept}, J=6.7 \mathrm{~Hz}, 1 \mathrm{H})$, $1.03(\mathrm{~d}, J=6.8 \mathrm{~Hz}, 3 \mathrm{H}), 0.98(\mathrm{~d}, J=6.8 \mathrm{~Hz}, 4 \mathrm{H}), 0.92(\mathrm{~s}, 1 \mathrm{H}), 0.09(\mathrm{~s}, 9 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( 126 MHz , $\left.\mathrm{C}_{6} \mathrm{D}_{6}\right) \delta 160.6,103.1,70.5,32.0,24.4,21.7,-3.00$.

## Synthesis of 3-methyl-2-methylene-1-(triethylsilyl)butan-1-ol (5.1f) and (E)-4-methyl-1-(triethylsilyl)pent-2-en-1-ol (5.3)



Following general procedure B , a solution of 3-methyl-2-methylenebutan-1-ol 5.9e and $(E)$ -4-methylpent-2-en-1-ol 5.10e ( $4.01 \mathrm{~g}, 40 \mathrm{mmol}$, 1 equiv.) in THF ( 80 mL ) was cooled to $-78{ }^{\circ} \mathrm{C}$. $n-\operatorname{BuLi}(2.5 \mathrm{M}$ in hexanes, $19.2 \mathrm{~mL}, 48 \mathrm{mmol}, 1.2$ equiv.) was added dropwise and the mixture stirred at $-78{ }^{\circ} \mathrm{C}$ for 1 h . Chlorotriethylsilane ( $8.4 \mathrm{~mL}, 50 \mathrm{mmol}, 1.25$ equiv.) was then added slowly from a syringe and the resulting mixture was stirred at $-78^{\circ} \mathrm{C}$ to room temperature for 18 hours resulting in the formation of a white suspension. The mixture was cooled back to $-78{ }^{\circ} \mathrm{C}$ followed by dropwise addition of sec-butyllithium (1.4 M in cyclohexane, $39.3 \mathrm{~mL}, 55 \mathrm{mmol}, 1.38$ equiv.) and the reaction stirred for an additional 2 hours at $-78^{\circ} \mathrm{C}$ to $-30^{\circ} \mathrm{C}$. The reaction mixture was cooled back to $-78^{\circ} \mathrm{C}$ and quenched by the addition of aqueous $\mathrm{NH}_{4} \mathrm{Cl}$ and diluted with $\mathrm{Et}_{2} \mathrm{O}$, and the mixture was warmed to room temperature. After the layers were separated, the aqueous phase was washed with $\mathrm{Et}_{2} \mathrm{O}(3 \times 20 \mathrm{~mL})$. Then all the organic phases were combined, washed with $\mathrm{H}_{2} \mathrm{O}(20 \mathrm{~mL})$ and brine ( 20 mL ) respectively, and dried over anhydrous $\mathrm{MgSO}_{4}$. After filtration and concentration, the residue was purified by column chromatography ( $15 \% \mathrm{Et}_{2} \mathrm{O}$ in hexanes) to afford a colorless liquid $4037 \mathrm{mg}, 18.8 \mathrm{mmol}$ ( $47 \%$ isolated yield) of compounds $\mathbf{5 . 1 f}$ and 5.3 as $2: 1$ mixture. This mixture was further purified by column chromatography ( $2.5 \% \mathrm{Et}_{2} \mathrm{O}$ in hexanes) to afford both compound $\mathbf{5 . 1 f}$ and $\mathbf{5 . 3}$ as pure compounds.

Spectroscopic data of 5.1f in $\mathrm{C}_{6} \mathrm{D}_{6}:{ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{C}_{6} \mathrm{D}_{6}$ ) $\delta 4.99(\mathrm{dd}, J=1.6,1.0 \mathrm{~Hz}$, $1 \mathrm{H}), 4.83(\mathrm{q}, J=1.0 \mathrm{~Hz}, 1 \mathrm{H}), 3.94(\mathrm{t}, J=1.3 \mathrm{~Hz}, 1 \mathrm{H}), 1.91(\mathrm{heptd}, J=6.8 \mathrm{~Hz}, 1 \mathrm{H}), 1.08-1.01$
$(\mathrm{m}, 12 \mathrm{H}), 1.00(\mathrm{~d}, J=6.8 \mathrm{~Hz}, 3 \mathrm{H}), 0.95(\mathrm{~s}, 1 \mathrm{H}), 0.74-0.60(\mathrm{mq} 6 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $126 \mathrm{MHz}, \mathrm{C}_{6} \mathrm{D}_{6}$ ) $\delta 161.0,103.3,68.8,31.7,24.4,21.9,8.2,2.8$.

Spectroscopic data of $\mathbf{5 . 1 f}$ in $\mathrm{CDCl}_{3}$ stored over $\mathrm{K}_{2} \mathrm{CO}_{3}:{ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 4.90$ $(\mathrm{dd}, J=1.6,0.9 \mathrm{~Hz}, 1 \mathrm{H}), 4.83(\mathrm{q}, J=0.9 \mathrm{~Hz}, 1 \mathrm{H}), 4.04(\mathrm{t}, J=1.2 \mathrm{~Hz}, 1 \mathrm{H}), 1.94$ (heptd, $J=6.8$, $1.0 \mathrm{~Hz}, 1 \mathrm{H}), 1.21(\mathrm{~s}, 1 \mathrm{H}), 1.09(\mathrm{~d}, J=6.8 \mathrm{~Hz}, 3 \mathrm{H}), 1.05(\mathrm{~d}, J=6.8 \mathrm{~Hz}, 3 \mathrm{H}), 0.98(\mathrm{t}, J=8.0 \mathrm{~Hz}$, 9H), $0.62(\mathrm{qd}, J=7.9,3.0 \mathrm{~Hz}, 6 \mathrm{H}) .{ }^{13} \mathrm{C} \operatorname{NMR}\left(126 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 160.1,102.7,68.3,31.2,23.9$, 21.2, 7.5, 1.9.

Spectroscopic data of 5.3: ${ }^{1} \mathrm{H}$ NMR $\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 5.58(\mathrm{ddd}, J=15.4,6.7,1.2 \mathrm{~Hz}, 1 \mathrm{H})$, $5.45(\mathrm{ddd}, J=15.4,6.6,1.6 \mathrm{~Hz}, 1 \mathrm{H}), 4.06(\mathrm{ddd}, J=6.6,1.6,0.9 \mathrm{~Hz}, 1 \mathrm{H}), 2.29(\mathrm{dqt}, J=13.4,6.7$, $1.1 \mathrm{~Hz}, 1 \mathrm{H}), 1.25(\mathrm{~s}, 1 \mathrm{H}), 1.03-0.91(\mathrm{~m}, 18 \mathrm{H}), 0.59(\mathrm{qd}, J=8.0,1.7 \mathrm{~Hz}, 6 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR (126 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 134.1,128.8,66.7,31.0,22.6,7.4,1.6$.

## Synthesis of 3-methyl-2-methylene-1-(triethylsilyl)butan-1-ol (5.1g) and (E)-4-methyl-1-(triethylsilyl)pent-2-en-1-ol (5.14)



Following general procedure B , to a 50 mL round bottom flask was weighed 600 mg of (E)-2-methylbut-2-en-1-ol 5.9g 66\% W/W with but-2-yn-1-ol (a total of $7.51 \mathrm{mmol}, 1$ equiv). 20 mL of freshly distilled THF was added to the flask, purged with nitrogen and the solution was cooled to $-78{ }^{\circ} \mathrm{C} . n-\mathrm{BuLi}(2.4 \mathrm{M}$ in hexanes, $3.75 \mathrm{~mL}, 9.0 \mathrm{mmol}, 1.2$ equiv.) was added dropwise and the mixture stirred at $-78{ }^{\circ} \mathrm{C}$ for 1 h . Chlorotriethylsilane ( $1.4 \mathrm{~mL}, 8.3 \mathrm{mmol}, 1.1$ equiv.) was then added slowly from a syringe and the resulting mixture was stirred at $-78^{\circ} \mathrm{C}$ to room temperature
for 18 hours resulting in the formation of a white suspension. The mixture was cooled back to -78 ${ }^{\circ} \mathrm{C}$ followed by dropwise addition of sec -butyllithium (1.4 M in cyclohexane, $6.4 \mathrm{~mL}, 9 \mathrm{mmol}, 1.2$ equiv.) and the reaction stirred for an additional 2 hours at $-78^{\circ} \mathrm{C}$ to $-30^{\circ} \mathrm{C}$. The reaction mixture was cooled back to $-78^{\circ} \mathrm{C}$ and quenched by the addition of aqueous $\mathrm{NH}_{4} \mathrm{Cl}$ and diluted with $\mathrm{Et}_{2} \mathrm{O}$, and the mixture was warmed to room temperature. After the layers were separated, the aqueous phase was washed with $\mathrm{Et}_{2} \mathrm{O}(3 \times 20 \mathrm{~mL})$. Then all the organic phases were combined, washed with $\mathrm{H}_{2} \mathrm{O}(20 \mathrm{~mL})$ and brine ( 20 mL ) respectively, and dried over anhydrous $\mathrm{MgSO}_{4}$. After filtration and concentration, the residue was purified by column chromatography ( $10 \% \mathrm{Et}_{2} \mathrm{O}$ in hexanes) to afford $463 \mathrm{mg}, 2.31 \mathrm{mmol}$ of $\mathbf{5 . 1 g}$ ( $50 \%$ based on starting alcohol $\mathbf{5 . 1 0 g}$ ) and 102 mg , 0.55 mmol of 5.14 ( $19 \%$ based on starting propargyl alcohol) as colorless liquids. Spectroscopic data of $\mathbf{5 . 1 g}:{ }^{1} \mathrm{H}$ NMR $\left(500 \mathrm{MHz}, \mathrm{C}_{6} \mathrm{D}_{6}\right) \delta 5.31(\mathrm{qt}, J=6.7,1.3 \mathrm{~Hz}, 1 \mathrm{H}), 3.81(\mathrm{~s}$, $1 \mathrm{H}), 1.56(\mathrm{dt}, J=6.7,1.3 \mathrm{~Hz}, 3 \mathrm{H}), 1.53(\mathrm{~s}, 3 \mathrm{H}), 1.04(\mathrm{t}, J=7.9 \mathrm{~Hz}, 9 \mathrm{H}), 0.86(\mathrm{~s}, 1 \mathrm{H}), 0.65(\mathrm{qq}, J$ $=16.0,7.9 \mathrm{~Hz}, 6 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $126 \mathrm{MHz}, \mathrm{C}_{6} \mathrm{D}_{6}$ ) $\delta 140.0,116.1,71.5,14.9,13.5,8.2$, 3.0. Spectroscopic data for 5.14: ${ }^{1} \mathrm{H} \operatorname{NMR}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 4.17(\mathrm{q}, J=2.7 \mathrm{~Hz}, 1 \mathrm{H}), 1.86(\mathrm{~d}, J=$ $2.5 \mathrm{~Hz}, 3 \mathrm{H}), 1.42(\mathrm{~s}, 1 \mathrm{H}), 1.00(\mathrm{t}, J=8.0 \mathrm{~Hz}, 9 \mathrm{H}), 0.68(\mathrm{q}, J=8.0 \mathrm{~Hz}, 6 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( 126 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta 83.8,79.8,54.6,7.3,3.8,1.6$.

### 4.7.3. The $[1,2]$-carbon to carbon silyl migration: general procedure $C$



Alcohol 5.1 was dissolved in $\mathrm{CDCl}_{3}$ and the solution transferred into NMR tube. The reaction in the NMR tube was monitored by NMR analysis until over $90 \%$ conversion had been achieved,
obtaining aldehyde 5.2.

## Synthesis of 2-methyl-2-(trimethylsilyl)propanal (5.2a)



Aldehyde 5.2a was synthesized from alcohol 4.3b following general procedure C in $97 \%$ conversion after 12 hours: ${ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 9.56(\mathrm{~s}, 1 \mathrm{H}), 1.19(\mathrm{~s}, 6 \mathrm{H}), 0.06(\mathrm{~s}, 9 \mathrm{H})$. ${ }^{13} \mathrm{C}$ NMR ( $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 206.8,42.1,17.3,-4.2$.

## Synthesis of 2-methyl-2-(triethylsilyl)propanal (5.2b)



Aldehyde 5.2b was synthesized from alcohol 5.1b following general procedure C in $92 \%$ conversion after 48 hours: ${ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 9.59(\mathrm{~s}, 1 \mathrm{H}), 1.21(\mathrm{~s}, 6 \mathrm{H}), 0.97(\mathrm{t}, J=$ $7.9 \mathrm{~Hz}, 9 \mathrm{H}), 0.65(\mathrm{q}, J=7.9 \mathrm{~Hz}, 6 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 206.3,43.0,18.2,7.7,1.8$. Synthesis of 2-methyl-2-(tripropylsilyl)propanal (5.2c)


Aldehyde 5.2c was synthesized from alcohol 5.1c following general procedure C in $95 \%$ conversion after 72 hours: ${ }^{1} \mathrm{H}$ NMR $\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 9.56(\mathrm{~s}, 1 \mathrm{H}), 1.38-1.27(\mathrm{~m}, 6 \mathrm{H}), 1.19$ $(\mathrm{s}, 6 \mathrm{H}), 0.95(\mathrm{t}, J=7.2 \mathrm{~Hz}, 9 \mathrm{H}), 0.64-0.58(\mathrm{~m}, 6 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 206.4,42.8$,
18.7, 18.3, 17.6, 13.5.

## Synthesis of 2-methyl-2-(methyldiphenylsilyl)propanal (5.2d)



Aldehyde 5.2d was synthesized from alcohol 5.1d following general procedure C in $97 \%$ conversion after 96 hours: ${ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 9.64(\mathrm{~s}, 1 \mathrm{H}), 7.64-7.59(\mathrm{~m}, 4 \mathrm{H}), 7.46$ $-7.38(\mathrm{~m}, 6 \mathrm{H}), 1.32(\mathrm{~s}, 6 \mathrm{H}), 0.70(\mathrm{~s}, 3 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $\left.126 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 206.7,135.2,133.2$, $129.8,128.0,41.9,18.5,-6.3$.

## Synthesis of 2,3-dimethyl-2-(trimethylsilyl)butanal (5.2e)



Aldehyde 5.2e was synthesized from alcohol 5.1e following general procedure C in $95 \%$ conversion after 24 hours. NMR analysis was done on a mixture of 5.1e:5.2e (1.4:1) ${ }^{1} \mathrm{H}$ NMR ( 500 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 9.53(\mathrm{~s}, 1 \mathrm{H}), 4.85(\mathrm{dt}, J=1.5,0.8 \mathrm{~Hz}, 1 \mathrm{H}), 4.83(\mathrm{q}, J=0.9 \mathrm{~Hz}, 1 \mathrm{H}), 3.90(\mathrm{~s}, 1 \mathrm{H})$, 2.53 (hept, $J=6.8 \mathrm{~Hz}, 1 \mathrm{H}), 1.94(\mathrm{~h}, J=6.8 \mathrm{~Hz}, 1 \mathrm{H}), 1.27(\mathrm{~s}, 2 \mathrm{H}), 1.10-1.02(\mathrm{~m}, 12 \mathrm{H}), 1.00(\mathrm{~d}$, $J=7.0 \mathrm{~Hz}, 4 \mathrm{H}), 0.85(\mathrm{~d}, J=6.9 \mathrm{~Hz}, 3 \mathrm{H}), 0.08(\mathrm{~s}, 9 \mathrm{H}), 0.05(\mathrm{~s}, 9 \mathrm{H}) .{ }^{13} \mathrm{C} \mathrm{NMR}\left(126 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$ $\delta 207.0,160.1,102.5,70.2,50.9,31.5,29.4,23.8,21.0,19.6,19.2,8.8,-2.3,-3.6$.

## Synthesis of 2,3-dimethyl-2-(triethylsilyl)butanal (5.2f)



Aldehyde 5.2f was synthesized from alcohol 5.1f following general procedure C in $100 \%$ conversion after 72 hours: ${ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 9.56(\mathrm{~s}, 1 \mathrm{H}), 2.50$ (hept, $J=6.9 \mathrm{~Hz}, 1 \mathrm{H}$ ), $1.04(\mathrm{~s}, 3 \mathrm{H}), 1.00(\mathrm{~d}, J=7.0 \mathrm{~Hz}, 3 \mathrm{H}), 0.96(\mathrm{t}, J=7.9 \mathrm{~Hz}, 9 \mathrm{H}), 0.82(\mathrm{~d}, J=6.8 \mathrm{~Hz}, 3 \mathrm{H}), 0.66(\mathrm{qd}$, $J=7.9,3.4 \mathrm{~Hz}, 6 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR (126 MHz, $\left.\mathrm{CDCl}_{3}\right) \delta 206.1,52.3,29.5,19.6,19.4,9.4,7.8,3.0$. ${ }^{29} \mathrm{Si}$ NMR $\left(99 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 8.71$.

## Synthesis of 2-methyl-2-(triethylsilyl)butanal (5.2g)



Aldehyde 5.2g was synthesized from alcohol $\mathbf{5 . 1 g}$ following general procedure C in $100 \%$ conversion after 144 hours: ${ }^{1} \mathrm{H}$ NMR $\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 9.57(\mathrm{~s}, 1 \mathrm{H}), 2.16(\mathrm{dq}, J=13.9,7.0 \mathrm{~Hz}$, $1 \mathrm{H}), 1.53(\mathrm{dq}, J=14.7,7.4 \mathrm{~Hz}, 1 \mathrm{H}), 1.15(\mathrm{~s}, 3 \mathrm{H}), 0.96(\mathrm{t}, J=7.9 \mathrm{~Hz}, 9 \mathrm{H}), 0.82(\mathrm{t}, J=7.4 \mathrm{~Hz}, 3 \mathrm{H})$, $0.64(\mathrm{q}, J=8.1 \mathrm{~Hz}, 6 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 206.5,48.3,24.5,13.2,9.9,7.7,1.8$.

### 5.8.4. Synthesis of enantioenriched 5.1f

Compound 5.1f* was synthesized from 5.1f in two steps as indicated on scheme 5.9.

## Synthesis of 3-methyl-2-methylene-1-(triethylsilyl)butan-1-one (5.4f)



Following a reported procedure; a dry 100 mL round bottom flask fitted with a magnetic stir bar was sealed with a rubber septum and cooled under nitrogen. 25 mL of dry DCM was then added to the flask via syringe followed by 1.04 mL trifluoroacetic anhydride $(7.5 \mathrm{mmmol}, 1.5$ equiv). The mixture was cooled on a dry ice-acetone bath to $-78^{\circ} \mathrm{C}$ followed by dropwise addition of 0.71 mL of DMSO ( $10 \mathrm{mmol}, 2.0$ equiv). After complete addition of DMSO, the resulting mixture was stirred at $-78{ }^{\circ} \mathrm{C}$ for additional 30 minutes after which a solution of $\alpha$ hydroxyallylsilane 5.1f 1072.1 mg in dry DCM ( 5.0 mmol , 1.0 equiv) was added dropwise and resulting solution stirred at $-78^{\circ} \mathrm{C}$ for one hour. This was followed by slow addition of 2.09 mL of freshly distilled triethylamine ( $15 \mathrm{mmol}, 3.0$ equiv) and the resulting mixture stirred at $-78{ }^{\circ} \mathrm{C}$ for another one hour. The reaction mixture was quenched with water and allowed to warm up slowly to room temperature. The layers were separated and the aqueous phase was extracted twice with DCM. The combined organic phases were dried over anhydrous magnesium sulfate filtered and concentrated in vacuo followed by column chromatography ( $5 \% \mathrm{EtOAc}$ in hexanes) to afford $722 \mathrm{mg}, 3.4 \mathrm{mmol}$ ( $68 \%$ isolated yield) of $\mathbf{5 . 4 f}$ as a colorless oil and 95 mg of $\mathbf{5 . 1 f}$ ( $9 \%$ recovery). Spectroscopic data for 5.4f: ${ }^{1} \mathrm{H}$ NMR $\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 5.94(\mathrm{~d}, J=1.2 \mathrm{~Hz}, 1 \mathrm{H}), 5.88(\mathrm{~s}, 1 \mathrm{H})$, 2.83 (hept, $J=6.9,1.2 \mathrm{~Hz}, 1 \mathrm{H}), 0.98-0.89(\mathrm{~m}, 15 \mathrm{H}), 0.76(\mathrm{qd}, J=7.8,1.1 \mathrm{~Hz}, 6 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR $\left(126 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 237.9,161.5,124.7,26.1,21.6,7.3,3.8 .{ }^{29} \mathrm{Si}^{\mathrm{N}} \mathrm{MR}\left(99 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta-0.88$.

## Synthesis of (S)-3-methyl-2-methylene-1-(triethylsilyl)butan-1-ol (5.1f*)



Following a reported procedure, ${ }^{58}$ to a solution of $659 \mathrm{mg} \mathbf{5 . 4 f}(3.1 \mathrm{mmol}, 1.0$ equiv) in 10 mL of freshly distilled anhydrous THF was added 2 g of $4 \AA$ molecular sieves at room temperature under nitrogen atmosphere and the mixture stirred at room temperature for 2 h . To the mixture was added $4.3 \mathrm{~g}(R)$-CBS as a solution in 5 mL dry THF $5(15.5 \mathrm{mmol}, 5.0$ equiv) and reaction mixture was cooled to $-30^{\circ} \mathrm{C}$ and stirred at this temperature for 10 min . This was followed by dropwise dropwise addition of $1.5 \mathrm{~mL} \mathrm{BH}_{3} \cdot \mathrm{SMe}_{2}\left(15.5 \mathrm{mmol}, 5.0\right.$ equiv) at $-30^{\circ} \mathrm{C}$. The resulting mixture was stirred at $-30{ }^{\circ} \mathrm{C}$ for 15 min , quenched with MeOH and saturated $\mathrm{NH}_{4} \mathrm{Cl}$, and extracted with $\mathrm{Et}_{2} \mathrm{O}$ (20 mL x 2). The combined organic layers were washed with brine, dried over MgSO 4 , and filtered. The filtrate was concentrated under reduced pressure. The residue was purified by flash column chromatography on silica gel ( $30 \% \mathrm{Et}_{2} \mathrm{O}$ in hexanes) to give $165 \mathrm{mg}, 0.78 \mathrm{mmol}$ of $\mathbf{5 . 1} \mathbf{f}^{*}$ ( $25 \%$ isolated yield) as a colorless oil. The optical purity and absolute configuration of $5.1 \mathrm{f}^{*}$ were not determined experimentally but compared to similar structures in the literature. ${ }^{58}$ The other spectral data were identical with those of $\mathbf{5 . 1 f}$.

## Synthesis of 2,3-dimethyl-2-(triethylsilyl)butanal (5.2f*?)



Aldehyde 5.2f*? was synthesized from alcohol 5.1f* following general procedure C in $100 \%$ conversion after 72 hours: Spectroscopic data was identical to those of $\mathbf{5 . 2 f}$.

Attempted derivatization of $5.2 \mathrm{f} *$ ?: Synthesis of $(E)$-1-(2,3-dimethylbutylidene)-2-(2,4dinitrophenyl)hydrazine


Compound $\mathbf{5 . 6}$ was synthesized following the same procedure for the synthesis of $\mathbf{3 . 1 6}$ in Chapter 3. Spectroscopic data for 5.6: ${ }^{1} \mathrm{H}$ NMR $\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 11.00(\mathrm{~s}, 1 \mathrm{H}), 9.11(\mathrm{~d}, J=$ $2.6 \mathrm{~Hz}, 1 \mathrm{H}), 8.29(\mathrm{ddd}, J=9.6,2.6,0.8 \mathrm{~Hz}, 1 \mathrm{H}), 7.93(\mathrm{~d}, J=9.6 \mathrm{~Hz}, 1 \mathrm{H}), 7.46(\mathrm{dd}, J=6.1,0.7$ $\mathrm{Hz}, 1 \mathrm{H}), 2.41(\mathrm{~h}, J=7.0 \mathrm{~Hz}, 1 \mathrm{H}), 1.90-1.78(\mathrm{~m}, J=6.8 \mathrm{~Hz}, 1 \mathrm{H}), 1.16(\mathrm{~d}, J=6.8 \mathrm{~Hz}, 3 \mathrm{H}), 0.98$ $(\mathrm{d}, J=2.7 \mathrm{~Hz}, 3 \mathrm{H}), 0.97(\mathrm{~d}, J=2.7 \mathrm{~Hz}, 3 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR $\left(126 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 156.3,145.2,137.6$, $129.9,128.7,123.5,116.6,43.0,31.6,20.1,19.5,14.3$. Compound 5.6 is known. ${ }^{59}$

## Synthesis of trimethyl((2-phenylallyl)oxy)silane (5.7)



To a 100 mL round bottom flask was weighed 1342 mg ally alcohol $\mathbf{5 . 9 h}$ ( $10 \mathrm{mmol}, 1.0$ equiv) and 50 mL of dry THF added. The flask was sealed with a rubber septum and purged with nitrogen. This was followed by addition of 0.89 mL of pyridine ( $11 \mathrm{mmol}, 1.1$ equiv) and the resulting mixture stirred in dry THF for 5 minutes. This was followed by addition of 1.27 mL of TMSCl
( $10 \mathrm{mmol}, 1.0$ equiv) and the resulting mixture stirred for an additional one hour at room temperature. The THF was removed by rotavap and the resulting residue diluted with hexanes. The mixture was filtered, and the filtrate was concentrated under reduced pressure to give the expected product. The product was further purified by column chromatography ( $10 \% \mathrm{EtOAc}$ in hexanes) to give $2046 \mathrm{mg}, 9.9 \mathrm{mmol}\left(99 \%\right.$ isolated yield) of $5.7 .{ }^{1} \mathrm{H} \mathrm{NMR}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ $7.49-7.41(\mathrm{~m}, 2 \mathrm{H}), 7.39-7.33(\mathrm{~m}, 2 \mathrm{H}), 7.33-7.27(\mathrm{~m}, 1 \mathrm{H}), 5.47(\mathrm{q}, J=1.4 \mathrm{~Hz}, 1 \mathrm{H}), 5.41(\mathrm{q}, J$ $=1.7 \mathrm{~Hz}, 1 \mathrm{H}), 4.54(\mathrm{t}, J=1.5 \mathrm{~Hz}, 2 \mathrm{H}), 0.19(\mathrm{~s}, 9 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $\left.126 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 146.8,139.0$, $128.3,127.6,126.0,111.9,64.3$. Spectroscopic data were in agreement with those reported in the literature. ${ }^{60}$

### 5.8.5. Unexpected $S_{N} 2$-like reactions: general procedure $D$



To a 50 mL round bottom flask was added as a solution of $\mathbf{5 . 7} \mathrm{in} 12 \mathrm{~mL}$ dry THF. The mixture was cooled to $-78{ }^{\circ} \mathrm{C}$ then butyllithium was added dropwise resulting in a colored solution. The mixture was stirred at $-78{ }^{\circ} \mathrm{C}$ for two hours then quenched with saturated ammonium chloride solution and diluted with 20 mL of ether 10 mL of water. The layers were separated and the aqueous layer was extracted with ether ( 20 mL X 3 ). Combined organics were washed with saturated ammonium chloride, water and brine respectively and then dried over anhydrous magnesium sulfate. Filtration and concentration gave the substitution product.

## Synthesis of (4,4-dimethylpent-1-en-2-yl)benzene 5.8a



Applying general procedure D to silyl ether 5.7412 .72 mg ( $2.0 \mathrm{mmol}, 1.0$ equiv) and 1.5 mL of tert-butyllithium (1.6 M in pentane, $2.4 \mathrm{mmol}, 1.2$ equiv), compound $\mathbf{5 . 8 a}$ was synthesized in quantitative yield: ${ }^{1} \mathrm{H}$ NMR $\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.44-7.38(\mathrm{~m}, 2 \mathrm{H}), 7.36-7.30(\mathrm{~m}, 2 \mathrm{H}), 7.29$ - $7.24(\mathrm{~m}, 1 \mathrm{H}), 5.28(\mathrm{~d}, J=2.1 \mathrm{~Hz}, 1 \mathrm{H}), 5.05(\mathrm{dd}, J=2.0,0.9 \mathrm{~Hz}, 1 \mathrm{H}), 2.50(\mathrm{~s}, 2 \mathrm{H}), 0.84(\mathrm{~s}, 9 \mathrm{H})$. ${ }^{13} \mathrm{C}$ NMR (126 MHz, $\mathrm{CDCl}_{3}$ ) $\delta 147.6$, 143.7, 128.1, 126.9, 126.5, 116.3, 48.9, 31.7, 30.1. Spectroscopic data were in agreement with those reported in the literature. ${ }^{61}$

## Synthesis of (4-methylhex-1-en-2-yl)benzene 5.8b



Applying general procedure D to silyl ether $\mathbf{5 . 7} 412.72 \mathrm{mg}$ ( $2.0 \mathrm{mmol}, 1.0$ equiv) and 1.71 mL of sec-butyllithium (1.4 M in cyclohexane, 2.4 mmol, 1.2 equiv). After workup and column chromatography ( $1.5 \%$ EtOAc in hexanes), compound $\mathbf{5 . 8 b}$ was synthesized in $292.81 \mathrm{mg}, 1.68$ $\mathrm{mmol}\left(84 \%\right.$ isolated yield): ${ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.45-7.40(\mathrm{~m}, 2 \mathrm{H}), 7.37-7.26(\mathrm{~m}$, $3 \mathrm{H}), 5.29(\mathrm{~d}, J=1.8 \mathrm{~Hz}, 1 \mathrm{H}), 5.06(\mathrm{q}, J=1.2 \mathrm{~Hz}, 1 \mathrm{H}), 2.62(\mathrm{ddd}, J=14.1,5.8,1.3 \mathrm{~Hz}, 1 \mathrm{H}), 2.25$ (ddd, $J=14.0,8.3,1.0 \mathrm{~Hz}, 1 \mathrm{H}), 1.52-1.43(\mathrm{~m}, 1 \mathrm{H}), 1.22-1.14(\mathrm{~m}, 1 \mathrm{H}), 0.89(\mathrm{t}, J=7.4 \mathrm{~Hz}, 3 \mathrm{H})$, $0.85(\mathrm{~d}, J=6.6 \mathrm{~Hz}, 4 \mathrm{H}), 0.80(\mathrm{ddd}, J=14.1,7.0,2.1 \mathrm{~Hz}, 1 \mathrm{H}) .{ }^{13} \mathrm{C} \operatorname{NMR}\left(126 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ $147.8,141.5,128.2,127.2,126.3,113.6,43.0,32.6,29.4,18.9,11.3$. Spectroscopic data were in
agreement with those reported in the literature. ${ }^{62}$
Synthesis of hept-1-en-2-ylbenzene 5.8c


Applying general procedure D to silyl ether 5.7412 .72 mg ( $2.0 \mathrm{mmol}, 1.0$ equiv) and 1 mL of $n$-butyllithium (2.4 M in cyclohexane, $2.4 \mathrm{mmol}, 1.2$ equiv). After workup compound $\mathbf{5 . 8} \mathbf{c}$ and 5.9h as a mixture $100 \%$ conversion by NMR of the crude material: ${ }^{1} \mathrm{H}$ NMR $\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ $7.51-7.43(\mathrm{~m}, 4 \mathrm{H}), 7.42-7.28(\mathrm{~m}, 6 \mathrm{H}), 5.51(\mathrm{q}, J=1.0 \mathrm{~Hz}, 1 \mathrm{H}), 5.38(\mathrm{q}, J=1.4 \mathrm{~Hz}, 1 \mathrm{H}), 5.31$ $(\mathrm{d}, J=1.5 \mathrm{~Hz}, 1 \mathrm{H}), 5.10(\mathrm{q}, J=1.4 \mathrm{~Hz}, 1 \mathrm{H}), 4.56(\mathrm{dd}, J=1.6,0.9 \mathrm{~Hz}, 2 \mathrm{H}), 2.57-2.50(\mathrm{~m}, 1 \mathrm{H})$, $1.91(\mathrm{~s}, 1 \mathrm{H}), 1.55-1.44(\mathrm{~m}, 2 \mathrm{H}), 1.43-1.31(\mathrm{~m}, 4 \mathrm{H}), 0.98-0.86(\mathrm{~m}, 4 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( 126 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta 148.7,147.2,141.4,138.4,128.5,128.2,127.9,127.2,126.0,126.0,112.5,111.9,64.9$, 35.3, 31.5, 27.9, 22.4, 14.0. Spectroscopic data of $\mathbf{5 . 8} \mathbf{c}$ were in agreement with those reported in the literature. ${ }^{63}$

## REFERENCES

(1) Brook, A. G. Isomerism of some $\alpha$-hydroxysilanes to silyl ethers. J. Am. Chem. Soc. 1958, 80, 1886.
(2) Brook, A. G. Molecular rearrangements of organosilicon compounds. Acc. Chem. Res. 1974, 7, 77.
(3) Page, P. C. B.; Klair, S. S.; Rosenthal, S. Synthesis and chemistry of acyl silanes. Chem. Soc. Rev. 1990, 19, 147.
(4) Moser, W. H. The Brook rearrangement in tandem bond formation strategies. Tetrahedron 2001,57, 2065.
(5) West, R.; Gornowicz, G. A. New anionic rearrangements XIII. Reactions of tertbutyllithium with organosilanes. J. Organomet. Chem. 1971, 28, 25.
(6) Thadani, A. N.; Huang, Y.; Rawal, V. H. Expedient, high-yielding synthesis of silylsubstituted salen ligands. Org. Lett. 2007, 9, 3873.
(7) He, Y.; Hu, H.; Xie, X.; She, X. Regioselective lithiation/retro-Brook rearrangement via direct deprotonation. Tetrahedron 2013, 69, 559.
(8) He, Y.; Ma, B.; Yang, J.; Xie, X.; She, X. Regioselective lateral or vinyl C-H lithiation/1,5-retro-Brook rearrangement via quinolyl or pyridyl ring directed deprotonation. Tetrahedron 2013, 69, 5545.
(9) Kapeller, D. C.; Brecker, L.; Hammerschmidt, F. Configurational stability of oxymethyllithiums as intermediates in intramolecular rearrangements. Chem. Eur. J. 2007, 13, 9582.
(10) Bariak, V.; Malastova, A.; Almassy, A.; Sebesta, R. Retro-Brook rearrangement of ferrocene-derived silyl ethers. Chem. Eur. J. 2015, 21, 13445.
(11) Duhamel, L.; Gralak, J.; Ngono, B. Aldéhydes $\alpha$-silylés: Préparation et propriétés nouvelles. J. Organomet. Chem.1989, 363 C4.
(12) Larson, L. G. $\alpha$-Silyl carbonyl compounds. Pure \& Appl. Chem.1990, 62, 2021.
(13) Enders, D.; Poiesz, C.; Joseph, R. Enantioselective synthesis of protected $\alpha$-aminoketones via electrophilic amination of $\alpha$-silylketones with an oxaziridine. Tetrahedron Asymmetry 1998, 9, 3709.
(14) Peterson, D. J. Carbonyl olefination reaction using silyl-substituted organometallic compounds. J. Org. Chem. 1968, 33, 780.
(15) Pulido, F.; Barbero, A. Peterson olefination from $\alpha$-silyl aldehydes. Nat Protoc. 2006, 1, 2068.
(16) Denise C.; Chauret, J.; Chong, M.; Ye, Q. A route to enantiomerically-enriched $\alpha$-silyl aldehydes from 2,3-epoxy alcohols. Tetrahedron Asymmetry, 1999, 10, 3601.
(17) Kim, J. Y.; Kang, B. C.; Ryu, D. H. Catalytic asymmetric Roskamp reaction of silyl diazoalkane: Synthesis of enantioenriched $\alpha$-silyl ketone. Org. Lett. 2017, 19, 5936.
(18) Novikov, Y. Y.; Sampson, P. 1-Bromo-1-lithioethene: A practical reagent in organic synthesis. J. Org. Chem. 2005, 70, 10247.
(19) Simpson, J. H. Instrumental Considerations. Organic Structure Determination Using 2-D NMR Spectroscopy (Second Edition). 2012, 21.
(20) Saunders, M.; Jimenez-Vazquez, A. Recent studies of carbocations.Chem. Rev. 1991, 91, 375.
(21) Saunders, M.; Jimenez-Vazquez, A. ChemInform abstract: Recent studies on carbocations. ChemInform 1991, 22.
(22) Hansen, T.; Vermeeren, P.; Bickelhaupt, F. M.; Hamlin, T. A. Stability of alkyl carbocations. Chem. Commun. 2022, 58, 12050.
(23) Ushakav, S. N.; Itenberg, A. M. The synthesis of triethylvinylsilicane. Zh. Obshch. Khim. 1937, 7, 2495.
(24) Sommer, L. H.; Dorfman, E.; Goldberg, G. M.; Whitmore, F. C. The reactivity with alkali of chlorine-carbon bonds alpha, beta and gamma to silicon. J. Am. Chem. Soc. 1946, 68, 488.
(25) Sommer, L. H.; Bailey, D. L.; Whitmore, F. C. Further studies of $\beta$-eliminations involving silicon. J. Am. Chem. Soc. 1948, 70, 2869.
(26) Sommer, L. H.; Baughman, G. A. Siliconium ions and carbonium ions as reaction intermediates. J. Am. Chem. Soc. 1961, 83, 3346.
(27) Traylor, T. G.; Hanstein, W.; Berwin, H. J.; Clinton, N. A.; Brown, R. S. Vertical stabilization of cations by neighboring $\sigma$ bonds. General considerations. J. Am. Chem. Soc. 1971, 93, 5715.
(28) Davis, D. D.; Jacocks, H. M., III. Deoxymetalation reactions. The mechanisms of deoxysilylation of mono-trimethylsilyl-and bis-trimethylsilyl-substituted alcohols and a comparison to the mechanism of deoxystannylation and deoxyplumbylation. J. Organomet. Chem. 1981, 206, 33.
(29) Colvin, E. W. Silicon in Organic Synthesis; Butterworths: London, 1981.
(30) Lambert, J. B.; Wang, G. T.; Finzel, R. B.; Teramura, D. H. Stabilization of positive charge by $\beta$-silicon. J. Am. Chem. Soc. 1987, 109, 7838.
(31) Lambert, J. B.; Chelius, E. C. The $\beta$ effect of silicon in the synperiplanar geometry. J. Am. Chem. Soc. 1990, 112, 8120.
(32) Lambert, J. B. Tetrahedron report number 273 : The interaction of silicon with positively charged carbon. Tetrahedron 1990, 46, 2677.
(33) White, J. M. Reactivity and ground state effects of silicon in organic chemistry. Aust. J. Chem. 1995, 48, 1227.
(34) Lambert, J. B.; Liu, X. The $\beta$-effect of silicon in the orthogonal geometry. J. Organomet. Chem. 1996, 521, 203.
(35) Lambert, J. B.; Zhao, Y.; Robert W. Emblidge, R. W.; Salvador, L. A.; Liu, X.; So, J-H.; Chelius, E. C. The $\beta$ effect of silicon and related manifestations of $\sigma$ conjugation. Acc. Chem. Res. 1999, 32, 183.
(36) Corey, E. J.; Bakshi, R. K.; Shibata S. Highly enantioselective borane reduction of ketones catalyzed by chiral oxazaborolidines: Mechanism and synthetic implications. J. Am. Chem. Soc. 1987, 109, 5551.
(37) Terao, J.; Todo, H.; Begum, S.A.; Kuniyasu, H. and Kambe, N. Copper-catalyzed crosscoupling reaction of grignard reagents with primary-alkyl halides: remarkable effect of 1phenylpropyne. Angew. Chem. Int. Ed. 2007, 46, 2086.
(38) Ren, P.; Stern, L.A.; Hu, X. Copper-Catalyzed Cross-Coupling of Functionalized Alkyl Halides and Tosylates with Secondary and Tertiary Alkyl Grignard Reagents. Angew. Chem. Int. Ed. 2012, 51, 9110.
(39) Yang, C.-T.; Zhang, Z.-Q.; Liang, J.; Liu, J.-H.; Lu, X.-Y.; Chen, H.-H.; Liu L. Coppercatalyzed cross-coupling of nonactivated secondary alkyl halides and tosylates with secondary alkyl Grignard reagents. J. Am. Chem. Soc. 2012, 134, 11124.
(40) Iwasaki, T.; Takagawa, H.; Singh, S. P.; Kuniyasu, H.; Kambe, N. Co-catalyzed crosscoupling of alkyl halides with tertiary alkyl Grignard reagents using a 1, 3-butadiene additive. J. Am. Chem. Soc. 2013, 135, 9604.
(41) Lauer, A. M.; Mahmud, F.; Wu, J. Cu (I)-catalyzed, $\alpha$-selective, allylic alkylation reactions between phosphorothioate esters and organomagnesium reagents. J. Am. Chem. Soc. 2011, 133, 9119.
(42) Han, X.; Zhang, Y.; Wu, J. Mild two-step process for the transition-metal-free synthesis of carbon-carbon bonds from allylic alcohols/ethers and Grignard reagents. J. Am. Chem. Soc. 132, 4104.
(43) Breit, B.; Demel, P.; Grauer, D.; Studte, C. Stereospecific and stereodivergent construction of tertiary and quaternary carbon centers through switchable directed/nondirected allylic substitution. Chem. Asian J. 2006, 1, 586.
(44) Ghorai, S. K.; Jin, M.; Hatakeyama, T.; Nakamura, M. Cross-coupling of non-activated chloroalkanes with aryl Grignard reagents in the presence of iron/N-heterocyclic carbene catalysts. Org. Lett. 2012, 14, 1066.
(45) Someya, H.; Ohmiya, H.; Yorimitsu, H.; Oshima, K. Silver-catalyzed benzylation and allylation reactions of tertiary and secondary alkyl halides with Grignard reagents. Org. Lett. 2008, 10, 969.
(46) Mitamura, Y.; Asada, Y.; Murakami, K.; Someya, H.; Yorimitsu, H.; Oshima, K. Silvercatalyzed benzylation and allylation of tertiary alkyl bromides with organozinc reagents. Chem. Asian J. 2010, 5, 1487.
(47) Tsuji, T.; Yorimitsu, H.; Oshima, K. Cobalt-catalyzed coupling reaction of alkyl halides with allylic grignard reagents. Angew. Chem. Int. Ed. 2002, 41, 4137.
(48) Ohmiya, H.; Tsuji, T.; Yorimitsu, H.; Oshima, K. Cobalt-catalyzed cross-coupling reactions of alkyl halides with allylic and benzylic Grignard reagents and their application to tandem radical cyclization/cross-coupling reactions. Chem. Eur. J. 2004, 10, 5640.
(49) Pupo, G.; Properzi, R.; List, B. Asymmetric catalysis with $\mathrm{CO}_{2}$ : The direct $\alpha$-allylation of ketones. Angew. Chem. Int. Ed. 2016, 55, 6099.
(50) Braun, M.; Meier, T. Tsuji-trost allylic alkylation with ketone enolates. Angew. Chem. Int. Ed. 2006, 45, 6952.
(51) Gieuw, M. H.; Leung, V. M-Y.; Ke, Z.; Yeung, Y-Y. Electrophilic bromolactonization of cyclopropyl diesters using lewis basic chalcogenide catalysts. Adv. Synth. Catal. 2018, 360, 4306.
(52) Mori-Quiroz, L. M.; Maleczka, R, E., Jr. Stereoconvergent [1,2]- and [1,4]-wittig rearrangements of 2-silyl-6-aryl-5,6-dihydropyrans: A tale of steric vs electronic regiocontrol of divergent pathways. J. Org. Chem. 2015, 80, 1163.
(53) Jing, C.; Jones, B. T.; Adams, R. J.; Bower, J. F. Cyclopropane-fused $N$-heterocycles via aza-Heck-triggered $\mathrm{C}\left(\mathrm{sp}^{3}\right)-\mathrm{H}$ functionalization cascades. J. Am. Chem. Soc. 2022, 144, 16749.
(54) Tosatti, P.; Horn, J.; Campbell, A. J.; House, D.; Nelson, A.; Marsden, S. P. Iridiumcatalyzed asymmetric allylic amination with polar amines: Access to building blocks with lead-like molecular properties. Adv. Synth. Catal. 2010, 352, 3153.
(55) Fotiadou, A. D.; Zografos, A. L. Accessing the structural diversity of pyridone alkaloids: Concise total synthesis of rac-citridone A. Org. Lett. 2011, 13, 4592.
(56) Wegmann, M.; Bach, T. Stereoselective synthesis of a highly oxygenated $\delta$-lactone related to the core structure of (-)-Enterocin. Synthesis 2017, 49, 209.
(57) He, T.; Klare, H. F. T.; Oestreich, M. Catalytically generated Meerwein's salt-type oxonium ions for Friedel-Crafts $\mathrm{C}\left(\mathrm{sp}^{2}\right)-\mathrm{H}$ methylation with methanol. J. Am. Chem. Soc. 2023, 145, 3795.
(58) Higashino, M.; Ikeda, N.; Shinada, T.; Sakaguchi, K.; Ohfune, Y. Stereoselective anti-SN2' Mitsunobu reaction of $\alpha$-hydroxy- $\alpha$-alkenylsilanes. Tetrahedron Lett. 2011, 52, 422.
(59) Müller, E.; Böttcher, A. E. Photooximierung von Methylgruppen in gesättigten Kohlenwasserstoffen. Chem. Ber. 1975, 108, 1475.
(60) Zhang, Q.; Zhu, S.-F.; Cai, Y.; Wang, L.-X.;Zhou, Q.-L. Nickel-catalyzed enantioselective hydrovinylation of silyl-protected allylic alcohols: An efficient access to homoallylic alcohols with a chiral quaternary center. Sci. China Chem. 2010, 53, 1899.
(61) Chen, H.; Jia, X.; Yu, Y.; Qian, Q.; Gong, H. Nickel-catalyzed reductive allylation of tertiary alkyl halides with allylic carbonates. Angew. Chem. Int. Ed. 2017, 129, 13103.
(62) Chen, X.; Luo, X.; Wang, P. Electrochemical-induced radical allylation via the fragmentation of alkyl 1,4-dihydropyridines. Tetrahedron Lett. 2022. 91, 153646.
(63) Lu, P.; Ren, X.; Xu, H.; Lu, D.; Sun, Y.; Lu, Z. Iron-catalyzed highly enantioselective hydrogenation of alkenes. J. Am. Chem. Soc. 2021, 143, 12433.

## APPENDIX

## Copies of NMR spectra



640


5.11
${ }^{1} \mathrm{H}$ NMR, $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$








$\stackrel{\cong}{i}$

5.12i
${ }^{1} \mathrm{H}$ NMR, $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$





5.1b
${ }^{13} \mathrm{C}$ NMR, $126 \mathrm{MHz}, \mathrm{C}_{6} \mathrm{D}_{6}$


[^8]
5.1c
${ }^{1} \mathrm{H}$ NMR, $500 \mathrm{MHz}, \mathrm{C}_{6} \mathrm{D}_{6}$



[^9]웅



5.1d
${ }^{1} \mathrm{H}$ NMR, $500 \mathrm{MHz}, \mathrm{C}_{6} \mathrm{D}_{6}$





[^10]
5.1 f
${ }^{1} \mathrm{H}$ NMR, $500 \mathrm{MHz}, \mathrm{C}_{6} \mathrm{D}_{6}$







5.3
${ }^{13} \mathrm{C}$ NMR, $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$





5.14
${ }^{1} \mathrm{H}$ NMR, $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$


5.14
${ }^{13} \mathrm{C}$ NMR， $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$



${ }^{13} \mathrm{C}$ NMR, $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$

$-9.59$


${ }^{1} \mathrm{H}$ NMR, $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$


${ }^{13} \mathrm{C}$ NMR, $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$



5.2c


${ }^{13} \mathrm{C}$ NMR, $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$




5.2d
${ }^{1} \mathrm{H}$ NMR, $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$




5.1e $1.4: 1 \quad 5.2 e$
${ }^{13} \mathrm{C}$ NMR, $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$

© ${\underset{\sim}{n}}_{\substack{n}}^{\substack{m \\ i}}$

[^11]
$$
\stackrel{M}{⿱ 艹 乙}
$$
(
${ }^{13} \mathrm{C}$ NMR， $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$


$5.2 f$
${ }^{29} \mathrm{Si} \mathrm{NMR}, 99 \mathrm{MHz}, \mathrm{CDCl}_{3}$


5.2g
${ }^{1} \mathrm{H}$ NMR, $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$




-

//

5.4
${ }^{1} \mathrm{H}$ NMR, $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$


5.4
${ }^{29}$ Si NMR, $99 \mathrm{MHz}, \mathrm{CDCl}_{3}$





5.6
${ }^{1} \mathrm{H}$ NMR, $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$



| 200 | 190 | 180 | 170 | 160 | 150 | 140 | 130 | 120 | 110 | 100 | 90 | 80 | 70 | 60 | 50 | 40 | 30 | 20 | 10 | 0 | -1C |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  |  |  |  |  |  |  |  |  |  | f1 |  |  |  |  |  |  |  |  |  |  |  |



5.7
${ }^{1} \mathrm{H}$ NMR, $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$



5.8a
${ }^{1} \mathrm{H}$ NMR, $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$



5.8b


## 



| 200 | 190 | 180 | 170 | 160 | 150 | 140 | 130 | 120 | 110 | 100 | 90 | 80 | 70 | 60 | 50 | 40 | 30 | 20 | 10 | 0 | -10 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  |  |  |  |  |  |  |  |  |  | f1 (p |  |  |  |  |  |  |  |  |  |  |  |




## CHAPTER 6.

## FUTURE WORK

### 6.1. Future work on Wittig rearrangements

### 6.1.1. Wittig rearrangement on a contracted ring

So far, we have seen the Wittig rearrangement of dihydropyrans and tetrahydrooxepins whereby regioselectivity is controlled by having a silyl group either at the 2- or 4-position.The future work on Wittig rearrangements will entail reducing the ring system to 2 -silyl-2,5dihydrofuran 6.1. This might lead to formation of 2-silylcyclobutenols 6.2 via [1,2]-Wittig rearrangement and/or the two ring opened products 6.3 and 6.4 (Scheme 6.1).


Scheme 6.1: Proposed Wittig rearrangement of 2-silyl-2,5-dihydrofuran
The above transformation has not been able to be accomplished due to difficulties in accessing the dihydrofuran via ring closing metathesis. This problem was solved in 2005 by Grubbs (Hong, S. H.; Sanders, D. P.; Lee, C. W.; Grubbs, R. H. Prevention of undesirable isomerization during olefin metathesis. J. Am. Chem. Soc. 2005, 127, 17160).

### 6.1.2. Use of other heteroatoms on Wittig rearrangement

All the Wittig rearrangements discussed in this dissertation use oxygen as the heteroatom. The future work will involve the use of other atoms in place of oxygen (Scheme 6.2).


Scheme 6.2: Proposed Wittig rearrangement on a six-member ring with other heteroatoms

### 6.2. Future Work on silyl migrations

For silyl migrations, future work will involve use of other electrophiles apart from $m$-CPBA (Scheme 6.3a). For the acid catalyzed silyl migration, future work will entail the use of Lewis acid (Scheme 6.3b). Lastly, for the unusual $\mathrm{S}_{\mathrm{N}} 2$-like reaction, the future work will involve expanding the substrate scope by modifying the aromatic ring and/or use of other alkyl/aryl lithiums in place of butyllithiums (Scheme 6.3c).


Scheme 6.3: Future work on silyl migrations


[^0]:    21020019018017016015014013012011010090 f1（ppm）

[^1]:    $21020019018017016015014013012011010090 \quad 80$ f1 (ppm)

[^2]:    $21020019018017016015014013012011010090 \quad 80$ f1 (ppm)

[^3]:    21020019018017016015014013012011010090

[^4]:    $21020019018017016015014013012011010090 \quad 80$ f1 (ppm)

[^5]:    $210200190180170160150140130120110100 \quad 90$ f1 (ppm)

[^6]:    $\begin{array}{llllllllllllllllllllllllllllllllll}230 & 220 & 210 & 200 & 190 & 180 & 170 & 160 & 150 & 140 & 130 & 120 & 110 & 100 & 90 & 80 & 70 & 60 & 50 & 40 & 30 & 20 & 10 & 0 & -10\end{array}$

[^7]:    

[^8]:    $\begin{array}{lllllllllllllllllllllllllll}200 & 190 & 180 & 170 & 160 & 150 & 140 & 130 & 120 & 110 & 100 & 90 & 80 & 70 & 60 & 50 & 40 & 30 & 20 & 10 & 0 & -10\end{array}$

[^9]:    

[^10]:    

[^11]:    $\begin{array}{lllllllllllllllllllllllllllllllllll}230 & 220 & 210 & 200 & 190 & 180 & 170 & 160 & 150 & 140 & 130 & 120 & 110 & 100 & 90 & 80 & 70 & 60 & 50 & 40 & 30 & 20 & 10 & 0 & -10\end{array}$

